

1 **Title:**

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3 **Denitrification and Nitrous Oxide Emissions from Riparian Forests Soils Exposed to**  
4 **Prolonged Nitrogen Runoff**

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32 **Key words:** Chronic nitrogen loading, Denitrification, Nitrous oxide emissions, Nitrogen  
33 saturation; Nursery runoff, Riparian wetlands, Phosphorus loading, Water quality

34

**1 ABSTRACT**

2 Compared to upland forests, riparian forest soils have greater potential to remove nitrate  
3 ( $\text{NO}_3$ ) from agricultural run-off through denitrification. It is unclear, however, whether  
4 prolonged exposure of riparian soils to nitrogen (N) loading will affect the rate of  
5 denitrification and its end products. This research assesses the rate of denitrification and  
6 nitrous oxide ( $\text{N}_2\text{O}$ ) emissions from riparian forest soils exposed to prolonged nutrient  
7 run-off from plant nurseries and compares these to similar forest soils not exposed to  
8 nutrient run-off. Nursery run-off also contains high levels of phosphate ( $\text{PO}_4$ ). Since there  
9 are conflicting reports on the impact of  $\text{PO}_4$  on the activity of denitrifying microbes, the  
10 impact of  $\text{PO}_4$  on such activity was also investigated. Bulk and intact soil cores were  
11 collected from N-exposed and non-exposed forests to determine denitrification and  $\text{N}_2\text{O}$   
12 emission rates, whereas denitrification potential was determined using soil slurries.  
13 Compared to the non-amended treatment, denitrification rate increased 2.7- and 3.4-fold  
14 when soil cores collected from both N-exposed and non-exposed sites were amended  
15 with 30 and 60  $\mu\text{g NO}_3\text{-N g}^{-1}$  soil, respectively. Net  $\text{N}_2\text{O}$  emissions were 1.5 and 1.7  
16 times higher from the N-exposed sites compared to the non-exposed sites at 30 and 60  $\mu\text{g}$   
17  $\text{NO}_3\text{-N g}^{-1}$  soil amendment rates, respectively. Similarly, denitrification potential  
18 increased 17 times in response to addition of 15  $\mu\text{g NO}_3\text{-N g}^{-1}$  in soil slurries. The  
19 addition of  $\text{PO}_4$  (5  $\mu\text{g PO}_4\text{-P g}^{-1}$ ) to soil slurries and intact cores did not affect  
20 denitrification rates. These observations suggest that prolonged N loading did not affect  
21 the denitrification potential of the riparian forest soils; however, it did result in higher  
22  $\text{N}_2\text{O}$  emissions compared to emission rates from non-exposed forests.

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24

## 1 **Introduction**

2           Extensive agricultural activities accompanied by the use of nitrogen (N) fertilizer  
3 have resulted in higher concentration of nitrate ( $\text{NO}_3$ ) in surface waters in the U.S.  
4 (Vitousek et al. 1997; Mitsch et al. 2001; Turner and Rabalais 2003). Among agricultural  
5 activities, ornamental plant nurseries use more fertilizer than is used to cultivate row  
6 crops in the U.S. (Colangelo and Brand 2001). Both  $\text{NO}_3$  and ammonium ( $\text{NH}_4$ ) are  
7 highly prone to leaching from soilless growing media in plant nurseries under intensive  
8 irrigation regimes (Harris et al. 1997). Loss of mineral N from nurseries occurs  
9 intermittently after irrigation or heavy rainfall (Harris et al. 1997; Colangelo and Brand  
10 2001). The N-laden runoff often flows across the nursery to finally reach bodies of water,  
11 contributing to the increasing reactive N load of surface and groundwater resources of the  
12 country (Galloway et al. 2004). Higher  $\text{NO}_3$  concentration in the rivers of the U.S. is a  
13 major cause of eutrophication in coastal waters (Turner and Rabalais 1994; Day et al.  
14 2003).

15           Denitrification, or reduction of  $\text{NO}_3$  to  $\text{N}_2\text{O}$  and  $\text{N}_2$  gases, is one of the major  
16 microbial processes in riparian forest soils (Hunter and Faulkner 2001). It occurs under  
17 anaerobic conditions in which organic carbon is used as an energy source and  $\text{NO}_3$  as the  
18 terminal electron acceptor by heterotrophic soil bacteria (Tiedje, 1982). Riparian forest  
19 soils have greater potential to denitrify  $\text{NO}_3$  than surrounding agricultural lands (Lindau  
20 et al. 1994; Delaune et al. 1996). Use and restoration of riparian forests as a nutrient  
21 management tool for removing  $\text{NO}_3$  from agricultural and urban runoff is highly  
22 recommended to protect and improve water quality in the U.S. (Mitsch et al. 2001; Day et  
23 al. 2003).

1           Although riparian soils denitrify  $\text{NO}_3$  at higher rates due to saturated soil  
2 conditions and greater quantities of microbially available carbon,  $\text{NO}_3$  content under  
3 normal conditions can be limiting (Lowrance et al. 1995). Thus, an external source of  
4  $\text{NO}_3$  is needed to maintain high denitrification rates (Ullah et al. 2005) in these soils.  
5 Such loading of runoff  $\text{NO}_3$  into N-limited riparian forests markedly enhances  
6 denitrification rates (DeLaune et al. 1996), but it is not clear whether chronic exposure to  
7 higher  $\text{NO}_3$  runoff has a positive or negative impact on denitrifier activity in soils  
8 (Smolander et al. 1994; Hanson et al. 1994a; Ettema et al. 1999). Bowden et al. (2004),  
9 Compton et al. (2004), and Wallenstein et al. (2006), observed significantly reduced  
10 microbial biomass carbon and activity in N-enriched temperate forest soils compared to  
11 control plots. This suggests that prolonged exposure of natural ecosystems to N can  
12 influence important microbial functions in soil. Discerning the effects of chronic  $\text{NO}_3$   
13 loading on denitrifier activity in riparian forest soils is crucial to quantify the potential of  
14 riparian buffers to remove  $\text{NO}_3$ . As denitrification is extremely variable both temporally  
15 and spatially (Groffman et al. 1991), it would be useful to investigate the effects of  
16 episodic higher  $\text{NO}_3$  loading, as occurs from plant nursery runoff after irrigation or  
17 rainfall, on denitrification rates of riparian forest soils (Groffman, et al. 1991). Such  
18 information would help to develop nutrient management strategies for agricultural runoff.

19           The relative amounts of  $\text{N}_2\text{O}$  and  $\text{N}_2$  gases produced during denitrification in soils  
20 (Skiba et al. 1998) depends mainly on soil moisture, available carbon substrate, and  $\text{NO}_3$   
21 concentration (Breitenbeck et al. 1980; Linn and Doran 1984; Skiba et al. 1998). Higher  
22 soil moisture and available organic carbon substrate promote complete reduction of low  
23 to moderate levels of  $\text{NO}_3$  to  $\text{N}_2$  gas, thus reducing the net amount of  $\text{N}_2\text{O}$  produced

1 (Linn and Doran 1984; Ullah et al. 2005). Higher levels of soil  $\text{NO}_3$ , however, result in  
2 higher net  $\text{N}_2\text{O}:\text{N}_2$  gas emission ratios, since reduction of  $\text{NO}_3$  compared to  $\text{N}_2\text{O}$  is more  
3 energy efficient and is favored by denitrifiers (Breitenbeck et al. 1980; Ullah et al. 2005).  
4 Thus, denitrification in riparian forest soils exposed to prolonged  $\text{NO}_3$  runoff may result  
5 in higher net  $\text{N}_2\text{O}$  emissions (Fenn et al. 1998).  $\text{N}_2\text{O}$  is a 'greenhouse gas' that can induce  
6 310 times more global warming than  $\text{CO}_2$  on a mole-per-mole basis and thus can upset  
7 the credits gained from atmospheric  $\text{CO}_2$  sequestration in these ecosystems (IPCC 1996;  
8 Yu et al. 2004). Moreover,  $\text{N}_2\text{O}$  is also a major contributor in depleting stratospheric  
9 ozone (IPCC 1996). Current efforts to sequester atmospheric  $\text{CO}_2$  into restored riparian  
10 wetland soils may be jeopardized by increased  $\text{N}_2\text{O}$  emissions from these same  
11 ecosystems. There is an acute paucity of data on  $\text{N}_2\text{O}$  emissions from riparian forests in  
12 the northeastern U.S. (Groffman et al. 2000a), particularly from those exposed to  
13 prolonged  $\text{NO}_3$  loading. Lack of data on the dynamics of  $\text{N}_2\text{O}$  emissions from riparian  
14 forests has hampered efforts to accurately measure and model  $\text{N}_2\text{O}$  emission factors from  
15 riparian zones for nitrogen cycling budgeting on a landscape scale (Groffman et al.  
16 2000a).

17 In addition to  $\text{NO}_3$ , agricultural runoff also carries phosphorus (P), which, as a  
18 pollutant, can affect water quality and other factors in aquatic ecosystems (Silvan et al.  
19 2003; Sudareshwar et al. 2003). Since P is an integral part of the microbial biomass in  
20 soils, prolonged P loading into riparian forest soils may affect the activity of soil  
21 microbes, including denitrifiers (Silvan et al. 2003; Meyer et al. 2005). There are  
22 conflicting reports on the effect of soil P level on the activity of denitrifiers. Sudareshwar  
23 et al. (2003) observed a decrease in denitrification rates when coastal wetland soils were

1 amended with P compared to soils with limited P; alternatively, Federer and Klemetsson  
2 (1988) and White et al. (2001) did not observe any effect of additional P on denitrifier  
3 activity in upland forest and Florida Everglade wetland soils, respectively. It would of  
4 interest to know if prolonged P loading of riparian forest soils impacts denitrifier activity.

5 In this study, we compared the effect of additional  $\text{NO}_3$  on denitrification and net  
6  $\text{N}_2\text{O}$  emission rates from riparian forest soils exposed to prolonged mineral N loading  
7 from plant nurseries. In addition, the impact of phosphate amendments on denitrification  
8 rates at selected sites was also evaluated.

## 9 **Material and Methods**

### 10 *Study sites*

11 Four riparian forest sites were identified in southern New Jersey in the upper  
12 Cohansey River watershed (located between  $75^\circ 5'$  to  $75^\circ 20'$  W longitude and  $39^\circ 22'$  to  
13  $39^\circ 35'$  N latitude). Two of the sites, Loew forest (LF) and Centerton forest (CF), were  
14 exposed to nutrient runoff from surrounding plant nurseries for a period of 10 years. The  
15 other two sites, Natural forest (NF) and Harmony forest (HF), are located within 0.5 and  
16 3 miles of the LF site and did not receive runoff from surrounding nurseries or landscapes  
17 for this period. As such, these sites are considered as non-exposed in terms of chronic  
18 mineral N loading from the surrounding acreage. Atmospheric N deposition in New  
19 Jersey range from  $3.6$  to  $7.8 \text{ kg N ha}^{-1} \text{ y}^{-1}$  (Dighton et al. 2004). This range of  
20 atmospheric N deposition in the region is considered elevated due to increased fossil fuel  
21 combustion and fertilizer production and use in the past 50 years (Fenn et al. 1998;  
22 Venterea et al. 2003). This may have deleterious impacts on soil N cycling in riparian

1 forest soils in southern New Jersey, in addition to the nursery run-off N entering into  
2 some of the riparian buffers.

3         Runoff reaching the N-exposed sites arose mainly from frequent over-head  
4 sprinkler irrigation (at least twice-weekly from May to September) and rainfall from 150  
5 acres of container grown and field nursery crops (LF) or 200 acres of container grown  
6 crops (CF). The runoff entered the LF site through a drainage PVC pipe and the CF site  
7 through a drainage ditch. Four replicate samples of runoff water were analyzed for NO<sub>3</sub>  
8 concentration at both locations in May and June, 2005 using the Flow Injection Analyzer  
9 at the Rutgers University Soil Analysis laboratory. The average NO<sub>3</sub> load of drainage  
10 entering the LF site was 15.0 and 8.2 mg L<sup>-1</sup> while that entering the CF site was 3.0 and  
11 12.5 mg NO<sub>3</sub> L<sup>-1</sup>, which in some cases exceeded the EPA water quality standard of 10 mg  
12 L<sup>-1</sup> (EPA 2004).

13         Due to lack of availability of analytical data on the extent and duration of run-off  
14 nitrate entering these sites, an indirect approach was adopted. Pools of N in soil and  
15 foliar litter were investigated for signs of prolonged nitrogen exposure and saturation. An  
16 increase in foliar nitrogen content, nitrification rates and NO<sub>3</sub> leaching from forests in  
17 response to chronic N loading are the established primary indicators of N saturation  
18 (Aber et al. 1989; Magill et al. 2000).

19         The soils in the four sites range in texture from silty clay loam to loamy sand. All  
20 supported mature forests, not used for commercial forestry, that were dominated by  
21 mature stands of hardwood tree species of white oak (*Quercus alba*), northern red oak (*Q.*  
22 *rubra*), red maple (*A. rubrum*), silver maple (*A. saccharinum*), willow oak (*Q. phellos*),  
23 pin oak (*Q. palustris*), and American holly (*Ilex opaca*). Other non-dominant tree species

1 in these forests are green ash (*Fraxinus pennsylvanica*), white ash (*F. americana*), yellow  
2 poplar (*Liriodendron tulipifera*), sweet gum (*Liquidamber styraciflua*), American elm  
3 (*Ulmus americana*), and bitternut hickory (*Carya cordiformis*). The LF site was infested  
4 with reeds (*Phragmites australis*), growing as a sub-canopy under the hardwood trees,  
5 that were concentrated along the nursery runoff flow path within the site. The CF site had  
6 relatively higher snag density and woody debris biomass than the other sites. Selected  
7 physico-chemical properties of the four sites are shown in Table 1. Consistently higher  
8 potential nitrification rates, % foliar N and soil mineral N, and lower C:N ratios in the N-  
9 exposed sites compared to the non-exposed sites shows that the LF and CF sites were  
10 exposed to prolonged mineral N loading (Table 1).

### 11 ***Soil sampling***

12 Four replicate 1 m<sup>2</sup> sampling plots were randomly located at each site. Plots at the  
13 LF and CF sites were located in forest areas inundated by the nursery runoff sheet flow.  
14 To avoid edge effects on soil characteristics, the randomly placed plots were situated in a  
15 line at least 16 m down the boundary of the surrounding land uses and the forest. Unusual  
16 features such as hoof prints, small depressions, large surface debris, and other unusual  
17 micro-features were avoided during sampling.

18 Soil cores and bulk soil samples used for determination of denitrification, net N<sub>2</sub>O  
19 emission rates, microbial biomass C and N and other relevant physico-chemical  
20 properties were collected on May 19, 20, 30, and June 18, 2005 from the LF, NF, HF, and  
21 CF sites respectively. To avoid high initial soil NO<sub>3</sub> concentration, cores from the LF and  
22 CF sites were collected on dates when no nursery runoff was entering the sampling plots.  
23 At each sampling plot, 9 intact soil cores (6 cm dia. x 10 cm length) were collected in

1 plastic liners (6 cm dia. x 15 cm length) using a slide hammer (AMS core sampler®,  
2 American Falls, Idaho). The collected cores were capped at both ends. An additional soil  
3 core (0-10 cm soil depth) was collected from each plot in bronze liners (6 cm dia. x 10  
4 cm length) for determination of bulk density and moisture content. Finally, 4 soil cores  
5 (0-10 cm soil depth) were collected and composited using a mud auger (4.4 cm dia.) for  
6 analysis of physico-chemical properties, a potential denitrification enzyme assay, and  
7 concentrations of nitrate and ammonium. The % water-filled pore space (WFPS) of all  
8 the cores collected from the LF, NF, CF and HF sites was 100, 100, 80 and 83%,  
9 respectively, at the time of sampling. The %WFPS of the soil samples were determined  
10 according to Ullah et al. (2005). The intact cores and bulk soil samples were transferred  
11 to the laboratory on ice and refrigerated until use.

12 Soil cores used for potential net N mineralization and nitrification rates were  
13 collected from all sampling plots during the last week of October, 2005. Duplicate, intact  
14 soil cores (10 cm long) were obtained as described above and transferred to the  
15 laboratory on ice, where they were refrigerated until use.

#### 16 ***Potential denitrification assay***

17 Potential denitrification was determined using soil slurries according to Hunter  
18 and Faulkner (2001). Field moist soils (10 g dry-soil weight basis) were weighed into  
19 four 150 ml serum bottles from each bulk soil sample and were assigned randomly to one  
20 of the four treatments – unamended control, 5  $\mu\text{g PO}_4 \text{ g}^{-1}$  soil, 15  $\mu\text{g NO}_3\text{-N g}^{-1}$  soil, and  
21 15  $\mu\text{g NO}_3\text{-N} + 5 \mu\text{g PO}_4 \text{ g}^{-1}$  soil in a factorial design. For each treatment 4 replicates  
22 were used. After weighing soils in serum bottles, 10 ml of  $\text{PO}_4$  solution delivering 5  $\mu\text{g}$   
23  $\text{PO}_4 \text{ g}^{-1}$  soil (as  $\text{KH}_2\text{PO}_4$ ) was added to 4 bottles each labeled as  $\text{PO}_4$  only and  $\text{PO}_4 + \text{NO}_3$ .

1 The remaining 8 bottles received 10 ml of DI water. The bottles were closed with rubber  
2 stoppers and shaken for 10 minutes to make slurry. After shaking, the rubber stoppers  
3 were removed and the bottles were wrapped in aluminum foil and allowed to equilibrate  
4 for 48 hours. It was assumed that 48 hours duration would be sufficient to expose  
5 microbes in the slurry to the added  $\text{PO}_4$  for cellular incorporation, keeping in mind the  
6 rapid turnover (in the order of hours) and assimilation of  $\text{PO}_4$  by the phosphate  
7 accumulating microbes in the soil (Meyer et al. 2005).

8 After 48 hours, 10 ml of a  $\text{NO}_3$  solution (as  $\text{KNO}_3$ ) was administered to 4 bottles  
9 each labeled as  $\text{NO}_3$  only and  $\text{PO}_4 + \text{NO}_3$  treatments, while 10 ml DI water was added to  
10 the remaining 8 bottles. Bottles were then capped using serum septa and purged with  $\text{O}_2$ -  
11 free  $\text{N}_2$  gas for 25 minutes to induce anaerobic conditions. After purging, 10% of the  
12 headspace was replaced with acetylene ( $\text{C}_2\text{H}_2$ ) gas that had been purified in concentrated  
13  $\text{H}_2\text{SO}_4$  solution and DI water sequentially for the removal of acetone. After the addition  
14 of  $\text{C}_2\text{H}_2$ , the bottles were wrapped in aluminum foil and shaken continuously for 6 hours  
15 on a reciprocating shaker at room temperature (appx. 22 °C). Headspace gas samples (9  
16 ml) were collected from the bottles after 0 and 6 hours using a hypodermic needle  
17 attached to a syringe. The gas samples were injected into 5 ml Becton Dickinson  
18 Vacutainers to maintain a high internal pressure to avoid any diffusion of outside air into  
19 the Vacutainers. The gas samples were analyzed within one week of collection on a  
20 Shimadzu GC-14A gas chromatograph equipped with an electron capture detector. The  
21 rate of  $\text{N}_2\text{O}$  production, determined from the rate of accumulation of  $\text{N}_2\text{O}$  in the  
22 headspaces of the bottles, was corrected for dissolved  $\text{N}_2\text{O}$  in the slurry using the Bunsen  
23 absorption coefficient of 0.54 (Tiedje 1982). Denitrification potential was converted to an

1 area basis (while accounting for differences in bulk density of the four sites) and is  
2 reported as  $\mu\text{g N m}^{-2} \text{ h}^{-1}$ .

### 3 *Denitrification and net N<sub>2</sub>O emission rates from soil cores*

4 Denitrification and net N<sub>2</sub>O emission rates were determined on intact soil cores  
5 brought to room temperature and incubated for 24 hours. The purpose was to quantify the  
6 response of these soils in terms of denitrification and net N<sub>2</sub>O emissions within the first  
7 24 hours of NO<sub>3</sub> loading. The 24 hours duration was chosen to simulate a hydrologic  
8 retention time of 24 hours of the loaded NO<sub>3</sub> into the riparian soils due to runoff. The 9  
9 cores collected from each sampling plot were randomly assigned to groups of three cores  
10 each. One set was randomly selected for measuring net N<sub>2</sub>O flux while the remaining 2  
11 sets were prepared for measuring denitrification rate with and without an added PO<sub>4</sub>  
12 amendment. The set to receive additional PO<sub>4</sub> was amended with a 5 ml phosphorus  
13 solution to deliver 5  $\mu\text{g PO}_4 \text{ g}^{-1}$  soil, while the remaining cores received 5 ml DI water.  
14 All sets of cores were covered and equilibrated for 48 hours to give sufficient time for  
15 microbes in the PO<sub>4</sub> amended treatment to be exposed to the added PO<sub>4</sub>. After 48 hours, a  
16 5 ml solution containing 0, 30, or 60  $\mu\text{g NO}_3\text{-N g}^{-1}$  was administered to one core within  
17 each set. A syringe was used to evenly distribute the NO<sub>3</sub> solution to the surface of the  
18 core. The WFPS of each core was brought to 100% by adding DI water to the cores  
19 where WFPS was less than 100%. This was done to simulate a sudden increase in NO<sub>3</sub>  
20 loading of the riparian soil under saturated soil conditions, delivered by nursery runoff  
21 after an irrigation or rainfall event. After amendment with NO<sub>3</sub>, purified C<sub>2</sub>H<sub>2</sub> gas was  
22 injected into the two sets of cores selected for determination of denitrification rate.  
23 Approximately 10 ml C<sub>2</sub>H<sub>2</sub> gas was injected directly into the cores at the liner and soil

1 column interface in small aliquots using a syringe fitted with a 16 gauge 10-cm long  
2 needle. This was done to ensure a rapid and even diffusion of  $C_2H_2$  gas into the soil pore  
3 space. The purpose of injection of  $C_2H_2$  at the liner and soil column interface instead of  
4 the middle of the columns was to avoid disturbance to the soil column. After  $C_2H_2$   
5 injection, the cores were sealed with airtight seals fitted with rubber septa for gas  
6 sampling. The headspace in the closed column was replaced with an additional 5 ml  $C_2H_2$   
7 gas to achieve an approximate 10%  $C_2H_2$  gas concentration in the column. The last set of  
8 cores selected for net  $N_2O$  emission were sealed with airtight caps without the addition of  
9  $C_2H_2$  gas. Soil cores incubated with and without additional  $C_2H_2$  gas were used to  
10 estimate denitrification and net  $N_2O$  emission rates. Gas samples, collected after 0 and 24  
11 hours of incubation from the closed column headspace using a syringe, were analyzed on  
12 a gas chromatograph for concentration of  $N_2O$  as described in the previous section. The  
13 rates of denitrification and net  $N_2O$  emissions determined are reported as  $\mu g N m^{-2} h^{-1}$ .

#### 14 ***Microbial biomass carbon and nitrogen***

15 Bulk soil samples collected from the four sites were used for the determination of  
16 microbial biomass C according to Voroney et al. (1993). Four replicate (25 g field-moist  
17 soils) soil samples were fumigated in a desiccator for 24 hours to kill and lyse microbial  
18 cells in the soil. The fumigated and a similar set of non fumigated soils (4 replicates each  
19 for each forest site) were extracted with 0.5 M  $K_2SO_4$  solution for soluble organic carbon  
20 (C) concentration at 1:8 soil to  $K_2SO_4$  solution ratio . The extracts were filtered through  
21 No. 42 Whatman filter paper into 20 ml vials and analyzed using a Shimadzu TOC  
22 analyzer for determination of soluble organic C. Before analysis, samples were diluted by  
23 a factor of 4 to reduce the concentration of  $K_2SO_4$  salts in the extracted samples because

1 salt passing through the TOC analyzer can clog the beaded column. The amount of  
2 microbial biomass C was calculated as the difference of soluble organic C between  
3 fumigated and unfumigated soils divided it by a correction factor ( $K_{EC} = 0.40$ ) to account  
4 for the efficiency of fumigation-extraction of the microbial C. Microbial biomass N was  
5 determined using the chloroform fumigation-incubation technique according to Voroney  
6 and Paul (1984). Four replicate (25g field-moist soils) samples from each forest site were  
7 fumigated in a desiccator for 24 hours as described above. The fumigated samples were  
8 inoculated with fresh soil for 10 days at room temperature ( $\sim 22^{\circ}\text{C}$ ) to allow  
9 mineralization of organic N in the sample including that in the lysed microbial cells. A  
10 similar set of non fumigated samples (4 replicates for each forest site) were also  
11 incubated with the fumigated samples. After the 10 days incubation, the samples were  
12 extracted with 2M KCL for mineral N concentration determination. Microbial biomass N  
13 was calculated as the difference in mineral N in fumigated and non fumigated soils  
14 divided by a correction factor ( $K_{EN} = 0.30$ ) to account for the efficiency of microbial N  
15 extraction. Both the microbial biomass carbon and nitrogen are reported as  $\mu\text{g C or N g}^{-1}$   
16 dry soil.

### 17 *Selected physico-chemical properties of soils*

18 Gravimetric soil moisture content, bulk density, total porosity, water-filled pore  
19 space, soil particle size distribution, soil pH, mineral nitrogen, water-soluble organic  
20 carbon, and total soil C and N were determined on bulk soil samples according to Ullah  
21 et al. (2005). Total soil P content was determined using Mehlich 3 method of soil  
22 extractable nutrients.

23

### 1 ***Potential net N mineralization and nitrification rates***

2           One of the duplicate soil cores from each sampling plot collected in October,  
3 2005 was homogenized thoroughly by hand, and a 5 g sub-sample was extracted with 2  
4 M KCL solution for the determination of initial mineral N concentration. The WFPS of  
5 the remaining soil cores was adjusted to 100% by adding DI water to the top of the cores.  
6 The cores were covered with a loose cap to allow for air exchange and to reduce the loss  
7 of water vapor and were then placed in a box to incubate in the dark at 20 °C for 28 days  
8 (Hart et al. 1994). These cores were incubated at 100% WFPS to simulate conditions  
9 similar to the cores incubated for the determination of denitrification rates. Following the  
10 incubation period, the cores were removed from the plastic liners and homogenized  
11 thoroughly by hand. A 5 g sub-sample of the homogenized soil was extracted with 2 M  
12 KCL solution for the determination of mineral N. Net nitrogen mineralization and  
13 nitrification rates were calculated from the difference in the amount of initial and final  
14 mineral N content (Hart et al. 1994). Net nitrogen mineralization and nitrification rates,  
15 are reported as  $\text{ng N g}^{-1} \text{ dry soil h}^{-1}$ .

### 16 ***Foliar Nitrogen***

17           Eight replicate samples of fresh leaf litter were collected from each 1 m<sup>2</sup> plots at  
18 the four forest sites on October 30, 2005. The samples were oven-dried at 65 °C for 5  
19 days. The dried samples were pulverized and analyzed on a LECO N analyzer using a  
20 thermoconductivity detector for the determination of foliar N, which is reported as % N  
21 on dried mass basis (Table 1).

22

23

## 1 *Statistical Analysis*

2 All data were analyzed using SAS V-8.3 (SAS Inc. 2000). Within-site differences  
3 in denitrification and net N<sub>2</sub>O emission rates of soils amended at 0, 30, and 60 μg NO<sub>3</sub> g<sup>-1</sup>  
4 soil were done using analysis of variance (ANOVA) using the General Linear Model.  
5 Fisher's protected LSD was used for post hoc comparisons at  $\alpha = 0.05$ . Similarly,  
6 ANOVA was also used for between-site comparison of denitrification, net N<sub>2</sub>O emission  
7 and N mineralization and nitrification rates. To elucidate any effect of PO<sub>4</sub> amendment  
8 on denitrification rate, a two-sample T test was done using the pooled variance technique  
9 at  $\alpha = 0.05$ . A multiple regression model using the backward-selection option was used  
10 to identify predictor variables that significantly affect denitrification and net N<sub>2</sub>O  
11 emission rates from the selected sites. The data was analyzed to meet the normal  
12 distribution assumption of ANOVA and regression using the Proc Univariate procedure  
13 at Shapiro-Wilk significance of  $p > 0.05$ . Pearson correlation coefficients between  
14 various microbial and physio-chemical characteristics of the sites were determined using  
15 SAS.

## 16 **Results**

### 17 *Potential denitrification assay*

18 The potential denitrification rate of riparian soils either exposed or not exposed to  
19 mineral N loading from nursery runoff increased significantly ( $p < 0.05$ ) when amended  
20 with 15 μg NO<sub>3</sub> g<sup>-1</sup> soil alone or in combination with PO<sub>4</sub> (Figure 1). The addition of PO<sub>4</sub>  
21 had no effect on potential denitrification in soils from any of the sites. A significant  
22 response of these soils to added NO<sub>3</sub> in terms of increased denitrification depicts a

1 limitation of this process by available  $\text{NO}_3$  even after prolonged exposure of the LF and  
2 CF sites to mineral N loading.

### 3 *Denitrification and net $\text{N}_2\text{O}$ emission rates from soil cores*

4       When intact soil cores were amended with  $30 \mu\text{g NO}_3 \text{ g}^{-1}$  soil, samples from all  
5 the sites responded with a significant increase in denitrification rate compared to non  
6 amended soils (Table 2), showing that denitrification in these sites is limited by  $\text{NO}_3$  in a  
7 manner similar to that found in Figure 1. The denitrification rates observed among sites  
8 amended with  $30 \mu\text{g NO}_3 \text{ g}^{-1}$ , however, did not significantly differ ( $p > 0.05$ ). Although  
9 denitrification rate was further increased in soils amended with  $60 \mu\text{g NO}_3 \text{ g}^{-1}$ , this was  
10 not significant except in soil from the NF site. The addition of  $5 \mu\text{g PO}_4 \text{ g}^{-1}$  soil made  
11 little difference in denitrification rate (Table 3

12       The addition of  $30 \mu\text{g NO}_3 \text{ g}^{-1}$  soil to soil cores collected from all riparian sites  
13 increased net  $\text{N}_2\text{O}$  emissions by an average of 15-fold compared to the unamended  
14 treatment (Table 4). However,  $\text{N}_2\text{O}$  emission rates averaged from soils collected from the  
15 N-exposed sites ( $22.5 \mu\text{g N m}^{-2} \text{ h}^{-1}$ ) were 1.5 times those of the non-exposed sites ( $14.5$   
16  $\mu\text{g N m}^{-2} \text{ h}^{-1}$ ) at  $30 \mu\text{g NO}_3 \text{ g}^{-1}$  amendment level. With  $60 \mu\text{g g}^{-1}$  additional  $\text{NO}_3$ , net  $\text{N}_2\text{O}$   
17 emissions increased significantly ( $p < 0.05$ ) compared to the  $30 \mu\text{g NO}_3 \text{ g}^{-1}$  treatment in  
18 soils from the N-exposed sites. Moreover,  $\text{N}_2\text{O}$  emission rates from the N exposed sites  
19 were on average 1.6 times higher ( $p < 0.05$ ) than  $\text{N}_2\text{O}$  emission rates from the non-  
20 exposed sites (Table 4).

21       Soluble organic carbon (SOC) was a key predictor variable of denitrification  
22 (multiple linear regression) in soils from the four riparian forest sites when amended with  
23  $30$  and  $60 \mu\text{g NO}_3 \text{ g}^{-1}$  soil, respectively (Figures 2 and 3). SOC accounted for 30% of the

1 variability in denitrification rate (denitrification in  $\mu\text{g N m}^{-2} \text{h}^{-1} = 294 + 0.58 \text{ SOC in } \mu\text{g}$   
2  $\text{C g}^{-1} \text{ soil}$ ) for the  $30 \mu\text{g NO}_3 \text{ g}^{-1}$  treatment, whereas this factor accounted for only 55% of  
3 the variability at the  $60 \mu\text{g NO}_3 \text{ g}^{-1}$  amendment level (denitrification in  $\mu\text{g N m}^{-2} \text{h}^{-1} = 199$   
4  $+ 1.70 \text{ SOC in } \mu\text{g C g}^{-1} \text{ soil}$ ). SOC controls denitrification rates in these sites once the  
5 process is not limited by  $\text{NO}_3$  availability. Unlike denitrification, no single strong  
6 predictor variable of  $\text{N}_2\text{O}$  flux from these forests was identified due to greater variability  
7 of the flux rates and the complex interactions of the predictor variables in regulating the  
8 flux- a condition encountered by other researchers (Smith et al. 1995; Groffman, et al.  
9 2000b). The combination of various predictor variables accounted for 93%, 48% and  
10 83% variability in net  $\text{N}_2\text{O}$  emissions at zero, 30 and  $60 \mu\text{g NO}_3 \text{ g}^{-1}$  amendment levels,  
11 respectively. Among these variables, microbial biomass nitrogen, total soil nitrogen and  
12  $\text{NH}_4$  concentration correlated positively with net  $\text{N}_2\text{O}$  emissions in the regression models.  
13 This suggests that an increases in different pools of soil nitrogen due to chronic N loading  
14 can increase  $\text{N}_2\text{O}$  emissions during denitrification.

### 15 ***Microbial biomass carbon and nitrogen***

16 Compared to soils from sites exposed to nursery runoff, relatively higher soil C:N  
17 ratio and microbial biomass C in the soils from sites not exposed to nursery runoff (Table  
18 1) indicates a higher pool of labile C available to denitrifiers, resulting in higher  
19 denitrification and lower net  $\text{N}_2\text{O}$  emission rate. Microbial biomass carbon, SOC, and  
20 total soil C correlated significantly with denitrification rate, whereas microbial biomass  
21 N, total soil N,  $\text{NH}_4$ , and C:N ratios correlated significantly with net  $\text{N}_2\text{O}$  emission (Table  
22 5).

### 23 ***Potential net N mineralization and nitrification rates***

1 Potential net nitrogen mineralization rates were not significantly different in soils  
2 collected from the four riparian forest sites ( $p > 0.05$ ). Potential net nitrification rate,  
3 however, differed significantly ( $p < 0.05$ ) between N-exposed and non-exposed sites  
4 (Table 1). The N-exposed sites had 8.4 times higher nitrification rates than those  
5 observed in the non-exposed sites. Total foliar nitrogen content was 1.2 times higher in  
6 leaf litter collected from sample plots on the N-exposed sites than litter collected from  
7 non-exposed sites (Table 1).

## 8 **Discussion**

9 Denitrification rate in soils collected from riparian forest sites either exposed or  
10 not exposed to mineral N loading, increased significantly in all the sites when amended  
11 with  $\text{NO}_3$ . This observation clearly demonstrates that denitrification in soils from these  
12 sites was limited by  $\text{NO}_3$  (Figure 1; Tables 2 and 3) and that prolonged mineral N loading  
13 did not affect the activity of denitrifying microbes in the soils collected from exposed  
14 sites (LF and CF sites). Hanson et al. (1994a and 1994b) also observed higher  
15 denitrification rates in a N-enriched riparian forest in Rhode Island, and they concluded  
16 that higher denitrification capacity is a key process that moderates the effects of chronic  
17 mineral N enrichment. Average lower soil  $\text{NO}_3$  (Table 1) concentration ( $2.9 \mu\text{g N g}^{-1}$   
18 soil) in the N-exposed sites in spite of chronic run-off input support the observation that  
19  $\text{NO}_3$  removal capacity of these sites is not exhausted by chronic N loading. In a study in  
20 Europe, lower  $\text{NO}_3$  concentrations in groundwater beneath a riparian forest receiving  
21 chronic N run-off was ascribed to higher denitrification rates (Hefting and de Klein  
22 1998), which is in agreement with our results.

1           The observed rates of denitrification (Tables 2 and 3) in soils from all sites were  
2 within the range of denitrification rates in riparian forest soils reported elsewhere in  
3 literature (Lowrance et al. 1995; Jordan et al. 1998; Hefting et al. 1998 and 2003).  
4 However, caution needs to be exercised when extrapolating denitrification rates of the  
5 current study to bigger spatial and temporal scales, since these rates were determined  
6 under controlled laboratory conditions of soil NO<sub>3</sub>, temperature and moisture and thus  
7 may not reflect actual field conditions.

8           As the addition of NO<sub>3</sub> to soil cores increased denitrification, the rate limiting  
9 factor shifted from NO<sub>3</sub> availability to available organic C substrate, especially at 60 µg  
10 NO<sub>3</sub> g<sup>-1</sup> soil treatment. For example, soil from the non-exposed NF site with significantly  
11 higher SOC and total soil C (Table 1) denitrified more NO<sub>3</sub> than the rest of the sites at 60  
12 µg NO<sub>3</sub> g<sup>-1</sup> amendment level. This apparent control of denitrification rates by available C  
13 substrate was found significant using the multiple regression and Pearson's correlation  
14 analyses (Figures 2 and 3; Table 5). Significant control of denitrification rates by  
15 available C substrate in riparian wetlands has been reported elsewhere in the literature  
16 (Lindau, et al. 1994; Lowrance, et al. 1995; DeLaune et al. 1996; Hefting et al. 2003).

17           Microbial biomass C also correlated significantly with denitrification rates (Table  
18 5) supporting the argument that available C exerts a regulatory control on denitrification  
19 rate, as biomass C is one of the sources of the labile C pools in soil. However, it is  
20 noteworthy that the microbial biomass carbon content (Table 1) of the N-exposed sites  
21 was significantly lower than those of the non-exposed sites ( $p < 0.05$ ). Lower microbial  
22 biomass C in the N-exposed sites is thought to be due to the negative effects of  
23 prolonged N exposure. This finding is in agreement with those of Compton et al. (2004),

1 Bowden et al. (2004) and Wallenstein et al. (2006), who observed lower microbial  
2 biomass carbon and activity in N-enriched temperate forest soils in the northeastern U.S.  
3 Wallenstein et al. (2006) also reported a 59 and 52% reduction in microbial biomass C  
4 and substrate-induced respiration, respectively, in soils of a N-saturated temperate forest  
5 compared to a non-saturated forest in New England. Ettema et al. (1999) observed similar  
6 effects of N enrichment on biomass C and activity in riparian forest soils in Georgia.  
7 These authors feared that the denitrifying microbes in riparian forests may be threatened  
8 by the cumulative negative effects of N saturation. Although we found significantly  
9 lower soil microbial biomass C in the N-exposed sites, the current study did not observe  
10 significant differences in denitrification rates among the N-exposed and non-exposed  
11 sites, showing that riparian forests can sustain a high and persistent capacity to denitrify  
12  $\text{NO}_3$  even if exposed to prolonged mineral N loading (Hanson et al. 1994b). Given the  
13 limited temporal coverage of this experiment under optimum laboratory soil moisture and  
14 temperature regimes, further temporally intensive field denitrification assessment studies  
15 of these sites is recommended to validate the current observations.

16 We found no effect of  $\text{PO}_4$  addition on denitrifier activity (Figure 1; Tables 2 and  
17 3), which is commensurate with the results of Federer and Klemetsson (1988) and White  
18 and Reddy (1999). However, our findings are in contrast to those of Sudareshwar et al.  
19 (2003) who reported that P-enrichment of coastal wetland soils reduced denitrification  
20 potential compared to similar non-enriched soils. None of these studies were conducted  
21 on riparian forest soils. Our data suggests that P input to riparian forests from agricultural  
22 run-off will not affect denitrifier activity.

1           Even though denitrification rate in soils amended with additional NO<sub>3</sub> (30 and 60  
2   μg NO<sub>3</sub> g<sup>-1</sup>) varied little among sites (Table 2), net N<sub>2</sub>O emission rates were higher from  
3   soils collected from the N-exposed sites (Table 4). It appears that these differences were a  
4   result of prolonged exposure of the N-exposed sites to nursery run-off. This result is  
5   consistent with the findings of Hefting et al (2003) who reported that N<sub>2</sub>O emissions from  
6   riparian forests receiving chronic N loads were higher compared to emissions from  
7   riparian grasslands, even though denitrification rates of the two ecosystems were similar.  
8   Higher soil N pools, greater potential nitrification rates, and lower soil and microbial  
9   biomass C:N ratios (Table 1) resulting from prolonged N loading in the N-exposed soils  
10   appeared to have reduced soil N<sub>2</sub>O reductase activity, which eventually led to higher N<sub>2</sub>O  
11   emissions compared to emissions from the non-exposed sites. Moreover, prolonged N  
12   exposure resulted in higher nitrification rates in the N-exposed sites (Magill et al. 2000)  
13   compared to the non-exposed sites. This observation is similar to those in other studies  
14   that evaluated N<sub>2</sub>O emissions from temperate forest soils after N fertilization in the  
15   northeastern U.S. (Bowden et al. 1991; Brumme and Beese 1992; Sitaula and Bakken,  
16   1993; Barnard et al. 2005).

17           In findings similar to ours, Hanson et al. (1994b) reported significantly higher  
18   microbial biomass N in a N-enriched riparian forest soil compared to a non-enriched site  
19   (Hanson et al. 1994b), suggesting that prolonged exposure of riparian forests to mineral  
20   N is saturating different soil N pools. The soil N saturation phenomena, including  
21   increases in microbial biomass N and net nitrification rates, may be resulting in relatively  
22   higher N<sub>2</sub>O emissions from riparian forests when loaded with mineral N from agricultural  
23   run-off. Although a significant relationship ( $r = 0.50$ ;  $p < 0.04$ ) found between microbial

1 biomass N and N<sub>2</sub>O emissions from cores amended with 60 μg NO<sub>3</sub>-N g<sup>-1</sup> soil (Table 5),  
2 this does not likely represent a cause and effect relationship. Further studies are needed to  
3 define the relationship between an increase in microbial biomass N and higher N<sub>2</sub>O  
4 emissions in riparian forest soils.

5         In this study, microbial biomass C was significantly lower ( $p < 0.05$ ) in the N-  
6 exposed sites (Table 1) compared to the non-exposed sites, which is in agreement with  
7 the findings of Ettema et al. (1999), Bowden et al. (2004), and Compton et al. (2004).  
8 Concomitant decrease in biomass C with increasing biomass N and increased net  
9 nitrification rates due to prolonged exposure of riparian forests to mineral N loading  
10 strongly suggests that episodic, high levels of NO<sub>3</sub> input into N-saturated riparian forest  
11 soil leads to higher net N<sub>2</sub>O emissions.

12         Soil texture affects N<sub>2</sub>O flux from soils by influencing gas diffusion rates in the  
13 soil profile (Weitz et al. 2001). Compared to coarse-textured soils, fine-textured soils  
14 limit gas diffusion rates, thus enhancing the probability that N<sub>2</sub>O is reduced to N<sub>2</sub> gas by  
15 soil denitrifying organisms (Weitz et al. 2001). Although the N-exposed sites (CF and  
16 LF) were higher in clay (Table 1), net N<sub>2</sub>O emissions from these soils exceeded those of  
17 sites not exposed to additional mineral N loading, supporting our finding that that  
18 prolonged exposure of riparian forest soils to mineral N may have reduced N<sub>2</sub>O reductase  
19 activity. Soil water can also reduce N<sub>2</sub>O diffusion by approximately 4 orders of  
20 magnitude by filling and blocking up soil air pores. This increases the time for microbial  
21 reduction of N<sub>2</sub>O to N<sub>2</sub> gas before its emission into the air (Clough et al. 2005). Saturated  
22 soil conditions of the soil cores at the time of incubation may have obscured the effect of  
23 soil texture on N<sub>2</sub>O emissions from the four sites. We recommend further studies to

1 elucidate the interactive effects of soil moisture and texture on N<sub>2</sub>O emission from soils  
2 to better understand the fate of N<sub>2</sub>O in soils.

3         In our study, N<sub>2</sub>O emission rates in treatments that did not receive additional NO<sub>3</sub>  
4 were within the range or lower than the N<sub>2</sub>O emission rates reported by other studies  
5 from temperate forests in the northeastern U.S. (Bowden et al. 1990, 1991, 2000; Hafner  
6 and Groffman 2005). However, when additional NO<sub>3</sub> is loaded into riparian forests,  
7 which are considered as ‘hotspots’ of denitrification and N<sub>2</sub>O production (Groffman et al.  
8 2000a), N<sub>2</sub>O emission rate increases by a factor of at least 12 or more even under  
9 saturated soil conditions. The increase in N<sub>2</sub>O emissions due to NO<sub>3</sub> loading needs to be  
10 considered when calculating N<sub>2</sub>O emission factors for riparian forests by concerned  
11 agencies (Groffman et al. 2000a) like the Intergovernmental Panel on Climate Change  
12 and the U.S. Department of Energy-National Commission on Carbon Sequestration.

13         In summary, the results of this research show that the denitrification potential of  
14 riparian forest soils is not compromised after chronic exposure to mineral N run-off for  
15 10 years. Moreover, addition of PO<sub>4</sub> does not seem to affect the activity of denitrifying  
16 microbes in these soils. Although riparian soils can substantially contribute to the  
17 reduction of NO<sub>3</sub> loading into water bodies in watersheds dominated by plant nurseries,  
18 these forests will emit relatively more N<sub>2</sub>O into the atmosphere compared to similar soils  
19 not exposed to chronic mineral N run-off. This should be accounted for at the landscape  
20 scale within the wetlands potential carbon-sequestration context. We recommend that  
21 riparian forests be considered as an integral component in developing strategies for NO<sub>3</sub>  
22 removal from nursery run-off in New Jersey and other similar eco-zones in the country.

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1 **List of Figures**

2

3 Figure 1. Mean potential denitrification rate and standard error of soil slurries from  
4 riparian forest soils exposed (LF, CF) or not exposed (NF, HF) to mineral N loading from  
5 nursery runoff.

6

7 Figure 2. Relationship between denitrification rate and soluble organic carbon in soils  
8 from riparian forest soils amended with  $30 \mu\text{g NO}_3 \text{ g}^{-1}$  soil. ( $Y = 294 + 0.58 X$ ).

9

10 Figure 3. Relationship between denitrification rate and soluble organic carbon in soils  
11 from riparian forest soils amended with  $60 \mu\text{g NO}_3 \text{ g}^{-1}$  soil ( $Y = 199 + 1.70 X$ ).

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1 Table 1. Selected soil (0-10cm depth) properties of riparian forest sites exposed to  
 2 mineral N loading from nursery runoff (mean  $\pm$  standard error)

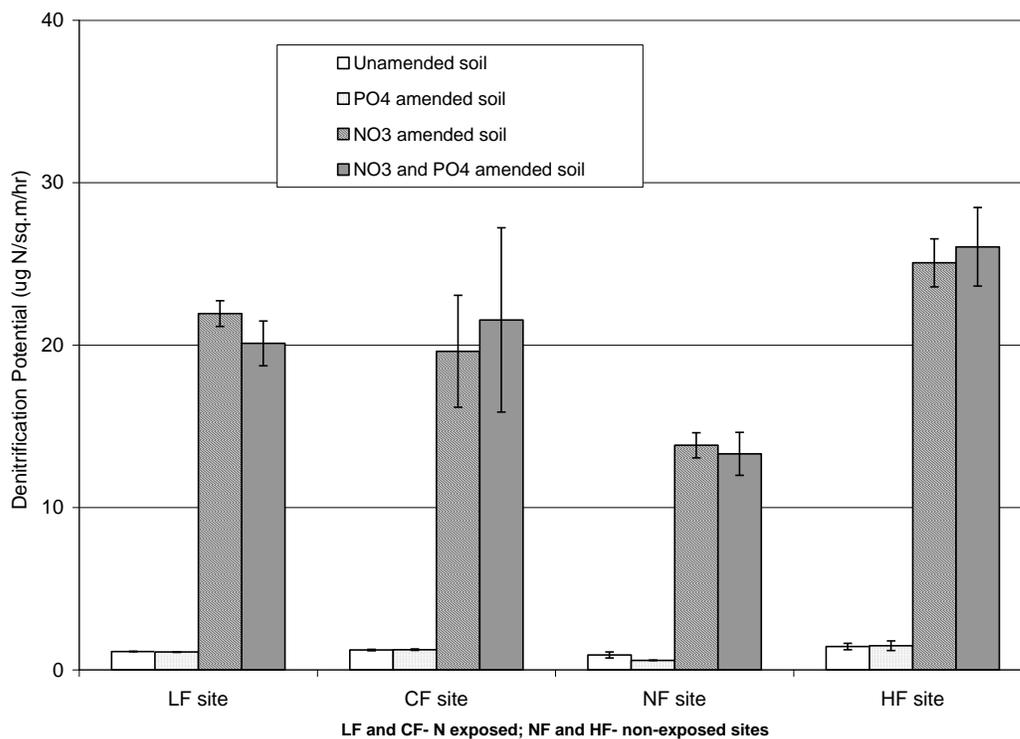
Soil properties	N Exposed sites		Non-exposed sites	
	LF	CF	NF	HF
Clay (%)	39 $\pm$ 1.7	33 $\pm$ 7	8 $\pm$ 1	23 $\pm$ 1.5
Silt (%)	51 $\pm$ 1.3	29 $\pm$ 3	9 $\pm$ 1	54 $\pm$ 9
Soil texture	Silty clay loam	Clay loam	Loamy sand(organic)	Silt loam
Approximate area (acres)	5	15	10	5
Bulk density (g cm <sup>-3</sup> )	0.90 $\pm$ 0.16	0.96 $\pm$ 0.07	0.46 $\pm$ .03	1.05 $\pm$ .05
Porosity (cm <sup>3</sup> cm <sup>-3</sup> )	0.61 $\pm$ 0.06	0.63 $\pm$ 0.02	0.82 $\pm$ 0.01	0.60 $\pm$ .02
Water-filled pore space (%)	100 $\pm$ 27	80 $\pm$ 4	100 $\pm$ 0.20	83 $\pm$ 12
pH	6.3 $\pm$ 0.1	5.4 $\pm$ 0.2	4 $\pm$ 0.1	5.7 $\pm$ 0.2
Soluble organic C ( $\mu$ g g <sup>-1</sup> )	108 $\pm$ 5	163 $\pm$ 18	300 $\pm$ 32	158 $\pm$ 15
Microbial biomass C ( $\mu$ g g <sup>-1</sup> )	713 $\pm$ 65	978 $\pm$ 94	2578 $\pm$ 351	1238 $\pm$ 132
Microbial biomass N ( $\mu$ g g <sup>-1</sup> )	394 $\pm$ 70	383 $\pm$ 75	315 $\pm$ 54	165 $\pm$ 29
Total P ( $\mu$ g g <sup>-1</sup> )	177 $\pm$ 4	222 $\pm$ 36	27 $\pm$ 13	87 $\pm$ 26
NO <sub>3</sub> -N ( $\mu$ g N g <sup>-1</sup> )	2.7 $\pm$ 1.8	3.1 $\pm$ 0.6	0.92 $\pm$ 0.32	1.9 $\pm$ 1.16
NH <sub>4</sub> -N ( $\mu$ g N g <sup>-1</sup> )	41 $\pm$ 5	23 $\pm$ 2	14 $\pm$ 1	8 $\pm$ 1
Total C (% of dry soil)	4.6 $\pm$ 0.60	3.7 $\pm$ 0.50	8.3 $\pm$ 0.64	3.9 $\pm$ 0.20
Total N (% of dry soil)	0.37 $\pm$ 0.03	0.23 $\pm$ 0.03	0.38 $\pm$ 0.03	0.20 $\pm$ 0.01
C:N ratio	12.1	16.0	22.0	19.0
N mineralization rate ( $\mu$ g N g <sup>-1</sup> h <sup>-1</sup> )	74 $\pm$ 28	91 $\pm$ 7	156 $\pm$ 79	98 $\pm$ 45
Nitrification rate ( $\mu$ g N g <sup>-1</sup> h <sup>-1</sup> )	18 $\pm$ 6.1	41 $\pm$ 8.4	4 $\pm$ 1.2	3 $\pm$ 0.9
Foliar N (% mass basis)	1.36 $\pm$ 0.11	1.32 $\pm$ 0.08	1.11 $\pm$ 0.6	1.11 $\pm$ 0.11

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Figure 1. Mean potential denitrification rate and standard error of soil slurries from riparian forest soils exposed (LF, CF) or not exposed (NF, HF) to mineral N loading from nursery runoff.

1  
2 Table 2. Denitrification rate (mean  $\pm$  standard error) of soil from riparian sites exposed  
3 (LF, CF) or not exposed (NF, HF) to N from nursery runoff.

Additional NO <sub>3</sub> ( $\mu\text{g}$ NO <sub>3</sub> g <sup>-1</sup> )	N exposed sites		Non-exposed sites	
	LF	CF	NF	HF
	.....Denitrification rate ( $\mu\text{g N m}^{-2} \text{ h}^{-1}$ ).....			
0	163 $\pm$ 30 a <sup>a</sup>	136 $\pm$ 35 a	147 $\pm$ 09 a	150 $\pm$ 26 a
30	362 $\pm$ 55 b	431 $\pm$ 28 b	458 $\pm$ 21 b	346 $\pm$ 45 b
60	398 $\pm$ 76 b	474 $\pm$ 105 b	674 $\pm$ 104 c	515 $\pm$ 80 b

4 <sup>a</sup> Means followed by same letters in a column show no significant difference ( $p > 0.05$ )  
5 using an ANOVA test.

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2 Table 3. Denitrification rate (mean  $\pm$  standard error) of soil from riparian sites exposed  
 3 (LF, CF) or not exposed (NF, HF) to N from nursery runoff and amended with 5  $\mu\text{g PO}_4$   
 4  $\text{g}^{-1}$  soil.

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Additional $\text{NO}_3$ ( $\mu\text{g NO}_3$ $\text{g}^{-1}$ )	N exposed sites		Non-exposed sites	
	LF	CF	NF	HF
	.....Denitrification rate ( $\mu\text{g N m}^{-2} \text{ h}^{-1}$ ).....			
0	152 $\pm$ 23 a <sup>a</sup>	152 $\pm$ 35 a	90 $\pm$ 12 a	97 $\pm$ 34 a
30	351 $\pm$ 56 b	424 $\pm$ 28 b	425 $\pm$ 35 b	357 $\pm$ 60 b
60	451 $\pm$ 37 b	505 $\pm$ 105 b	625 $\pm$ 37 c	459 $\pm$ 64 b

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<sup>a</sup> Means followed by same letters in a column show no significant difference

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( $p > 0.05$ ) using an ANOVA test.

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Table 4. Net N<sub>2</sub>O emission rates (mean ± standard error) of soil from riparian sites exposed (LF, CF) or not exposed (NF, HF) to N from nursery runoff.

Additional NO <sub>3</sub> (µg NO <sub>3</sub> g <sup>-1</sup> )	N exposed sites		Non-exposed sites	
	LF	CF	NF	HF
	.....Net N <sub>2</sub> O emission rate (µg N m <sup>-2</sup> h <sup>-1</sup> ).....			
0	3 ± 0.6 a	1 ± 1.3 a	1.20 ± 0.5 a	0.8 ± 0.9 a
30	25 ± 1.8 b	20 ± 2.7 b	17 ± 4.7 b	12 ± 2.1 b
60	33 ± 2.7 c	32 ± 3.1 c	22 ± 2.2 b	17 ± 2.8 b

<sup>a</sup> Means followed by same letters in a column show no significant difference (p > 0.05) using an ANOVA test.

1 Table 5. Relationship between denitrification rate and N<sub>2</sub>O emission rate to various soil  
 2 factors (Pearson correlation analysis) in riparian forest soils amended with 0, 30, and 60  
 3 µg NO<sub>3</sub> g<sup>-1</sup> soil.

Additional NO <sub>3</sub> g <sup>-1</sup> soil	Denitrification rate (µg N m <sup>-2</sup> h <sup>-1</sup> )			..N <sub>2</sub> O emission rate (µg N m <sup>-2</sup> h <sup>-1</sup> )..		
	0	30	60	0	30	60
Variables						
Soluble organic C	0.07 <sup>a</sup> (0.78) <sup>b</sup>	0.55* (0.02)	0.74* (.0009)	-0.41 (0.10)	-0.10 (0.70)	0.15 (0.55)
Microbial biomass C	0.08 (0.77)	0.54* (0.03)	0.72* (0.001)	-0.38 (0.14)	0.16 (0.53)	0.10 (0.69)
Microbial biomass N	-0.15 (0.57)	0.20 (0.45)	-0.35 (0.18)	-0.01 (0.94)	0.38 (0.14)	0.50* (0.04)
Total C	0.08 (0.75)	0.26 (0.32)	0.46** (0.07)	-0.06 (0.82)	0.11 (0.67)	-0.04 (0.86)
Total N	0.20 (0.44)	0.11 (0.67)	0.16 (0.54)	0.55* (0.02)	0.24 (0.35)	0.11 (0.66)
C:N ratio	-0.10 (0.66)	0.22 (0.40)	0.48 (0.05)	-0.72* (0.001)	-0.16 (0.54)	0.10 (0.70)
pH	-0.02 (0.92)	-0.41 (0.11)	-0.52* (0.04)	0.59* (0.01)	0.02 (0.92)	-0.18 (0.48)
Total P	-0.22 (0.42)	-0.08 (0.76)	-0.48** (0.06)	0.29 (0.29)	0.43 (0.10)	0.41 (0.12)
NO <sub>3</sub>	0.22 (0.40)	-0.29 (0.26)	-0.43 (0.09)	0.15 (0.57)	0.06 (0.80)	0.25 (0.34)
NH <sub>4</sub>	-0.04 (0.85)	0.05 (0.83)	-0.35 (0.18)	0.84* (0.0001)	0.22 (0.40)	0.02 (0.92)

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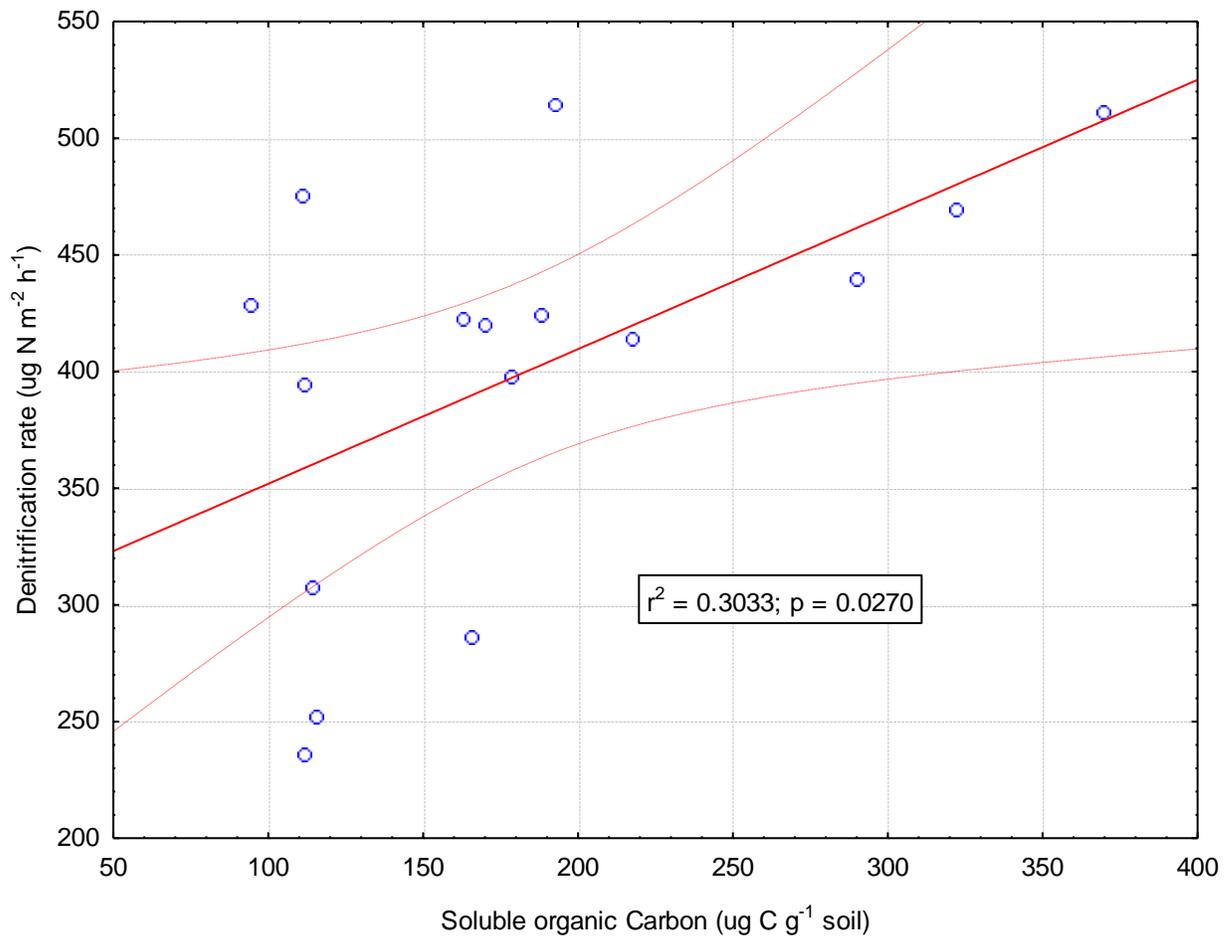
5 <sup>a</sup> Pearson correlation coefficient.

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7 <sup>b</sup> Significance (n =16) at  $p < 0.05$  (\*),  $p < 0.10$  (\*\*), or not significant (no asterisk).

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Figure 2. Relationship between denitrification rate and soluble organic carbon in soils from riparian forest soils amended with 30  $\mu\text{g NO}_3 \text{ g}^{-1}$  soil ( $Y = 294 + 0.58 X$ ).

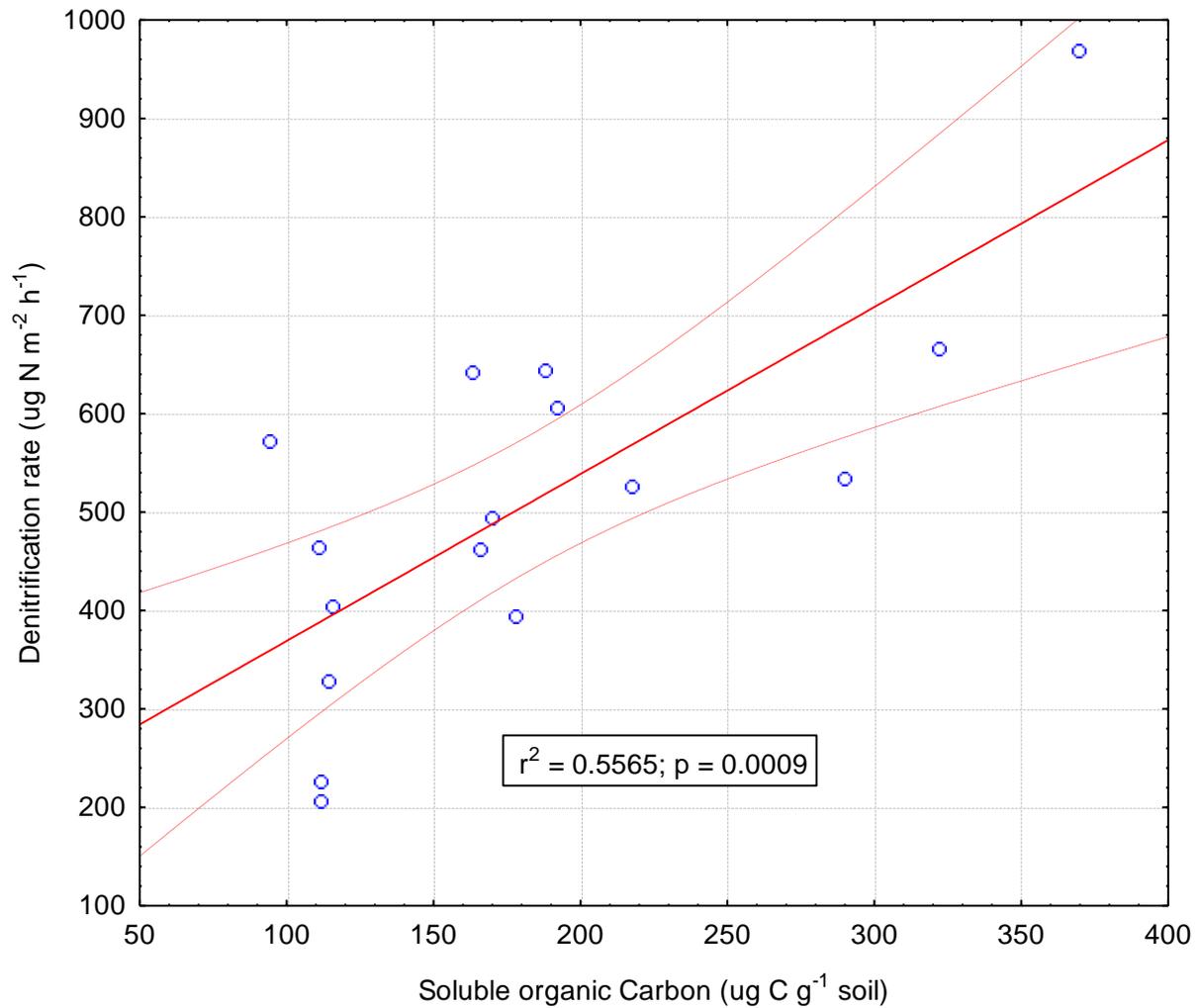
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Figure 3. Relationship between denitrification rate and soluble organic carbon in soils from riparian forest soils amended with 60  $\mu\text{g NO}_3 \text{ g}^{-1}$  soil ( $Y = 199 + 1.70 X$ ).