

1 **Simulating behavior of perfluorooctane sulfonate (PFOS) in**
2 **the mainstream of a river system with sluice regulations**

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24 **Abstract:** Perfluorooctane sulfonate (PFOS) is a persistent, long-range transport,
25 anionic, and ubiquitous contaminant. Its behavior has attracted wide-ranging scientific
26 and regulatory attention. In this article, the concentrations of PFOS in both freshwater
27 and sediment, and the behavior of PFOS between freshwater and sediment were
28 simulated based on a mass balance approach in the mainstream of Haihe river system
29 combining with sluices and artificial rivers. Annual emission discharged into each river
30 was estimated and North Canal was the maximum receiver in the study. The simulated
31 concentrations of PFOS in both steady and unsteady states agreed with the measured
32 concentrations in surveys carried out in Nov. 2019, July 2020, Oct. 2020, and June 2021.
33 Every year, approximately 23.2 kg PFOS was discharged into the Bohai Sea with
34 Chaobai New River being the largest contributor for 44%. Moreover, the transport of
35 PFOS in the original rivers would be restricted by sluices and replaced by artificial rivers.
36 Monte Carlo analysis showed the model predicting PFOS concentrations in sediment is
37 subject to greater uncertainty than that in freshwater as the former is impacted by more
38 parameters. It provides a scientific basis for local government to manage and control
39 PFOS in the study area.

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43 **Keyword:** PFOS, behavior, measured concentrations, mass fluxes, uncertainty

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46 **1. Introduction**

47 Perfluorinated products have been widely used in a range of specialized consumer
48 and industrial products as polymers, surfactants, refrigerants, etc. (Zhang et al., 2019;
49 Cui et al., 2020; Dixit et al., 2022; Liagkouridis et al., 2022). Perfluorooctane sulfonate
50 (PFOS, $C_8F_{17}SO_3$), as one of the main perfluorinated products, has been identified in a
51 wide range of global media, including the Arctic region (Charbonnet et al., 2021;
52 Charbonnet et al., 2022; Evich et al., 2022; Young et al., 2022). It is released from direct
53 use in products or as an end product of the transformation and metabolism of its
54 precursors (Prevedouros et al., 2006; Paul et al., 2009; Wang et al., 2014b; Wang et al.,
55 2014a; Boucher et al., 2018). Different from polycyclic aromatic hydrocarbons (PAHs)
56 or polychlorinated biphenyls (PCBs), PFOS is a non-volatile surfactant that has
57 required a precursor hypothesis to explain its long-term atmospheric transport
58 (Schenker et al., 2008; Armitage et al., 2009; Cousins et al., 2017; Boucher et al., 2018).
59 Furthermore, there is no evidence that it appears to be subject to any degradation or
60 metabolism, due to its highly stable C-F bonds (Barrett et al., 2021; Ruyle et al., 2021;
61 Hao et al., 2022; Yu et al., 2022).

62 Previous studies have investigated the behavior of PFOS based on mass balance
63 models, and assumed that the study area was divided into several regions or segments
64 where the compartments were well-mixed or homogeneous (Armitage et al., 2009;
65 Earnshaw et al., 2014; Liu et al., 2015; Cousins et al., 2017; Su et al., 2018). For
66 example, the spatial distribution of PFOS and its concentrations in freshwater and no
67 sediment in the Aire and Calder Rivers were accurately estimated by a model that was

68 reparametrized from the the Rhine River (Earnshaw et al., 2014). In the previous studies
69 of our research group, the fate of PFOS in the Bohai coastal region of China was
70 simulated with the BETR model and improved BETR-UR, with the Bohai coastal
71 region divided into 56 grid cells (Liu et al., 2015; Su et al., 2018). The trend of PFOS
72 and its volatile precursors with the major production phase-out was modeled and the
73 actual emissions of PFOS and its volatile precursors was capture by CliMoChem where
74 the global is divided into a series of latitudinal bands (Armitage et al., 2009; Cousins et
75 al., 2017).

76 The mass balance model approach for rivers was proposed (Mackay et al., 1983b)
77 based on model approaches for lakes, with a river being considered as consisting of a
78 series of lakes (Mackay et al., 1983a). It has been widely and successfully applied to
79 research the fate of pollutants in a river, such as PAHs in the Saguenay Fjord, Canada,
80 and PCBs in the Altamaha River (Lun et al., 1998; Kilic and Aral, 2009). It was
81 modified, and then successfully applied to explain the fate of four textile chemicals,
82 silver, and TiO₂ Nanoparticles in the Rhine River (Beck et al., 2000; Scheringer et al.,
83 2000), or PFOS in the Aire and Calder of UK (Earnshaw et al., 2014). G-CIEMS that
84 river was divided according to river node was successful to research the concentration
85 of molinate in Shinano River of Japan (Suzuki et al., 2005).

86 However, the behavior of PFOS in the mainstream of Haihe River system
87 combined with sluices and artificial rivers, and the effect of sluices and artificial rivers
88 on the behavior of PFOS in rivers hasn't been fully researched by using the mass
89 balance approach for rivers. Although PFOS was unremittingly discharged into the

90 mainstream of Haihe River system, the concentration of PFOS would sustain a certain
91 range in the natural environment. The Haihe River system is one of the seven major
92 rivers in China and the largest water system in North China, mainly flowing through
93 Beijing and Tianjin, two municipalities directly under the central government in China.
94 There are a large number of PFOS sources that would discharge directly into the Haihe
95 River (Song et al., 2007; Luo et al., 2010; Luo et al., 2011; Wang et al., 2012a).
96 Compared with other rivers, there are lots of sluices and artificial rivers in the
97 mainstream of Haihe River system excavated in both recently and in the past (Wang et
98 al., 2019). It is important to note that there are clear uncertainties with the mass balance
99 model approach which include simplifications of complex and irreversible problems
100 and unique environmental systems (Meent et al., 1999; Gag et al., 2004; Schenker et
101 al., 2009).

102 The objectives of this article are to simulate the concentration of PFOS checked
103 by the measured concentrations from Nov. 2019, July 2020, Oct. 2020, and June 2021,
104 explain why PFOS was unremittingly discharged into rivers but the concentration stay
105 within a certain range instead of rising all the time, illustrate the behavior of PFOS
106 between rivers combined with sluices and artificial rivers and within a river, and
107 analyze the effects of sluices and artificial rivers on the behavior of PFOS and the
108 uncertainty of the river model in the mainstream of Haihe river system. It provides a
109 scientific basis for the local government to manage PFOS in the mainstream of Haihe
110 River system, and also provides a reference for the fate of pollutants in other rivers.

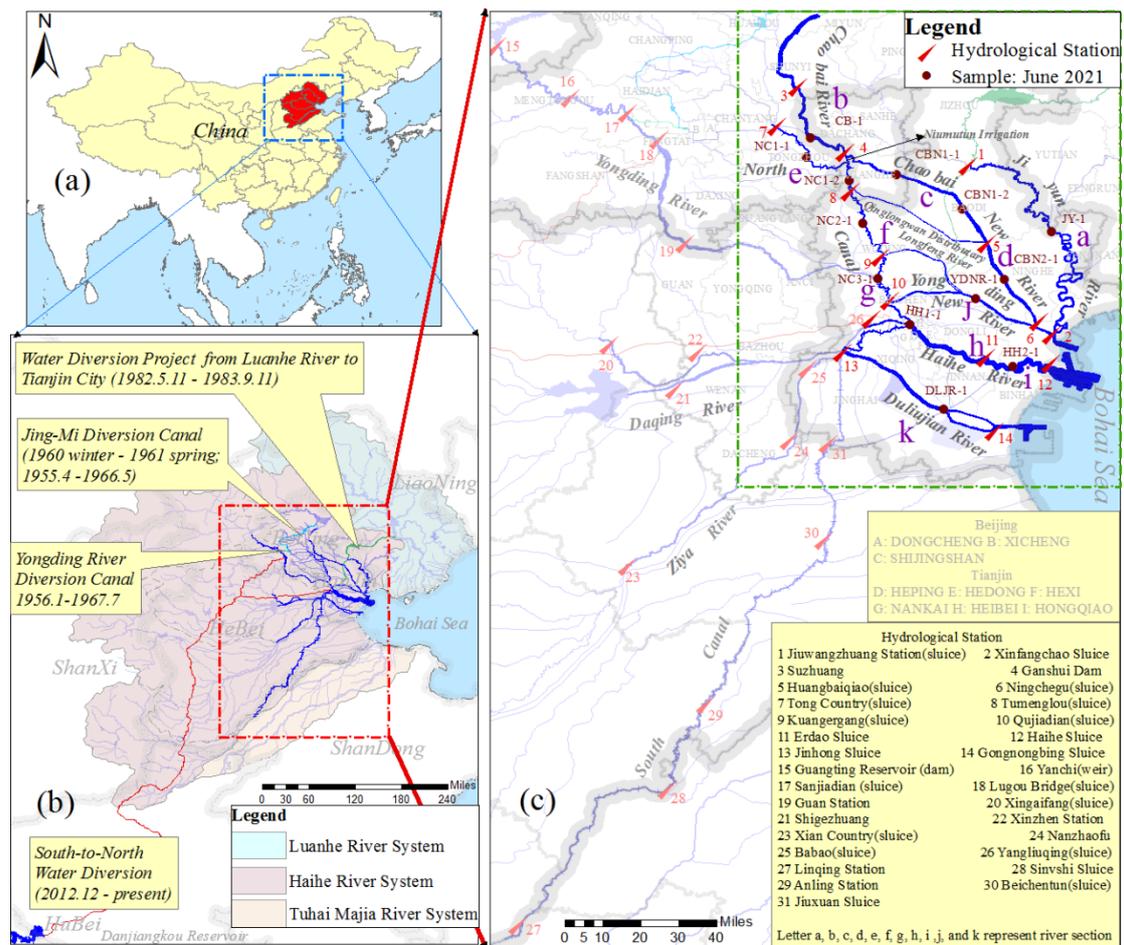
111 **2. Materials and methodology**

112 2.1 Study area

113 The Haihe River basin (Fig. 1(a)), one of the seven river basins in China, contains
114 Luanhe River water system, Haihe River water system and Tuhai Majia Rivers water
115 system (Fig.1(b)). Among them, its main body is the Haihe River water system. There
116 are a number of water conservancy projects to ensure water supply to Beijing and
117 Tianjin, such as Yongding River Diversion Canal, Ji-Mi Diversion Canal, Water
118 Diversion Project from Luanhe River to Tianjin City, and South-to-North Water
119 Diversion in Haihe Rivers' system. In order to protect Beijing and Tianjin from flood
120 disaster, a number of artificial rivers have been excavated and sluices have been built.

121 As shown in Fig.1(c), there are 11 rivers in the mainstream of Haihe River water
122 system: Jiyun River, Chaobai River, Chaobai New River, North Canal, Yongding New
123 River, Haihe River, Duliujian River, Yongding River, Daqing River, Ziya River, and
124 South Canal. Among them, North Canal, South Canal, Chaobai New River, Yongding
125 New River, and Duliujian River are artificial rivers. North and South Canals were
126 excavated to transport grain, while others to alleviate flooding. Each of them has a
127 number of hydrological stations, and the majority were constructed with sluices, as
128 shown in Fig.1(c).

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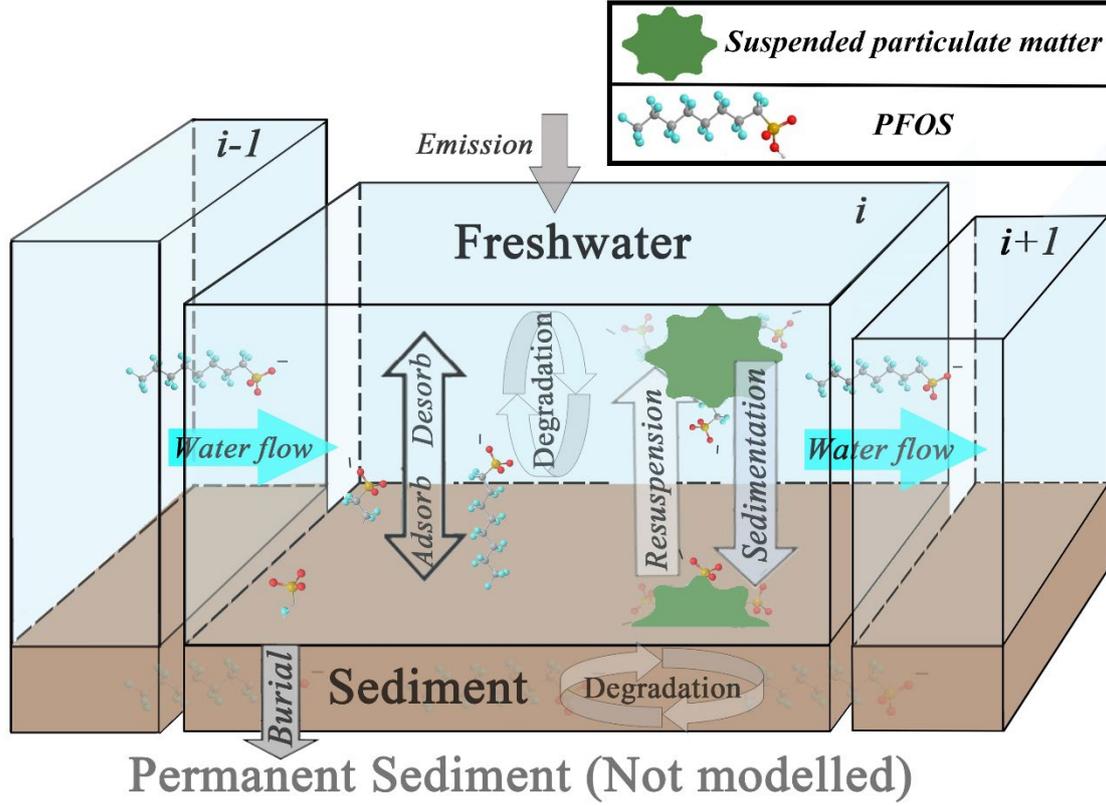
131 Fig. 1 (a) Location of Haihe River basin; (b) the river water systems in Haihe River
 132 basin and major water conservancy projects; (c) Detailed information about the
 133 mainstream of the Haihe River system (excluding other artificial rivers), hydrological
 134 stations, sampling sites from June 2021, mainstream divisions according to
 135 hydrological stations in the model, and districts of Beijing, Tianjin, Langfang and
 136 Tangshan.

137

138 As shown in *Annual Hydrological Report of China: Hydrological Data of Haihe*
 139 *River Basin in 2017 and 2018*, the Yongding River (Table S1), Daqing River (Table S2,
 140 S3, S4 and S5), Ziya River (Table S6), and South River (Table S7) have stopped flowing
 141 or only have an occasional flow during wet seasons. In contrast, the other rivers exhibit
 142 flows of water due to water conservancy projects, such as the South-to-North Water
 143 Diversion. They flow through Beijing, Tianjin, and parts of Langfang and Tangshan
 144 (both belong to Hebei province), as shown in Fig. 1(c).

145 2.2 River model

146 As shown in Fig. 2, a river can be considered as being divided into a series of
147 sections linked by water flow (Mackay et al., 1983b). Freshwater flows from section
148 $i - 1$ to section i , and then from section i to $i + 1$. There exist two well-mixed
149 compartments in each section i : freshwater and sediment (RIVM, 1996). Among them,
150 the suspended particulate matter belongs to the freshwater compartment. Pollutants,
151 such as PFOS, would be discharged into freshwater directly not into the sediment. In
152 each section i , pollutants would be adsorbed from freshwater to sediment, whilst also
153 desorbed from sediment to freshwater. The suspended particulate matter with pollutants
154 in freshwater would undergo gravity settlement, while pollutants in sediment would be
155 resuspended to freshwater due to biological activity or currents. Some of the pollutant
156 loadings in sediment could be lost from the freshwater-sediment system via a
157 permanent burial. Generally, pollutants would be degraded or transformed in both
158 freshwater and sediment but PFOS wouldn't (Beck et al., 2000; Scheringer et al., 2000).
159



160

161 Fig. 2 Each river consists of sections linked by water flow, and the detailed processes
 162 between freshwater and sediment in each section.

163

164 The mass balance equation of freshwater at section i was shown as follow:

$$\begin{aligned}
 165 \quad V_i^{FW} \frac{dC_i^{FW}}{dt} = & \text{Emis}_i^{FW} + \sum_{i-1=1}^j \text{Imp}_{i-1 \rightarrow i}^{FW} - \sum_{i+1=1}^k \text{Exp}_{i \rightarrow i+1}^{Sed} - \text{Deg}_i^{FW} - \text{Diff}_i^{FW \rightarrow Sed} \\
 166 \quad & + \text{Diff}_i^{Sed \rightarrow FW} - \text{Sed}_i^{FW \rightarrow Sed} + \text{Resusp}_i^{Sed \rightarrow FW}
 \end{aligned}$$

$$167 \quad i = 1, 2, 3, \dots, n; \tag{1}$$

168 The mass balance for sediment at section i ,

$$\begin{aligned}
 169 \quad V_i^{Sed} \frac{dC_i^{Sed}}{dt} = & -\text{Deg}_i^{Sed} + \text{Diff}_i^{FW \rightarrow Sed} - \text{Diff}_i^{Sed \rightarrow FW} + \text{Sed}_i^{FW \rightarrow Sed} \\
 170 \quad & - \text{Resusp}_i^{Sed \rightarrow FW} - \text{Bur}_i^{Sed}
 \end{aligned}$$

$$171 \quad i = 1, 2, 3, \dots, n; \tag{2}$$

172 where,

173 V_i^{FW} : the volume of freshwater at section i ;

174 C_i^{Fw} : the concentration of PFOS in freshwater at section i ;

175 $Emis_i^{Fw}$: the emission of PFOS discharged into freshwater at section i ;

176 $\sum_{i-1=1}^j Imp_{i-1 \rightarrow i}^{Fw}$: impart mass flow from all sections (assuming total j number) $i - 1$

177 to i at section i ;

178 $\sum_{i+1=1}^k Exp_{i \rightarrow i+1}^{Sed}$: export mass flow from section i to all $i + 1$ (assuming total k

179 number) at section i ;

180 Deg^{Fw} : degradation in freshwater at section i ;

181 $Diff_i^{Fw \rightarrow Sed}$: adsorb from freshwater to sediment at section i ;

182 $Diff_i^{Sed \rightarrow Fw}$: desorb from sediment to freshwater at section i ;

183 C_i^{Sed} : the concentration of PFOS in sediment at section i ;

184 $Sed_i^{Fw \rightarrow Sed}$: sedimentation from freshwater to sediment by suspended particular matter

185 at section i ;

186 $Resusp_i^{Sed \rightarrow Fw}$: resuspension from sediment to freshwater at section i ;

187 Deg^{Sed} : degradation in sediment at section i ;

188 Bur_i^{Sed} : sediment burial at section i ;

189 The detailed description of $Emis_i^{Fw}$, $Imp_{i-1 \rightarrow i}^{Fw}$, $Exp_{i \rightarrow i+1}^{Fw}$, Deg_i^{Fw} , Deg_i^{Sed} ,

190 $Diff_i^{Fw \rightarrow Sed}$, $Diff_i^{Sed \rightarrow Fw}$, $Sed_i^{Fw \rightarrow Sed}$, $Resusp_i^{Sed \rightarrow Fw}$, and Bur_i^{Sed} was provided

191 in Supplementary Material S1.

192 **2.3 Parameterization of sections, including estimation of volume and emission**

193 As almost all hydrological stations were built on sluices that impacted the water

194 flow and water level on both sides, compartments between two sluices were considered

195 well-mixed in each section. As shown in Fig. 1(c), the study area was divided into 11

196 sections according to hydrological stations. The Niumutun Irrigation Channel,

197 Qinglongwan Distribution, and Longfeng River, artificial rivers, were excavated to help
198 North Canal to flood initially, which has an important function linking these sections.

199 The Jiyun River was divided into section *a* from Jiuwangzhuang (sluice) to
200 Xinfangchao Sluice; Chaobai River section *b* from Suzhuang to Ganshui Dam;
201 Chaobai New River sections *c* and *d* from Ganshui Dam to Huangbaiqiao (Sluice)
202 and Huangbaiqiao (sluice) to Ningchegu (sluice); North Canal sections *e* , *f* , and *g*
203 from Tong country to Tumenlou (sluice), Tumenlou (sluice) to Kuangergang (sluice),
204 and from Kuangergang (sluice) to Qujiadian (sluice); Haihe River sections *h* and *i*
205 from Qujiadian (sluice), Yangliuqing (sluice) and Jinhong sluice to Erdao sluice, and
206 Erdao sluice to Haihe Sluice; Yongding New River section *j* from Qujiadian (sluice)
207 to Ningchegu (sluice); and Duliujian River section *k* from Jinhong Sluice to
208 Gongnongbing Sluice, as shown in Fig.1(c).

209 The volume of freshwater V_{FW}^i in each section was estimated according to the
210 cross-sectional area (Fig. S2) and the length of sections (Table S8). Generally, water
211 levels were different between inflow and outflow sides of a sluice. If a hydrological
212 station did not give both water levels, the water levels on both sides were assumed equal.
213 Each $Area_i$ between freshwater and sediment was estimated by the length of cross-
214 section (Fig. S2) and section (Table S8). The volume of sediment V_S^i was estimated by
215 $Area_i$ and parameters in Table S9 in each section *i*.

216 The emission of PFOS to each section in the mainstream of Haihe River system
217 was estimated using formula (S1), Table S10, S12-S16 and Fig. S3-S11 in
218 Supplementary Material. If a district is served by two rivers, the emission discharged

219 into one of them was approximately 50%. For example, Baodi and Ninggu is served by
220 Jiyun River and Chaobai New River (Fig. S3, S6). The Binhai district was the result of
221 the merger of Tangu, Hangu, and Dagang in Nov. 2009. The emission of PFOS
222 discharged into Jiyun River (*a*), Haihe River (*i*) and Duliujian River (*k*) from Binhai
223 (Fig. S4) was therefore estimated based the three districts before the merger (Table S13-
224 2).

225 Tong country station has two sluices: North sluice and Yun sluice. The emission
226 of PFOS discharge from upstream of Chaobai River *b* and North Canal *e* was
227 divided according to the discharge into two sluices (Table S16). As the discharge of a
228 number of pollutants has been restricted into Haihe River by controlling sluices and
229 artificial rivers, such as Yongding New River and Duliujian River etc., the emission
230 discharged into section *h* of Haihe River was approximately estimated by the
231 population of Heping and Hongqiao districts (Fig. S9) timed parameter in Table S10.

232 **2.4 Sampling campaign and chemical analysis**

233 As shown in Fig.1(c), each section has at least one representative sampling site to
234 evaluate the river model performance. Freshwater and sediment were collected
235 separately at each site from Nov. 2019, July 2020, Oct. 2020, and June 2021. The
236 locations of sample sites from June 2021 are shown in Fig. 1(c), and the remaining
237 shown in Fig. S1. Freshwater was stored in 1L clean polypropylene bottles and
238 sediment packed in double-layer polyethylene (PE) bags at -4 temperature before
239 extraction. Freshwater samples were stored at -20 °C and sediment was dried in a room
240 sheltered from light, ground in a mortar and then passed through a 100 mesh sieve, and

241 packed in PE bags.

242 The concentrations of linear PFOS in freshwater and sediment samples were
243 measured using previously published methods (Zhang et al., 2019; Wang et al., 2020;
244 Zhang et al., 2021). In brief, 400ml freshwater sample was extracted by cartridge (Oasis
245 WAX) and PFOS in 2g sediment was extracted by acetonitrile and then cleaned up by
246 Supelco ENVI-Carb cartridge and Oasis WAX cartridge. After purification and
247 concentration, the samples were filtered and stored in autosampler vials. The
248 concentrations of PFOS in freshwater and sediment in sample sites were quantified by
249 the Agilent 1290 Infinity HPLC System coupled with an Agilent 6460 Triple
250 Quadrupole LC/MS System (Agilent Technologies, Palo Alto, CA).

251 A 10-point internal standard calibration curve consisting of 0.01, 0.05, 0.1, 0.5, 1,
252 5, 10, 100, 500 ng/mL concentration gradients with 5 ng/mL mass-label standard as
253 internal standard was to ensure the accurate quantification of PFOS. The R^2 value of
254 calibration curves of PFOS was greater than 0.99. Then 10 ng native standards of PFOS
255 were added to freshwater and sediment to calculate the matrix spike recoveries (MSRs).
256 The MSRs of PFOS ((mean value \pm standard deviation, $n = 4$) in freshwater and
257 sediment were 105.2 ± 12.5 and 97.5 ± 11 , respectively. The limits of quantification of
258 PFOS in freshwater and sediment were 0.06 ng/L and 0.03 ng/g. The limit of detection
259 of PFOS in freshwater and sediment were 0.03 ng/L and 0.01 ng/g, respectively.

260 **2.5 Uncertainty analysis**

261 The simulated concentrations and mass fluxes by mass balances models exhibit
262 uncertainty due to the error of parameter estimations (Meent et al., 1999; Macleod et

263 al., 2002; Gag et al., 2004). Monte Carlo simulation is an effective method to analyze
 264 the produced uncertainty. In our Monte Carlo analysis, the probability distribution of
 265 each parameter is assumed to be log-normal distribution, since the values of parameters
 266 are positive and the log-normal distribution of parameters is both warranted and
 267 advantageous (Mckone, 1994; Slob, 1994; Luo and Yang, 2007). The stochastic data
 268 for each parameter was generated from the distribution of parameters with MATLAB,
 269 then input into the mass balance model to produce simulated concentrations and
 270 uncertainty (Gag et al., 2004; Schenker et al., 2009).

271 If the logarithmic of a random variable is normal distribution, the random variable
 272 is log-normal distribution, shown as follows:

$$273 \quad \quad \quad \ln(X) \sim N(\mu, \sigma^2) \quad (3)$$

274 where parameters μ and σ in log-normal distribution are the mean and variance of the
 275 corresponding normal distribution on a log scale. Both of them are dimensionless
 276 because they are estimated after taking logarithms (Slob, 1994; Mackay et al., 2014;
 277 Niwitpong and Somkhuean, 2016). In the original scale, the log-normal distribution is
 278 defined by the population median (M) and the population coefficient of variation (CV)
 279 that is the arithmetic standard deviation divided by its arithmetic mean (Mckone, 1994),
 280 as follows:

$$281 \quad \quad \quad M = b^\mu \quad (4)$$

$$282 \quad \quad \quad CV = \sqrt{b^{\sigma^2 \ln b} - 1} \quad (5)$$

283 The 95% interval confidence can be shown as follows:

$$284 \quad \quad \quad \text{Prob} \left\{ \frac{M}{k} < X < M \times k \right\} = 95\% \quad (6)$$

285 and k is the dispersion factor, as follows

$$286 \quad k = e^{1.96\sqrt{\ln(CV^2+1)}} \quad (7)$$

287 Among, population (Table S12-S15) and rate of water flow (Table S17) were
288 assumed with a fixed log-transformed CV value 5% (Xu et al., 2006). And the assumed
289 CV of other parameters were shown in Table S17.

290 **3. Results and discussion**

291 **3.1 Estimated emissions**

292 The divisions of PFOS potentially discharged into Jiyun River are shown in Fig
293 S3, S4; and for the Chaobai River in Fig. S5, S6; the Chaobai New River in Fig. S6, S7;
294 the North Canal in Fig. S5, S7 - S9; the Haihe River in Fig. S4, S10; the Yongding New
295 River in Fig. S8; and the Duliujian River in Fig. S4, S11. The annual emissions of PFOS
296 were approximately 1.8, 5.0, 1.4, 6.7, 3.5, 2.2, and 2.9 kg in the Jiyun River, Chaobai
297 River, Chaobai New River, North Canal, Haihe River, Yongding New River, and
298 Duliujian River, respectively, and the maximum receiver was North Canal, as shown in
299 Fig.5. For Jiyun River, Chaobai River, North Canal, Haihe River, and Duliujian River,
300 the mass fluxes of PFOS transported by water flow from upstream were also taken part
301 of their emission (Fig.5).

302 In the whole mainstream of Haihe River system, the total annual emission of PFOS
303 was approximately 23.6 kg. Among them, the top three rivers with the highest
304 contribution to the emissions of PFOS were 28% for North Canal, 21% for Chaobai
305 River, and 15% for Haihe River, as North Canal and Chaobai River were affected by
306 Beijing City (Zhang et al., 2016; Wang et al., 2019), and Haihe River by Tianjin City.

307 The emission contributed to the Jiyun river accounted for 8% of total emission, which
308 is consistent with its distance far from Beijing and Tianjin shown in Fig.1(c).

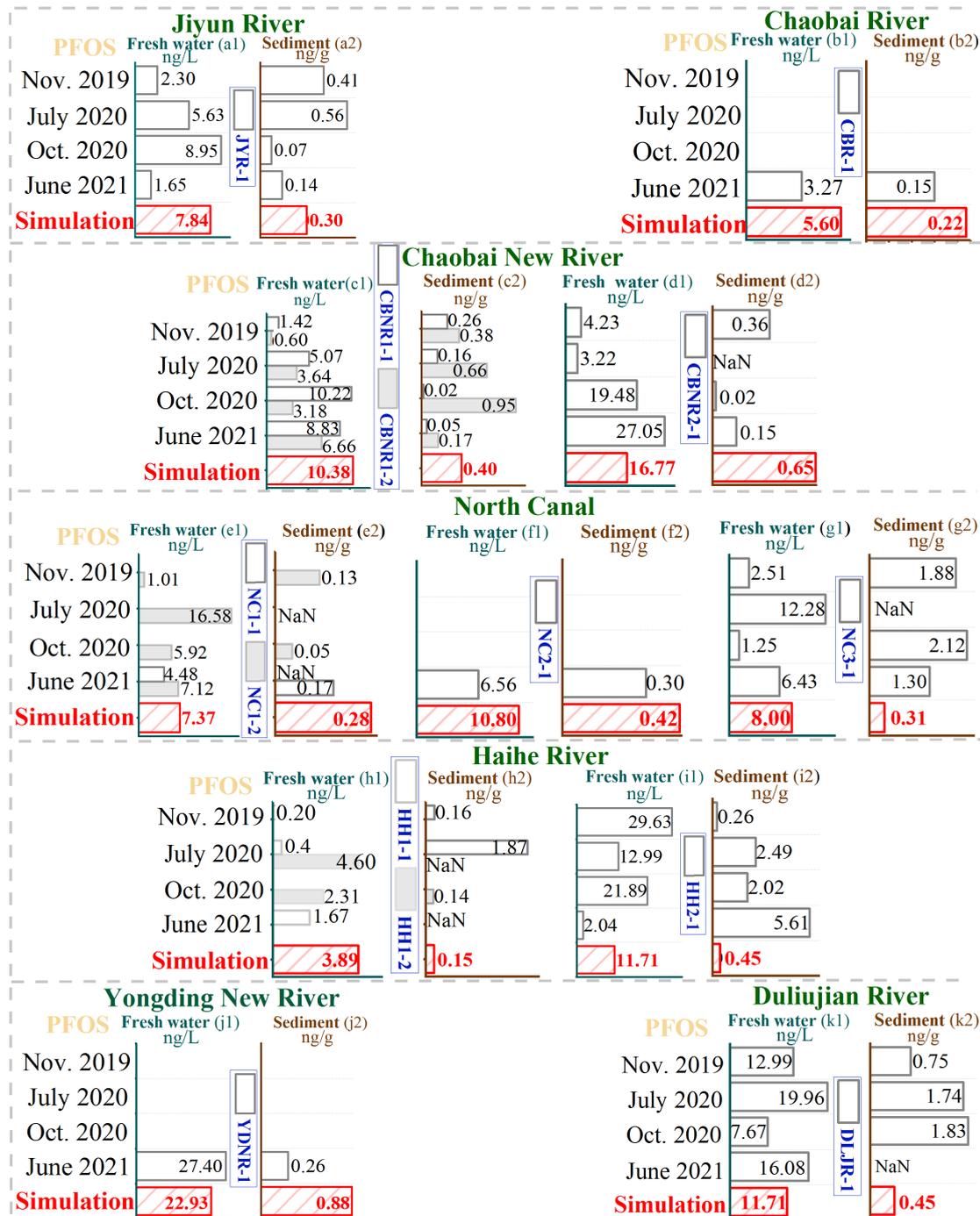
309 **3.2 Comparison of measured concentrations and predicted values at a steady state**

310 The predicted concentrations of PFOS in freshwater and sediment under steady-
311 state conditions simulated by the river model shown in section 2.1 with parameters in
312 Table S8-S16, were compared to the measured data collected in Nov. 2019, July 2020,
313 Oct. 2020, and June 2021 (some only in June 2021) using the method described in
314 section 2.4. Fig. 3(a-k) shows the simulated results versus the four measured
315 concentrations in sections *a – k* , suggesting that the simulated concentrations agreed
316 well with the measurements.

317 The predicted concentrations were lower than the maximum of the corresponding
318 measured concentration and greater than the minimum of the corresponding measured
319 ones in both compartments at almost all sections (Fig. 3a1-3k2). For example, in section
320 *a* of Jiyun River, the predicted freshwater concentration was 7.84 ng/L lower than 8.95
321 ng/L (sampling date: Oct. 2020) and higher than 1.65 ng/L (June 2021) (Fig. 3a1).
322 Although there was only a sampling site in June 2021 in sections *b*, *f*, and *j*, it is not
323 significantly different. The predicted concentrations of PFOS in freshwater were more
324 accurate than for those in sediment, especially in sections *d*, and *g*, as the predicted
325 concentration of PFOS in sediment was affected by more parameters than that in fresh
326 water, such as organic carbon content of sediment, density of solid phase, etc.

327

328



329

330 Fig. 3 The concentration of PFOS in freshwater (left) and sediment (right) measured in
 331 Nov. 2019, July 2020, Oct. 2020, and June 2021 and the simulation values under steady
 332 state in Jiyun River (a1, and a2), Chaobai River (b1, and b2), Chaibai New River (c1,
 333 c2, d1, and d2), North Canal (e1, e2, f1, f2, g1, and g2), Haihe River (h1, h2, i1, and
 334 i2), Yongding New River (j1, and j2), and Duliujian River (k1, and k2) (some sections
 335 only have sample sites from June 2021).

336

337 The predicted concentrations were 7.37, 10.50, and 8.00 ng/L in fresh water in

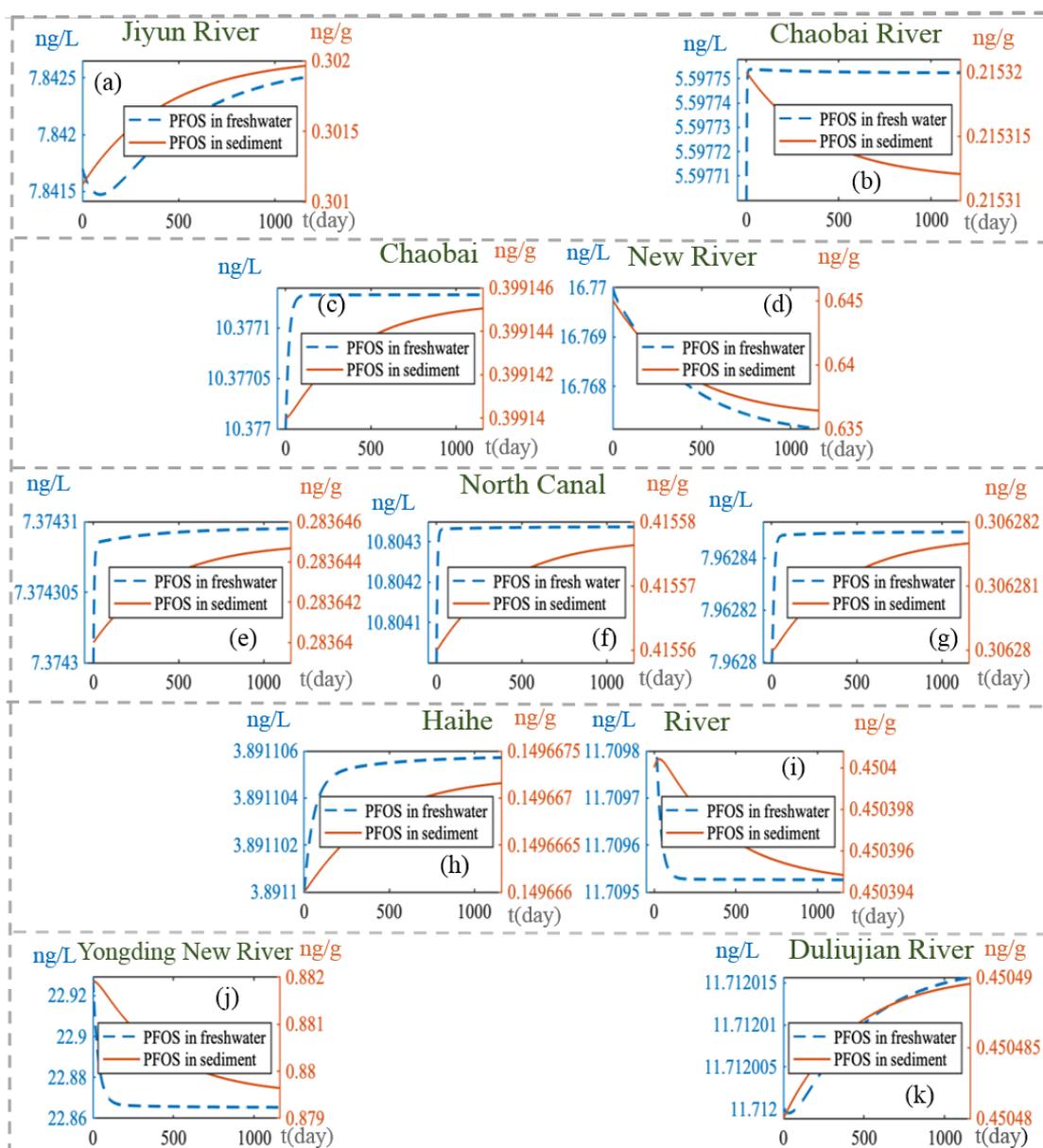
338 North Canal, which was consistent with 9.44 and 11.02 ng/L from the high flow period
339 reported by (Zhang et al., 2016). The predicted concentrations of PFOS in freshwater
340 were 7.84, 22.93 and 11.71 ng/L in the Jiyun River, Yongding New River, and
341 Duliujian River, respectively, and were significantly higher than the measured
342 concentrations of 0.74, 2.4 and 1.7 ng/L in 2012 (Wang et al., 2012a), respectively.
343 The predicted sediment concentrations were 0.88 and 0.45 ng/g in Yongding New
344 River and Duliujian River, respectively, which was almost equal to 0.94 and 0.53 ng/g
345 measured in 2012 by (Wang et al., 2012b).

346 Predicted concentrations were 3.89 and 11.71 ng/L in freshwater, and 0.15 and
347 0.45 ng/g in sediment in the Haihe River, which agreed well with that measured data
348 ranging from 1.11 to 7.22 ng/L and 0.062 to 0.79 ng/g reported by (Pan et al., 2011),
349 with 2.02-7.62 ng/L and 1.76 - 7.32 ng/g reported by (Li et al., 2011), and 1.5 and 10
350 ng/L (Wang et al., 2012a) and 1.2 and 4.3 ng/g by (Wang et al., 2012b), from separate
351 studies. These comparisons showed that the concentrations of PFOS in Haihe River
352 haven't changed much over the years, revealing Haihe River was protected and the
353 discharge of PFOS was restricted by local government.

354 **3.3 Simulated concentrations of PFOS with an unsteady-state**

355 In the natural environment, emission, volumes of freshwater and sediment,
356 parameters of PFOS, and parameters of river change over time. As they are affected by
357 lots of factors, such as season, temperature, rainfall, etc., the concentration of PFOS in
358 both compartments will respond as a function of time. Assuming that the parameters
359 are fixed for simplification and taking the results of steady-state (Fig.3) as the initial

360 condition of the unsteady state, the concentrations of PFOS in freshwater and sediment
 361 under unsteady-state conditions have been estimated by the method in section 2.2. The
 362 results in both compartments with the unsteady-state were shown in Fig. 4(a-k).
 363



364
 365 Fig. 4 The results of PFOS with the unsteady state in freshwater (left) and sediment
 366 (right) in Jiyun River (a), Chaobai River (b), Chaobai New River (c and d), North Canal
 367 (e, f, and g), Haihe River (h and i), Yongding New River (j), and Duliujian River (k).
 368

369 As shown in Fig. 4(a-k), the concentration curves of PFOS in freshwater and

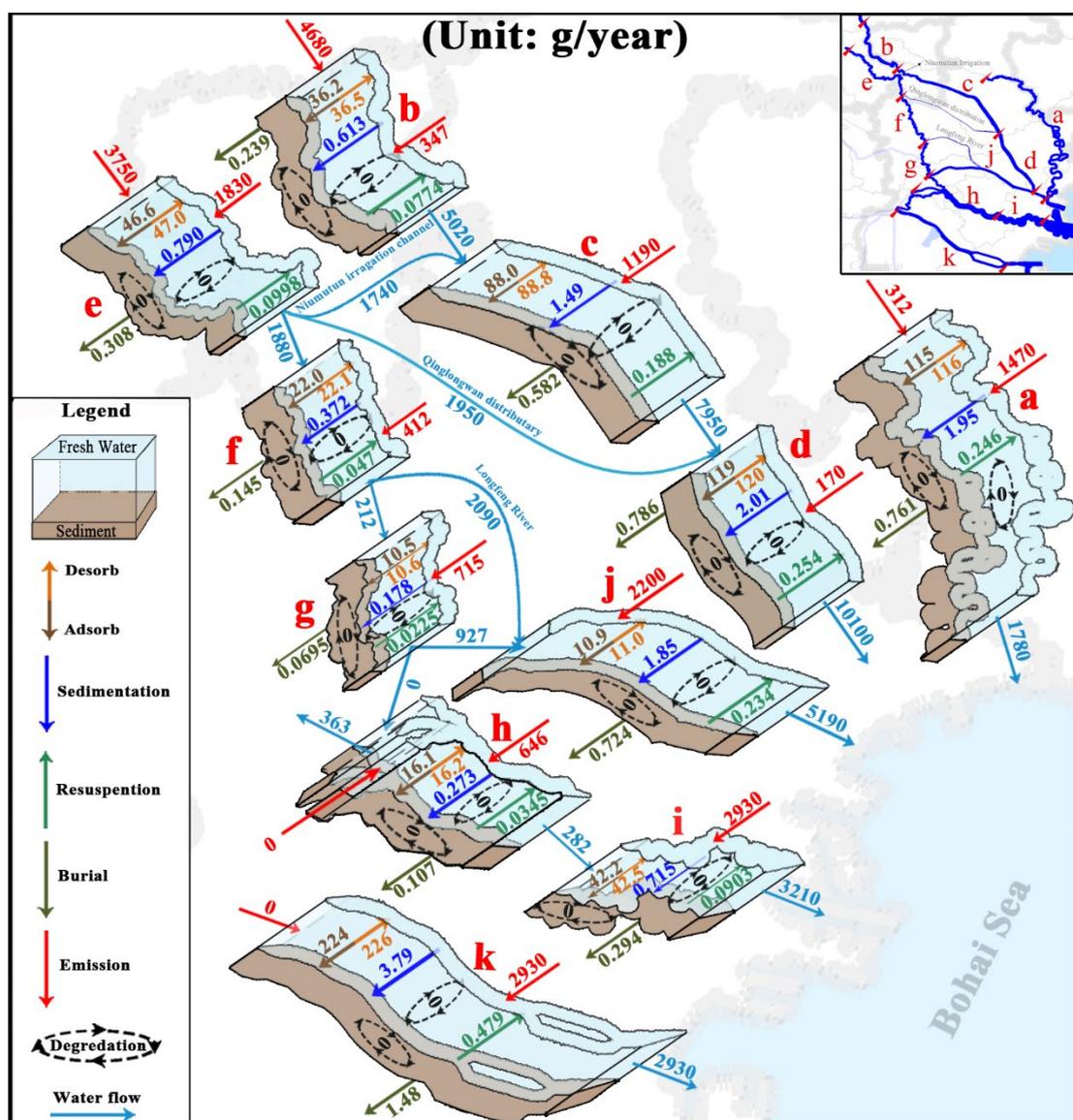
370 sediment would tend to a certain value when the exchanged of PFOS reached
371 thermodynamic equilibrium in the freshwater-sediment system. Under thermodynamic
372 equilibrium, the concentration of PFOS in both compartments would be sustained at a
373 certain value, although PFOS was unremittingly discharged into fresh water, imported
374 and outported by water flow between sections, exchanged between compartments in
375 every section, and buried in sediment. The results with unsteady state explain why
376 PFOS was discharged into the mainstream of Haihe River system but the concenraion
377 of PFOS sustained a certain range and wouldn't increase all the time in the natural
378 environment.

379 **3.4 The mass fluxes of PFOS in the mainstream of Haihe River system**

380 The annual mass fluxes of PFOS in the mainstream of the Haihe River system
381 were estimated by formulations (1, 2), and (S1-S12), shown in Fig.5. The Chaobai River,
382 North Canal, Chaobai New River, Yongding New River, HaiHe River, and Duliujian
383 River as well as Niumutun Irrigation Channel, Qinglongwan Distribution, and
384 Longfeng River form a complex structure, while Jiyun River forms an independent, and
385 simple water system by itself.

386 Under dynamic equilibrium between fresh water and sediment, the mass fluxes of
387 adsorption were almost equal to those of the corresponding desorption in each river as
388 they were derived by differences in chemical potentials in freshwater and sediment
389 (RIVM, 1996), approximately from 109 ~ 797 g in one of the rivers, shown in Fig.5.
390 Nevertheless, the values of sedimentation were totally unequal to the corresponding
391 value of resuspension in one of rivers in the mainstream, as both processes are the single

392 direction (RIVM, 1996). Each year, approximately 0.613 ~ 3.79 g PFOS were removed
 393 from freshwater to sediment, while approximately 0.0774 ~ 0.479 g PFOS were
 394 resuspended from sediment to fresh water in one of them. The mass fluxes of
 395 degradation in both freshwater and sediment were 0 kg in all rivers, as the highly stable
 396 C-F bond in PFOS.
 397



398
 399 Fig. 5 The mass fluxes of PFOS in each section ins Jiyun River (a), Chaibai River (b),
 400 Chaibai New River (c & d), North Canal (e & f & g), Haihe River (h & i), Yongding
 401 New River (j), and Duliujian River (k), and between sections. The Niumutun Irrigation
 402 Channel, Qinglongwan Distribution and Longfeng River have a function to link rivers.

403

404

405 A total of 5.48 g per year PFOS were estimated to be buried in sediment in the
406 mainstream of Haihe River system. Among them, the top two rivers that contributed
407 the highest burial were Duliujian River (27%) and Chaobai New River (25%), as their
408 area between sediment and freshwater was greater than the others. The rank of rivers
409 based on the mass value of sediment burial was Duliujian River > Chaobai New River >
410 Jiyun River > Yongding New River > North Canal > Haihe River > Chaobai River.

411 Every year, approximately 23 kg of PFOS was estimated to have been discharged
412 from Jiyun River, Chaobai New River, Haihe River, and Duliujian River to the Bohai
413 Sea. The most important contributor to the total was the Chaobai New River at 44% of
414 the total, the second was Yongding New River at 22%, the third was Haihe River at
415 14%, the fourth was Duliujian River at 13%, and the last was Jiyun River at 8%. As the
416 Chaobai New River is responsible for the sewage discharge from Beijing, it has the
417 maximum value of PFOS discharged into the Bohai Sea. The mass balance approach
418 for river has been used to estimated an annual mass discharged of PFOS the Aire and
419 Calder rivers were from 215 to 310 kg in UK (Earnshaw et al., 2014). The mass fluxes
420 of PFOS deposition to the Arctic were approximately 2.0 to 200kg/year over the period
421 1996-2000 range by CliMoChem (Armitage et al., 2009).

422 For comparison, a mass flux of PFOS was estimated to be 26 kg/year from the
423 Yonding River, Haihe River, and Duliujian River to Bohai sea by (Zhou et al., 2018)
424 which was comparable to this study; emissions from Jinyun River, Yongding New River,
425 Haihe River, and Duliujian River to the Bohai sea were 10 kg/year, which was lower

426 than the results from this study as emissions of PFOS were lower a decade ago (Wang
427 et al., 2012a). In comparison, the summary of PFOS discharged into the Bohai Sea from
428 rivers in LiaoNing, HeBei, Tianjin City, and Shandong was estimated to be from 80.0
429 to 80.5 kg/year which was higher than the results from this study (Chen et al., 2016).

430 **3.5 Sluices and artificial rivers**

431 Sluices and artificial rivers played an important role in the transport of PFOS in
432 the mainstream of Haihe River system. Water flow is restricted by the presence of
433 sluices which would be replaced by the flow from an artificial river. A representative
434 sluice is Qujiandian (North Sluice) which totally restricts the transport of the PFOS
435 from North Canal to Haihe River to protect the health of Haihe River (0 kg shown in
436 Fig. 5), while pollutants are entirely transported from North Canal to Yongding New
437 River through Quijiandian (Yongxin Sluice).

438 Representative artificial rivers are Nummutun Irrigation Channel, Qinglongwan
439 Distribution, and Longfeng River. Approximately 31% and 35% of the PFOS in the
440 original North Canal (section *e*) was replaced by Numutun Irrigation Channel, and
441 Qinglongwan Distribution through Numutun Sluice and Tumenlou (Qing Sluice),
442 respectively. In combination, they approximately replaced 66% of the function of North
443 River (section *e*). Approximately 91% of the PFOS in the original North Canal in
444 section *g* was replaced by Longfeng River through Kangergang (Jiezhi Sluice).

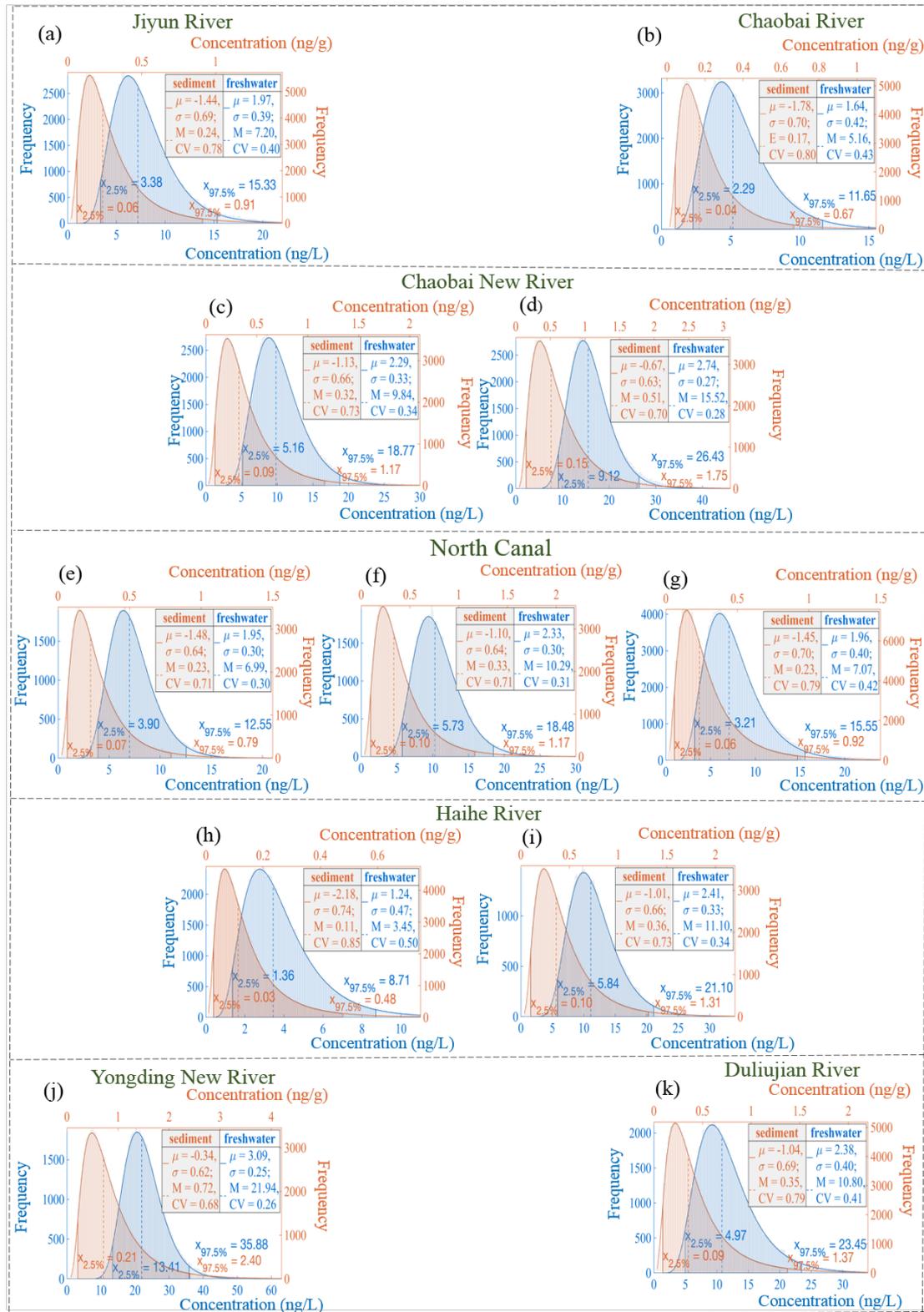
445 Although Chaobai New River, Yongding New River, and Duliujian River have
446 been parts of the mainstream of Haihe River system, they are actually artificial rivers
447 that replace the function of Haihe River. Approximately 46%, 24%, and 16% of the

448 total PFOS transported in the original Haihe River were replaced by Chaobai New River,
449 Yongding New River and Duliujian River, respectively. Of course, North Canal, an
450 ancient artificial river, played an important role in transporting PFOS in the mainstream
451 of Haihe River system, accounting for approximately 28% of the PFOS in the
452 mainstream.

453 **3.6 Uncertainty of model estimates**

454 The predicted concentrations of PFOS in the two compartments are shown in Fig.6
455 after 100000 simulations by Monte Carlo using MATLAB with the assumed
456 distribution of parameters in section 2.5 and Table S17. The probability distribution of
457 simulated PFOS concentration in both freshwater and sediment followed the log-
458 normal distribution, and the values of their parameters μ and σ in each section were
459 shown in Fig.6(a-k) with the median values, 95% confidence intervals, and coefficients
460 of variance.

461



462

463 Fig. 6 Both probability distribution of PFOS in freshwater (left) and sediment (right)
 464 with Monte Carlo in Jiyun River (a), Chaobai River (b), Chaobai New River (c and d),
 465 North Canal (e, f and g), Haihe River (h and i), Yongding New River (j), and Duliujian
 466 River (k), respectively. μ and σ are parameters of lognormal distribution ($Y = \ln(X) \sim N$
 467 (μ, σ^2)). Capital M is median value, and CV is coefficient of variable.

468

469 Among, the simulated concentration of PFOS in compartments in sections 3.2 and
470 3.3 were within the 95% interval confidence of the corresponding compartments, and
471 most of the measured concentrations in section 3.2 fall into 95% interval confidence.
472 The median value of PFOS concentration in freshwater and sediment in each section
473 was slightly lower than that corresponding one simulated with both steady-state and
474 unsteady-state in sections 3.2 and 3.3, showing the structure of the model was stable to
475 a certain extent and no individual parameter was identified as having a significant
476 impact on the predictions.

477 The values of CV in freshwater ranged from 0.25 to 0.50, while those in sediment
478 ranged from 0.69 to 0.86 in Fig. 6(a-k). The former values were significantly lower than
479 those in the corresponding one showing there existed more uncertainty in sediment than
480 that in freshwater, as the simulated concentrations in sediment were propagated by more
481 parameters, such as density of sediment (ρ) (Gag et al., 2004). That was consistent with
482 the results that the predicted concentrations of PFOS in freshwater were more accurate
483 than for those in sediment in section 3.1. Additionally, this study has not considered
484 that PFOS is degraded from its precursors (Wang et al., 2014b; Wang et al., 2014a).

485 **4. Conclusion**

486 The mainstream of Haihe River system was analyzed and divided according to
487 hydrological data and hydrological stations. The annual emissions were estimated in
488 each river based on the division, and the top three rivers with the highest contribution
489 were identified as North Canal (28%), Chaobai River (21%), and Haihe (15%). The
490 predicted concentration of PFOS with both steady state and unsteady state based on

491 mass balance equations were agreed with the measured concentration of PFOS at
492 sampling sites from Nov. 2019, July 2020, Oct. 2020, and June 2021. The unsteady
493 state results explain why PFOS was discharged but the concentration would sustain a
494 certain range instead of increasing all the time.

495 The transport of PFOS between rivers combined with sluices and artificial rivers,
496 and transport and transformation within a river between freshwater and sediment were
497 illustrated. Among, total mass fluxes of 23.2 kg and 5.48 g PFOS were estimated to
498 flow into the Bohai Sea and were buried in the mainstream of Haihe River system,
499 respectively. And sluice would restrict the transport of PFOS in rivers, such as
500 Qujiandian (North sluice); artificial rivers, including some become part of the
501 mainstream of Haihe water system, would replace the function of original rivers to
502 transport PFOS. The uncertainty analysis estimated the distribution and its
503 parameters, median value, 95% interval confidence and CV value, indicating there was
504 more uncertainty in sediment than freshwater.

505

506

507

508 **References:**

- 509 Armitage, J. M., Schenker, U., Scheringer, M., Martin, J. W., Macleod, M., Cousins, I. T. 2009.
510 Modeling the Global Fate and Transport of Perfluorooctane Sulfonate (PFOS) and
511 Precursor Compounds in Relation to Temporal Trends in Wildlife Exposure. *Environmental*
512 *Science & Technology*. 43(24): 9274-9280.
- 513 Barrett, H., Du, X., Houde, M., Lair, S., Verreault, J., Peng, H. 2021. Suspect and Nontarget Screening
514 Revealed Class-Specific Temporal Trends (2000–2017) of Poly- and Perfluoroalkyl
515 Substances in St. Lawrence Beluga Whales. *Environmental Science & Technology*. 55(3):
516 1659-1671.
- 517 Beck, A., Scheringer, M., Hungerbühler, K. 2000. Fate modelling within LCA. *International Journal*
518 *of Life Cycle Assessment*. 5(6): 335-344.
- 519 Boucher, Justin, M., Cousins, Ian, T., Scheringer, Martin, Hungerbühler, Konrad, Wang, Zhanyun.
520 2018. Toward a Comprehensive Global Emission Inventory of C4-C10
521 Perfluoroalkanesulfonic Acids (PFSA) and Related Precursors: Focus on the Life Cycle of
522 C6- and C10-Based Products. *Environmental Science & Technology Letters*. 6(1): 1-7.
- 523 Charbonnet, J. A., McDonough, C. A., Xiao, F., Schwichtenberg, T., Cao, D., Kaserzon, S., Thomas,
524 K. V., Dewapriya, P., Place, B. J., Schymanski, E. L., Field, J. A., Helbling, D. E., Higgins, C. P.
525 2022. Communicating Confidence of Per- and Polyfluoroalkyl Substance Identification via
526 High-Resolution Mass Spectrometry. *Environmental Science & Technology Letters*.
- 527 Charbonnet, J. A., Rodowa, A. E., Joseph, N. T., Guelfo, J. L., Field, J. A., Jones, G. D., Higgins, C. P.,
528 Helbling, D. E., Houtz, E. F. 2021. Environmental Source Tracking of Per- and
529 Polyfluoroalkyl Substances within a Forensic Context: Current and Future Techniques.
530 *Environmental Science & Technology*. 55(11): 7237-7245.
- 531 Chen, H., Wang, X., Zhang, C., Sun, R., He, X. 2016. Occurrence and inputs of perfluoroalkyl
532 substances (PFASs) from rivers and drain outlets to the Bohai Sea, China. *Environmental*
533 *Pollution*. 221: 234.
- 534 Cousins, Ian, T., Hungerbühler, Konrad, Boucher, Justin, M., Wang, Zhanyun. 2017. Toward a
535 Comprehensive Global Emission Inventory of C4-C10 Perfluoroalkanesulfonic Acids
536 (PFSA) and Related Precursors: Focus on the Life Cycle of C8-Based Products and
537 Ongoing Industrial Transition. *Environmental Science & Technology*. 51: 4482-4493.
- 538 Cui, J., Gao, P., Deng, Y. 2020. Destruction of Per- and Polyfluoroalkyl Substances (PFAS) with
539 Advanced Reduction Processes (ARPs): A Critical Review. *Environmental Science &*
540 *Technology*. 54(7): 3752-3766.
- 541 Dixit, F., Munoz, G., Mirzaei, M., Barbeau, B., Liu, J., Duy, S. V., Sauvé, S., Kandasubramanian, B.,
542 Mohseni, M. 2022. Removal of Zwitterionic PFAS by MXenes: Comparisons with Anionic,
543 Nonionic, and PFAS-Specific Resins. *Environmental Science & Technology*. 56(10): 6212-
544 6222.
- 545 Earnshaw, M. R., Paul, A. G., Loos, R., Tavazzi, S., Paracchini, B., Scheringer, M., Hungerbühler, K.,
546 Jones, K. C., Sweetman, A. J. 2014. Comparing measured and modelled PFOS
547 concentrations in a UK freshwater catchment and estimating emission rates. *Environment*
548 *International*. 70: 25-31.
- 549 Evich, M. G., Davis, M. J. B., McCord, J. P., Acrey, B., Awkerman, J. A., Knappe, D. R. U., Lindstrom,
550 A. B., Speth, T. F., Tebes-Stevens, C., Strynar, M. J., Wang, Z., Weber, E. J., Henderson, W.
551 M., Washington, J. W. 2022. Per- and polyfluoroalkyl substances in the environment.

552 Science. 375(6580): eabg9065.

553 Gag, H., SHU FULIU, X. U., Jr., R. M. C., Cao, J., Bengang, L. I., Liu, W., Wang, X., Jianying, H. U., Shen,
554 W., Qin, B. 2004. Multimedia Fate Model for Hexachlorocyclohexane in Tianjin, China.
555 Environmental Science & Technology. 38(7): 2126-2132.

556 Hao, S., Choi, Y. J., Deeb, R. A., Strathmann, T. J., Higgins, C. P. 2022. Application of Hydrothermal
557 Alkaline Treatment for Destruction of Per- and Polyfluoroalkyl Substances in
558 Contaminated Groundwater and Soil. Environmental Science & Technology. 56(10): 6647-
559 6657.

560 Kilic, S. G., Aral, M. M. 2009. A fugacity based continuous and dynamic fate and transport model
561 for river networks and its application to Altamaha River. Science of the Total Environment.
562 407(12): 3855-3866.

563 Li, F., Sun, H., Hao, Z., He, N., Zhao, L., Tao, Z., Sun, T. 2011. Perfluorinated compounds in Haihe
564 River and Dagu Drainage Canal in Tianjin, China. Chemosphere. 84(2): 265-271.

565 Liagkouridis, I., Awad, R., Schellenberger, S., Plassmann, M. M., Cousins, I. T., Benskin, J. P. 2022.
566 Combined Use of Total Fluorine and Oxidative Fingerprinting for Quantitative
567 Determination of Side-Chain Fluorinated Polymers in Textiles. Environmental Science &
568 Technology Letters. 9(1): 30-36.

569 Liu, S., Lu, Y., Xie, S., Wang, T., Jones, K. C., Sweetman, A. J. 2015. Exploring the fate, transport and
570 risk of Perfluorooctane Sulfonate (PFOS) in a coastal region of China using a multimedia
571 model. Environment International. 85: 15-26.

572 Lun, R., Lee, K., De Marco, L., Nalewajko, C., Mackay, D. 1998. A model of the fate of polycyclic
573 aromatic hydrocarbons in the Saguenay Fjord, Canada. Environmental Toxicology &
574 Chemistry. 17(2): 333-341.

575 Luo, Y., Mao, D., Rysz, M., Zhou, Q., Alvarez, P. 2010. Trends in Antibiotic Resistance Genes
576 Occurrence in the Haihe River, China. Environmental Science & Technology. 44(19): 7220-
577 7225.

578 Luo, Y., Xu, L., Rysz, M., Wang, Y., Zhang, H., Alvarez, P. 2011. Occurrence and Transport of
579 Tetracycline, Sulfonamide, Quinolone, and Macrolide Antibiotics in the Haihe River Basin,
580 China. Environmental Science & Technology. 45(5): 1827-1833.

581 Luo, Y., Yang, X. 2007. A multimedia environmental model of chemical distribution: Fate, transport,
582 and uncertainty analysis. Chemosphere. 66(8): 1396-1407.

583 Mackay, D., Hughes, L., Powell, D. E., Kim, J. 2014. An updated Quantitative Water Air Sediment
584 Interaction (QWASI) model for evaluating chemical fate and input parameter sensitivities
585 in aquatic systems: Application to D5 (decamethylcyclopentasiloxane) and PCB-180 in two
586 lakes. Chemosphere. 111: 359-365.

587 Mackay, D., Paterson, S., Joy, M. 1983a. A quantitative water, air, sediment interaction(QWASI)
588 fugacity model for describing the fate of chemicals in lakes. Chemosphere. 981-997.

589 Mackay, D., Paterson, S., Joy, M. 1983b. A quantitative water, air, sediment interaction(QWASI)
590 fugacity model for describing the fate of chemicals in river. Chemosphere. 1193-1208.

591 Macleod, M., Fraser, A. J., Mackay, D. 2002. Evaluating and expressing the propagation of
592 uncertainty in chemical fate and bioaccumulation models. Environmental Toxicology &
593 Chemistry. 21(4): 700-709.

594 Mckone, T. E. 1994. Uncertainty and Variability in Human Exposures to Soil Contaminants Through
595 Home-Grown Food: A Monte Carlo Assessment. Risk Analysis. 14(4): 449-463.

596 Meent, V. D., D., Etienne, R., S., Ragas, A., M. J., Willemsen, F., H. 1999. Assessing model uncertainty
597 for environmental decision making: a case study of the coherence of independently
598 derived environmental quality objectives for air and water. *Environmental Toxicology &*
599 *Chemistry*. 18(8): 1856–1867.

600 Niwitpong, S. A., Somkhuean, R. 2016. Confidence intervals for the median of lognormal
601 distribution with restricted parameter mean. *International Journal of Mathematics Trends*
602 *and Technology*. 29(1): 21-27.

603 Pan, Y., Shi, Y., Wang, J., Jin, X., Cai, Y. 2011. Pilot Investigation of Perfluorinated Compounds in
604 River Water, Sediment, Soil and Fish in Tianjin, China. *Bulletin of Environmental*
605 *Contamination and Toxicology*. 87(2): 152-157.

606 Paul, A. G., Jones, K. C., Sweetman, A. J. 2009. A first global production, emission, and
607 environmental inventory for perfluorooctane sulfonate. *Environmental Science &*
608 *Technology*. 43(2): 386-392.

609 Prevedouros, K., Cousins, I. T., Buck, R. C., Korzeniowski, S. H. 2006. Sources, Fate and Transport of
610 Perfluorocarboxylates. *Environmental Science & Technology*. 37(11): 32-44.

611 RIVM. 1996. SimpleBox 2.0: a nested multimedia fate model for evaluating the environmental fate
612 of chemicals. Report no. 719101029. National Institute for Public Health and the
613 Environment, Bilthoven, The Netherlands.

614 Ruyle, B. J., Thackray, C. P., McCord, J. P., Strynar, M. J., Mauge-Lewis, K. A., Fenton, S. E.,
615 Sunderland, E. M. 2021. Reconstructing the Composition of Per- and Polyfluoroalkyl
616 Substances in Contemporary Aqueous Film-Forming Foams. *Environmental Science &*
617 *Technology Letters*. 8(1): 59-65.

618 Schenker, U., Scheringer, M., Macleod, M., Martin, J. W., Cousins, I. T., Hungerbuhler, K. 2008.
619 Contribution of volatile precursor substances to the flux of perfluorooctanoate to the
620 arctic. *Environmental Science & Technology*. 42(10): 3710-3716.

621 Schenker, U., Scheringer, M., Sohn, M. D., Maddalena, R. L., Mckone, T. E., Hungerbuehler, K. 2009.
622 Using Information on Uncertainty to Improve Environmental Fate Modeling: A Case Study
623 on DDT. *Environmental Science & Technology*. 43(1): 128-134.

624 Scheringer, M., Halder, D., Hungerbuhler, K. 2000. Comparing the environmental performance of
625 fluorescent whitening agents with peroxide bleaching of mechanical pulp. *Journal of*
626 *Industrial Ecology*. 3(4): 77-95.

627 Slob, W. 1994. Uncertainty Analysis in Multiplicative Models. *Risk Analysis*. 14: 571-576.

628 Song, Y. U., Shao, M., Liu, Y., Sihua, L. U., Kuster, W., Goldan, P., Xie, S. 2007. Source apportionment
629 of ambient volatile organic compounds in Beijing. *Environmental Science & Technology*.
630 41(12): 4348-4353.

631 Su, C., Lu, Y., Wang, T., Lu, X., Song, S., Li, L., Khan, K., Wang, C., Liang, R. 2018. Dynamic multimedia
632 fate simulation of Perfluorooctane Sulfonate (PFOS) from 1981 to 2050 in the urbanizing
633 Bohai Rim of China. *Environmental Pollution*. 235: 235-244.

634 Suzuki, N., Murasawa, K., Nansai, K., Sakurai, T., Matsushashi, K., Moriguchi, Y., Tanabe, K., Nakasugi,
635 O., MORITA, M., Morita, M. 2005. Development of Geo-Referenced Environmental Fate
636 Model (G-CIEMS) for Chemical Contaminants Based on GIS (Geographic Information
637 System (In Japanese)). *Journal of Environmental Chemistry*. 15(2): 385-395.

638 Wang, G., Yan, D., Pan, T., He, X., Qi, J., Ren, M., Zhao, L., Wang, F., Zhang, Z., Jiang, X. 2019. Study
639 on Technical Schemes for Major Pollutants Emission Reduction in Beijing North Canal

640 River Basin Based on Watershed Water Quality Target Management. *Journal of Water*
641 *Resource and Protection*. 11(11): 1327-1350.

642 Wang, P., Lu, Y., Su, H., Su, C., Jenkins, A. 2020. Managing health risks of perfluoroalkyl acids in
643 aquatic food from a river-estuary-sea environment affected by fluorochemical industry.
644 *Environment International*. 138: 105621.

645 Wang, T., Khim, J. S., Chen, C., Naile, J. E., Giesy, J. P. 2012a. Perfluorinated compounds in surface
646 waters from Northern China: Comparison to level of industrialization. *Environment*
647 *International*. 42(1): 37-46.

648 Wang, T., Lu, Y., Chen, C., Naile, J. E., Khim, J. S., Giesy, J. P. 2012b. Perfluorinated compounds in a
649 coastal industrial area of Tianjin, China. *Environmental Geochemistry and Health*. 34(3):
650 301-311.

651 Wang, Z., Cousins, I. T., Scheringer, M., Buck, R. C., Hungerbühler, K. 2014a. Global emission
652 inventories for C4–C14 perfluoroalkyl carboxylic acid (PFCA) homologues from 1951 to
653 2030, Part I: production and emissions from quantifiable sources. *Environment*
654 *International*. 62-75.

655 Wang, Z., Cousins, I. T., Scheringer, M., Buck, R. C., Hungerbühler, K. 2014b. Global emission
656 inventories for C4–C14 perfluoroalkyl carboxylic acid (PFCA) homologues from 1951 to
657 2030, part II: The remaining pieces of the puzzle. *Environment International*. 69: 166-176.

658 Xu, S., Liu, W., Tao, S. 2006. Emission of polycyclic aromatic hydrocarbons in China. *Environmental*
659 *Science & Technology*. 40(3): 702-708.

660 Young, R. B., Pica, N. E., Sharifan, H., Chen, H., Roth, H. K., Blakney, G. T., Borch, T., Higgins, C. P.,
661 Kornuc, J. J., McKenna, A. M., Blotvogel, J. 2022. PFAS Analysis with Ultrahigh Resolution
662 21T FT-ICR MS: Suspect and Nontargeted Screening with Unrivaled Mass Resolving Power
663 and Accuracy. *Environmental Science & Technology*. 56(4): 2455-2465.

664 Yu, Y., Che, S., Ren, C., Jin, B., Tian, Z., Liu, J., Men, Y. 2022. Microbial Defluorination of Unsaturated
665 Per- and Polyfluorinated Carboxylic Acids under Anaerobic and Aerobic Conditions: A
666 Structure Specificity Study. *Environmental Science & Technology*. 56(8): 4894-4904.

667 Zhang, M., Wang, P., Lu, Y., Lu, X., Sarvajayakesavalu, S. 2019. Bioaccumulation and human
668 exposure of perfluoroalkyl acids (PFAAs) in vegetables from the largest vegetable
669 production base of China. *Environment International*. 135: 37-47.

670 Zhang, M., Wang, P., Lu, Y., Shi, Y., Wang, C., Sun, B., Li, X., Song, S., Yu, M. 2021. Transport and
671 environmental risks of perfluoroalkyl acids in a large irrigation and drainage system for
672 agricultural production. *Environment International*. 157: 106856.

673 Zhang, Y., Wang, B., Wang, W., Li, W., Deng, S., Wang, Y., Yu, G. 2016. Occurrence and source
674 apportionment of Per- and poly-fluorinated compounds (PFCs) in North Canal Basin,
675 Beijing. *Scientific Reports*. 6(1): 1-9.

676 Zhou, Yunqiao, Wang, Tiejun, Qifeng, Pei, Lei, Chen, Shuqin, Zhang. 2018. Spatial and vertical
677 variations of perfluoroalkyl acids (PFAAs) in the Bohai and Yellow Seas: Bridging the gap
678 between riverine sources and marine sinks. *Environmental Pollution*. 238: 111-120.

679