

1 **Measurements and modeling of the interhemispheric differences of**
2 **atmospheric chlorinated very short-lived substances**

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24 **Key Points:**

- 25 • Coincident global scale airborne data from multiple instruments of chlorinated very
26 short-lived substances (Cl-VSLS) provide insights into their distributions and inter-
27 comparability of measurements.
- 28 • The updated model reasonably simulates the global distribution of the major Cl-VSLS
29 species and their relative contributions to the tropospheric chlorine burden.
- 30 • Model results show a disproportionately high impact of Asian Cl-VSLS emissions
31 regionally and globally throughout the troposphere.

32 Abstract

33 Chlorinated very short-lived substances (Cl-VSLS) are ubiquitous in the troposphere and can
34 contribute to the stratospheric chlorine budget. In this study, we present measurements of
35 atmospheric dichloromethane (CH_2Cl_2), tetrachloroethene (C_2Cl_4), chloroform (CHCl_3), and 1,2-
36 dichloroethane (1,2-DCA) obtained during the National Aeronautics and Space Administration
37 (NASA) Atmospheric Tomography (ATom) global-scale aircraft mission (2016-2018), and use
38 the Community Earth System Model (CESM) updated with recent chlorine chemistry to further
39 investigate their global tropospheric distribution. The measured global average Cl-VSLS mixing
40 ratios, from 0.2 to 13 km altitude, were 46.6 ppt (CH_2Cl_2), 9.6 ppt (CHCl_3), 7.8 ppt (1,2-DCA),
41 and 0.84 ppt (C_2Cl_4) measured by the NSF NCAR Trace Organic Analyzer (TOGA) during
42 ATom. Both measurements and model show distinct hemispheric gradients with the mean
43 measured Northern to Southern Hemisphere (NH/SH) ratio of 2 or greater for all four Cl-VSLS.
44 In addition, the TOGA profiles over the NH mid-latitudes showed general enhancements in the
45 Pacific basin compared to the Atlantic basin, with up to ~18 ppt difference for CH_2Cl_2 in the mid
46 troposphere. We tagged regional source emissions of CH_2Cl_2 and C_2Cl_4 in the model and found
47 that Asian emissions dominate the global distributions of these species both at the surface (950
48 hPa) and at high altitudes (150 hPa). Overall, our results confirm relatively high mixing ratios of
49 Cl-VSLS in the UTLS region and show that the CESM model does a reasonable job of
50 simulating their global abundance but we also note the uncertainties with Cl-VSLS emissions
51 and active chlorine sources in the model. These findings will be used to validate future emission
52 inventories and to investigate the fast convective transport of Cl-VSLS to the UTLS region and
53 their impact on stratospheric ozone.

54 Plain Language Summary

55 The Montreal Protocol has phased down the consumption and production of a large number of
56 halogenated compounds such as CFCs, due to their potential for depleting stratospheric ozone.
57 However, the consumption and production of a class of halogenated compounds, referred to as
58 very short-lived substances (VSLS), is not controlled by the Montreal Protocol. Evidence is
59 growing that globally increasing emissions of human-produced chlorinated VSLS (Cl-VSLS)
60 could have an impact on stratospheric ozone. In this work, we present comprehensive aircraft
61 measurements coupled to modeling of the major speciated Cl-VSLS that show their present day

62 global distribution at altitudes up to 12 km and also show that Asian emissions are responsible
63 for the majority of observed Cl-VSLS throughout the troposphere including the Southern
64 Hemisphere.

65 **1. Introduction**

66 Very short-lived substances (VSLS) include the class of halogen (Cl, Br, and I) containing
67 substances that have atmospheric lifetimes typically less than six months. While the majority of
68 VSLS emissions are photochemically destroyed in the troposphere during their lifetime, the
69 *Scientific Assessment of Ozone Depletion* reports – as early as the 1998 edition – have shown that
70 these species do enter the stratosphere and affect the stratospheric ozone layer, with the latest
71 findings reported in the 2022 edition (hereafter called SAOD-2022) by Laube et al. (2022).
72 Unlike brominated VSLS (Br-VSLS) which are primarily emitted by natural sources (Carpenter
73 and Liss, 2000; Keng et al., 2021), chlorinated VSLS (Cl-VSLS) have significant anthropogenic
74 sources (Keene et al., 1999; Claxton et al., 2020) and their relative importance is increasing as
75 their estimated contribution to the total stratospheric chlorine abundance has increased from ~2%
76 in 2000 to ~3.4% in 2017 (Hossaini et al., 2019). This increase is attributed to both reduction of
77 long-lived chlorocarbons and growth of Cl-VSLS abundance (Hossaini et al., 2019).
78 Specifically, the sum of total chlorine from VSLS in the stratosphere (through source gas
79 injection + product gas injection) has increased from 80 parts per trillion (ppt) in 1993 to 130 ppt
80 in 2020 (Laube et al., 2022). In addition, although chlorine is reported to be 45-74 times less
81 efficient than bromine in destroying ozone based on several modeling studies (Daniel et al.,
82 1999; Sinnhuber et al., 2009; Klobas et al., 2020), Cl-VSLS could be as important as Br-VSLS
83 due to their high mixing ratios. For example, Oram et al. (2017) observed high Cl-VSLS mixing
84 ratios at the surface over eastern Asia (median of 755.8 ppt of chlorine in Taiwan) which are
85 about 116 times higher than reported Br-VSLS in the SAOD-2022 (Laube et al., 2022). Oram et
86 al. (2017) also reported high concentrations of Cl-VSLS in tropical regions over the western
87 Pacific Ocean, where they can be rapidly uplifted to the upper troposphere. Additionally, it is
88 shown that keeping the Cl-VSLS flux constant (compared to a zero Cl-VSLS flux scenario) will
89 delay the return date of total column ozone to 1980 levels by approximately seven years
90 (Dhomse et al., 2019; Chipperfield et al., 2020; Laube et al., 2022). Long-lived chlorinated
91 ozone depleting substances (ODSs; i.e., CFCs, HCFCs, etc.) have been overall successfully

92 controlled under the Montreal Protocol. Hence, a better understanding of the sources, global
93 abundance, distribution, and transport of Cl-VSLS to the stratosphere has become more crucial
94 to ensure that their trends do not slow down the progress made in reducing atmospheric Cl and
95 Br abundance and significant progress has been made in addressing this issue (Pan et al., 2017;
96 Feng et al., 2018; Hossaini et al., 2019; Claxton et al., 2020; Adcock et al., 2021; An et al., 2021;
97 Keng et al., 2021; Lauther et al., 2022; Thompson et al., 2022).

98 The most abundant Cl-VSLS in the atmosphere are dichloromethane (CH_2Cl_2), chloroform
99 (CHCl_3), 1,2-dichloroethane ($\text{CH}_2\text{ClCH}_2\text{Cl}$, herein referred to as 1,2-DCA) and tetrachloroethene
100 (C_2Cl_4). The atmospheric lifetimes of these Cl-VSLS differ regionally with the longest lifetimes
101 at high latitudes and altitudes. The SAOD-2022 reported both an average (and a range) for their
102 lifetimes: 1,2-DCA = 81.3 days (41–555 days) and C_2Cl_4 = 109 days (66–245 days), CHCl_3 =
103 178 days (97–1145 days), and CH_2Cl_2 = 176 days (95–1070 days). CHCl_3 , has significant
104 emissions by natural sources (Worton et al., 2006; Feng et al., 2018), but the other three
105 atmospherically abundant Cl-VSLS compounds are largely (about 90%) anthropogenic (Laube et
106 al., 2022) and primarily used in industry. CHCl_3 also has industrial sources and is used in
107 chemical manufacturing; its global emission trend between 2010 and 2015 is explained entirely
108 by increasing emissions in eastern Asia (Fang et al., 2019). Nevertheless, long-term ground
109 observations by the Advanced Global Atmospheric Gases Experiment (AGAGE) network
110 suggest CHCl_3 is currently stable; its annual mean mole fraction remained constant at 8.7 ppt in
111 2019 and 2020 (Laube et al., 2022).

112 CH_2Cl_2 is a co-product of industrial production of CHCl_3 and is used as a versatile chemical
113 solvent and degreasing agent (Feng et al., 2018). It had global annual mean mole fractions of
114 38.3 and 45.5 ppt in 2020 based on the AGAGE and National Oceanic and Atmospheric
115 Administration (NOAA) networks, respectively (Laube et al., 2022). Asia is currently recognized
116 to be the main source region of CH_2Cl_2 with the largest contributions from China and India
117 (Feng et al., 2018; Say et al., 2019; An et al., 2021). It has been estimated that Asian emissions
118 of CH_2Cl_2 increased annually by 51 Gg yr^{-1} from 2006 to 2017 (Claxton et al., 2020). Sustained
119 growth in the atmospheric CH_2Cl_2 mole fraction in coming decades could delay the recovery of
120 stratospheric ozone over Antarctica (Hossaini et al., 2017). However, the rate of increase of
121 global abundance of CH_2Cl_2 slowed after 2016, potentially due to regulatory control and
122 economic drivers (An et al., 2021; Laube et al., 2022).

123 C_2Cl_4 , with annual global mole fractions in the low ppt range, is used as both a dry cleaning
124 solvent and industrial precursor (Simpson et al., 2004; Laube et al., 2022). Unlike CH_2Cl_2 ,
125 emissions of C_2Cl_4 are of the same order in Asia, North America, and Europe (Claxton et al.,
126 2020). While C_2Cl_4 , with four Cl atoms, has the largest number of Cl atoms per molecule
127 amongst the Cl-VSLS discussed, it has a shorter lifetime than CH_2Cl_2 and the emissions of this
128 toxic compound have declined in recent years owing to controls in different countries on its use
129 (Claxton et al., 2020). Similarly, 1,2-DCA which is most commonly used in the production of
130 vinyl chloride (chloroethene), is also a toxic compound. Although not much is known about its
131 global budget as it is not reported by long-term observations networks, limited observations
132 suggest it is the second most abundant Cl-VSLS after CH_2Cl_2 and in some regions of the
133 atmosphere it has a very large abundance (Oram et al., 2017; Laube et al., 2022). Oram et al.
134 (2017) reported 1,2-DCA ground-level mole fractions up to 309 ppt in Taiwan and estimated its
135 annual Asian emissions at about 203 Gg, based on CH_2Cl_2 emissions and its positive correlation
136 with 1,2-DCA measurements.

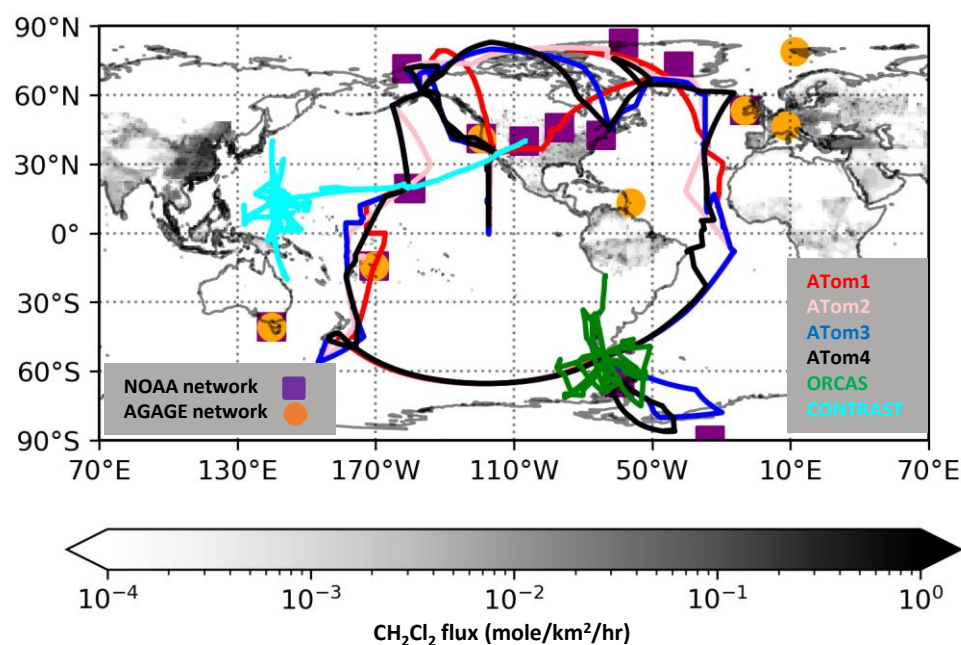
137 Due to their non-uniform emissions and relatively short lifetimes, Cl-VSLS have large variability
138 in their global tropospheric distribution (zonally, meridionally, and vertically), with higher mole
139 fractions close to their source regions. CH_2Cl_2 ground measurements show up to 3.5 times higher
140 mole fractions in the Northern Hemisphere (NH) than in the Southern Hemisphere (SH) (Laube
141 et al., 2022). Additionally, a large vertical gradient is reported over the tropics with sharper
142 slopes for shorter-lived VSLS species (e.g. C_2Cl_4) (Andrews et al., 2016; Pan et al., 2017). Also,
143 there is a significant difference between the global CH_2Cl_2 mean mole fractions reported by
144 AGAGE and NOAA networks, only some of which could be explained by calibration
145 differences, suggesting an impact due to differing sampling locations (Laube et al., 2022). Since
146 the measurements by global networks are also used for developing top-down emission estimates,
147 the limited spatial coverage of measurement stations leads to differences between top-down and
148 bottom-up inventories (An et al., 2021). Similarly, most airborne campaigns that measured VSLS
149 within the last decade focused on specific regions or times of the year (Leedham Elvidge et al.,
150 2015; Oelhaf et al., 2019; Keber et al., 2020; Adcock et al., 2021; Jesswein et al., 2022). In
151 particular, the CAST (Coordinated Airborne Studies in the Tropics), CONTRAST (CONvective
152 TRansport of Active Species in the Tropics), and ATTREX (Airborne Tropical Tropopause
153 EXperiment) airborne studies measured Cl-VSLS over the tropical western Pacific Ocean during

154 January and February 2014 (Andrews et al., 2016; Harris et al., 2017; Pan et al., 2017).
155 Additionally, the ORCAS (O₂/N₂ Ratio and CO₂ Airborne Southern Ocean Study) airborne study
156 (January–February 2016) measured Cl-VSLS over the Southern Ocean (Stephens et al., 2018). In
157 contrast, this study is focused on measurements made during the NASA Atmospheric
158 Tomography (ATom) mission (Thompson et al., 2022), a recent global-scale multi-year mission
159 (2016–2018), which included observations of Cl-VSLS from the NCAR Trace Organic Gas
160 Analyzer (TOGA), the University of California, Irvine Whole Air Sampler (UCI WAS), and the
161 NOAA Programmable Flask Package (PFP) instruments, throughout all four seasons, to
162 characterize their global chemistry and transport.

163 Halogens have a broad range of impacts on atmospheric chemistry, including catalytic loss of
164 tropospheric ozone. In particular, rapid cycling between halogen radicals and their reservoirs
165 converts NO_x to HNO₃, reducing ozone concentrations, which further affects the OH abundance
166 as the principal atmospheric oxidant (Simpson et al., 2015; Wang et al., 2021; Li et al., 2022).
167 The recent addition of halogen chemistry to global models has enabled scientists to better
168 understand the global impact of chlorine sources on atmospheric chemistry (Ordóñez et al.,
169 2012; Saiz-Lopez and von Glasow, 2012; Fernandez et al., 2014; Simpson et al., 2015; Wang et
170 al., 2021). Hossaini et al. (2016) updated the tropospheric chlorine chemistry in the TOMCAT
171 model, which was used later by Claxton et al. (2020) to constrain the global emissions of CH₂Cl₂
172 and C₂Cl₄ based on available atmospheric long-term and campaign-based measurements.
173 Hossaini et al. (2019), Chipperfield et al. (2018), and Chipperfield et al. (2020) used this updated
174 TOMCAT model to investigate the trend of stratospheric chlorine from Cl-VSLS, their impact
175 on lower stratospheric ozone, and ozone layer recovery, respectively. Li et al. (2022) updated the
176 halogen chemistry in the Community Earth System Model (CESM) and studied the impact of
177 halogens on the global methane burden. Nevertheless, prior to this study a comprehensive
178 evaluation of the global models in reproducing the global distribution and seasonal variation of
179 Cl-VSLS had not been performed.

180 In this work, we present the spatial and vertical distribution of CH₂Cl₂, C₂Cl₄, 1,2-DCA, and
181 CHCl₃ measured by the TOGA, WAS, and PFP instruments during ATom. We highlight the
182 latitudinal and seasonal variability of Cl-VSLS and use tracers to understand their distributions
183 and interhemispheric difference. The CESM model with updated chlorine chemistry is used to
184 simulate the Cl-VSLS during the ATom mission period and the model output is compared with

185 the measurements. We also evaluate the model results against co-located AGAGE and NOAA
 186 ground-site observations and use the reaction rate information from the model output to
 187 investigate the dominant photochemical destruction pathways of Cl-VSLS. Finally, we tag the
 188 emissions in different regions of the world to estimate their impact on global tropospheric
 189 abundance of Cl-VSLS.



190

191 **Figure 1.** Map of the flight tracks for individual ATom deployments, and ORCAS and CONTRAST missions. The
 192 ground measurement sites in the NOAA and AGAGE networks used in this study are also shown. The map is
 193 shaded by the 2016 annual average CH₂Cl₂ emissions developed by Claxton et al. (2020).

194 2. Methods

195 2.1. Measurements

196 This work is primarily based on observations made during the NASA ATom mission. ATom was
 197 a global-scale mission that flew the heavily instrumented NASA DC-8 aircraft over remote
 198 regions during four seasons between 2016 and 2018 (Thompson et al., 2022). Specifically,
 199 ATom collected measurements, with regular profiling between 200 m and ~13 km altitude, from
 200 28 July to 22 August 2016 (ATom-1), 26 January to 22 February 2017 (ATom-2), 28 September
 201 to 26 October 2017 (ATom-3), and 24 April to 21 May 2018 (ATom-4). In this work, we assign
 202 each ATom deployment to the month with the majority of data, e.g., August 2016 for ATom-1.
 203 This mission provides a comprehensive dataset to characterize the chemistry and transport of

204 chemical species in the remote regions of the troposphere and the UTLS (upper troposphere and
205 lower stratosphere). We specifically present the Cl-VSLS data measured during the ATom
206 mission. In addition, we compare ATom data with our TOGA measurements made during
207 ORCAS and CONTRAST and use surface observation data measured by ground monitoring
208 networks around the globe to complement the evaluation of the model performance.

209 Table 1 summarizes the measurements used in this study.

210 **2.1.1. ATom Cl-VSLS observations**

211 As shown in Figure 1, the ATom mission consisted of four deployments with very similar near-
212 pole-to-pole flight tracks over remote regions with regular profiling throughout much of the
213 troposphere, which provides a unique dataset to better understand the spatial and seasonal
214 variations of Cl-VSLS. Within these deployments, three systems provided measurements of Cl-
215 VSLS mixing ratios: 1) the NCAR Trace Organic Gas Analyzer (TOGA) instrument, 2) the
216 University of California, Irvine Whole Air Sampler (UCI WAS), and 3) the NOAA
217 Programmable Flask Package (PFP).

218 The three systems used in this study to measure Cl-VSLS, have been previously documented in
219 the literature and will only be briefly described here. TOGA is an *in-situ* fast gas chromatograph
220 quadrupole mass spectrometer (GC/MS) that continuously measures a wide range of VOCs.
221 During ATom, TOGA analyzed samples collected over 35 seconds every 2 minutes (Apel et al.,
222 2015). The UCI WAS (Simpson et al., 2010) had 168 electropolished stainless steel (SS)
223 canisters on board for each flight and collected pressurized whole air samples throughout the
224 flights, sampling every 2–5 minutes for between 20 seconds to 2 minutes in duration, pressure
225 altitude dependent. Post flight, the canisters were returned to the UCI laboratory where they were
226 analysed using a series of gas chromatographs. Likewise, the NOAA PFP (Hu et al., 2015;
227 Sweeney et al., 2015) system collected whole air samples in up to 24 glass flasks per flight (48
228 flasks during ATom-2), typically averaging one 10-s to 20-s sample every 10–20 minutes. These
229 flasks were returned to the NOAA laboratory for analysis by gas chromatography with mass
230 spectrometric detection. Data obtained from all three methodologies were used to analyze results
231 for a large suite of VOCs, including Cl-VSLS. TOGA uses NOAA calibration scales and on-
232 board standards and cross calibrates with a multi-component calibration standard from the
233 National Institute of Standards and Technology (NIST). Specifically, TOGA referenced their

234 measurements for CH_2Cl_2 and CHCl_3 to the NOAA ESX-3583 standard and referenced their
235 measurements for C_2Cl_4 and 1,2-DCA to the NIST CAL 014921 calibration mixture,
236 respectively. In addition, TOGA used NCAR prepared standards as in-flight working standards
237 for all species. NOAA PFP used their calibration scales for all the compounds including a
238 recently developed NOAA-2019 scale for 1,2-DCA. UCI-WAS data are reported based on their
239 own set of calibration standards and thus their calibration is independent of both NOAA PFP and
240 NCAR TOGA.

241 The TOGA instrument measured 12,168 samples during the four ATom deployments, and WAS
242 and PFP collected and analyzed 6,991 and 1,109 flask samples, respectively. For statistical
243 analyses of the data, we replaced observations below the lower limit of detection (LLOD) with
244 $0.5 \times \text{LLOD}$ based on the U.S. EPA guidance for data quality assessment (USEPA, 2000);
245 approximately 15% of TOGA samples were below the LLOD for C_2Cl_4 (Table S1). For
246 comparisons to the observations, the model output was sampled instantaneously along the ATom
247 flight tracks using TOGA sampling times and locations. We use normalized mean bias (NMB),
248 root mean squared error (RMSE), and Pearson correlation coefficient (r) as statistical metrics for
249 evaluation.

250 **2.1.2. Ground-based observations**

251 We used the monthly mean mole fractions of ground-based CH_2Cl_2 and C_2Cl_4 measurements for
252 a three-year period, 2016–2018, from the National Oceanic and Atmospheric Administration
253 (NOAA) and Advanced Global Atmospheric Gases Experiment (AGAGE) monitoring networks
254 (Prinn et al., 2018; Claxton et al., 2020). Specifically, we used data from 14 NOAA and 7
255 AGAGE sites measured using GC/MS systems (the stations used are listed in Table S2). As
256 shown in Figure 1, the majority of these sites are located in the NH, and co-located
257 measurements are made by these networks at four sites (two in each hemisphere). For CH_2Cl_2 ,
258 the NOAA and AGAGE networks use NOAA-2003 and AGAGE SIO-14 calibration scales; we
259 increased the AGAGE measurements by 10.38% following the known calibration offset between
260 results for this chemical in these networks (Claxton et al., 2020). It is worth mentioning that
261 AGAGE network uses *in-situ* measurement systems, while the NOAA network uses flasks,
262 which are transported and analyzed at the NOAA Global Monitoring Laboratory in Boulder,
263 Colorado.

264 2.1.3. Auxiliary data from ORCAS and CONTRAST

265 We complemented our analysis with Cl-VLSL observations from two additional field campaigns
 266 (Figure 1). We used CH₂Cl₂, CHCl₃, and C₂Cl₄ data measured by TOGA from the O₂/N₂ Ratio
 267 and CO₂ Airborne Southern Ocean Study (ORCAS) and the CONvective TRANsport of Active
 268 Species in the Tropics (CONTRAST) missions, both deployed on the NSF/NCAR Gulfstream V
 269 (GV) aircraft. ORCAS was a six-week study in 2016 (January–February) focused on the
 270 Southern Ocean (35–75°S) and collected data by frequently profiling the atmosphere up to ~13
 271 km altitude (Stephens et al., 2018). Similarly, during CONTRAST the GV profiled up to ~15 km
 272 altitude during a seven-week study in 2014 (January–February), focusing on deep convection in
 273 the tropical western Pacific (20°S–40°N) (Pan et al., 2017).

274 **Table 1.** Summary of airborne missions and measurement instruments used in this study.

Instrument	Mission (# of samples)	Methodology	Measurement Details	Inlet Setup	Reported Uncertainties ^a				Ref.
					CH ₂ Cl ₂	C ₂ Cl ₄	CHCl ₃	1,2-DCA	
NCAR TOGA	ATom (12,168), ORCAS (2573), CONTRAST (3568)	<i>in situ</i> Fast GC/MS	35-s samples every 2 min	Heated Sulfinert® tubing	± 15%	± 30%	± 15%	± 30%	Apel et al. (2015)
UCI WAS	ATom (6,991)	Stainless steel flask collection and laboratory analysis by multi-detector GC	Variable – flasks sampled on demand (average 45-s sample every 3.5 min)	Unheated stainless steel	± 20%	± 20%	± 20%	± 25%	Simpson et al. (2010)
NOAA PFP	ATom (1,121)	Glass flask collection and laboratory analysis by GC/MS	Variable – flasks sampled on demand (average 15-s sample every 10-20 min)	Chemically-passivated SS tubing	± 10%	± 15%	± 20%	± 15%	Sweeney et al. (2015); Hu et al. (2015)

275 ^a The uncertainties reported here represent the combination of estimated scale uncertainties and measurement
 276 precision.

277 2.2. Simulations

278 2.2.1. Model configuration

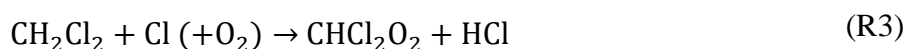
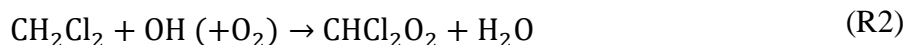
279 CESM is a coupled climate/Earth system modeling framework with different components
 280 including land and atmosphere (Hurrell et al., 2013). CAM-chem, the Community Atmosphere
 281 Model with Chemistry, is the atmospheric component of CESM that solves the physics,

282 dynamics, and chemistry of the troposphere and stratosphere (Lamarque et al., 2012). The online
283 interaction between meteorology, thermodynamic processes, and chemistry makes CESM an
284 appropriate model to study the global distribution and transport of VSLS (Wang et al., 2019a).
285 The version of CAM-chem (based on CESM1) used in this study is based on the Community
286 Atmosphere Model, version 4 (Neale et al., 2013; Tilmes et al., 2016), in which the hydroxyl
287 radical (OH) abundance is calculated online (i.e., not prescribed). The horizontal resolution is
288 $0.9^\circ \times 1.25^\circ$ (latitude \times longitude) with 56 vertical levels in a hybrid sigma-pressure system. The
289 top boundary of the model is at 3 hPa (\approx 40 km). We ran the model for three years 2016–2018
290 (after two years of spin-up) in specified dynamics mode, in which the temperature and horizontal
291 winds are nudged with a 50-h relaxation time towards data from NASA’s Modern-Era
292 Retrospective analysis for Research and Applications, Version 2 (MERRA-2) (Gelaro et al.,
293 2017). The anthropogenic emissions are from the Copernicus Atmosphere Monitoring Service
294 global emission inventory version 5.1 (CAMsv5.1) and Coupled Model Intercomparison Project
295 Phase 6 (Hoesly et al., 2018; Granier et al., 2019), biomass burning emissions are from the Fire
296 Inventory from NCAR, version 2.5 (FINN v2.5) based on MODIS fire detections (Wiedinmyer
297 et al., 2023), and emissions from vegetation are calculated online in the land component of the
298 model (CLM) based on the Model of Emissions of Gases and Aerosols from Nature (MEGAN;
299 Guenther et al. (2012)). For CH_2Cl_2 and C_2Cl_4 emissions, we used the recently-developed
300 emission inventory by Claxton et al. (2020). They constrained the long-term global emissions of
301 these two species up to 2017 (with no seasonality) based on a combination of long-term
302 observation data, recent bottom-up emissions for Asia, and Reactive Chlorine Emissions
303 Inventory (RCEI) framework for other regions, which led to improved modeling results (we
304 extrapolated the emissions for the year 2018). The RCEI was a pioneering effort to produce the
305 gridded global emission inventory of chlorinated species, which enabled assessment of
306 atmospheric chlorine cycle (Keene et al., 1999). A latitude-dependent lower boundary condition
307 (LBC) for CHCl_3 and 1,2-DCA was prescribed with higher values in the NH (Hossaini et al.,
308 2016).

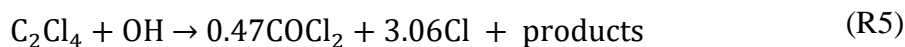
309 The tropospheric and stratospheric chemistry scheme is based on the MOZART-4 chemical
310 mechanism implemented in CAM4-chem (Kinnison et al., 2007; Emmons et al., 2010; Tilmes et
311 al., 2016). In addition, we used an updated VSLS chemistry scheme that includes the four Cl-
312 VSLS, specifically CH_2Cl_2 , C_2Cl_4 , CHCl_3 , and 1,2-DCA. This scheme is based on the halogen

313 chemistry mechanism implemented by Ordóñez et al. (2012) and includes updates on halogen
 314 heterogeneous reactions and rate constants (Fernandez et al., 2014; Saiz-Lopez et al., 2014; Li et
 315 al., 2022). These updates will be included in the upcoming CESM2.3 release. The photochemical
 316 destruction pathways for the four Cl-VSLS are through reactions with OH and chlorine atom (Cl)
 317 and photolysis ($h\nu$) as shown below (R1–R12) (Hossaini et al., 2016).

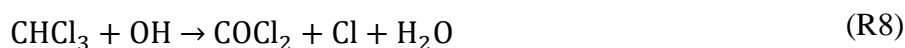
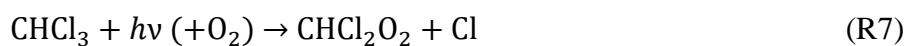
CH₂Cl₂:



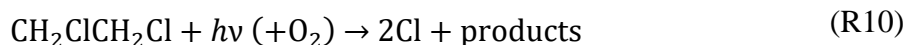
C₂Cl₄:



CHCl₃:



1,2-DCA:



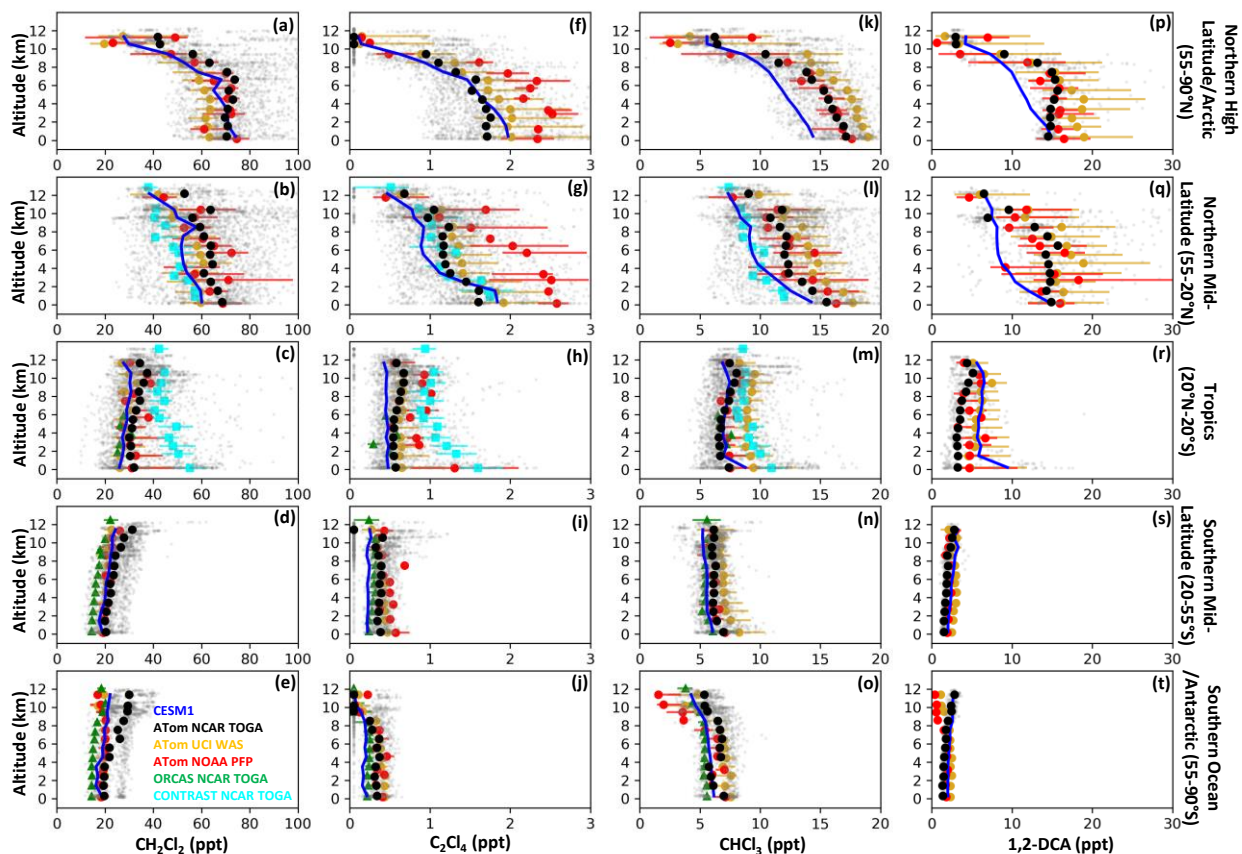
318 2.2.2. Tagged species

319 Tagged tracers are artificial emitted species treated as the replica of original tracers, going
 320 through similar reactions in the chemical mechanism and continuity equations as the original
 321 tracers, while not affecting the interactive chemistry of the atmosphere (Emmons et al., 2012).
 322 To study different source regions' contributions to the atmospheric abundance of anthropogenic

323 Cl-VSLS, we tagged CH_2Cl_2 and C_2Cl_4 emissions in the model simulation. Note that CHCl_3 and
324 1,2-DCA were based on LBC in this work and hence not tagged. We tagged CH_2Cl_2 and C_2Cl_4
325 emitted from Asia (AS), Europe (EU), North America (NA), South America (SA), and Africa
326 (AF). Additionally, we assigned emissions outside these regions as the rest of the world (ROW)
327 tracers. As a result, we implemented a total of 12 tagged tracers in the model. The emission
328 magnitudes from each region of the tagged tracer was equal to the emission of that specific
329 region in the original tracer emission inventory. Figure S1 shows that AS emissions are the
330 dominant emission source of CH_2Cl_2 while AS, EU, and NA emissions of C_2Cl_4 are of the same
331 order of magnitude. These tagged emissions go through photolysis, reaction with OH, and
332 reaction with Cl. The primary source of inorganic chlorine in the model is through the
333 heterogeneous reactions of halogenated species and sea salt aerosol (SSA) as well as
334 mobilization of Cl^- from SSA through acid displacement (following Li et al., 2022). We use
335 these emissions to estimate the tropospheric Cl-VSLS budgets due to different sources.

336 **3. Results and discussion**337 **3.1. Vertical and spatial distribution of Cl-VSLS**

338 The vertical profiles of measured CH_2Cl_2 , C_2Cl_4 , CHCl_3 , and 1,2-DCA mixing ratios in different
 339 latitude bins for all the samples analyzed during ATom-1 through ATom-4 are shown in



340
 341 Figure 2 (vertical profiles for individual ATom deployments are shown in the supporting
 342 information; Figures S2–S5). In this section, we focus only on the measured data and different
 343 instruments, and the model performance is discussed in Section 3.3. The general trend for all the
 344 species is that the mixing ratios in the NH are higher than the SH. For example, the average (\pm
 345 standard deviation) of the measured CH_2Cl_2 mixing ratios by TOGA, within the vertical profile
 346 and all latitudes of the NH and SH (including their subtropical regions), were 59 ± 22 and 26 ± 7
 347 ppt, respectively. Hossaini et al. (2017) found the NH to SH ratio (NH/SH) of 3 in mid-latitudes
 348 for CH_2Cl_2 , based on data from ground measurement networks, which is within the *in-situ*
 349 NH/SH possible range of our TOGA measurements (1.1–4.3).

350 Similarly, for C_2Cl_4 , the mean mixing ratios in the NH and SH were 1.1 ± 0.6 and 0.3 ± 0.2 ppt,
351 respectively. These C_2Cl_4 mixing ratios are significantly lower than those reported in the 20th
352 century (Simmonds et al., 2006), consistent with its declining emission trend (Claxton et al.,
353 2020). For example, Wiedmann et al. (1994) reported mean NH and SH C_2Cl_4 values of 21 ± 5
354 ppt and 2.2 ± 0.5 , respectively, for the period between 1982 and 1989. TOGA data for 1,2-DCA
355 (during ATom-4) shows mixing ratios of 12 ± 7 and 2 ± 1 in NH and SH, respectively,
356 supporting the idea of their predominant NH sources, which are believed to be dominated by
357 anthropogenic sources (Oram et al., 2017; Laube et al., 2022).

358 The interhemispheric gradient is also apparent for $CHCl_3$ with an NH/SH ratio of ~ 2 (1–3.1)
359 during ATom for TOGA data. While anthropogenic sources (i.e., solvents in the industrial
360 sector) are the main source of CH_2Cl_2 , 1,2-DCA, and C_2Cl_4 emissions, these sources have
361 historically accounted for less than 30% of emitted $CHCl_3$ (Worton et al., 2006) although a larger
362 anthropogenic contribution is more recently reported (Fang et al., 2019). An increase in NH
363 anthropogenic emissions of $CHCl_3$ between 2011 and 2017 (and a decrease after 2018) was also
364 reported in SAOD-2022 (Laube et al., 2022).

365 There are also regional differences for the Cl-VSLS, primarily over the NH mid-latitudes. Figure
366 3 shows the altitude dependence of Cl-VSLS differences measured over the Pacific and Atlantic
367 Oceans (Pacific minus Atlantic), in 1-km altitude bins. Our measurements show that the NH
368 mixing ratios over the Pacific Ocean are in general higher than those over the Atlantic Ocean.
369 This enhancement is strongest for CH_2Cl_2 with a difference of up to 18 ppt in the mid-
370 troposphere. In addition, the results show a seasonal difference. For example, the results for
371 ATom-2 (May 2017) and ATom-3 (October 2017) are the extremes of the measured differences,
372 in particular for C_2Cl_4 and $CHCl_3$, meaning the smallest and largest differences occurred during
373 these two deployments (see Figure 3b-c).

374 **3.1.1. CH_2Cl_2**

375 For CH_2Cl_2 in the northern high-latitude region ($55\text{--}90^\circ\text{N}$), measurements indicate that the
376 atmosphere is well mixed in the troposphere and there is a strong vertical gradient in the UTLS
377 region. In the northern mid-latitudes ($20\text{--}55^\circ\text{N}$), the measured data show a smooth vertical
378 gradient, where the median TOGA measurements decreased, overall, by 1.3 ppt km^{-1} with
379 increasing altitude in contrast with 2.5 ppt km^{-1} in the high latitudes (the mean slope is calculated
380 based on lowest and highest altitude bins). It should be noted that the slopes differ for each

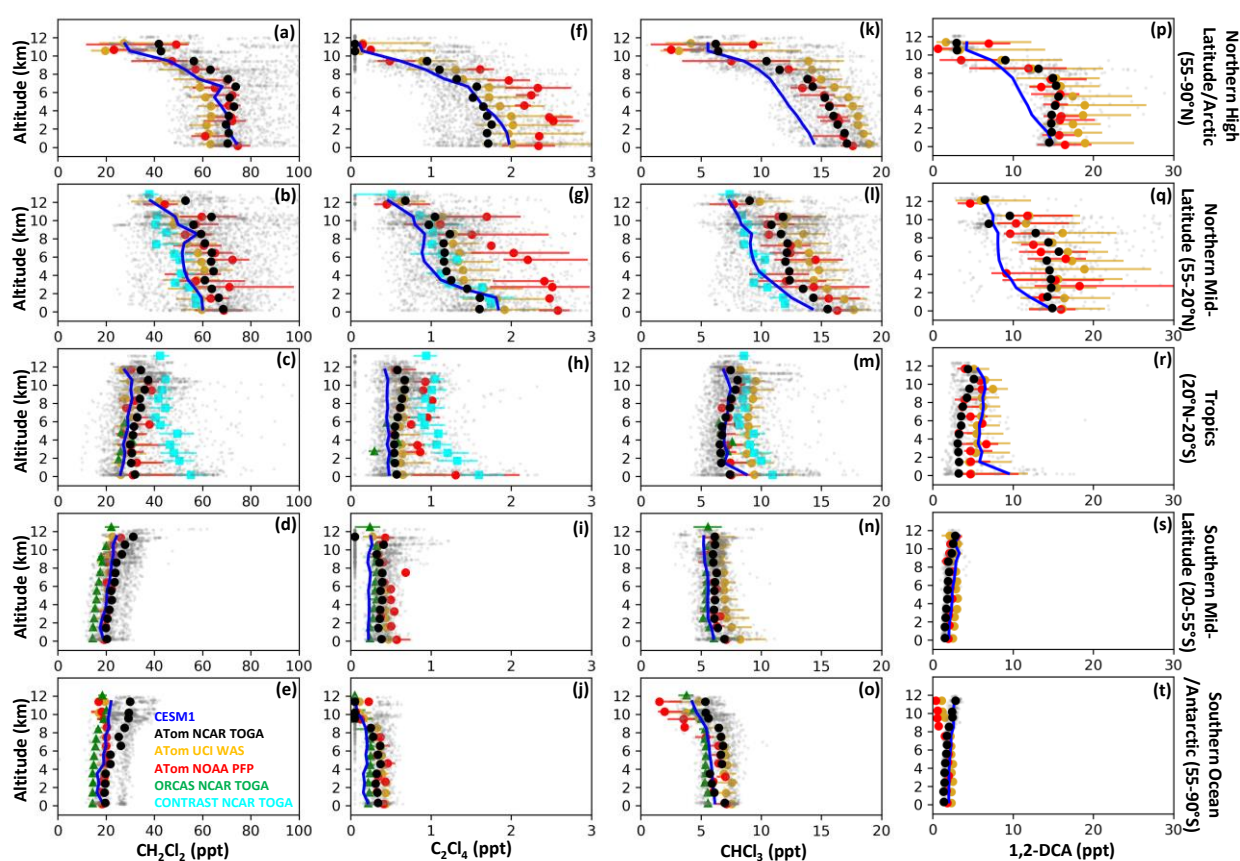
381 ATom deployment (Figure S2). The model captured the seasonal pattern for each ATom
382 deployment (note that emissions do not have seasonality (Claxton et al., 2020)), suggesting that
383 OH abundances control the seasonal trend (i.e., R2 is the main photochemical destruction
384 pathway for CH₂Cl₂ as will be discussed in Section 3.5). Similarly, Claxton et al. (2020) found
385 that their constant monthly emissions were able to generate CH₂Cl₂ seasonality. In the tropics,
386 the CH₂Cl₂ levels are about half of the values for the NH mid- and high latitudes. In addition, the
387 observed tropical mixing ratios are higher at upper altitudes such that the maximum median
388 CH₂Cl₂ value in the profile was at ~10 km. The relatively high mixing ratios in the upper
389 troposphere and reversal of the vertical gradient in the tropics suggest influential impacts of
390 atmospheric dynamics (i.e., transport and convection), pointing to the interhemispheric transport
391 from NH to SH. Similarly in the southern mid-latitudes (20–55°S) and over the Southern Ocean
392 (55–90°S) TOGA CH₂Cl₂ mixing ratios were higher in the upper troposphere than the lower
393 troposphere (inversion with slope ~ -0.9 ppt km⁻¹, where a negative slope indicates increasing
394 levels with increasing height).

395 The CH₂Cl₂ observations from all three instruments (TOGA, WAS, and PFP) compare quite well
396 with each other (Figure 2a-e) and lead to a similar conclusion although there are some minor
397 differences in some regions. For example, in the high-latitude Arctic region (Figure 2a), the
398 vertically-binned median CH₂Cl₂ values from TOGA and PFP are in good agreement with each
399 other yet the WAS data are about 10 ppt lower throughout the profile. The comparisons differ
400 somewhat for each ATom deployment (Figure S2). Specifically, during ATom-2, the medians
401 for all three instruments were about 10 ppt from each other within the marine boundary layer
402 (MBL) and very close in the UTLS region. On the other hand, during ATom-3 the median for
403 WAS and PFP CH₂Cl₂ levels were consistent while TOGA was as much as 15 ppt higher. In
404 addition, TOGA and WAS agreed well during ATom-4 while PFP was on the higher end of the
405 measurements. Some additional inconsistencies were observed between TOGA and the flask
406 measurements from PFP and WAS over the southern high latitudes. TOGA measurements above
407 8 km over the Southern Ocean increased with altitude while WAS and PFP measurements
408 showed a decrease, leading to large biases (for example, the CH₂Cl₂ mixing ratio within the 10-
409 km bin (10–11 km altitude) during ATom-4 was 31.5 ppt and 13.8 ppt measured by TOGA and
410 WAS, respectively). As will be discussed in Section 3.3, the model also showed a modest

411 increase, which is consistent with TOGA data although the TOGA data showed a substantially
 412 larger increase.

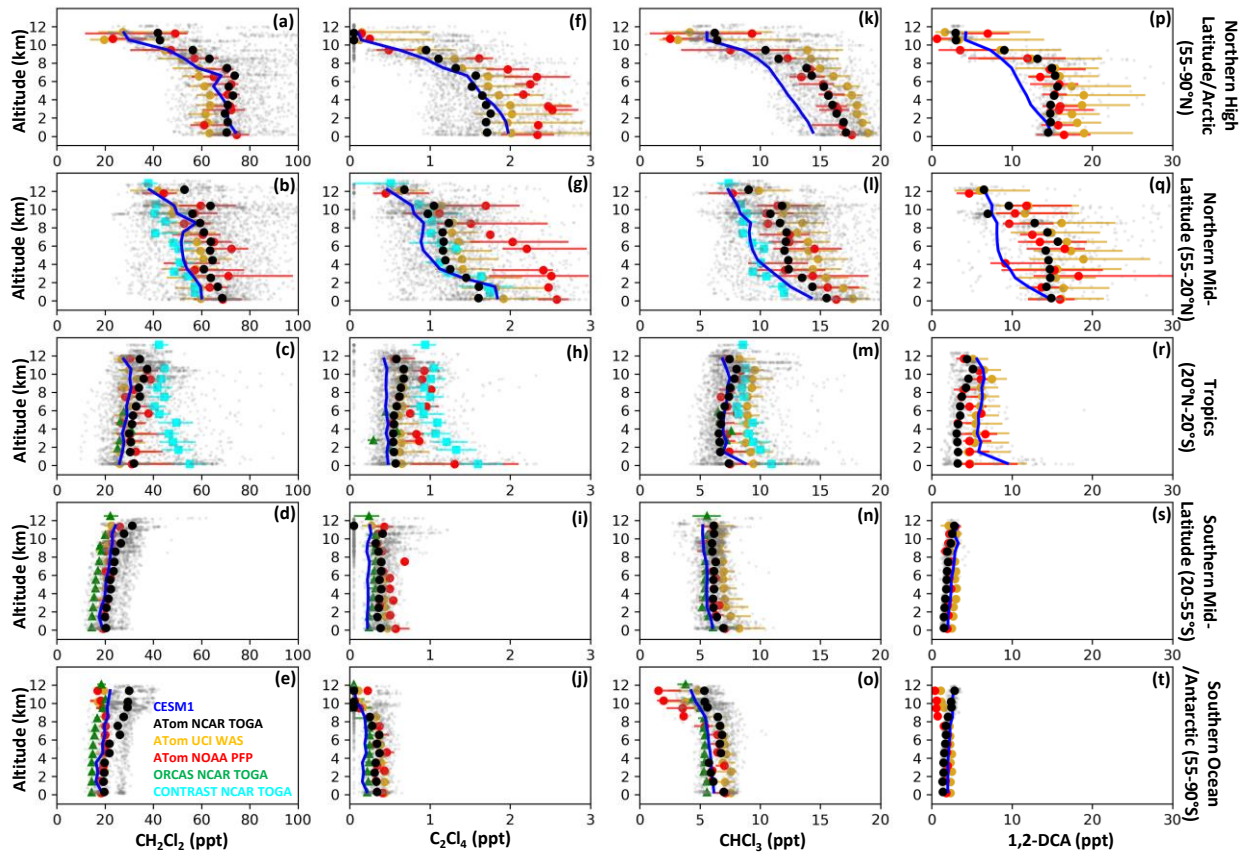
413 3.1.2. C_2Cl_4

414 C_2Cl_4 mixing ratios are very low (< 5 ppt) everywhere. Nevertheless, C_2Cl_4 has vertical and
 415 spatial gradients similar to CH_2Cl_2 . At high latitudes, median TOGA measurements show a
 416 smooth gradient in the troposphere, and transition to below LLOD with a strong gradient in the
 417 UTLS region. Similarly, both WAS and PFP measured very small C_2Cl_4 levels in the UTLS
 418 region. However,

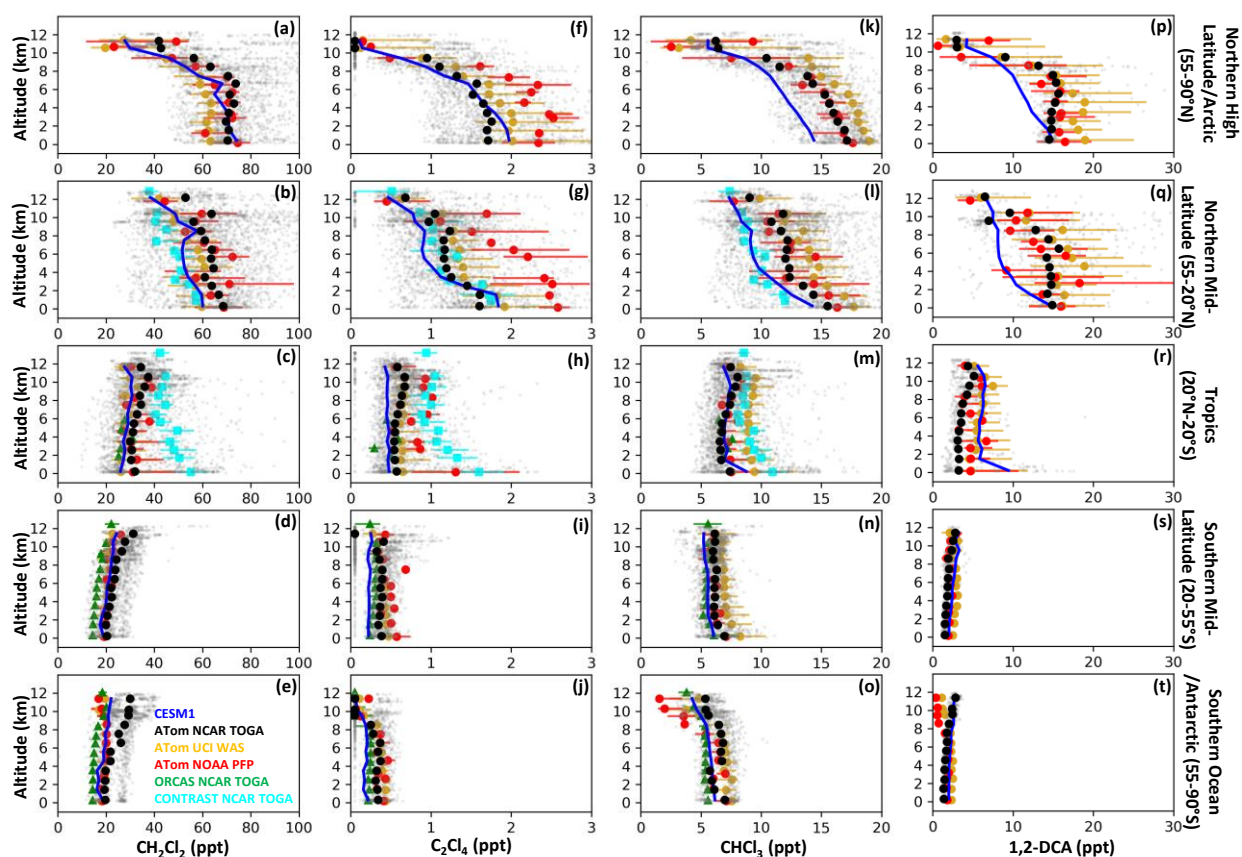


419
 420 Figure 2 shows fairly significant differences that are most pronounced in the NH between PFP
 421 and the other two instruments that measured C_2Cl_4 , even when only ATom-4 results are
 422 considered. This may be due to differences in calibration scales, which we have not pursued
 423 further in this work, and may relate to Teflon O-ring contamination of some PFP flask data. It is
 424 worth noting that PFP results were reported for C_2Cl_4 only during the ATom-4 deployment
 425 (Figure S3). In northern mid-latitudes, the median TOGA C_2Cl_4 levels decreased with increasing

426 altitude down to 0.7 ppt above 10 km. In the tropics, median high-altitude C_2Cl_4 mixing ratios
 427 were slightly higher than at low altitudes. However, the median data during the ATom mission
 428 (for all three measurement systems) did not show an inverted vertical gradient in the SH.
 429 Nevertheless, a wider range of mixing ratios (gray dots) was seen at high altitudes in the
 430 southern mid-latitudes (



431
 432 Figure 2i) with mixing ratios reaching > 1 ppt, suggesting an inversion, with less intensity but
 433 similar to CH_2Cl_2 data.

434 **3.1.3. CHCl₃**435 The observed CHCl₃ mixing ratios also showed a vertical gradient in the NH (

436

437 Figure 2k,l). However, unlike the predominantly anthropogenically-emitted CH₂Cl₂ and C₂Cl₄,
 438 the low-altitude CHCl₃ levels are modestly higher in the NH high latitudes compared with the
 439 mid-latitudes. Specifically, TOGA CHCl₃ levels below 3 km in the 60–90°N latitude bin were
 440 8% higher than in the 30–60°N latitude bins, with the highest differential measured during
 441 ATom-1 (18%). This is in contrast to constant CHCl₃ mixing ratios between 30–90°N shown by
 442 Prinn et al. (2000), which is used in our model. The global emissions of CHCl₃ showed a rapid
 443 increase between 2010 and 2017 (Laube et al., 2022), which is attributed to anthropogenic
 444 emission increases in China (Fang et al., 2019; An et al., 2023). On the other hand, CHCl₃ has
 445 major natural sources with higher emissions in the mid-latitude regions (Khalil and Rasmussen,
 446 1999). The combined impact of transport of emissions (natural and anthropogenic) and
 447 distribution of natural emissions could explain, in part, the higher mixing ratios at 60–90°N. In
 448 the tropics, TOGA observed a declining gradient within the MBL, with increasing altitude, and

449 an increasing gradient at higher altitudes (i.e., change of the gradient sign), similar to what was
450 observed for CH_2Cl_2 and C_2Cl_4 . This inverted “S” shape profile has been also observed for
451 brominated compounds during the CONTRAST mission and was attributed to the convective and
452 advective transport from the surface to the free troposphere (Butler et al., 2018). In the SH, the
453 overall trend suggests a small decreasing gradient although it differs for individual ATom
454 deployments. For example, in the southern mid-latitudes, TOGA measured a decreasing vertical
455 gradient during ATom-1 and ATom-3, and an increasing vertical gradient during ATom-2 and
456 ATom-4. Measurements by WAS and PFP were, in general, consistent with TOGA although
457 WAS reported up to 30% higher mixing ratios during ATom-1 compared with the other two
458 instruments.

459 **3.1.4. 1,2-DCA**

460 The measurements of 1,2-DCA mixing ratios show significant enhancements in the NH (Figure
461 2). In addition, the similar vertical profile shape of 1,2-DCA and CH_2Cl_2 suggests their emission
462 sources are co-located with each other (Oram et al., 2017). Similar to CH_2Cl_2 , the Southern
463 Ocean vertical profile of 1,2-DCA between TOGA, WAS, and PFP are different (Figure 2t). The
464 vertical profiles of measured 1,2-DCA for each deployment are in the SI (Figure S5). The
465 SAOD-2018 reported 1,2-DCA as the second most abundant anthropogenic Cl-VSLS, after
466 CH_2Cl_2 , in the tropical MBL with a median of 12.8 ppt based on measurements during CAST
467 and CONTRAST in 2013–2014 (Andrews et al., 2016; Pan et al., 2017; Engel et al., 2018).
468 TOGA measurements, Table 2, show that 1,2-DCA mixing ratios during ATom (e.g., 1.8–11.3
469 ppt in the tropical MBL) are lower than that of CHCl_3 (e.g., 5.5–12.3 ppt in tropical MBL). Our
470 observations with lower levels of 1,2-DCA compared to CHCl_3 may be due to the region studied
471 during CAST and CONTRAST being further west (and closer to the emission sources) and the
472 shorter atmospheric lifetime of 1,2-DCA compared to CHCl_3 , or due to either decreasing 1,2-
473 DCA emissions or increasing CHCl_3 emissions or a difference in measurement calibration.

474 **3.1.5. Total chlorine during ATom**

475 Overall, there is reasonable agreement between TOGA, PFP, and WAS Cl-VSLS measurements
476 during the ATom mission. In particular, all the measurements agreed within the reported
477 uncertainties. Nevertheless, TOGA provided better data coverage than PFP and WAS as was
478 mentioned in the methods section. As a result, we only use TOGA measurements for the

479 remainder of this paper. The measured TOGA annual global mean mixing ratios \pm standard
480 deviations for CH_2Cl_2 (46.6 ± 24.2 ppt), CHCl_3 (9.6 ± 4.2 ppt), 1,2-DCA (7.8 ± 7.3 ppt), and
481 C_2Cl_4 (0.84 ± 0.66 ppt) confirm that CH_2Cl_2 is the most abundant Cl-VSLS in the atmosphere.
482 Table 2 summarizes the TOGA-measured anthropogenic Cl-VSLS and total VSLS-sourced
483 chlorine, Cl^{VSLS} , during the ATom mission for different latitude bins and altitude ranges
484 (represented by potential temperature, θ). Here, total Cl^{VSLS} is calculated based on the available
485 chlorine from major Cl-VSLS (CHCl_3 , CH_2Cl_2 , C_2Cl_4 , and 1,2-DCA). In the NH and tropics, the
486 highest amount of total Cl^{VSLS} was observed within the MBL (i.e., $\theta < 320$ K). However, the
487 highest values in the SH were found in the upper troposphere (i.e., $\theta > 340$ K), suggesting an
488 interhemispheric transport as discussed below. Within the tropical upper troposphere, the median
489 total Cl^{VSLS} (and range) was 99.6 (76.7–130.5) ppt. Table S3 summarizes the ATom measured
490 mole fractions of Cl-VSLS, in the tropical upper troposphere, alongside the reported values in the
491 literature (Adcock et al., 2021; Oram et al., 2017; SAOD, 2022). Both the total Cl^{VSLS} and the
492 median mixing ratios of individual Cl-VSLS (i.e., CH_2Cl_2 , C_2Cl_4 , and CHCl_3) in the tropics
493 (Table S3) are close to the range of the reported values in the SAOD-2022 report (Laube et al.,
494 2022). However, the differences for 1,2-DCA were large; ATom data (median of 4.8 ppt) were
495 about 40% lower than SAOD (2022) data (median of 8.3 ppt). On the other hand, the upper
496 bound of measurements in South and East Asia (Oram et al., 2017; Adcock et al., 2021) were
497 significantly high, as they were close to the source regions. For example, the upper bound of
498 CH_2Cl_2 mixing ratios were 136 ppt in South Asia (Adcock et al., 2021), in contrast with 47.5 ppt
499 measured during ATom.

500 **3.1.6. Uncertainties between instruments**

501 As discussed above, there are some locations with poor agreement between the Cl-VSLS
502 measured by the three instruments, in particular over the Southern Ocean. From our analysis
503 using the five latitude bins shown in Figure 2, the Southern Ocean bin had the lowest average
504 temperatures and water vapor mixing ratios. Above 8 km, the average temperature was 214.8 K
505 and the average water vapor mixing ratio was 25.9 ppm. The latitude bin with next lowest mean
506 water vapor mixing ratio was over the northern high latitudes at 56.2 ppm followed by the
507 southern mid-latitudes at 153.0 ppm.

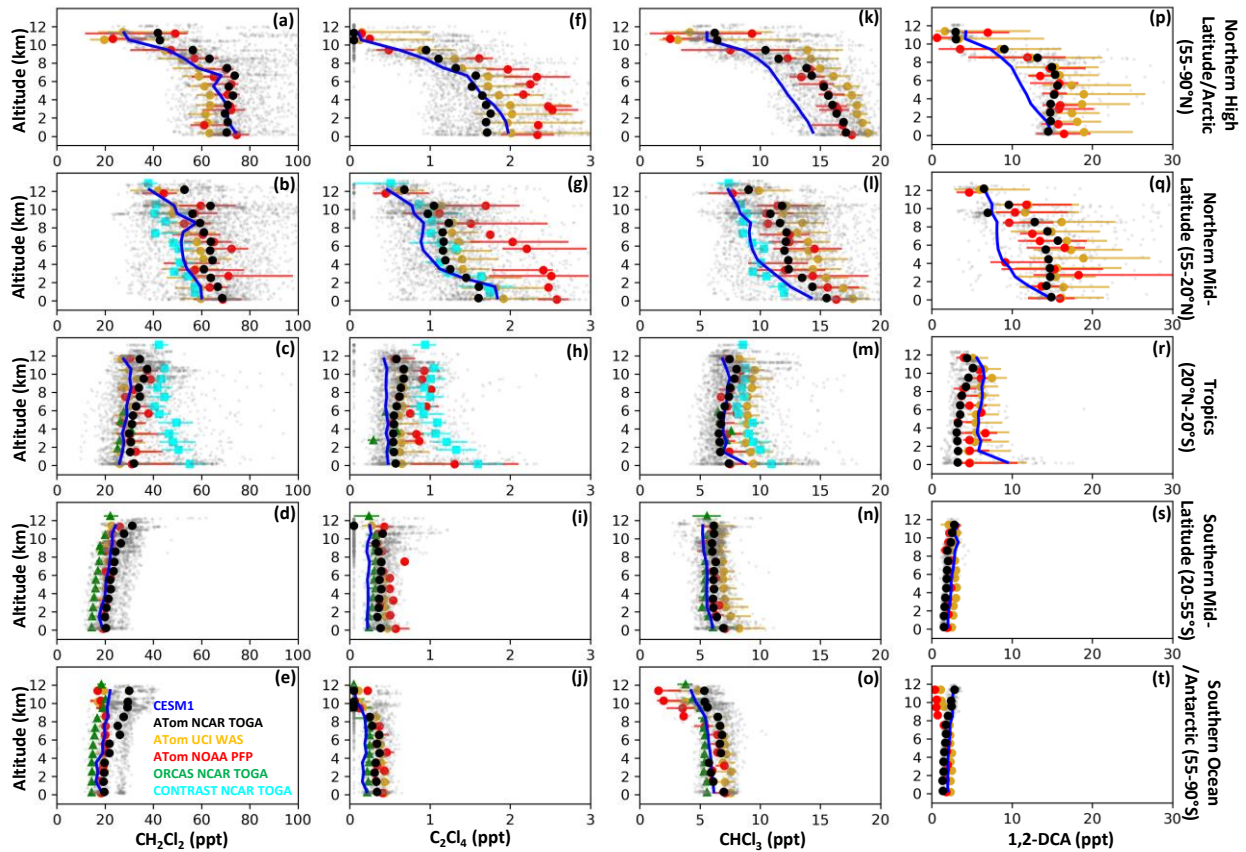
508 One hypothesis for the instrumental differences is that the measurement biases could be due to
509 losses on either the unheated inlet walls or the canister/flask walls for the WAS and PFP
510 measurements, which can be more pronounced at lower ambient humidities.

511 TOGA uses a heated Sulfinert™ inlet line and does not store the samples prior to analysis. The
512 flask samples (WAS and PFP) are collected by drawing air in through non-heated lines; WAS
513 uses a non-heated stainless steel (SS) line and the PFP uses a coated SS line. For most materials,
514 a threshold level of humidity is needed to keep surfaces properly passivated. Exceptions are in
515 the case of Teflon™ or Silonite™ (Deming et al., 2019) for which there is little absorptive effect
516 even at 0% humidity. Glass and SS also showed to be relatively non-adsorptive although not as
517 good as Teflon™ or Silonite™. In the Deming et al. (2019) study, Sulfinert™ tubing was not
518 tested but Silonite™, which is believed to be a very similar product, was tested. The UCI WAS
519 uses SS canisters that are humidified, thus potentially mitigating absorptive losses on the canister
520 surfaces. The PFP glass flasks have been tested for long-term storage of VSLS in surface-air
521 samples collected under a wide variety of humidity levels and storage times and no issues have
522 been found for CH₂Cl₂ measurements. An alternative hypothesis to explain measurement
523 discrepancies is that TOGA had a sampling artifact at these low humidities and that the
524 canister/flask measurements are more accurate. The rationale for this hypothesis is that
525 subtropical jets in the UTLS region inhibits the horizontal transport of tracers, leading to small
526 values measured by UCI WAS and PFP (Jesswein et al., 2022). One problem with this
527 hypothesis is that there would have to be a mechanism for artifact formation in the TOGA
528 analytical system that would result in higher CH₂Cl₂ and 1,2-DCA mixing ratios in this region
529 than were actually present and to date we have not seen evidence of artifact formation for this
530 species. Additionally, better correlation between the model and TOGA results increases the
531 likelihood of our first hypothesis.

532 **3.2. Cl-VSLS abundance in different field studies**

533 Cl-VSLS were also measured by TOGA during ORCAS and CONTRAST. These two missions
534 as well as the ATom-2 deployment occurred primarily during the month of February in different
535 years previous to ATom. The vertical profiles of CH₂Cl₂, CHCl₃, and C₂Cl₄ from ORCAS and

536 CONTRAST are also shown in

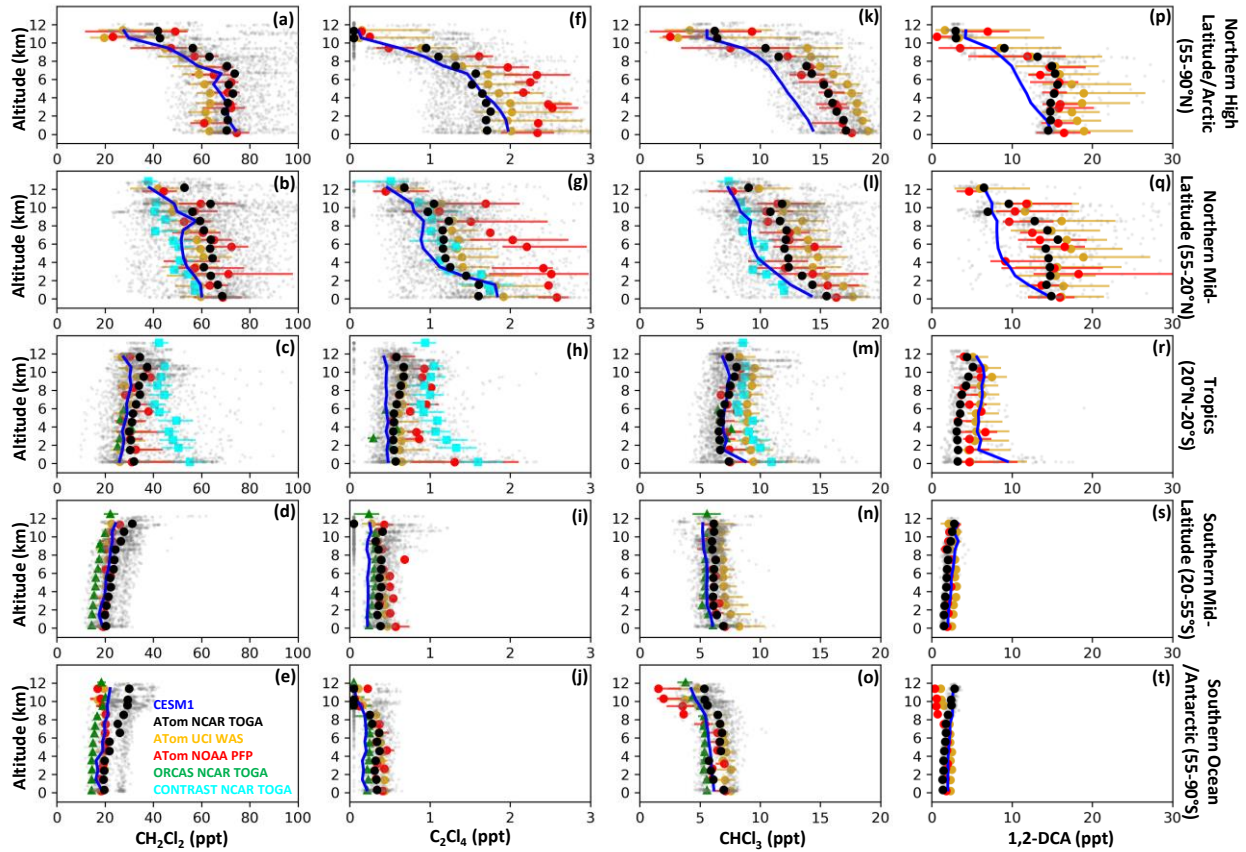


537

538 Figure 2 (and Figures S2-S4). Both ORCAS and ATom sampled air from similar regions in the
 539 SH (Figure 1) but different years and the similar shape of the vertical profiles (in Figures S2-S4)
 540 shows small interannual variability of the Cl-VSLS mixing ratios. On the other hand, the
 541 CONTRAST mission sampled air masses in the western tropical Pacific region closer to Asia,
 542 which led to higher mixing ratios compared with the ATom data (Figures S2-S4). In addition, the
 543 vertical profiles for the CONTRAST Cl-VSLS had sharper gradients in the MBL compared with
 544 ATom data, suggesting that low-level outflow from the Asian continent was sampled. Also, the
 545 fast convection over the Pacific increases the Cl-VSLS levels at higher altitudes.

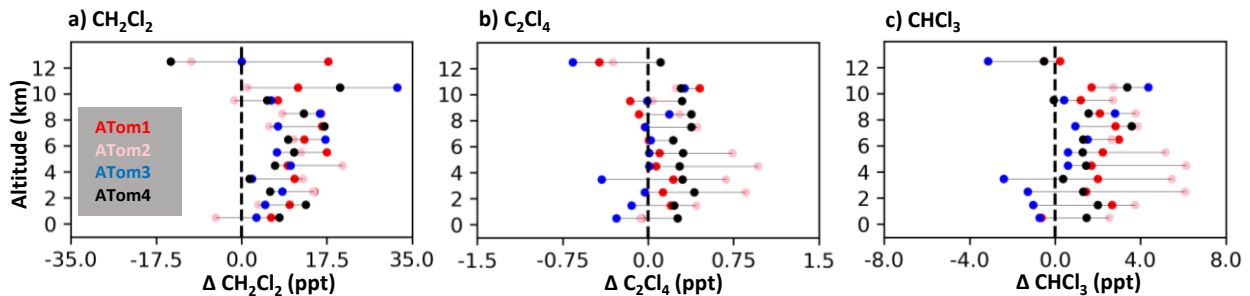
546 Figure 4 shows correlations of observed C_2Cl_4 vs. CH_2Cl_2 and $CHCl_3$ vs. CH_2Cl_2 for ORCAS,
 547 CONTRAST, and ATom-2. It is worth mentioning that only ATom-2 is considered in this
 548 analysis to match the seasons. C_2Cl_4 vs. CH_2Cl_2 for the ORCAS and CONTRAST regional
 549 missions are overlaid on ATom-2 global data and show similar positive correlations. In
 550 particular, the C_2Cl_4/CH_2Cl_2 ratio was 0.02 and 0.03 for the limited latitudinal-range ORCAS

551 (Southern Ocean region) and CONTRAST (tropics) data, respectively. The C_2Cl_4/CH_2Cl_2 ratio
552 for the global latitudinal-range ATom-2 data was 0.01-0.04, decreasing for SH latitude bins,
553 suggesting less transported C_2Cl_4 in the SH or faster loss in the SH. We will analyze
554 interhemispheric transport in more detail in Section 3.4. Figure 4 also shows similar tight
555 positive correlations between $CHCl_3$ and CH_2Cl_2 during CONTRAST and ATom-2 data, while
556 $CHCl_3$ does not correlate with CH_2Cl_2 in some of the ORCAS data. Within the Southern Ocean
557 region, this was also seen in the ATom-2 data (Figure S6). As shown in Figure 4d, there are two
558 clusters of data in the Southern Ocean/Antarctica latitude bins ($55-90^\circ S$). One cluster contains
559 low CH_2Cl_2 and relatively high $CHCl_3$ mixing ratios that are measured near the surface. We
560 found positive correlation between near the surface $CHCl_3$ and two major oceanic tracers (CH_3I
561 and $CHBr_3$) during ORCAS, confirming the role of natural oceanic emissions of $CHCl_3$ in the
562 Antarctic region (not shown). In contrast, the other cluster shows a positive correlation between
563 $CHCl_3$ and CH_2Cl_2 at high altitudes. Since the lifetime of CH_2Cl_2 and $CHCl_3$ are of the same
564 magnitude (Hossaini et al., 2019), the positive correlation suggests this cluster is being
565 transported from other regions; i.e., from the NH or tropics. The natural sources of $CHCl_3$
566 include both oceanic (macro and microalgae) and terrestrial (soil) sources, with more natural
567 emissions in the NH than the SH (Laternus et al., 2002). Nevertheless, the strong correlation
568 between $CHCl_3$ and CH_2Cl_2 , in particular in the tropics and NH (Figure S6), also suggests that
569 maybe anthropogenic sources dominate $CHCl_3$ emissions sources. Fang et al. (2019) and An et
570 al. (2023) have shown that the changes in anthropogenic emissions can explain the recent
571 changes in global concentrations.



572

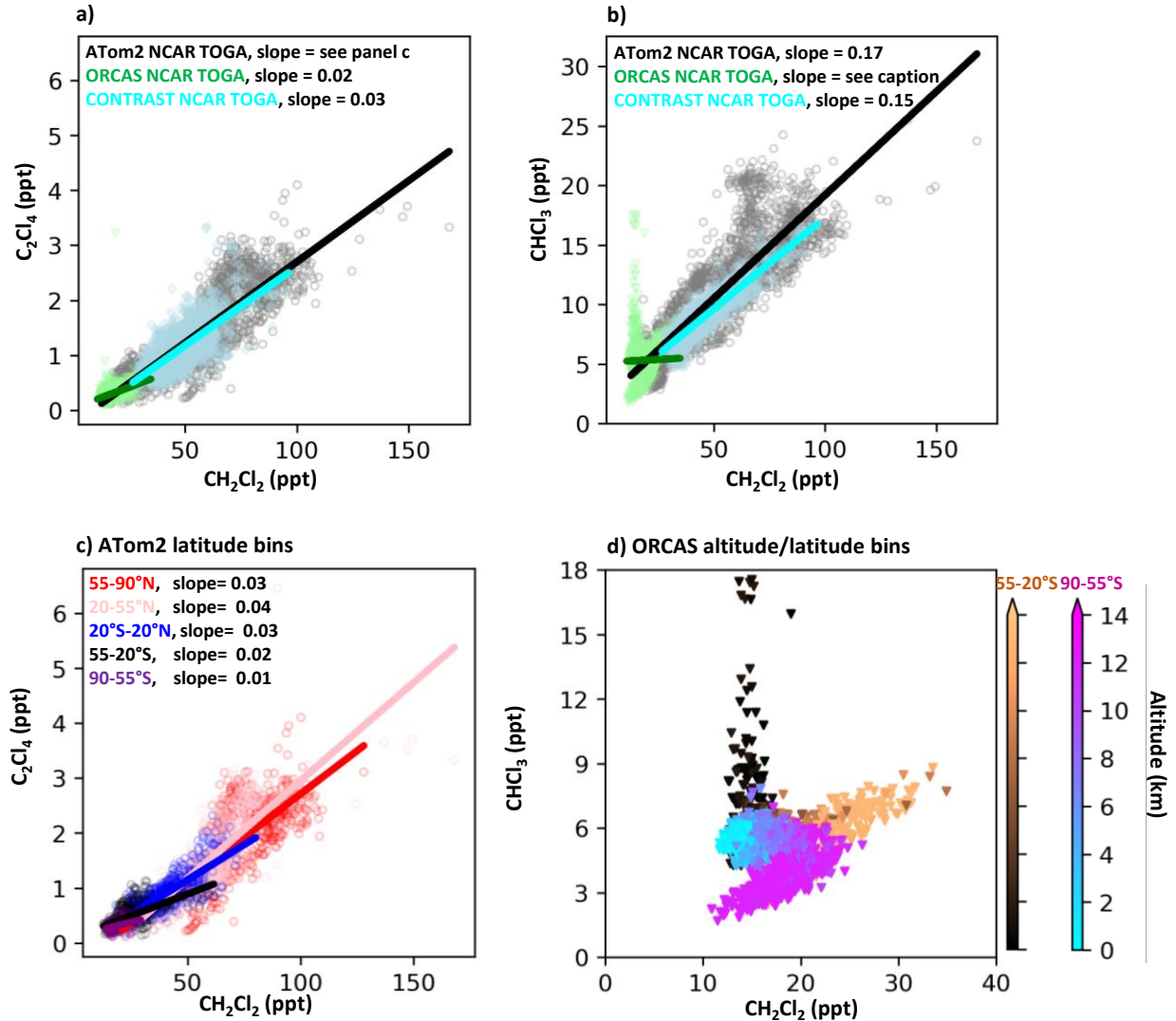
573 **Figure 2.** Binned vertical profiles of median measured CH_2Cl_2 , C_2Cl_4 , CHCl_3 , and 1,2-DCA mixing ratios for all
 574 ATom deployments from TOGA (black), WAS (gold), and PFP (red), separated into five different latitude bins; 55–
 575 90°N (a, f, k, p), 20–55°N (b, g, l, q), 20°S–20°N (c, h, m, r), 55–20°S (d, i, n, s), and 90–55°S (e, j, o, t),
 576 accompanied by the vertically-binned median CESM1 model results sampled along the ATom flight tracks (blue
 577 curve). Note that C_2Cl_4 data for PFP (red points) are only available for ATom-4. Additionally, 1,2-DCA data for
 578 PFP are for ATom-2 to ATom-4 and for TOGA are only for ATom-4 – see Figures S2-S5 for individual
 579 deployments. Vertically-binned median TOGA measurements made during ORCAS (green) and CONTRAST
 580 (cyan) are also shown. Individual ATom TOGA data points are included (light grey points) although the x-axes are
 581 limited for better clarity in the SH plots. Error bars for PFP, WAS, CONTRAST, and ORCAS data are the 25th to
 582 75th percentiles.
 583



584

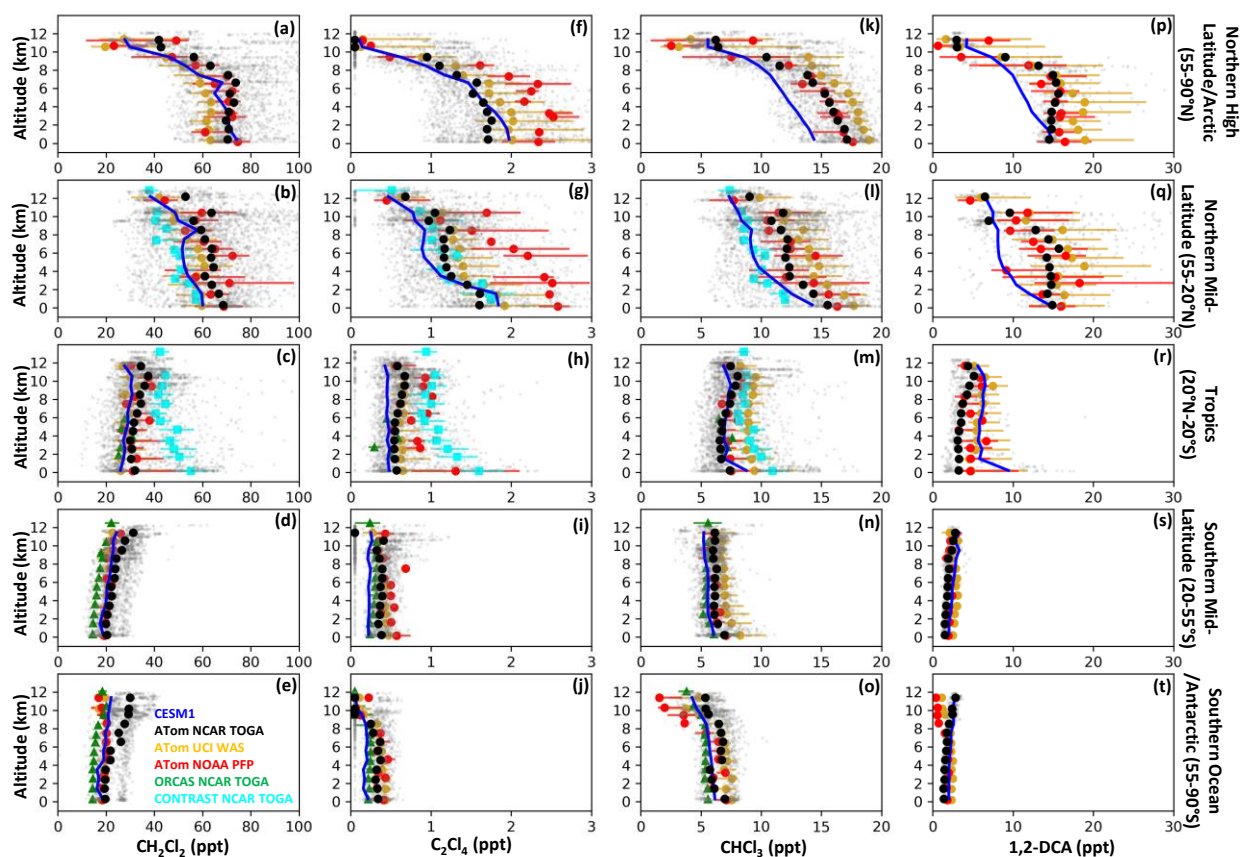
585 **Figure 3.** Vertical profiles of the median Pacific minus median Atlantic mixing ratios for (a) CH_2Cl_2 , (b) C_2Cl_4 , and
 586 (c) CHCl_3 from the TOGA observations at northern mid-latitudes (20–55°N). Marker colors show the differences
 587 during ATom-1 (red), ATom-2 (pink), ATom-3 (blue), and ATom-4 (black); negative values indicate that median

588 mixing ratios over the Atlantic were larger than the Pacific. Note the horizontal grey lines only connect the points
 589 for better visualization and do not show the range of the observations.



590

591 **Figure 4.** Correlation plots of Cl-VSLs measured by TOGA for (a) C_2Cl_4 vs. CH_2Cl_2 and (b) $CHCl_3$ vs. CH_2Cl_2
 592 from ATom-2, ORCAS, and CONTRAST, (c) C_2Cl_4 vs. CH_2Cl_2 from ATom-2, colored by latitude bins, and (d)
 593 $CHCl_3$ vs. CH_2Cl_2 from ORCAS colored by latitude bin and shaded by altitude. Lines show least square linear
 594 regressions. Note that although a line is shown for $CHCl_3$ vs. CH_2Cl_2 from ORCAS in (b), regional data in (d)
 595 indicate a non-linear relationship. Note that the axes are different in each panel. Note also that only ATom-2 data are
 596 considered in this analysis to match the seasons with the CONTRAST and ORCAS projects.

597 **3.3. Evaluation of CESM1 using ATom measurements**

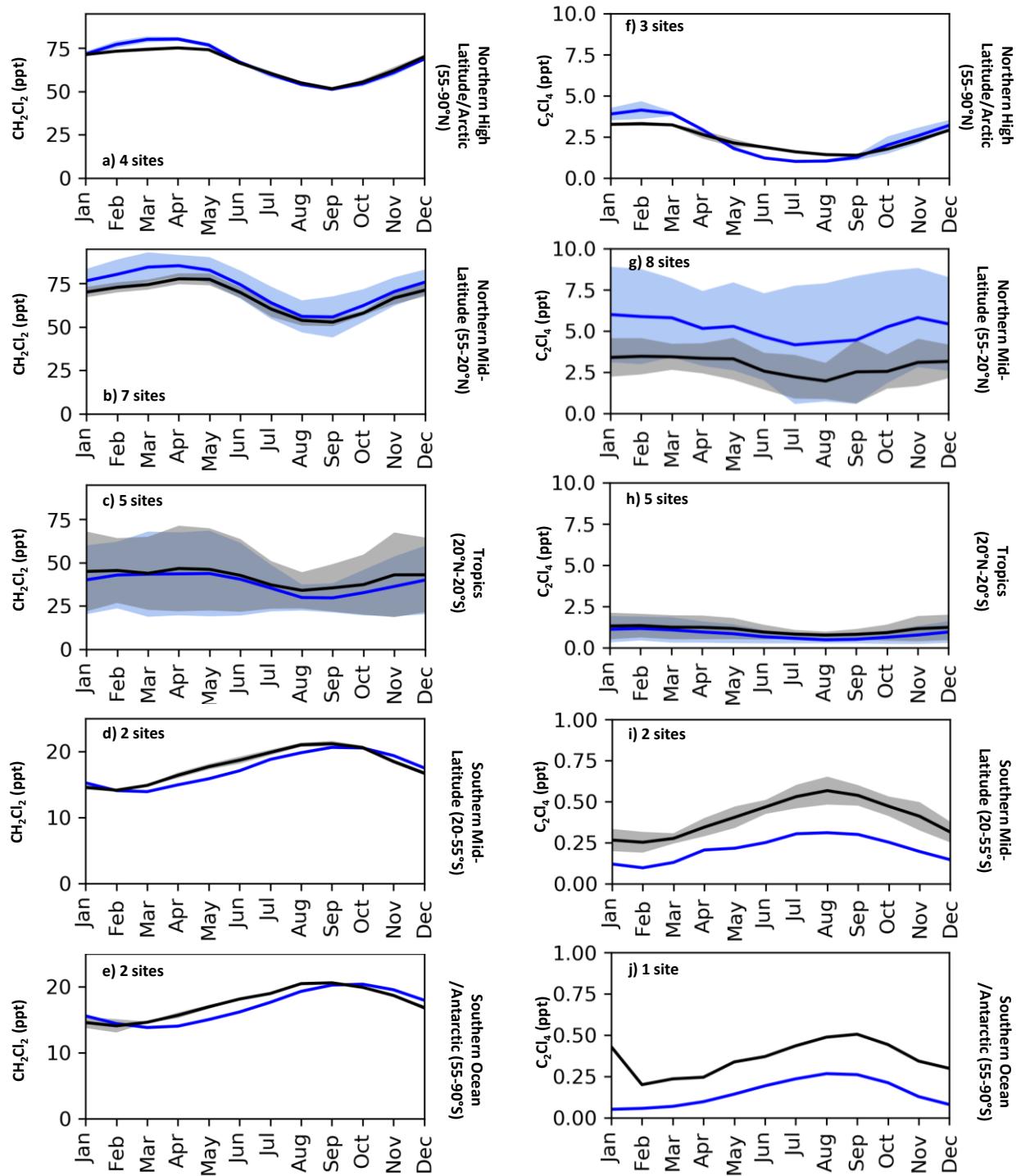
598

599 Figure 2 includes the vertical profiles of median modeled Cl-VSLS species in CESM1 sampled
 600 along ATom flight tracks and corresponding to TOGA measurement times. Similar to the
 601 measurements, the model showed substantially higher Cl-VSLS mixing ratios in the NH. For
 602 CH_2Cl_2 , the model captured the observed vertical gradients; i.e., mixing ratios decreasing with
 603 increasing altitude in the NH and a slight vertical inversion (higher levels at higher altitudes) in
 604 the tropics and SH. The model captured the magnitude of the median surface layer (i.e., < 1 km)
 605 CH_2Cl_2 mixing ratios in the high-latitude Arctic region, but was approximately 10 ppt lower than
 606 the measurements in the northern mid-latitudes. In addition, the modeled CH_2Cl_2 mixing ratios
 607 were lower than TOGA measurements in the tropics and SH throughout the troposphere. Overall,
 608 we report the NMB, RMSE, and r of -15%, 16 ppt, and 0.80, respectively, for modeled CH_2Cl_2
 609 against TOGA measurements in all ATom deployments (Table S4). For C_2Cl_4 , the model
 610 captured the surface mixing ratios over the extratropical regions in the NH. However, the model
 611 was biased low at higher altitudes in the northern mid-latitudes. Additionally, the model was

612 biased low in the tropics and SH. The NMB, RMSE, and r for C_2Cl_4 are -7%, 0.5 ppt, and 0.82,
613 respectively (Table S5). Our model used constant LBC values, rather than emissions, for $CHCl_3$
614 and 1,2-DCA. For $CHCl_3$, the model underestimated the concentrations with NMB, RMSE, and r
615 of -17%, 3 ppt, and 0.85, respectively. Similarly for 1,2-DCA, the model underestimated its
616 global abundance with NMB, RMSE, and r of -23%, 6 ppt, and 0.74, respectively. (Note that
617 TOGA measured 1,2-DCA only during ATom-4). The model captured the Cl-VSLS mixing
618 ratios in the tropics and SH, with UCI WAS generally showing higher values than TOGA and
619 PFP. On the other hand, the model bias for $CHCl_3$ and 1,2-DCA was larger than CH_2Cl_2 and
620 C_2Cl_4 over the NH high latitudes. Statistics for individual ATom deployments are also listed in
621 Tables S4–S7. In the following sections, we look in more detail on model performance and
622 transport of CH_2Cl_2 and C_2Cl_4 , which have anthropogenic sources with well-constrained
623 magnitudes (Claxton et al., 2020).

624 Figure 5 shows the measured and corresponding modeled profiles for CH_2Cl_2 and C_2Cl_4 mixing
625 ratios for each ATom deployment, binned zonally (Figure S7 shows similar plots for $CHCl_3$ and
626 1,2-DCA, as well as the OH observed by the Pennsylvania State University Airborne
627 Tropospheric Hydrogen Oxides Sensor, ATHOS). Both measurements and model output show
628 strong seasonal differences for Cl-VSLS in the NH. For CH_2Cl_2 , both the highest measured and
629 modeled mixing ratios were over the NH high latitudes. However, the highest modeled values
630 were for ATom-4 (May 2018) but the highest measured values were during ATom-2 (February
631 2017). In the NH, CH_2Cl_2 up to 100 ppt was measured during ATom-3 (October 2017), yet the
632 model values only reached 60 ppt. Since the model is biased high for OH concentration (Figure
633 S7), and Cl-VSLS lifetimes are a few months, the large bias of the model during ATom-3 could
634 be due to the integrated impact of high-biased OH during the highest OH exposure period in the
635 NH. The lowest modeled CH_2Cl_2 mixing ratios in the NH were for the ATom-1 (August 2016)
636 period, consistent with the measurements. In addition, the model predicted a homogenous NH
637 vertical gradient during ATom-1, which is also consistent with the measurements. For C_2Cl_4 ,
638 both model and measurements agreed with the highest NH mixing ratios during ATom-2
639 (February 2017), although the model overestimates the mixing ratios (NMB of 24% in high-
640 latitude Arctic region). Similar to CH_2Cl_2 , the lowest NH C_2Cl_4 mixing ratios were during
641 ATom-1, both measured and modeled. The high mixing ratios during ATom-2 (February 2017)
642 and low mixing ratios during ATom-1 (August 2016) reveal seasonal differences for both

643 CH₂Cl₂ and C₂Cl₄, which the model clearly captured. This seasonal variation can be explained
644 primarily by OH seasonal variations, as the emissions in the model were constant throughout the
645 year (Figure S7). In particular, lower [OH] during NH winter leads to greater Cl-VSLS
646 abundance, with the opposite true in NH summer. In addition, atmospheric dynamics play a role
647 in the observed mixing ratios. For example, the enhanced high-altitude Cl-VSLS mixing ratios
648 measured in the NH mid-latitudes during ATom-1 are potentially due to the Asian Summer
649 Monsoon deep convection. Wang et al. (2020) found that the CAM-chem model captured the
650 large-scale processes during ATom-1 and ATom-2 including the isentropic mixing from low
651 latitudes to high latitudes. Similar to Wang et al. (2020), the wind fields and temperature in the
652 model in this work are nudged towards the MERRA-2 reanalysis data to better capture the large-
653 scale processes.

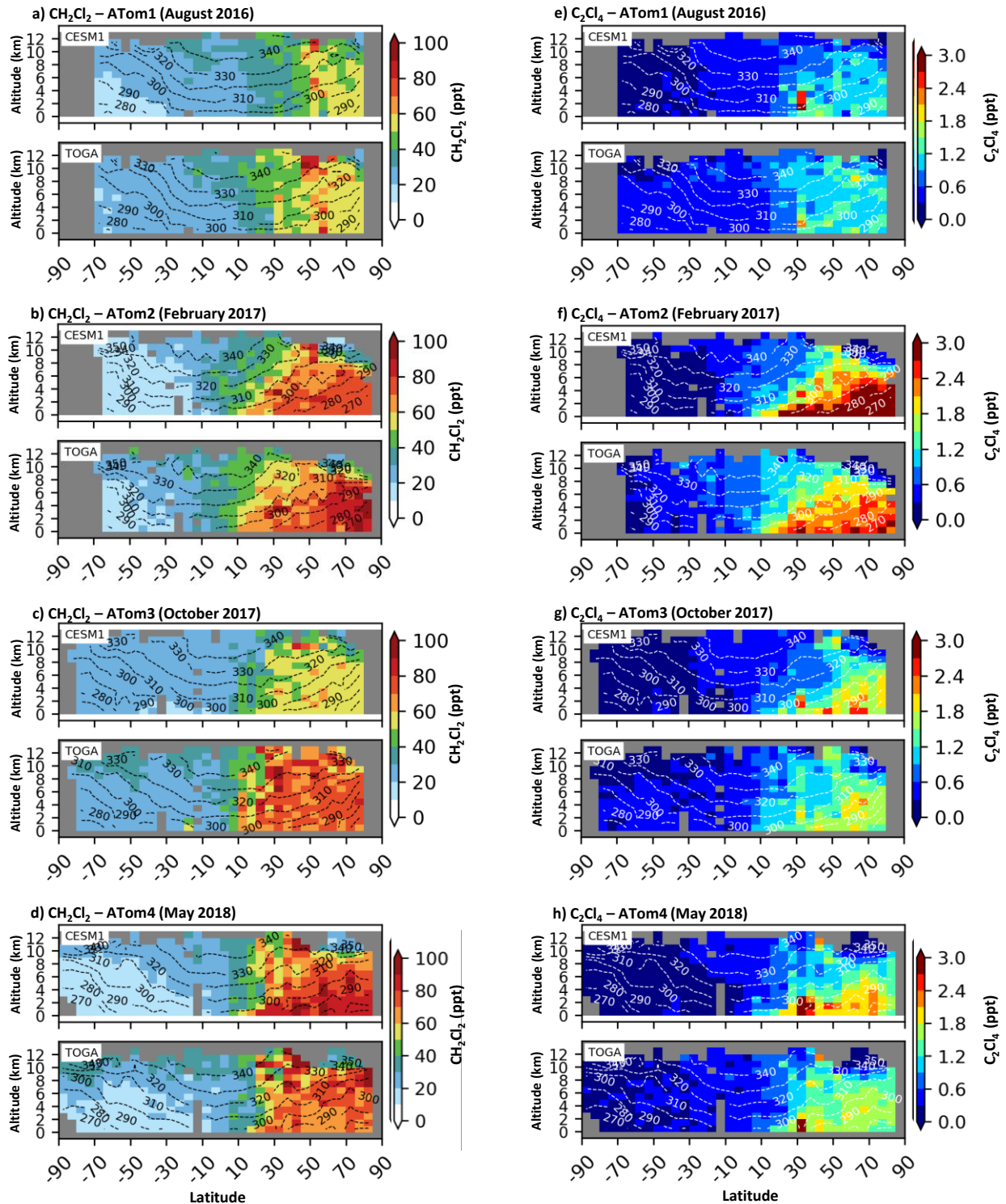


654

655 Figure 6 shows the monthly means of surface CH_2Cl_2 and C_2Cl_4 mixing ratios modeled and
 656 observed by ground measurement stations between 2016 and 2018 and using the corresponding
 657 model grid cells, divided into latitude bins. For CH_2Cl_2 , the model overestimated the maximum
 658 in the observations in NH mid-latitudes during winter and spring and captured the minimum

659 values during summer, and measurements were in the range of modeled data. Over other regions,
660 the model mostly underestimated the CH_2Cl_2 mixing ratios. The majority of stations were located
661 in the tropics (five stations) and northern mid-latitudes (seven stations), where the spread of
662 modeled and observed values overlapped with each other. The NMB, RMSE, and r for monthly
663 mean CH_2Cl_2 data against all ground measurement data are 2%, 8 ppt, and 0.95, respectively. It
664 should be noted that over the SH regions, the model minimum and maximum values are delayed
665 (by about 1 month) for CH_2Cl_2 compared to the measurements, while the model more accurately
666 predicts the timing of the maximum and minimum values of C_2Cl_4 . This delay difference can be
667 explained by different source regions for CH_2Cl_2 and C_2Cl_4 . In particular, Asian emission
668 dominate SH abundance of CH_2Cl_2 , while C_2Cl_4 abundance are attributed more evenly to
669 regions, as will be discussed in Section 3.4. For C_2Cl_4 , the model captured the maximum values
670 in the high-latitude Arctic region. In the northern mid-latitudes, a less pronounced seasonal
671 variability was observed, which was captured by the model, despite being always biased high
672 and with a larger standard deviation. Similar to CH_2Cl_2 , the model is biased low for C_2Cl_4 in the
673 tropics and Southern Hemisphere. The NMB, RMSE, and r of 45%, 2 ppt, and 0.79, respectively,
674 show the overestimation of C_2Cl_4 . This is in contrast with the model being biased low, in
675 general, against ATom data as mentioned above (NMB of -7%). The overestimation during all
676 the months over the NH mid-latitudes and strong underestimation in the SH (i.e., emission
677 sources) suggests either C_2Cl_4 emission magnitudes or global distributions require further
678 improvements.

679



680

681 **Figure 5.** Spatially-binned distribution of mean modeled (top subplot) and measured (bottom subplot) for (a-d)
 682 CH_2Cl_2 and (e-h) C_2Cl_4 mixing ratios during individual ATom deployments. Latitude and altitude bin intervals are 5
 683 degrees and 1 km, respectively. Corresponding TOGA sample locations for each ATom mission are shown in Figure
 684 S7.

685 **Table 2.** Summary of TOGA-observed median mole fractions (and 10th-90th percentiles) for four Cl-VSLs and total
 686 Cl^{VSLs} during ATom within different latitude and altitude bins.

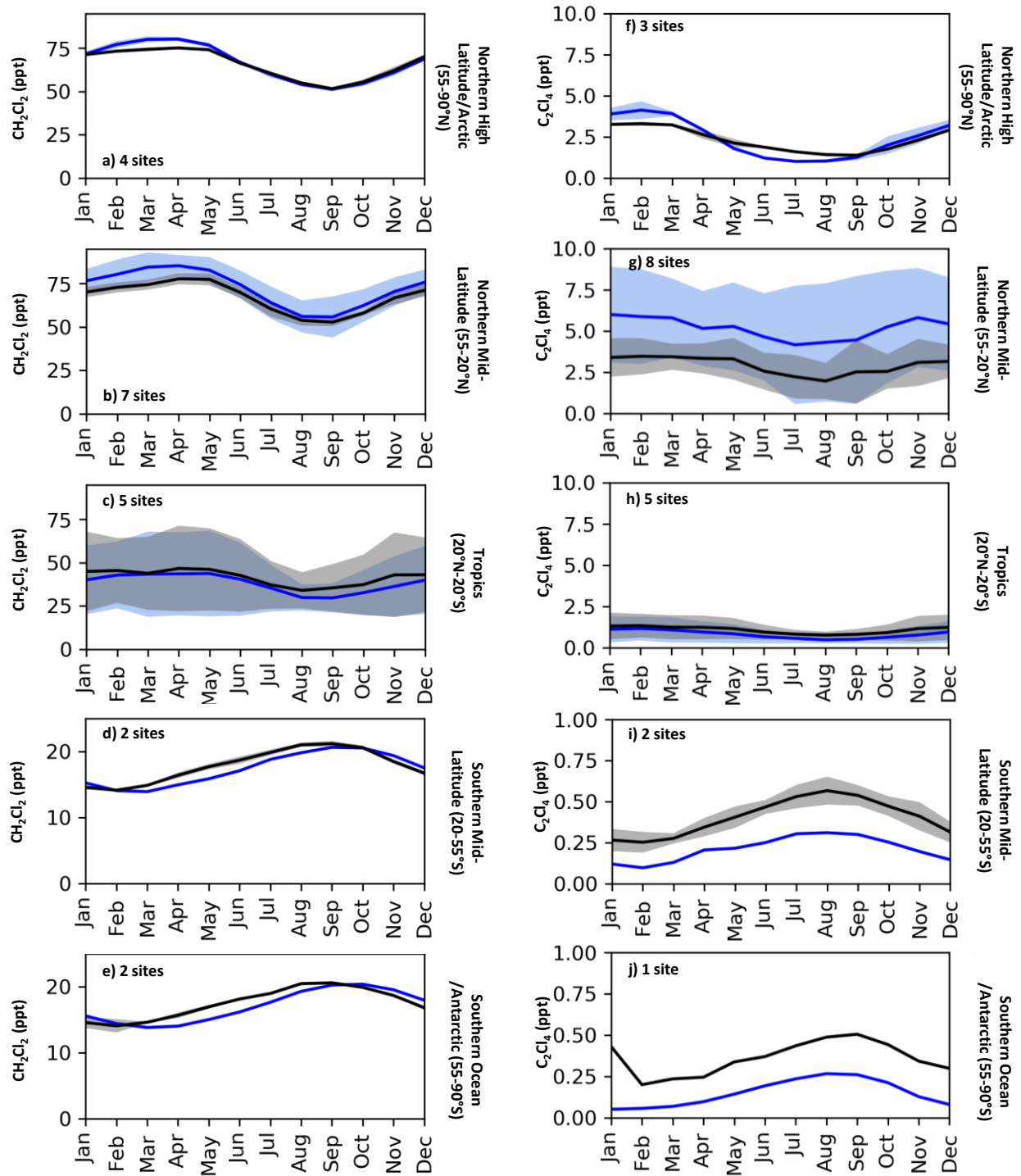
Latitude Region	Potential Temperature θ (K)	CH ₂ Cl ₂ ^a , ppt	C ₂ Cl ₄ ^a , ppt	CHCl ₃ ^a , ppt	1,2-DCA ^b , ppt	Σ Cl ^{VSLs} ^c , ppt
Northern High Latitudes/ Arctic	< 320	70.2 (54.1 - 91.6)	1.5 (1.0 - 2.4)	15.3 (11.4 - 18.7)	14.8 (12.6 - 18.2)	206.4 (150.0 - 251.2)
	320-340	59.4 (41.3 - 94.7)	1.1 (LLOD - 1.60)	11.4 (6.6 - 15.3)	11.1 (4.0 - 15.9)	166.1 (105.0 - 244.3)
	340-360	33.8 (25.9 - 46.3)	LLOD	5.2 (4.2 - 6.6)	2.9 (2.4 - 4.1)	86.7 (66.8 - 119)
	360-380	41.6 (33.1 - 57.5)	LLOD	5.8 (4.8 - 8.1)	3.1 (2.5 - 3.4)	106.6 (85.3 - 139.3)
Northern Mid-latitudes	< 320	66.2 (45.4 - 81.7)	1.4 (0.8 - 2.3)	14. (9.2 - 17.9)	14.7 (9.8 - 20.5)	189.9 (129.2 - 234.1)
	320-340	59.2 (39.7 - 96)	1.1 (0.6 - 1.6)	11.5 (7.8 - 15.2)	11.7 (5.7 - 27.4)	162.9 (107.2 - 258.7)
	340-360	41.6 (34.2 - 67.3)	0.7 (LLOD - 1.1)	8.3 (6.7 - 10.6)	6.5 (3.5 - 16.5)	110.8 (93.6 - 171.5)
	360-380	N.A. ^d	N.A.	N.A.	N.A.	N.A.
Tropics	< 320	31.1 (21.1 - 59.5)	0.6 (0.3 - 1.4)	7.1 (5.5 - 12.3)	3.3 (1.8 - 11.3)	86.4 (63.7 - 166.3)
	320-340	33.2 (23.4 - 48)	0.6 (0.4 - 1.0)	7.2 (5.8 - 9.8)	3.6 (2.5 - 6.6)	91.7 (69.5 - 130.2)
	340-360	35.7 (27.3 - 47.5)	0.6 (0.4 - 0.9)	7.7 (6.3 - 10.0)	4.8 (2.9 - 5.7)	99.6 (76.7 - 130.5)
	360-380	N.A.	N.A.	N.A.	N.A.	N.A.
Southern Mid-latitudes	< 320	21.9 (16.1 - 28.7)	0.4 (LLOD - 0.51)	6.3 (5.1 - 7.6)	1.7 (1.3 - 2.1)	65.2 (51.9 - 80.4)
	320-340	27.4 (20.8 - 33.6)	0.4 (LLOD - 0.6)	6.0 (4.6 - 6.9)	2.5 (2.0 - 3.0)	75.9 (58.8 - 88.8)
	340-360	30.5 (22.7 - 39.7)	0.5 (LLOD - 0.8)	6.1 (4.3 - 7.6)	3.2 (2.9 - 3.4)	81.2 (61.2 - 102.2)
	360-380	25.6 (22.4 - 27.9)	LLOD	3.6 (3.2 - 4.7)	N.A.	61.5 (55.5 - 70.2)
Southern Ocean/ Antarctic	< 320	24.8 (16.7 - 30.8)	0.3 (LLOD - 0.5)	6.5 (5.1 - 7.7)	1.6 (1.3 - 2.3)	71.2 (51.4 - 83.9)
	320-340	29.1 (22.5 - 35.5)	LLOD (LLOD - 0.4)	5.3 (3.8 - 5.9)	2.4 (2.0 - 2.9)	76.7 (58.8 - 91.1)
	340-360	34.2 (23.4 - 39.5)	LLOD	4.9 (3.5 - 6.0)	2.8 (2.3 - 3.2)	87.6 (57.5 - 102.5)
	360-380	29.3 (24.3 - 39.9)	LLOD	4.6 (3.3 - 5.9)	3.0 (2.3 - 3.4)	73.4 (60.2 - 104.9)

687 ^a The reported values for CH₂Cl₂, C₂Cl₄, and CHCl₃ are based on TOGA data for all ATom deployments.

688 ^b 1,2-DCA is only available from TOGA for ATom-4.

689 ^c Total Cl atoms from the sum of available chlorine in CH₂Cl₂, CHCl₃, C₂Cl₄, and 1,2-DCA.

690 ^d N.A. = data not available.



691
 692 **Figure 6.** 2016–2018 monthly means of ground-based measurements (black) and corresponding modeled (blue) \pm 1
 693 standard deviation (shading) of CH_2Cl_2 and C_2Cl_4 mixing ratios over five different latitude bins: 55–90°N, 20–55°N,
 694 20°S–20°N, 55–20°S, and 90–55°S. Ground-based measurements are from the stations within NOAA and AGAGE
 695 networks shown in Figure 1. The shading represents the \pm 1 standard deviation of the monthly means from multiple
 696 sites within each latitude range and not the spread over different years for individual sites. Note that for many
 697 months the standard deviations (both model and measurements) are very small compared with the y-axes scales.
 698 Also, note that smaller y-axes limits are used for SH panels.

699 **3.4. Impact of transport and convection from regionally tagged emissions of Cl-VSLS**

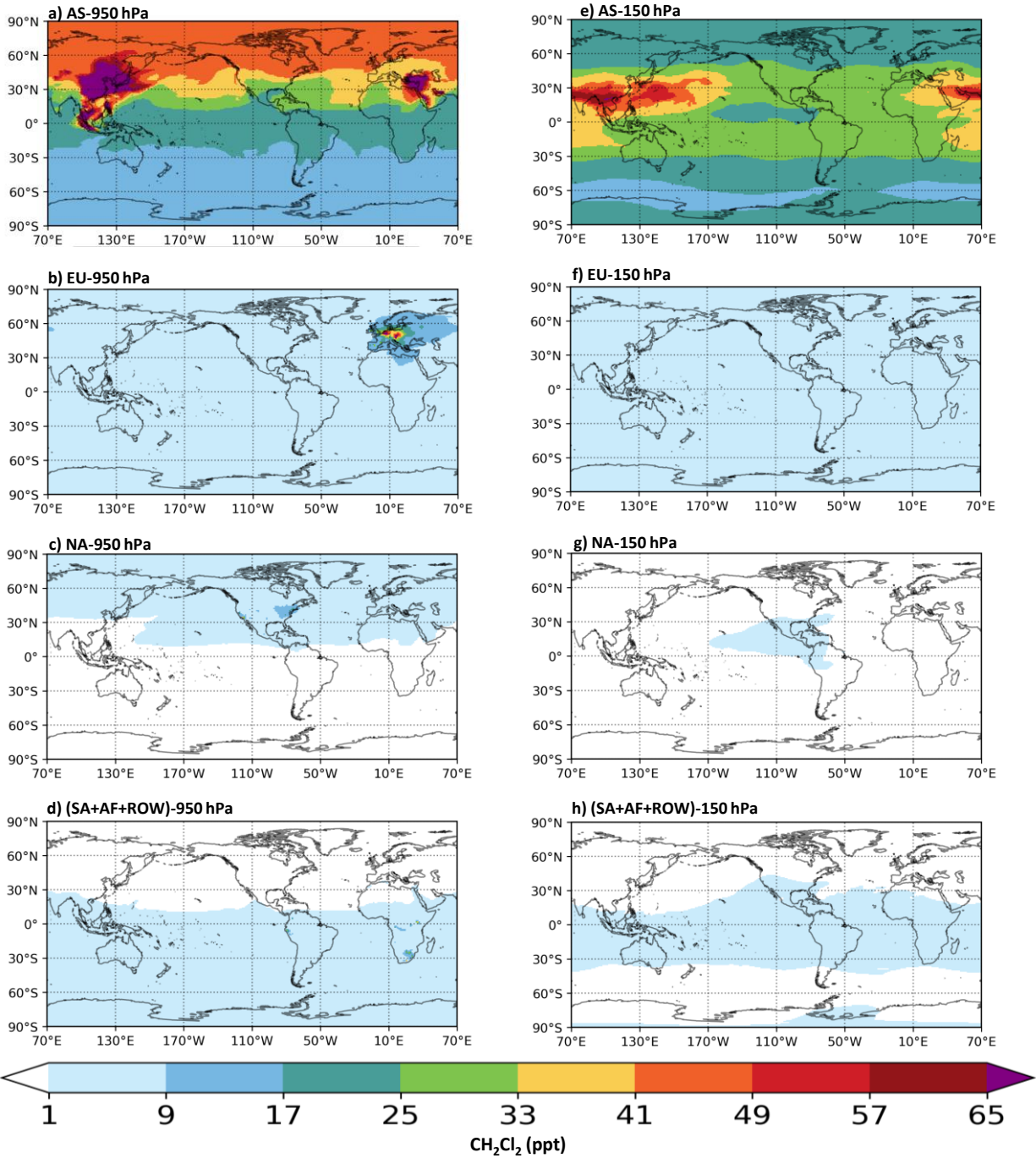
700 As shown in the previous sections, the model was mostly able to capture the distribution and
701 seasonal variability of four Cl-VSLS. We also observed the presence of these Cl-VSLS in the SH
702 remote regions and high altitudes despite the fact that the large majority of their emissions are in
703 the NH. In this section, we use the regional tagged tracers (described in Section 2.2.2) to estimate
704 the contribution of emissions from each region on Cl-VSLS abundance at various altitudes and
705 locations.

706 Figure 7 shows the spatial distribution of each regional tagged CH_2Cl_2 tracer at 950 hPa and 150
707 hPa during August 2017. The deep convection during August 2017 due to the Asian summer
708 monsoon (ASM) has been previously shown (Honomichl and Pan, 2020). Near the surface (i.e.,
709 950 hPa), the highest mixing ratios are near the source regions for the Asian (AS) and European
710 (EU) tagged tracers. The AS emissions are also the major contributor to the CH_2Cl_2 mixing ratios
711 over North America, South America, and Africa. Similarly, the AS emissions are the dominant
712 source of CH_2Cl_2 over the remote regions from the Arctic to the Antarctic (Figure 7a). In
713 addition, the NH mixing ratios over the high latitude regions that are contributed by the AS
714 region are larger than the mid-latitude regions. At high altitudes (i.e., 150 hPa), the AS emissions
715 are the dominant contributor around the globe (Figure 7e). In particular, AS mixing ratios above
716 50 ppt were modeled over their source regions in contrast with < 9 ppt mixing ratios from EU.
717 Furthermore, the model shows larger values at high altitudes than at the surface in the tropics and
718 subtropical regions in SH (Figure 7e). Figure 8, on the other hand, shows that AS, EU, and NA
719 emissions are the dominant contributors of C_2Cl_4 in their source regions and their impact on
720 other regions in the NH, near the surface, are within the same order. However, their contribution
721 in the SH is minimal compared to the cumulative impact of sources in the SH (Figure 8d). At
722 high altitudes, Asia is the dominant contributor to C_2Cl_4 , and its impact is primarily limited to the
723 tropical regions. Adcock et al. (2021) and Lauther et al. (2022) showed that the ASM anticyclone
724 transports large amounts of surface Cl-VSLS emissions to the UTLS region, where they can be
725 transported horizontally by jet streams to other regions.

726 Figure 9 and 10 show the annual zonal means of the total and tagged emissions of CH_2Cl_2 and
727 C_2Cl_4 , respectively, between 2016 and 2018. For CH_2Cl_2 , we estimate the annual mean
728 tropospheric mixing ratios of 36.9 ppt (i.e., 73.8 ppt Cl), with the Asian emissions being
729 responsible for 85% of the total. Within Asia, we further calculated that China's annual mean

730 tropospheric mixing ratio of CH_2Cl_2 is 17.1 ppt, which accounts for 46% of global tropospheric
731 mean abundance. On continental scale, EU (8%) and NA (4%) emissions ranked second and
732 third as sources of tropospheric CH_2Cl_2 , respectively. Table 3 shows the mean tropospheric
733 values in individual latitude bins, highlighting that AS emissions dominate in all regions.
734 Similarly, the vertical profiles of tagged emissions along the ATom flight tracks confirm the AS
735 dominance (Figure S8). For C_2Cl_4 , in contrast, emissions from EU, NA, and AS are all important
736 contributors to the global annual mean tropospheric mixing ratio of 0.73 ppt (i.e., 2.92 ppt Cl).
737 The modeling results show that total chlorine from CH_2Cl_2 is ≈ 25 times more than from C_2Cl_4 in
738 the troposphere. The similar source gas ratio for TOGA measurements along the ATom flight
739 tracks is ≈ 28 (using 46.6 and 0.84 ppt global means of CH_2Cl_2 and C_2Cl_4 , respectively, in
740 Section 3.1.5).

741 EU, NA, and AS contribute 32% (0.23 ppt), 27% (0.20 ppt), and 33% (0.24 ppt) to the global
742 mean C_2Cl_4 , respectively. As Figure 10 shows, the C_2Cl_4 mixing ratios are highest over their
743 individual source regions with very low values in other regions, due to its shorter atmospheric
744 lifetime, which we will discuss in Section 3.5. While AF and SA C_2Cl_4 emissions do not strongly
745 impact global concentrations, the tagged zonal plots shows clear sensitivity of the C_2Cl_4
746 distribution to regional sources. In particular, Figure 10 suggests that an increase in future AF
747 and SA emissions could have a more significant impact than other continental sources on the SH
748 region, although an increase in C_2Cl_4 emissions is not anticipated based on current emission and
749 concentration trends (Claxton et al., 2020; Laube et al., 2022). Figures 9 and 10 also show
750 contributions from SA and AF over the tropics, and highlights the significance of the emission
751 regions. This is important as these regions become increasingly industrialized. This could lead to
752 more widespread use of VSLs in the future, which could be transported to the stratosphere and
753 impact the stratospheric ozone layer (Oram et al., 2017). The tagged emissions along the ATom
754 flight tracks for C_2Cl_4 also show relatively higher contribution of AF and SA emissions in the SH
755 (Figure S9).



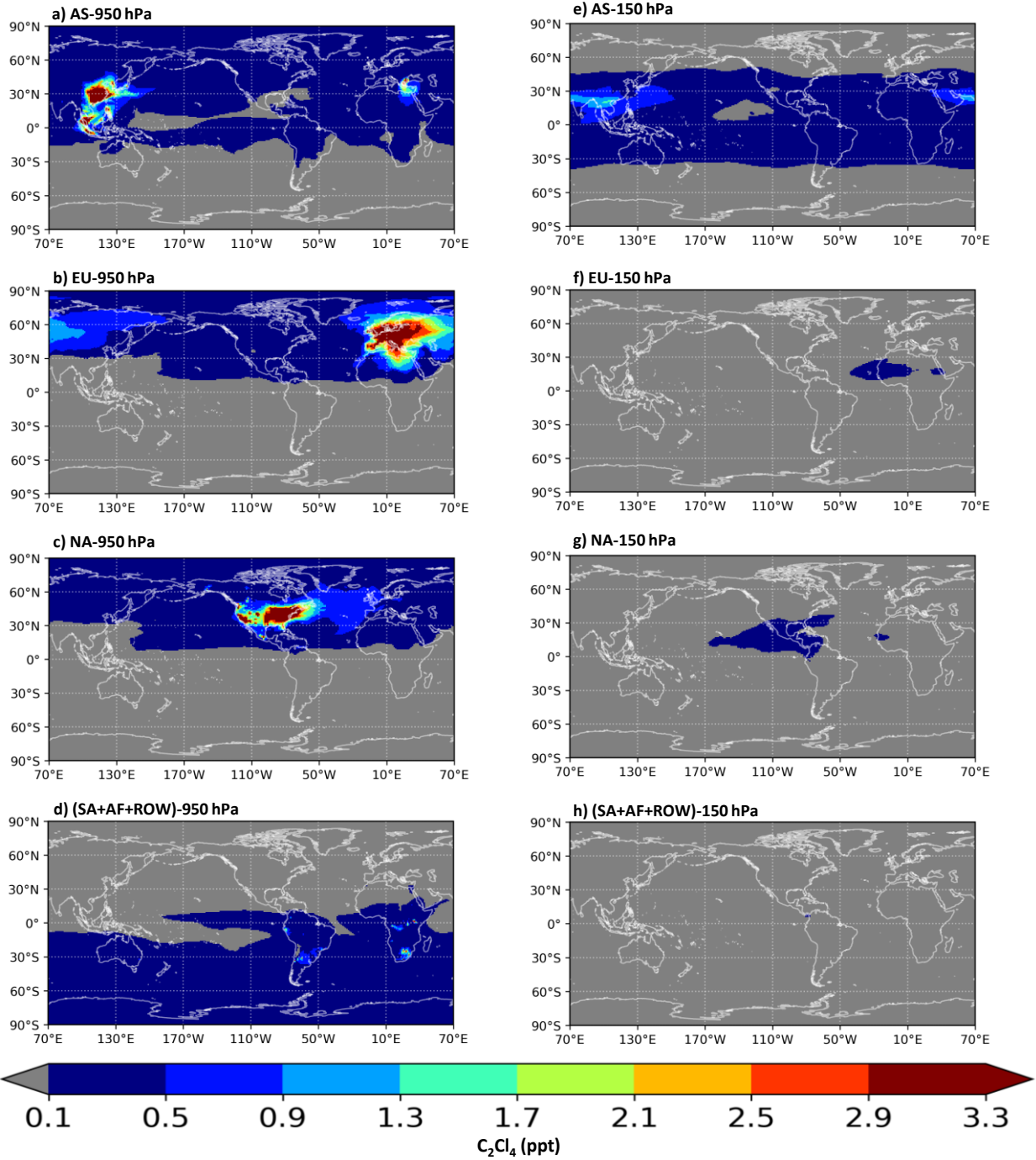
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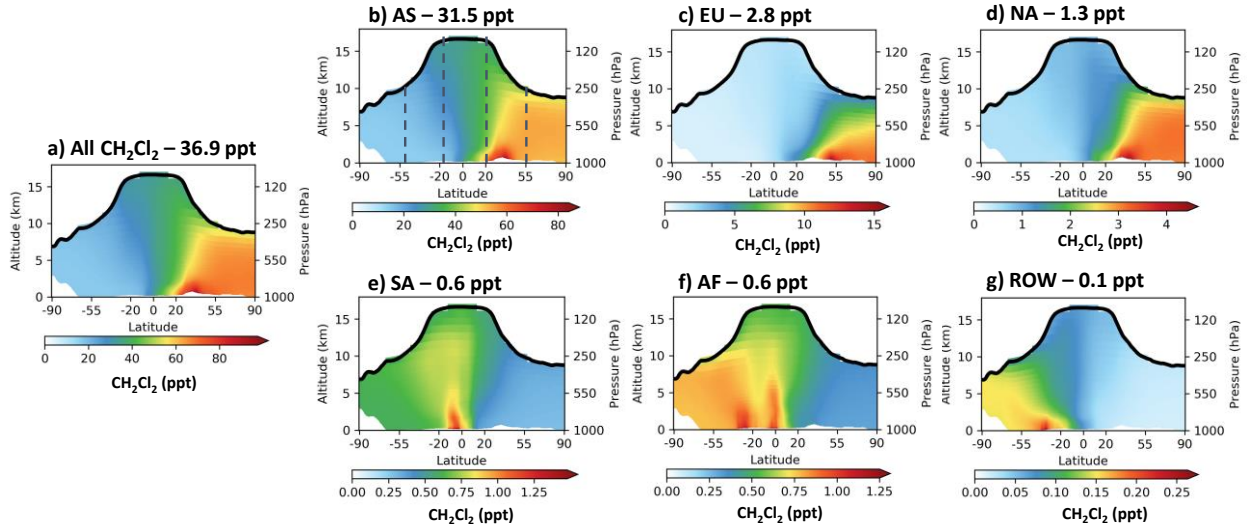
Figure 7. Maps of August 2017 mean CH_2Cl_2 mixing ratios of tagged emissions near the surface (950 hPa) and UTLS (150 hPa \approx 13 km). AS: Asia, EU: Europe, NA: North America, SA: South America, AF: Africa, and ROW: Rest of the world.



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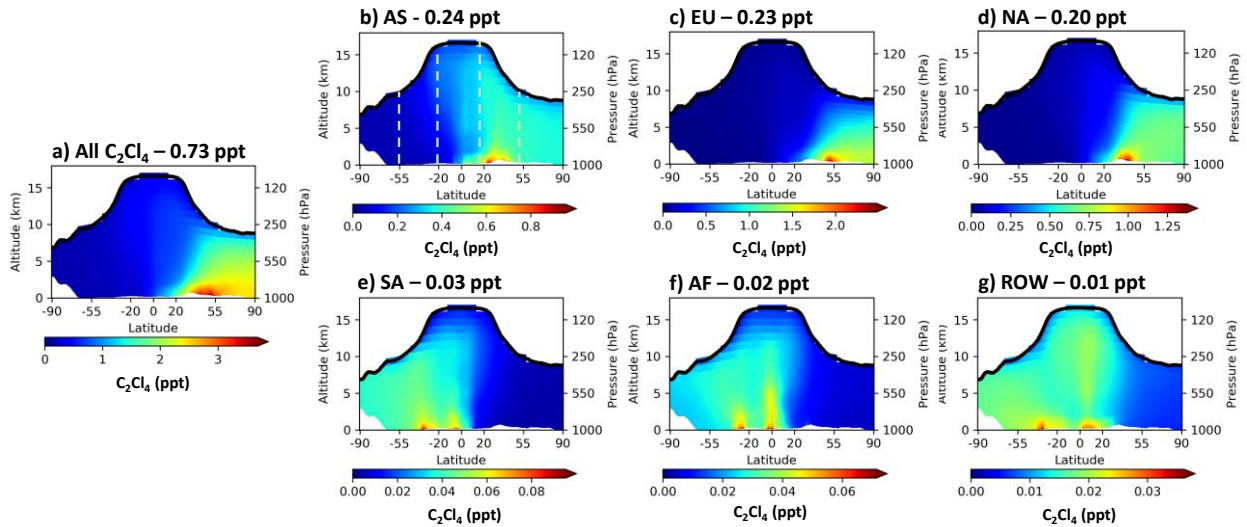
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Figure 8. Same as Figure 7, but for C_2Cl_4 . Note that all the values in (h) are below 0.1 ppt.



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Figure 9. Modeled zonal 2016–2018 annual mean mixing ratios of (a) CH_2Cl_2 and (b-g) zonal mixing ratios resulting from tagged CH_2Cl_2 emissions from each designated region. The solid line shows the location of the mean modeled tropopause. The global tropospheric mean mixing ratios of the (a) global and (b-g) regionally-sourced CH_2Cl_2 are shown on top of each panel, where AS: Asia, EU: Europe, NA: North America, SA: South America, AF: Africa, and ROW: Rest of the world. The dashed line on (b) shows the latitudinal bins used in Table 3. Note that the color scales are different in each panel.



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Figure 10. Same as Figure 9, but for C_2Cl_4 .

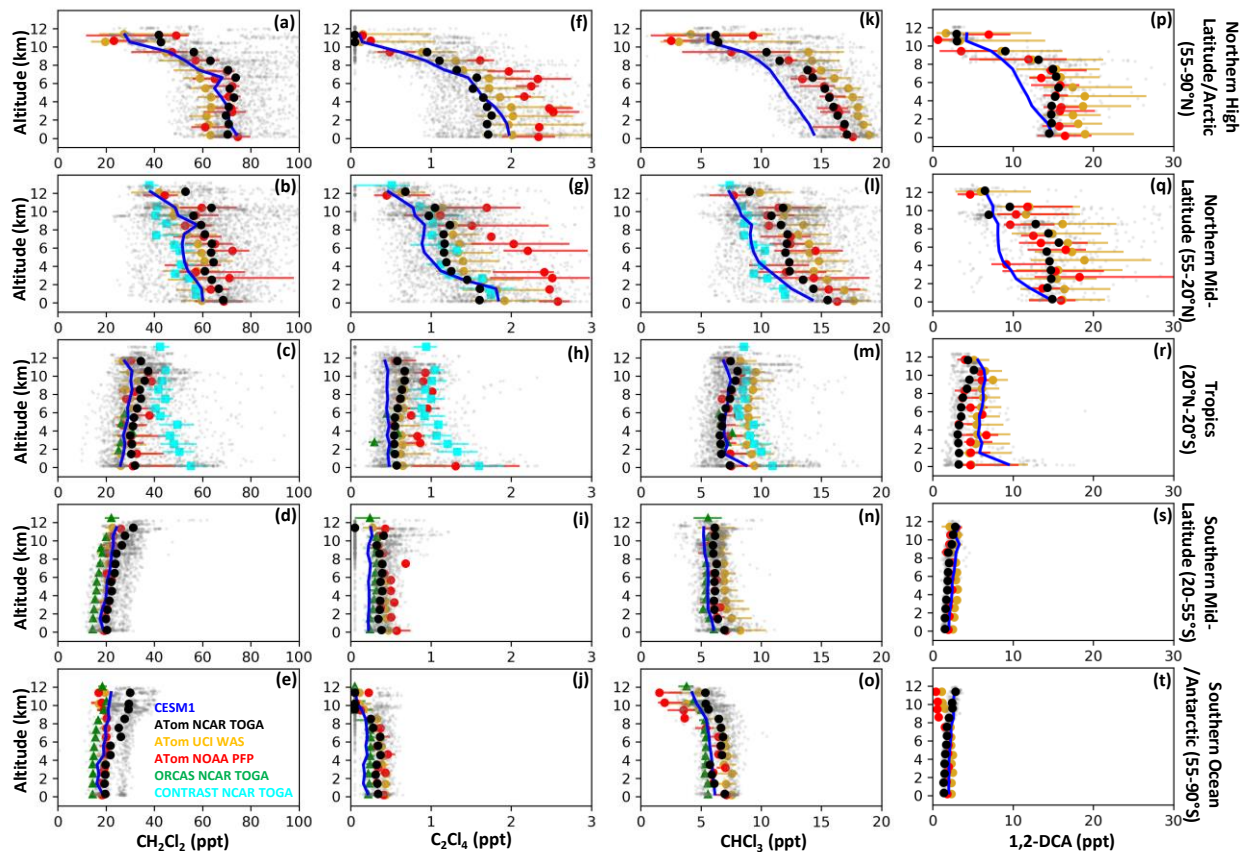
773 **Table 3.** The annual mean (2016–2018) modeled tropospheric source contribution of CH₂Cl₂ and C₂Cl₄, in ppt, due
 774 to tagged emissions in each source region.

Species	Region	Asia (China)	Europe	North America	South America	Africa	Rest of the World
CH ₂ Cl ₂	High latitudes/ Arctic	52.90 (31.95)	7.98	2.92	0.32	0.33	0.02
	Northern mid-latitudes	46.69 (28.39)	4.33	2.05	0.41	0.42	0.03
	Tropics	28.86 (14.13)	1.86	1.02	0.71	0.65	0.06
	Southern mid-latitudes	18.38 (8.41)	1.16	0.65	0.69	0.76	0.11
	Southern Ocean/Antarctic	14.74 (6.74)	0.96	0.54	0.62	0.78	0.14
C ₂ Cl ₄	High latitudes/ Arctic	0.36 (0.20)	0.94	0.62	0.004	0.005	0.007
	Northern mid-latitudes	0.39 (0.23)	0.42	0.40	0.01	0.01	0.01
	Tropics	0.26 (0.09)	0.10	0.11	0.03	0.03	0.02
	Southern mid-latitudes	0.10 (0.03)	0.03	0.03	0.04	0.03	0.02
	Southern Ocean/Antarctic	0.05 (0.01)	0.02	0.02	0.04	0.02	0.02

775 3.5. Modeled major photochemical destruction pathways for Cl-VSLS

776 Reactions with the OH radical, Cl atoms, and photolysis are known photochemical destruction
 777 pathways of Cl-VSLS in the atmosphere and are included in the model (R1–R12). Figure 11
 778 shows the mean rate of CH₂Cl₂ and C₂Cl₄ destruction via these reactions in different latitude bins
 779 between 2016 and 2018. For CH₂Cl₂, reaction with OH is the dominant destruction pathway in
 780 all latitude bins and within the vertical profile. Specifically, the OH reaction rate is 0.9–1.3
 781 orders of magnitude higher than the Cl reaction rate. In addition, photolysis plays a significant
 782 role only during summer and at high altitudes. Specifically, over the SH mid-latitudes and in the
 783 tropics, the mixing ratios of CH₂Cl₂ and C₂Cl₄ decrease at a faster rate with increasing altitude
 784 (sharper gradient) when photolysis (blue line) begins to compete with the Cl reaction.

785 As was shown in

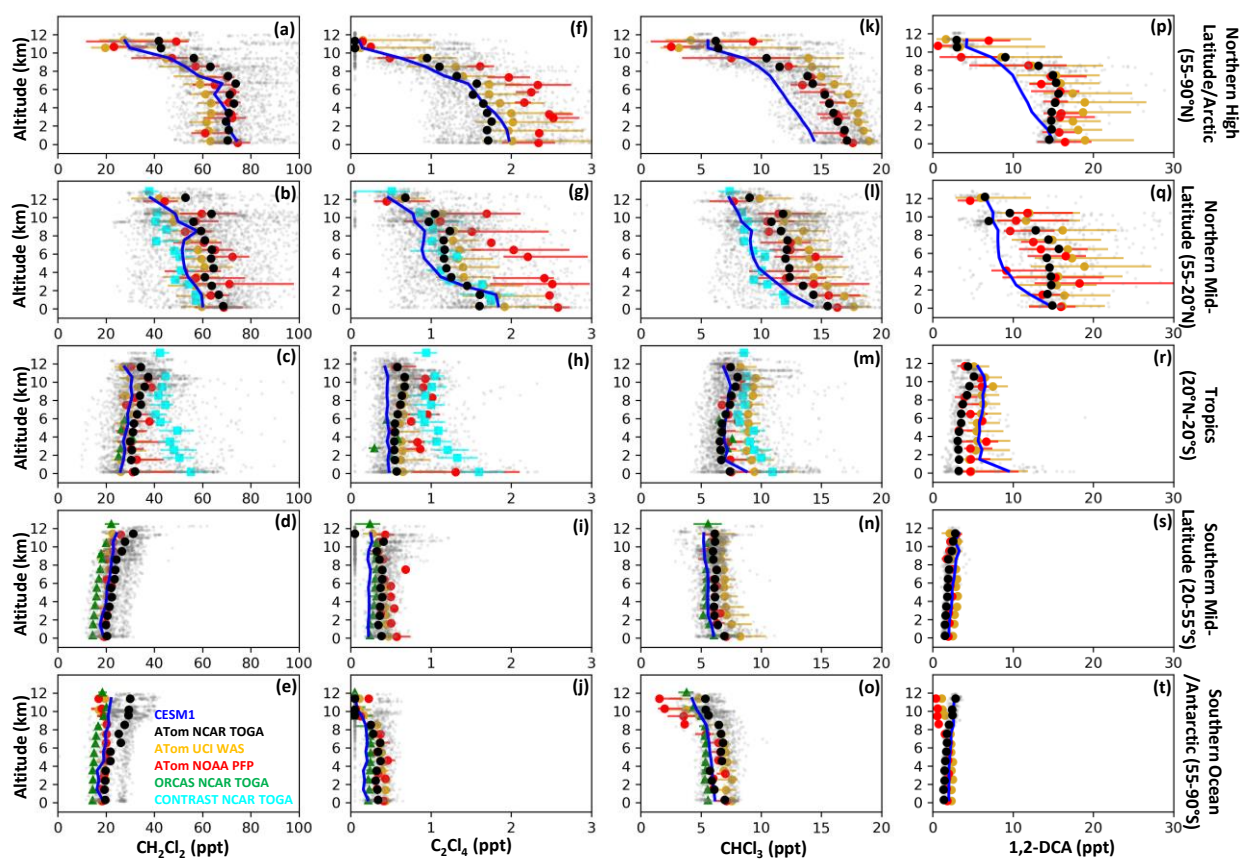


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787 Figure 2, the model underestimates CH_2Cl_2 mixing ratios. Reaction with OH is the main
 788 destruction process of CH_2Cl_2 . As OH is calculated online in the model, it is challenging to
 789 separate the drivers of the sources and sinks of OH, thus the methane lifetime is a good indicator
 790 of the average OH concentration. In our base simulation, the lifetime of methane was 8.4 years,
 791 which is close to the reported values by other studies using CESM1 model (Tilmes et al., 2015;
 792 Tilmes et al., 2016; Nicely et al., 2020), but lower than observational estimates (Prather et al.,
 793 2012). Our model includes halogen chemistry, which has shown to increase the methane lifetime
 794 by 6–9% (Li et al., 2022). Therefore, the uncertainties in the modeled OH (Stevenson et al.,
 795 2020) could explain the bias in part for CH_2Cl_2 . To investigate the OH sensitivity of the model
 796 without directly attempting to change the OH amounts, we ran a simulation in which we reduced
 797 the rate constant k_{OH} for all four Cl-VSLS (i.e., R2, R5, R8, and R11) by 10%, which decreased
 798 the model bias for CH_2Cl_2 by 7% (Figure S10). It is shown that the modeled OH is not more than
 799 ~10% too high (Stevenson et al., 2020), indicating the Cl-VSLS emissions inventories are likely
 800 too low based on the persistent bias with the adjusted OH rate constants. On the other hand,

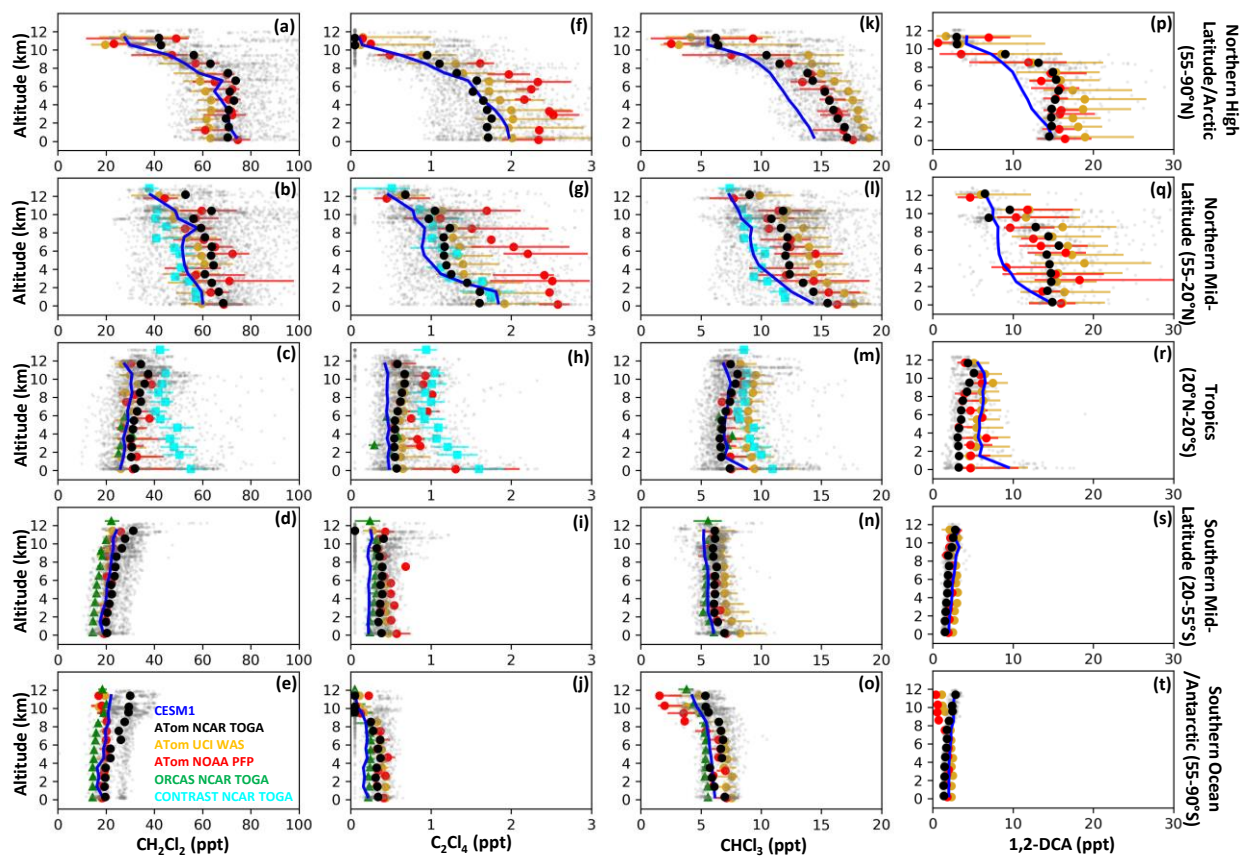
801 CHCl_3 and 1,2-DCA mixing ratios, which are based on LBC rather than emissions in our model
 802 – see Section 2.2 – are not sensitive to this 10% OH perturbation experiment. It shows that
 803 further work is required to update the LBC values in the model or to develop global emission
 804 inventories for CHCl_3 and 1,2-DCA.

805 For C_2Cl_4 , the OH reaction is the major destruction pathway within the lower troposphere except
 806 at high latitudes. As a result, the model NMB reduced by 5% in the OH perturbation experiment
 807 (Figure S10). However, the role of Cl oxidation relative to OH is substantially larger than for
 808 CH_2Cl_2 in the model. In the Arctic, there is a competition between OH and Cl oxidation
 809 pathways while the Cl reaction prevails in the Antarctic region. In other latitude bins, the Cl
 810 reaction is also found to be a dominant pathway in the free troposphere and UTLS region, in
 811 good agreement with the model study of Hossaini et al. (2019) – see their Figure 2. The
 812 photolysis reaction is not an important photochemical destruction pathway in the lower
 813 troposphere for C_2Cl_4 either. However, photolysis of C_2Cl_4 becomes influential in the UTLS
 814 region.



815

816 Figure 2 showed that modeled C_2Cl_4 was too low in the upper troposphere of the NH mid-latitude
 817 region. Our k_{OH} perturbation experiment (Figure S10) shows an infinitesimal improvement for
 818 C_2Cl_4 , indicating the Cl reaction as its main destruction pathway. The Cl reaction being the
 819 dominant pathway in high altitudes requires the presence of Cl atoms in the model. Inorganic
 820 chlorine species are the main sources of Cl atoms in the troposphere (Wang et al., 2019b). Our
 821 results suggest photolysis of molecular chlorine (Cl_2), formed due to acid displacement and
 822 heterogeneous reactions (Li et al., 2022) including stoichiometric ice recycling in the UT
 823 (Fernandez et al., 2014), as the potential source of Cl atoms in high latitudes in the model (Figure
 824 S11). However, ATom measurements made by the NOAA chemical ionization mass
 825 spectrometer (CIMS) instrument show a different spatial distribution (and mean abundance) of
 826 Cl_2 with higher mixing ratios in the tropics and subtropical lower troposphere (Figure S12). Also,
 827 the larger bias of modeled C_2Cl_4 in high altitudes over the mid-latitude region during the ATom
 828 mission does not support the modeled dominant contribution through reactions with Cl atoms (

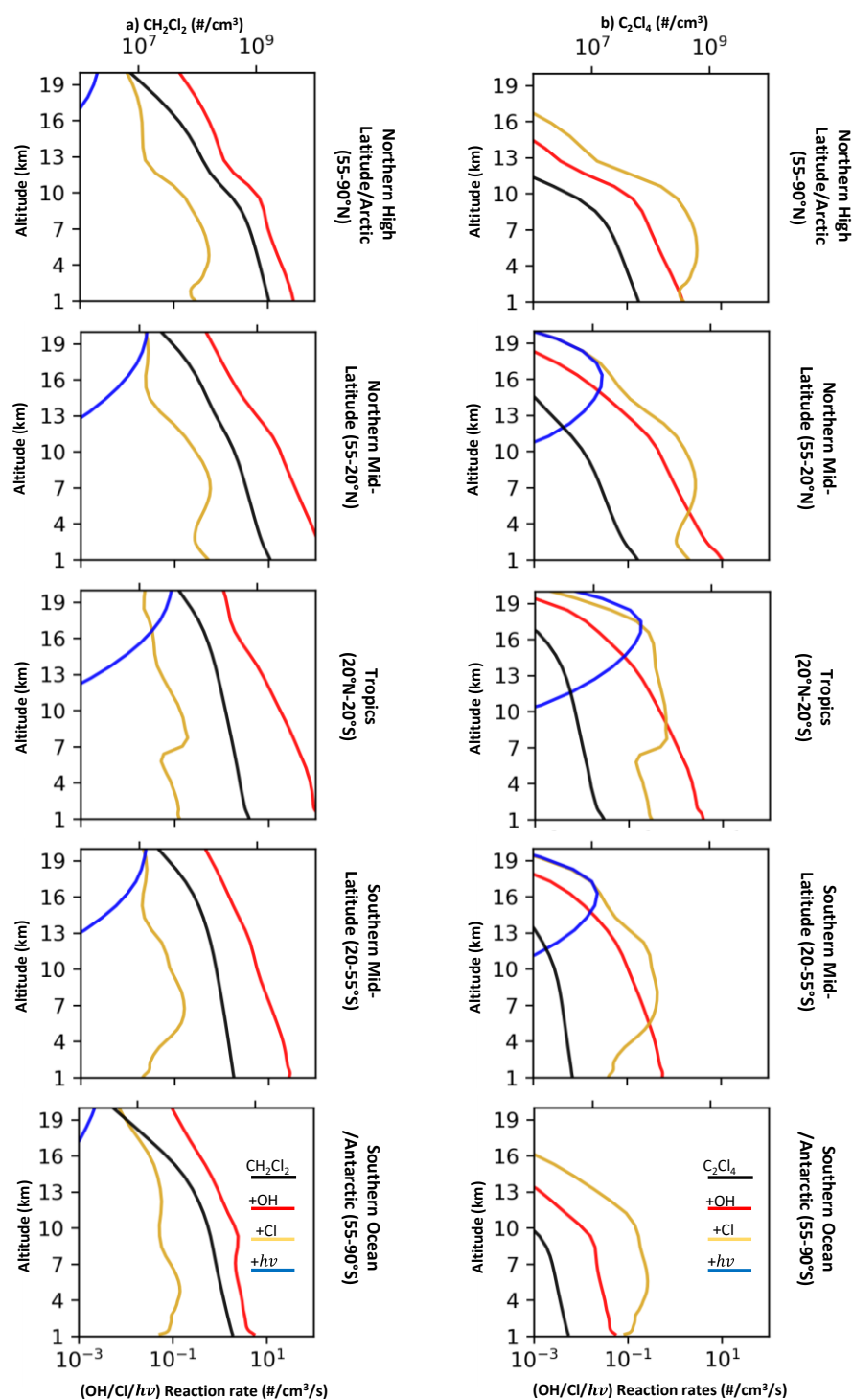


829

830 Figure 2). On the other hand, the source of active chlorine in the atmosphere has still large
831 uncertainties (Wang et al., 2021). Thus, we are investigating our model bias for Cl_2 but have not
832 resolved this.

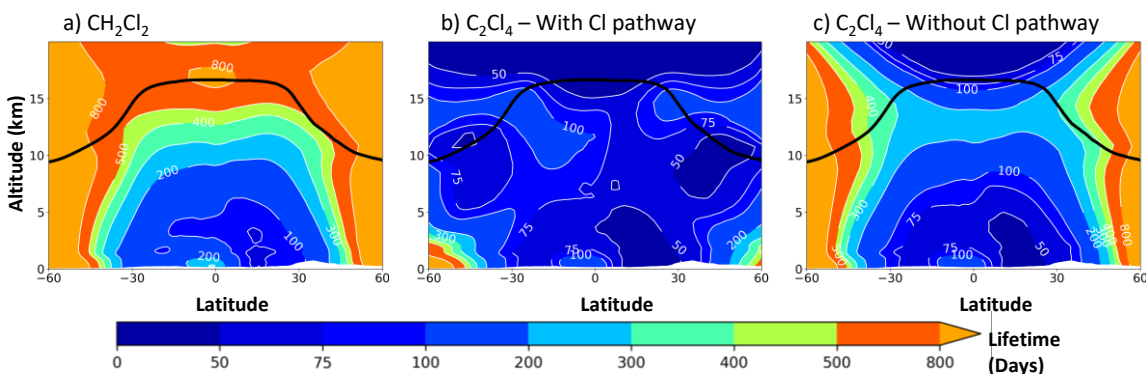
833 Current emission estimates suggest that anthropogenic C_2Cl_4 emissions are about ten times lower
834 than CH_2Cl_2 (Claxton et al., 2020). However, the NH low-altitude (< 4 km or $\theta < 320$ K)
835 measured C_2Cl_4 to CH_2Cl_2 ratio is 0.02 during the ATom mission, which is consistent with the
836 shorter lifetime of C_2Cl_4 (Hossaini et al., 2019). Figure 12 shows the local lifetime of CH_2Cl_2
837 and C_2Cl_4 averaged between 2016 and 2018, calculated based on their partial destruction reaction
838 rates. Overall, the CH_2Cl_2 lifetime increases when moving from tropics towards the mid-latitudes
839 and higher altitudes, from ~ 75 days over the tropical boundary layer up to ~ 500 days around the
840 tropical tropopause. In addition, the CH_2Cl_2 local lifetime has a seasonal variation (Figure S13):
841 it is shorter in the NH during warmer seasons (e.g., JJA) than colder seasons (e.g., DJF).
842 Nevertheless, the shortest boundary layer CH_2Cl_2 local lifetime is always predicted to be over the
843 NH mid-latitudes (Figure S13).

844 The local lifetime of C_2Cl_4 is shorter than CH_2Cl_2 , ranging between ~ 50 days in the tropical
845 boundary layer to ~ 100 days near the tropical tropopause. This is primarily due to the strong
846 oxidation rate of C_2Cl_4 with Cl atoms in our model as discussed above. Figure 12 also shows the
847 C_2Cl_4 local lifetime without considering the Cl destruction pathway, which resembles more to
848 the shape of CH_2Cl_2 local lifetime plots. Regardless, C_2Cl_4 without the Cl destruction pathway
849 still has a shorter lifetime than CH_2Cl_2 in the UTLS region. The shorter lifetime (and more
850 chlorine atoms per molecule) of C_2Cl_4 in the UTLS region suggests that in a case of deep
851 convection, on a molecule/molecule basis, C_2Cl_4 more rapidly liberates its Cl in the UTLS
852 compared to CH_2Cl_2 .



853

854 **Figure 11.** Vertical profiles of 2016–2018 annual mean modeled mixing ratios (black; top x-axis) and reaction rates
 855 (colors; bottom x-axis) for CH_2Cl_2 (left column) and C_2Cl_4 (right column) over five latitude bins around the world
 856 (not only along ATom flight tracks – see text). Note that x-axes are in logarithmic scale.



857

858 **Figure 12.** Modeled zonal 2016–2018 annual mean local lifetimes of (a) CH_2Cl_2 and (b-c) C_2Cl_4 . Panel (b) shows
 859 the C_2Cl_4 local lifetime based on all the reactions (OH, Cl, and $h\nu$) while (c) includes only OH reaction and
 860 photolysis (i.e., without Cl pathway). The solid line shows the mean tropopause location.

861 4. Summary and conclusions

862 In this study, measurements of four Cl-VSLS from surface and aircraft-based programs are
 863 compared to the CESM model that includes updated chlorine chemistry. The bulk of the analyses
 864 focus on ATom data, which provided geographical coverage throughout the Pacific and Atlantic
 865 troposphere and nearly pole to pole. The detailed emission inventories for CH_2Cl_2 and C_2Cl_4 by
 866 Claxton et al. (2020) are used in the model. The CHCl_3 is prescribed as surface boundary
 867 condition (versus emissions) as it is largely emitted by natural sources, and its emissions are less
 868 well known although recent studies have tried to constrain its growing anthropogenic emissions
 869 (Fang et al., 2019; An et al., 2023). Similarly, 1,2-dichloroethane (1,2-DCA) is prescribed as
 870 surface boundary conditions as its emissions are even less well known than CHCl_3 .

871 Aircraft-based measurement data from the NCAR TOGA, UCI WAS and NOAA PFP
 872 instruments are presented and compared. The agreement between the measurements is
 873 reasonably good and generally within reported uncertainties, but some significant differences
 874 exist, which have not yet been resolved. For example, a fairly large discrepancy (up to 60% = 17
 875 ppt for CH_2Cl_2) in the high altitudes of the Southern Ocean and Antarctic region during ATom-3
 876 and ATom-4.

877 The bulk of the model-measurement comparisons were made using the NCAR TOGA data for
 878 which there is the greatest spatial coverage. Detailed comparisons are presented for CH_2Cl_2 and
 879 C_2Cl_4 . The model does a reasonable job of simulating the vertical profiles of all four Cl-VSLS
 880 species. However, the model generally underestimated the mixing ratios of CH_2Cl_2 , most
 881 significantly in the northern mid-latitudes and tropics; it also generally underestimated C_2Cl_4 ,

882 particularly in the tropics and southern latitudes. The model comparisons with surface
883 observations from AGAGE and NOAA networks indicate a reasonable representation of the Cl-
884 VSLS mole fraction seasonal cycles in the model, with the global mean being biased low and
885 high for CH_2Cl_2 and C_2Cl_4 , respectively. These biases primarily point to uncertainties in the
886 emissions, in terms of both magnitudes and global distribution.

887 Measurement profiles over the northern mid-latitudes show a large enhancement of CH_2Cl_2
888 mixing ratios over the Pacific Ocean compared to the Atlantic Ocean. We attribute this to the
889 Pacific being closer to the largest source region for this species and the lifetime of the species of
890 approximately four months, which precludes uniform hemispheric mixing. This is consistent
891 with our analysis of the regionally-tagged CH_2Cl_2 emissions. C_2Cl_4 and CHCl_3 also showed
892 general enhancements for the Pacific vs. Atlantic Ocean. In the NH, Cl-VSLS have sources that
893 are generally co-located and strong correlations between them were noted. The correlations
894 break down in the SH because CHCl_3 sources in the SH have a significant natural component
895 whereas CH_2Cl_2 and C_2Cl_4 do not (Worton et al., 2006; Feng et al., 2018). Significant NH/SH
896 gradients were observed for CH_2Cl_2 and C_2Cl_4 (and CHCl_3 and 1,2-DCA) owing to their source
897 locations and lifetimes with respect to interhemispheric mixing. This was a good test for the
898 model oxidation rates and transport capability and showed that the model simulated the inter-
899 hemispheric distribution of these Cl-VSLS quite well.

900 To further investigate how Cl-VSLS are redistributed globally after release from their sources,
901 we separated the globe into specific regions, and “tagged” model emissions for CH_2Cl_2 and
902 C_2Cl_4 from each of these regions and investigated their global distribution. The annual zonal
903 means of the tagged emissions showed that Asian emissions (AS) dominate the global
904 distributions of these species both near the surface (950 hPa) and at high altitudes (150 hPa),
905 representing up to 70% of the annual mean tropospheric mixing ratios of 36.9 ppt CH_2Cl_2 (73.8
906 ppt Cl). For C_2Cl_4 , EU, NA, and AS are all important contributors to the global annual mean
907 modeled tropospheric mixing ratios of 0.73 ppt (2.92 ppt Cl). Therefore, the contribution of
908 source gas CH_2Cl_2 to the global Cl^{VSLS} burden is approximately 25 times larger than that arising
909 from C_2Cl_4 based on the modeling results averaged between 2016 and 2018. Similar calculations
910 using NCAR TOGA data, along the ATom flight tracks, led to a factor of ≈ 28 (using global
911 mean observed mixing ratios of 46.6 and 0.84 ppt for CH_2Cl_2 and C_2Cl_4 , respectively).

912 Finally, we investigate the primary loss pathways for CH_2Cl_2 and C_2Cl_4 as a function of altitude
913 and latitude within the model framework and we find that reaction with OH dominates the loss
914 rate of CH_2Cl_2 at all latitudes and altitudes while photolysis plays only a minor role and only at
915 high altitudes. For C_2Cl_4 , the reaction with OH represents the most significant loss process, but
916 given the Cl abundances calculated in the model, the reaction with chlorine atoms is competitive
917 with OH in the mid-troposphere and UTLS region, while photolysis is a significant loss process
918 in the LS. This conclusion hinges on the accuracy of the Cl calculated in the model, and further
919 analysis is required to determine if the Cl loss pathway is overestimated in the model given
920 ATom observations of high concentrations of Cl_2 in the remote troposphere and the new Cl atom
921 formation mechanism proposed by van Herpen et al. (2023).

922 In summary, the datasets reported here have provided the best global data coverage to date for
923 Cl-VSLS species and have provided us with the opportunity to test our understanding of global
924 distributions by comparing the measurements with model simulations. The model provided
925 insights into global distributions and pathways. Overall, the model did a reasonable job but
926 missed some observed features such as enhanced low-altitude concentrations of Cl-VSLS
927 (potentially due to both magnitudes and distribution of emissions) and Cl_2 concentrations in the
928 free troposphere that require further study. In our future work, we will study the impact of global
929 atmospheric circulation phenomena such as Asian Summer Monsoon Anticyclone and the North
930 American Monsoon Anticyclone as fast transport pathways of VSLS to the UTLS region, by
931 utilizing the data measured during the 2022 Asian Summer Monsoon Chemical and Climate
932 Impact Project (ACCLIP) and 2021–2022 Dynamics and Chemistry of the Summer Stratosphere
933 (DCOTSS) missions, respectively.

934 **Data Availability**

935 CAM-chem is a component of the NCAR CESM which is publicly available on the project
936 website (<http://www.cesm.ucar.edu/>). The NCAR TOGA, UCI WAS, NOAA PFP, NOAA
937 CIMS, and ATHOS chemical composition data, and NASA DC-8 positional and meteorological
938 measurements for the NASA ATom campaign are available from Wofsy et al. (2021). The
939 ORCAS and CONTRAST data are publicly available from Apel (2017) and Apel and Hornbrook
940 (2014), respectively. The NOAA ground-based measurements were supported in part by
941 NOAA's Atmospheric Chemistry, Carbon Cycle and Climate Program of its Climate Program

942 Office and are available from <https://gml.noaa.gov/hats/data.html> and
943 <https://gml.noaa.gov/aftp/data/hats/solvents/>. Data from the AGAGE monitoring network are
944 available from <https://agage.mit.edu/data/>. The model results used in this study are available
945 from Zenodo data repository (<https://doi.org/10.5281/zenodo.10021453>, Roozitalab et al., 2023)

946 **Competing Interests**

947 The authors declare that they have no conflict of interest.

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1424 **Figure S1 Annual mean (2017) emissions of CH₂Cl₂ and contribution of different tagged**
 1425 **source regions. AS: Asia, EU: Europe, NA: North America, SA: South America, AF:**
 1426 **Africa, and ROW: Rest of the world.**

1427 **Figure S2 Vertical profiles of median CH₂Cl₂ mixing ratios measured during individual**
 1428 **ATom deployments (columns) based on NCAR TOGA (black), UCI WAS (gold), and**
 1429 **NOAA PFP (green) instruments over five different bins, accompanied by the CESM1**
 1430 **modeling results (blue line). Also, NCAR TOGA measurements during ORCAS (green)**
 1431 **and CONTRAST (cyan) missions are shown. All individual data points are shown only for**
 1432 **TOGA instrument (only up to the upper limit of X-axis).**

1433 **Figure S3 Same as Figure S3 for C₂Cl₄.**

1434 **Figure S4 Same as Figure S3 for CHCl₃.**

1435 **Figure S5 Vertical profiles of median 1,2-DCA mixing ratios measured during Atom-4**
 1436 **deployments based on NCAR TOGA (black), UCI WAS (gold), and NOAA PFP (green)**
 1437 **instruments over five different bins, accompanied by the CESM1 modeling results (blue**
 1438 **line). All individual data points are shown only for TOGA instrument (only up to the upper**
 1439 **limit of X-axis).**

1440 **Figure S6 Scatter plots of different Cl-VSLS measured by NCAR TOGA instrument**
 1441 **during a,d) CONTRAST missions, b,e) ORCAS mission, and c,f) Atom-2 deployment color**
 1442 **coded by latitude bins. Note that axis limits are different in each panel.**

1443 **Figure S7 Spatial binned distribution of mean modeled (top subplot) and measured**
 1444 **(bottom subplot) for CHCl₃ (a-d) and OH (e-h) mixing ratios during individual ATom**
 1445 **missions. Latitude and altitude bin intervals are 5 degrees and 1km, respectively. Note that**
 1446 **the measured OH data are based on ATHOS instrument.**

1447 **Figure S8 Vertical profile contribution (percentage) of CH₂Cl₂ tagged emissions during**
 1448 **each individual ATom deployment over the Pacific and Atlantic sides. Percentages are**
 1449 **normalized values for each panel. AS: Asia, EU: Europe, NA: North America, SA: South**
 1450 **America, AF: Africa, and ROW: Rest of the world.**

1451 **Figure S9 Same as Figure S9 for C₂Cl₄.**

1452 **Figure S10 Vertical profiles of median CH₂Cl₂, C₂Cl₄, and CHCl₃ mixing ratios measured**
 1453 **during all ATom missions based on NCAR TOGA (black), base model (blue line), and base**
 1454 **model with 10% reduction for OH reaction constant of Cl-VSLS(red line) over five different**
 1455 **bins. The OH reaction constant is reduced by 10% for R2, R5, and R8 to compensate for the**
 1456 **model high OH abundance.**

1457 **Figure S11 Annual zonal mean (2016-2018) of Cl atom, Cl₂ molecule, and HOCl.**

1458 **Figure S12 Spatial binned distribution of mean modeled (top subplot) and measured**
 1459 **(bottom subplot) for Cl₂ during Atom-3 and Atom-4 deployments. Latitude and altitude**

1460 **bin intervals are 5 degrees and 1km, respectively. Note that the measured data are based**
 1461 **on NOAA CIMS instrument**

1462 **Figure S13 Zonal seasonal mean (2016-2018) local lifetime (days) of a-d) CH₂Cl₂, e-h)**
 1463 **C₂Cl₄. The solid line shows the mean tropopause.**

1464 **Table S1 Number of TOGA data used for each compound from ATom mission. LOD:**
 1465 **Limit of Detection, NaN: Not a Number.**

1466 **Table S2 Monthly mean (106-2018) model evaluation results against ground measurement**
 1467 **sites.**

1468 **Table S3 Summary of the Cl-VSLS data, over tropical upper troposphere, measured by TOGA**

1469 **Table S4 Model evaluation results for CH₂Cl₂ against the ATom deployments.**

1470 **Table S5 Model evaluation results for C₂Cl₄ against the ATom deployments.**

1471 **Table S6 Model evaluation results for CHCl₃ against the ATom deployments.**

1472 **Table S7 Model evaluation results for 1,2-DCA against ATom-4 deployment.**

1473

1474 **Figure 1** Map of the flight tracks for individual ATom deployments, and ORCAS and
 1475 CONTRAST missions. The ground measurement sites in the NOAA and AGAGE networks used
 1476 in this study are also shown. The map is shaded by the 2016 annual average CH₂Cl₂ emissions
 1477 developed by Claxton et al. (2020). 7

1478 **Figure 2.** Binned vertical profiles of median measured CH₂Cl₂, C₂Cl₄, CHCl₃, and 1,2-DCA
 1479 mixing ratios for all ATom deployments from TOGA (black), WAS (gold), and PFP (red),
 1480 separated into five different latitude bins; 55–90°N (a, f, k, p), 20–55°N (b, g, l, q), 20°S–20°N
 1481 (c, h, m, r), 55–20°S (d, i, n, s), and 90–55°S (e, j, o, t), accompanied by the vertically-binned
 1482 median CESM1 model results sampled along the ATom flight tracks (blue curve). Note that
 1483 C₂Cl₄ data for PFP (red points) are only available for ATom-4. Additionally, 1,2-DCA data for
 1484 PFP are for ATom-2 to ATom-4 and for TOGA are only for ATom-4 – see Figures S2-S5 for
 1485 individual deployments. Vertically-binned median TOGA measurements made during ORCAS
 1486 (green) and CONTRAST (cyan) are also shown. Individual ATom TOGA data points are
 1487 included (light grey points) although the x-axes are limited for better clarity in the SH plots.
 1488 Error bars for PFP, WAS, CONTRAST, and ORCAS data are the 25th to 75th percentiles. 21

1489 **Figure 3** Vertical profiles of the median Pacific minus median Atlantic mixing ratios for (a)
 1490 CH₂Cl₂, (b) C₂Cl₄, and (c) CHCl₃ from the TOGA observations at northern mid-latitudes (20–

1491 55°N). Marker colors show the differences during ATom-1 (red), ATom-2 (pink), ATom-3
 1492 (blue), and ATom-4 (black); negative values indicate that median mixing ratios over the Atlantic
 1493 were larger than the Pacific. Note the horizontal grey lines only connect the points for better
 1494 visualization and do not show the range of the observations..... 22

1495 **Figure 4.** Correlation plots of Cl-VSLS measured by TOGA for (a) C₂Cl₄ vs. CH₂Cl₂ and (b)
 1496 CHCl₃ vs. CH₂Cl₂ from ATom-2, ORCAS, and CONTRAST, (c) C₂Cl₄ vs. CH₂Cl₂ from ATom-
 1497 2, colored by latitude bins, and (d) CHCl₃ vs. CH₂Cl₂ from ORCAS colored by latitude bin and
 1498 shaded by altitude. Lines show least square linear regressions. Note that although a line is shown
 1499 for CHCl₃ vs. CH₂Cl₂ from ORCAS in (b), regional data in (d) indicate a non-linear relationship.
 1500 Note that the axes are different in each panel. Also, only ATom-2 is considered in this analysis to
 1501 match the seasons with the CONTRAST and ORCAS projects..... 22

1502 **Figure 5.** Spatially-binned distribution of mean modeled (top subplot) and measured (bottom
 1503 subplot) for (a-d) CH₂Cl₂ and (e-h) C₂Cl₄ mixing ratios during individual ATom deployments.
 1504 Latitude and altitude bin intervals are 5 degrees and 1 km, respectively. TOGA sample locations
 1505 for each ATom mission are shown in Figure S7..... 26

1506 **Figure 6.** 2016–2018 monthly means of ground-based measurements (black) and corresponding
 1507 modeled (blue) ± 1 standard deviation (shading) of CH₂Cl₂ and C₂Cl₄ mixing ratios over five
 1508 different latitude bins: 55–90°N, 20–55°N, 20°S–20°N, 55–20°S, and 90–55°S. Ground-based
 1509 measurements are from the stations within NOAA and AGAGE networks shown in Figure 1.
 1510 The shading represents the ± 1 standard deviation of the monthly means from multiple sites
 1511 within each latitude range and not the spread over different years for individual sites. Note that
 1512 for many months the standard deviations (both model and measurements) are very small
 1513 compared with the y-axes scale. Also, note that smaller y-axes limits are used for SH panels. .. 28

1514 **Figure 7.** Maps of August 2017 mean CH₂Cl₂ mixing ratios of tagged emissions near the surface
 1515 (950 hPa) and UTLS (150 hPa ≈13 km). AS: Asia, EU: Europe, NA: North America, SA: South
 1516 America, AF: Africa, and ROW: Rest of the world. 31

1517 **Figure 8** Same as Figure 7, but for C₂Cl₄. Note that all the values in panel h are below 0.1 ppt. 32

1518 **Figure 9.** Modeled zonal 2016–2018 annual mean mixing ratios of (a) CH₂Cl₂ and (b-g) zonal
 1519 mixing ratios resulting from tagged CH₂Cl₂ emissions from each designated region. The solid

1520 line shows the location of the mean modeled tropopause. The global tropospheric mean mixing
 1521 ratios of the (a) global and (b-g) regionally-sourced CH₂Cl₂ are shown on top of each panel,
 1522 where AS: Asia, EU: Europe, NA: North America, SA: South America, AF: Africa, and ROW:
 1523 Rest of the world. The dashed line on (b) shows the latitudinal bins used in Table 3. Note that
 1524 colorbar limits are different in each panel. 33

1525 **Figure 10.** Same as Figure 9, but for C₂Cl₄..... 33

1526 **Figure 11.** Vertical profiles of 2016–2018 annual mean modeled mixing ratios (black; top *x*-
 1527 axis) and reaction rates (colors; bottom *x*-axis) for CH₂Cl₂ (left column) and C₂Cl₄ (right
 1528 column) over five latitude bins around the world (not only over ATom deployments – see text).
 1529 Note *x*-axes are in logarithmic scale. 37

1530 **Figure 12.** Modeled zonal 2016–2018 annual mean local lifetimes of (a) CH₂Cl₂ and (b-c) C₂Cl₄.
 1531 Panel (b) shows the C₂Cl₄ local lifetime based on all the reactions (OH, Cl, and *hν*) while (c)
 1532 includes only OH reaction and photolysis (i.e., without Cl pathway). The solid line shows the
 1533 mean tropopause location. 38

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