

Tunable Electrical Field-Induced Metal-Insulator Phase Separation in LiCoO₂ Synaptic Transistor Operating in *Post-Percolation* Region

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Abstract

While mixed ionic-electronic conductors with metal-insulator transition (MIT) are promising candidates for designing neuromorphic computing hardware, the fundamentals of resistive switching in these materials are yet to be well understood. This work studies the switching mechanism of the three-terminal nonvolatile redox transistor (NVRT) containing the LiCoO₂ (LCO) channel layer with tunable preferred crystallographic orientation. We used atomic force microscope nanotomography to reconstruct the 3D conductance map of NVRTs, that reveals the applied gate electric-field induces the MIT via reversible phase separation in the LCO channel layer, with the nonequilibrium thermodynamics analytical model providing validation to this mechanism. By operating in the *post-percolation* region, the memory properties can continuously adjust the conductance states of NVRTs. The percolation conductance mechanism via the metallic LCO phase ensures the exceptional linearity and reproducibility of conductance modulation, whereas the field-, rather than current-, induced transition results in the low energy consumption replicating key features of the living neural cells.

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Keywords

Synaptic transistor, phase separation, insulator-to-metal transition, LiCoO₂

1. Introduction

Resistive memristors (RMs) can process information stored in form of electrical resistance which depends on the history of the electrical stimulation signals. The neuromorphic architectures based on RM units in a crossbar array architecture can therefore enable the one-shot operation of matrix-matrix multiplications [1] and the parallel reading of stored data through ‘compute-in-physics’ [2], avoiding the shuttling of data between the processing and memory units in the conventional von Neumann chip architectures [3,4]. By taking advantage of multi-level analog states, in-memory computing capability, and parallel processing – all key characteristics of a living brain, integrated chips that use RM units continue to dramatically improve in terms of memory density, data bandwidth and energy efficiency. However, the multi-level analog states have to rely on the linear nature of the tunable conductance/resistance in RM units, which is still a major challenge for the hardware neural network system where highly repeatable operation is necessary.

The linearity of conductance change in RMs stems from their resistive switching mechanisms. RMs fall into different categories according to the switching mechanisms [5], among which the valence change memories (VCMs) and electrochemical metallization memories (ECMs) are the two most investigated ion migration-based electrochemical redox memristors [6,7]. These ion migration-based RMs can mimic the percolation processes [8-10]. The intriguing nature of the percolation model is that in the *post-percolation* region where a scaling relationship holds, the conductivity can be varied through smooth and continuous transitions with small variations in the percolation network [11,12]. This fundamental physical law provides a great opportunity for ion migration-based RMs to deliver good and predictable conductance linearity. However, in conventional VCMs, the migration of oxygen vacancies often required a high electric field to overcome the energy barriers [13]. Meanwhile, in ECMs, the reversible formation and rupture of the conductive filaments rely on the migration of foreign cations (e.g. Cu²⁺/Ag⁺) [14], and thereby the uncontrollable formation and rupture of metallic filaments in ECMs deteriorate the resistive switching performance [15-16]. As most electrochemical redox memristors rely on the complex filament formation process within the switching layer (or in the interface), the uncertainty of migration ion species makes the switching performance difficult to predict [6]. These hinder the reversibility, accuracy, and symmetry of the

1 transition, that are detrimental to the performance of neuromorphic devices, where analog switching
2 with finely adjustable resistive states is desirable.

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4 Lithium-ion migration-based memristors have recently become emerging RMs. Lithium-ion has the
5 smallest size among all metal cations, which has made it a perfect charge carrier for energy storage
6 devices such as lithium-ion batteries. Compared to other metal cations or oxygen anion migration
7 that creates a switching filament [17], much lower energies are required to drive Li ions'
8 intercalation into battery materials. Additionally, one of the most striking outcomes of lithium
9 intercalation/de-intercalation within transition-metal-oxide crystal lattices is that the resultant metal-
10 insulator transition (MIT) [18] can change its conductivity by several orders of magnitude [19-21].
11 Therefore, recently, lithium-transition metal oxide materials, including Li_xCoO_2 [22], $\text{Li}_x\text{Ti}_y\text{O}_2$ [23],
12 LiFePO_4 [24], LiLaTiO_3 [25], and LiNiO_x [26], have been studied as the resistive switching layer for
13 RMs simply by adjusting the lithium stoichiometry and phase spatial pattern. These studies have
14 shown that Li ions-based memristors have enormous potential as artificial synapses for
15 neuromorphic computing with high density, low power consumption, and fast operating devices,
16 offering applications beyond energy storage. The resistance switching behaviors are suggested to be
17 controlled by the migration and accumulation of Li ions without large structural distortion, and
18 therefore it is highly analogous to the transmission of neurotransmitters from the presynaptic neuron
19 to the postsynaptic neuron. However, understanding of the resistive switching mechanism in LiCoO_2
20 and $\text{Li}_x\text{Ti}_5\text{O}_{12}$, which show an MIT transition upon the lithium intercalation/extraction, has
21 undergone long historical debates since the initial report of analog battery materials (NaCoO_2 /
22 LiCoO_2) as resistive switches [27,28].

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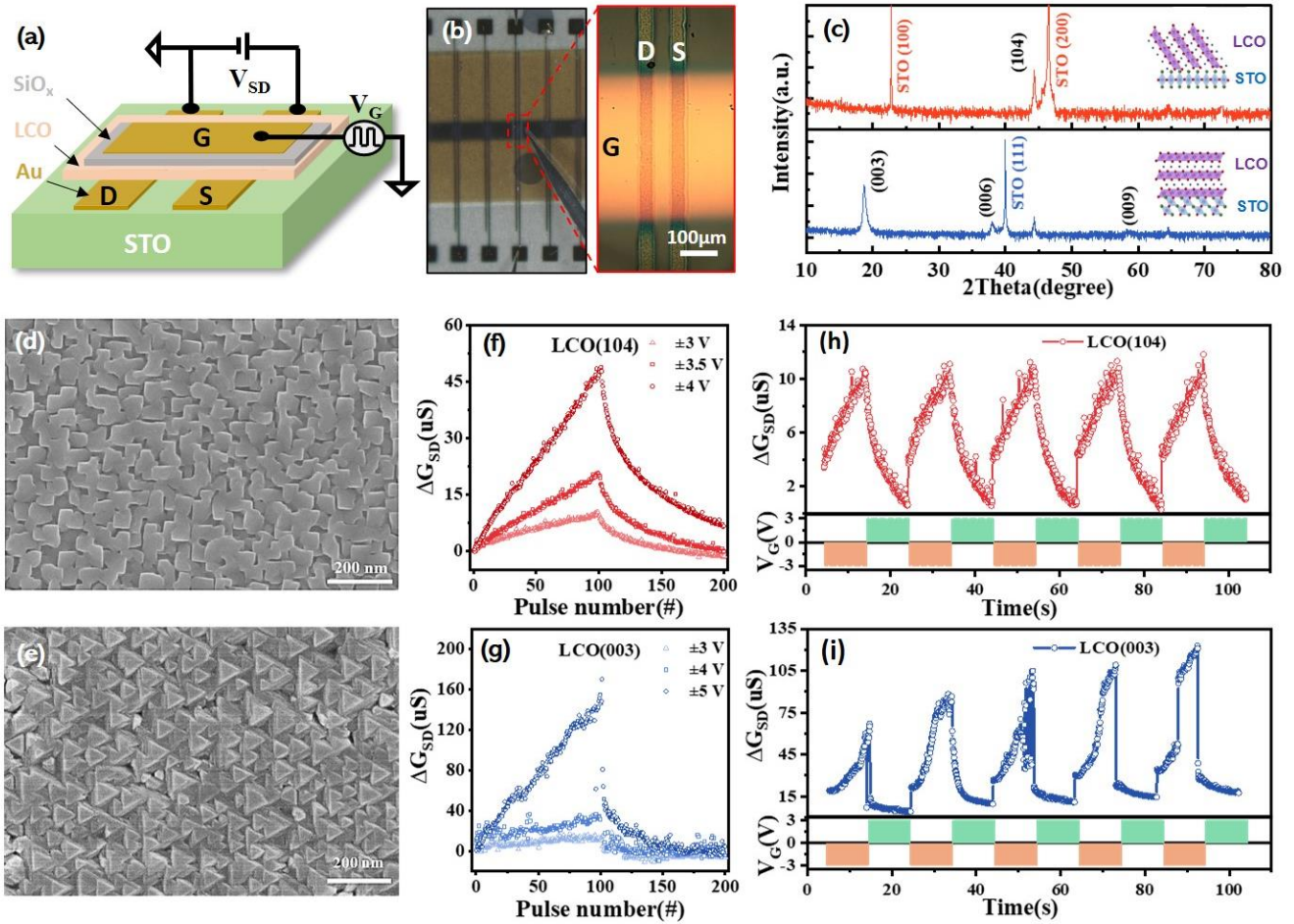
24 In 2011, Moradpour et al. reported the possibility of LiCoO_2 (LCO) as a resistive memory material
25 for the first time [28]. Conductive Atomic Force Microscope (c-AFM) was used to measure the
26 resistive switching behavior of conductive AFM tip/ $\text{LCO}/\text{Si}_{p++}$ micro-nano battery memristor [29-
27 30], and the authors proposed that the resistive switching phenomenon originates from the redox
28 reaction between Co and H_2O inside the LCO-tip meniscus. This mechanism was also accepted in
29 the follow-up reports [29,31]. However, Fuller et al. fabricated the first three-terminal LiCoO_2 -based
30 RM device to mimic the stimuli-responsive nature of neural networks [32], claiming that the resistive
31 switching originated from the MIT of LCO switching layer due to the change of the lithium content
32 that was locally driven by the electric field. This contradicted the previously proposed mechanism of
33 interfacial redox reaction between Co and H_2O . Later, the diffusion of Li ions driven by the electric
34 field and the interfacial reaction between the LCO switching layer and silicon oxide trapping layer

1 were confirmed by Nguyen et al. [30], revealing the necessity of a dielectric silicon oxide trap layer
2 for the LCO resistive switching. Hu et al. [33] also found that a suitable thickness of the SiO₂
3 trapping layer can reduce the total electromotive force field [34] and prevent the backward migration
4 of lithium ions to LCO. Recently, the symmetric two-terminal memristor based on the MIT of
5 lithium titanate oxide has been demonstrated as a promising artificial synapse [23], in which an
6 electric field-induced non-equilibrium phase separation was proposed as the underlining switching
7 mechanism. Although this provides new insights toward the MIT phase separation switching model
8 for LCO [35], the phase separation pattern hasn't yet been verified experimentally in real space and
9 nanoscale spatial resolution due to the lack of analytical tools. Moreover, the sharp MIT transition
10 during the phase separation seems to be contradictive to the obtained decent tunability and linearity
11 of conductance [23]. Significantly, the interplay between ion migration/redistribution dynamics and
12 metal-insulator transition [36], which defines the memristive behavior and directly affects the
13 performance of Li ions-based devices [17,23], has not yet been explained by existing switching
14 theoretical models.

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16 In this work, we fabricate the three-terminal memristive transistor with LCO switching layer as a
17 close-to-ideal model to explore the switching fundamentals in lithium-ion based RMs. The system
18 allows also control of the lithium diffusion dynamic and permittivity of the switching layer by
19 changing the LCO crystal orientation. We found that the devices with different LCO crystal
20 orientations show different switching threshold voltage (switching energy barrier), conductance
21 linearity and retainability. The resistive switching of the NVTRs derives from the percolation and
22 MIT phase separation of the LCO layer (bulk effect) and the trapping of lithium ions by the dielectric
23 SiO_x layer (interfacial effect). The percolation that drives the conductivity gain, as well as the
24 interfacial ion trapping, stems from the localized redistribution of lithium ions. This fundamentally
25 eliminates the internal stress variation caused by the repetitive formation and rupture of conductive
26 filaments within the switching layer, eliminating the strain-induced damage to devices during
27 cycling. The designed NVTRs operating in the *post-percolation* region provide good linearity of
28 conductance modulations. Significantly, the proposed electric field-induced nanoscale MIT phase
29 separation and percolation mechanism are directly verified by 3D nano-tomography AFM in good
30 agreement with the analytical solutions via nonequilibrium thermodynamics. These findings pave the
31 way for mixed ion-electron conductors with MIT as a powerful platform for the development of
32 synaptic transistors with tunable electrical properties.

33 34 **2. Results and discussion**

1 2.1 Three-terminal NVRT with tunable switching characteristics



2 **Figure 1.** (a) The sketch of a three-terminal NVRT unit. (b) Top view of the NVRT unit array. (c) XRD patterns of LCO channel layer
3 with (104) and (003) preferred orientation. (Insets - the crystal structure models showing epitaxial relationships) (d) and (e) SEM
4 images of surface morphology of LCO channel layer with different orientations. (f) and (g) The long-term potentiation (LTP) and
5 long-term depression (LTD) curves of LCO-(104) and LCO-(003) NVRT, where training pulses consisting of 100 negative pulses (200
6 ms) followed by 100 positive pulses (200 ms) at the voltages specified in the graph. (h) and (i) Reproducible analog channel
7 conductance modulation of LCO-(104) and LCO-(003) NVRT, where 50 negative (-3 V, 200 ms) and 50 positive pulses (3 V, 200 ms)
8 are performed.

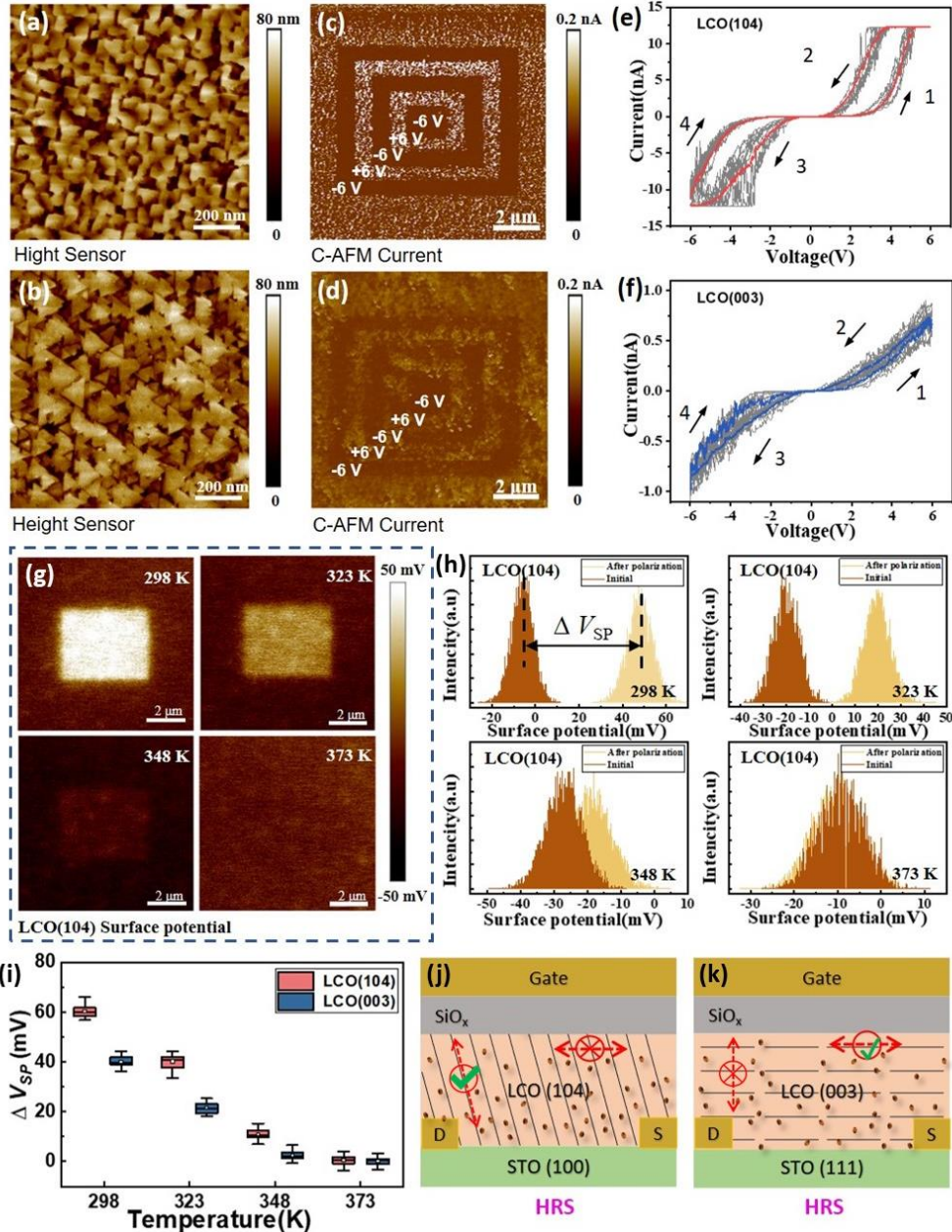
9 We fabricated the NVRT device as shown in **Figure 1a** using LCO as the switching channel layer,
10 and silica as the gate ion trapping layer. The trapping effect [30,33] of the amorphous silica gate
11 dielectric layer was evaluated by testing the robust switching behavior in the $\text{LiCoO}_2/\text{SiO}_x/\text{Si}$ stack
12 (see supplementary information, SI, section 1). The advantage of this NVRT device structure is that
13 the direction of the applied ‘write’ electric field is perpendicular to the direction of the ‘read’ electric
14 field (between source and drain electrodes). This drastically reduces the crosstalk of data writing and
15 reading processes and hence considerably improves the device's robustness. Figure 1b is the top view
16 photo of an NVRT unit during the electrical performance tests, the drain electrode is grounded, the

1 voltage pulses (V_G) are applied to the gate electrodes, and the small DC voltage (V_{SD}) was applied
2 between the source and drain for the conductivity measurements. Additionally, to understand the
3 effects of anisotropic lithium-ion transportation on the switching performance of this LCO-based
4 memristor [22,37], we employed two orientations of SrTiO₃ (STO), (100) and (111) signal crystal
5 substrates [38-40], to control the orientation of LCO channel layers. The thickness of the channel
6 layer was targeted to be ~70 nm (SI, Figure S4). The XRD pattern in Figure 1c shows that the LCO
7 channel layer with (104) and (003) preferred orientations are obtained on the STO (100) and (111)
8 substrates respectively. The SEM images in Figures 1d and 1e further confirmed the characteristic
9 crystal faces on different LCO crystal planes. The LCO channel layer surface consists of ‘cubic’
10 crystal facets has (104) preferred orientation, while the one with ‘triangle’ crystal facets is the (003)
11 preferred orientation [41]. This is consistent with the XRD results in Figure 1c. Moreover, this
12 epitaxial relation between LCO and STO substrate is well maintained at the nanoscale boundary
13 between the Au source/drain bottom electrodes and channel (SI, Figures S6 and S7), facilitating the
14 further scaling down of the NVRT device.

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16 Figures 1f and 1g are the long-term potentiation (LTP) and long-term depression (LTD) curves of
17 LCO-(104) and LCO-(003) NVRT. In each cycle, the changes in source-drain conductance (ΔG_{SD})
18 were recorded after applying the negative “set” voltage pulses during the LTP, followed by the
19 positive “reset” pulses with the same amplitude and width during the LTD. It is worth noting in
20 Figure 1g that, at the pulse amplitude of 4 V, the ΔG_{SD} of LCO-(003) is about 1.88 times compared
21 to its initial conductance value G_{SD0} ($\Delta G_{SD}/G_{SD0} \sim 1.88$) after the 100th negative voltage pulse.
22 However, when the voltage pulse amplitude increases to 5 V, a drastic modulation of NVRT
23 conductance can be observed in Figure 1g, the $\Delta G_{SD}/G_{SD0}$ value increases to about 10 after the 100th
24 negative voltage pulse. By contrast, as shown in Figure 1f, the LCO-(104) NVRT not only has a
25 lower pulse amplitude threshold ($\Delta G_{SD}/G_{SD0} \sim 16$ under the plus amplitude of 4 V) but also shows a
26 better linearity and retention capability for conductance modulations compared to the LCO-(003)
27 (see SI, section 3). Importantly, the “linearity coefficient” of LTP of LCO-(104) is about 0.11 (see
28 SI, section 4), which is smaller than the values reported recently in metal oxide analogue memristors,
29 such as NaNbO₃ [42] and WO₃-based [43]. We also noticed that the LTD process shows much less
30 linearity compared to the LTP. This nonlinearity is closely related to the different lithium-ion
31 migration dynamics in the NVRTs, which are discussed in detail in SI, section 5. Figures 1h and 1i
32 are the five cycles of LTP and LTD curves. The obtained channel conductance in LCO-(003) NVRT
33 shows poor symmetry and an abrupt increase of conductance during LTP due to the lithium
34 congestion effect [43]. The LCO-(104) device shows good linearity and symmetry even when the

1 identical gate pulses were used in both potentiation and depression processes. This high linearity and
 2 symmetry are essential indicators for synaptic transistors to accurately control and predict learning
 3 processes in hardware neural networks. Overall, we demonstrate that the LCO NVRT with a low
 4 switching voltage, as well as a highly linear switching operation, can be obtained by adjusting the
 5 crystal orientation of the LCO channel layer.

6 2.2 The effects of anisotropic lithium migration within the switching layer



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 8 **Figure 2.** (a, b) AFM surface topography images, (b, c) C-AFM conductivity switching patterns, (e, f) multi-cycled *I-V* curves. (g)
 9 KPFM surface potential images of a square polarization region on LCO surface at temperatures of 298K, 323K, 348K, 373K. (h) The
 10 statistics of four surface potential distributions of polarized and unpolarized regions after different temperatures initiated the
 11 destruction of polarization corresponding to Figure g. (i) The changes of the average value of ΔV_{SP} at different temperatures for LCO-

1 (104) and LCO-(003) surface. (j) and (k) The sketching of ion migration/redistribution models of LCO-(104) channel layer and LCO-
2 (003) channel layer.

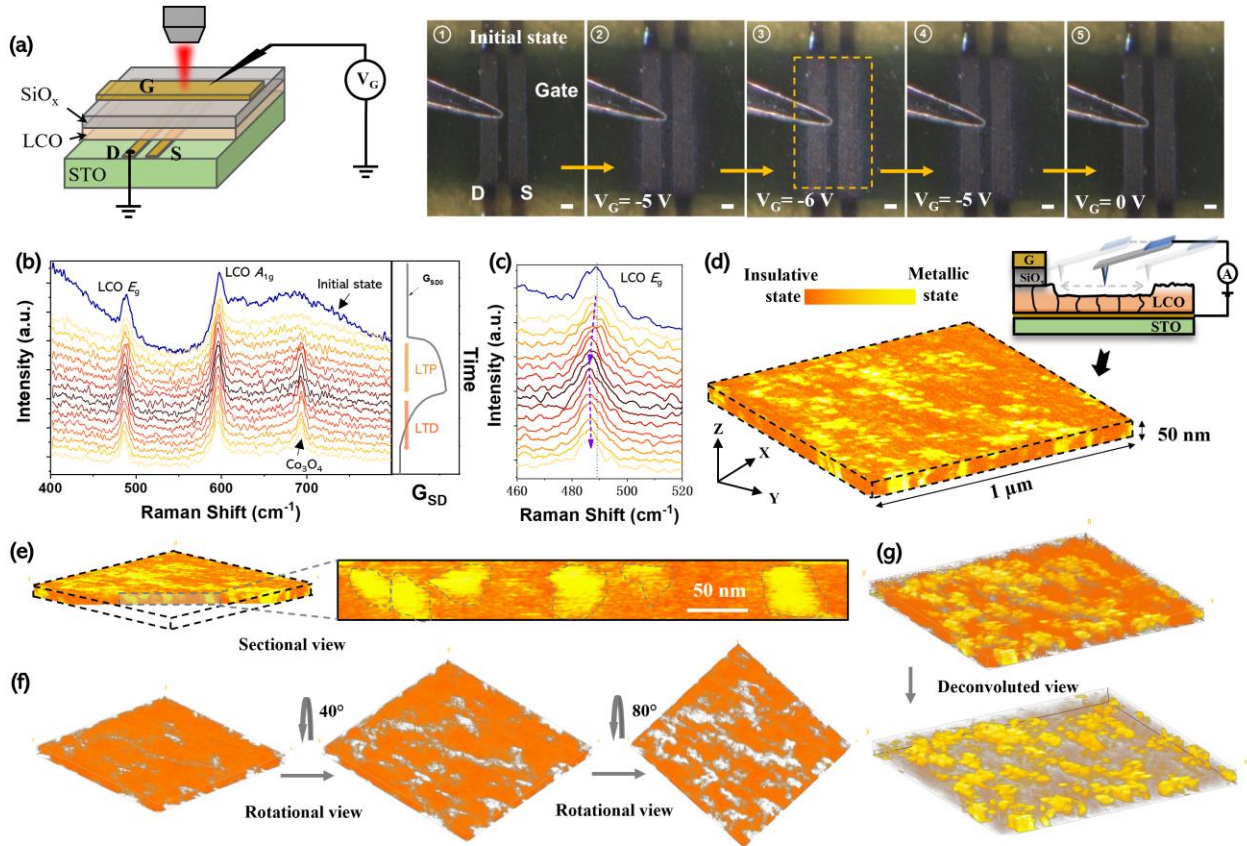
3 The electrical performance characterizations in Figure 1 show that the NVRT unit with LCO-(104)
4 channel layer exhibits a smaller switching voltage and better retention capability compared to LCO-
5 (003). To further understand this difference, c-AFM and temperature-dependent KPFM were
6 introduced to study the lithium diffusion dynamic under the applied electric field and thermal field in
7 LCO-(104) and LCO-(003).

8
9 The AFM topography images of LCO-(104) and LCO-(003) in **Figures 2a** and **2b** show numerous
10 cuboidal and triangular domains respectively, which is consistent with the SEM images in Figures
11 1d, e. The c-AFM images in Figures 2c and 2d were recorded after electrical tip-writing using the
12 concentric square method [28]. As shown in Figure 2c, the c-AFM image of LCO-(104) has a clear
13 boundary of high and low resistance state areas, indicating good stability and repeatability in
14 resistance switching behaviour. However, the c-AFM image on the LCO-(003) surface, as seen in
15 Figure 2d, shows a blurred conductivity distribution map. Meanwhile, as shown in the *I-V* curves in
16 Figures 2e and 2f, both films show similar and robust bipolar resistance switching behaviours, and
17 the switching is but LCO-(104) has a higher current response (higher conductivity) along the
18 direction perpendicular to the film surface compared to LCO-(003). These results confirm that the
19 conductivity modification is easier to achieve along the out-of-plane direction in LCO-(104). We
20 present the localized Li-ion migration models under the applied electric field in Figures 2j and 2k.
21 The LCO-(104) film has Li-ion transport channels (Co-O octahedra interlayers) almost perpendicular
22 to the film surface, and the low lithium/electron diffusion energy barriers within the two-dimensional
23 Co-O layers [44] result in the low switching write voltage of LCO-(104) NVRT. In contrast, the
24 LCO-(003) film surface is parallel to the Li-ion channels, the local redistribution of Li-ions needs to
25 be realized via the diffusion of Li-ions along the grain boundaries, as shown in Figure 2k. Along the
26 grain boundaries, the high diffusion energy barrier of lithium-ions and low conductivity of electrons
27 [45] both contribute to the high switching voltage of LCO-(003) NVRT.

28
29 The surface potential retention was assessed to understand the stability of the NVRT devices toward
30 thermal destruction. Inside a $10\times 10\ \mu\text{m}$ scanning region, a $5\times 5\ \mu\text{m}$ square region was polarized by
31 applying a +6 V bias to the substrate. Afterwards, the surface potential images of the LCO-(104)
32 films were recorded at different destructive temperatures, as shown in Figure 2g. As the temperature
33 increases, the average value of the surface potential difference (ΔV_{SP}) between the central region and

1 marginal region tends to reduce leading to the destruction of the recorded pattern. The statistical
 2 distributions of the surface potential of polarized and unpolarized regions under different
 3 temperatures were shown in Figure 2h (see also SI, Figure S4). Figure 2i shows the temperature-
 4 dependent ΔV_{SP} for LCO-(104) and LCO-(003). From the figures, one can find that the LCO-(104)
 5 film displays better retention and higher ΔV_{SP} at each temperature than LCO-(003) film. For
 6 example, at 348 K, the average value of ΔV_{SP} of LCO-(003) is close to zero eV, while that of LCO-
 7 (104) is nearly 10 mV. This indicates the LCO-(104) NVRT has higher retainability than LCO-(003)
 8 NVRT. The schematic diagrams are shown in Figures 2j and 2k. Under the thermally induced
 9 destruction of the recorded potential pattern, the Li-ions in LCO-(104) NVRT redistribute slowly in
 10 the horizontal direction (between source and drain) due to the pinning effect of Co-O layers, while
 11 the Li-ions flow relatively ‘free’ between the source and drain in the LCO-(003) NVRT. Therefore,
 12 the LCO-(104) NVRT shows much more symmetric LTP and LTD curves in Figures 1f-i. Most
 13 importantly, a much better retention capability can be found in the LCO-(104) NVRT during the
 14 conductance depression training.

15 2.3 The electric field-induced MIT phase separation and percolation of metallic phase



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 17 **Figure 3.** (a) Schematic diagram of the optical microscope and optical images of the device with LCO-(003) during the LTP test at
 18 different gate voltages. The gate voltage pulse is a triangular pulse with an amplitude of -6 V and a width of 2 seconds. The inset scale
 19 bars are 50 μm. (b) *in situ* Raman full range (400-800 cm⁻¹) spectra and (c) E_g peak shifting of LCO-(104) channel layer during the

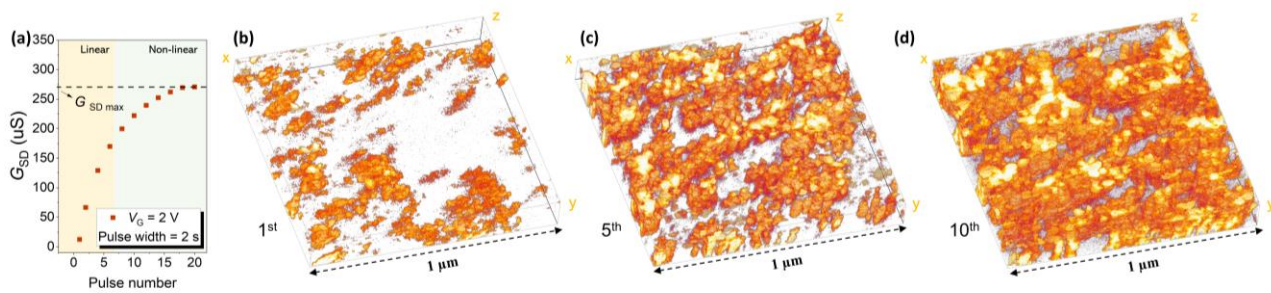
1 LTP and LTD tests. A_{1g} band shifting can be found in SI, section 7. (d) Reconstructed 3D distribution of metallic and insulating LCO
2 phases for LCO-(104) channel layer. (e) Section view, (f) rotation view, and (g) deconvoluted view of metallic and insulative phases
3 inside the LCO-(104) channel layer corresponding to Figure 3d. (The full video recording can be found in SI).

4 To reveal the microscopic nanoscale switching mechanism and dynamics in the LCO transistors, we
5 recorded the switching process of an NVRT unit using *in situ* optical spectroscopy as shown in
6 **Figure 3a**. From the optical images, one can find that a metallic phase with high reflectivity [46] (the
7 shining area) underneath the gate electrode reversibly appears during the large negative voltage (-6
8 V) polarization cycles (The full video can be found in the supplementary information, VIDEO 1).
9 This indicates that the MIT phase transition and relaxation [19,47] of LCO layer are happening under
10 voltage polarization. *In situ* Raman spectroscopy was used to further reveal the phase transition of
11 the LCO-(104) NVRT under gate voltage polarizations during LTP and LTD tests. As seen in
12 Figures 3b and 3c, the initial LCO E_g and A_{1g} peaks blue shift to the lower wavenumber region
13 during the LTP test (negative gate voltage polarization), and then moving back to the high
14 wavenumber region during the LTD, unambiguously indicating the reversible phase transition
15 between the lithium-deficient LCO metallic phase and lithium-rich LCO insulative phase [48,49].
16 The peak position shifting of E_g and A_{1g} bands are summarized in SI, Figure S15. Besides, it's also
17 worth noting that, when the device conductance is approaching the saturation value in LTP training,
18 further applying the negative gate pulses results in the formation of the Co_3O_4 phase (Raman peak ~
19 691 cm^{-1}). This implies that the NVRTs should be operated at the linear LTP conductance
20 modulation region to avoid the over-extraction of lithium from the LCO lattice which will cause the
21 irreversible phase transition.

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23 This reversible phase transition in the LCO materials can be realized by either the intercalation/de-
24 intercalation of lithium through the solid-state electrochemical reaction with the SiO_x layer (long
25 rang diffusion), or the localized redistribution of lithium (inside or between the LCO grains). To
26 verify which of these assumptions is correct, we introduced a 3D tomography conductive AFM [50]
27 to explore the subsurface conductive properties in the LCO-(104) channel layer. As shown in Figure
28 3d, 3D tomography conductive AFM uses a relatively stiff (spring constant of $\sim 100\text{ N m}^{-1}$)
29 conductive diamond-coated tip to 'scratch' the LCO channel layer after negative gate voltage
30 polarization. During the scratching, a high force of $\sim 10\text{ }\mu\text{N}$ was applied to the tip to 'dig' into the
31 bulk region of the LCO layer, and the Z scanner feedback gain was reduced to achieve a 'loose'
32 feedback loop. The tomographic conductivity maps were recorded during the tip milling through the
33 LCO layer eventually reaching the substrate. By reconstructing the tomographic conductive maps
34 along the Z direction, the conductivity distribution inside the LCO layer can be revealed. The

1 obtained 3D reconstructed conductivity mapping is shown in Figure 3d. From the figure, it is found
 2 that after the LCO layer was polarized and switched to the low resistance state by a negative gate
 3 voltage, some active grains (bright yellow colour) show a high current signal (metallic LCO phase),
 4 while the rest of regions show low conductivity (insulative LCO grains and the grain boundaries that
 5 are also of low conductivity) with sparse conductive areas. The 3D sectional view in Figure 3e shows
 6 the internal distribution of the conductive grains. Interestingly, each of the metallic grains serves as a
 7 “conductive hotspot” of the LCO channel layer (see the full video in the SI, VIDEO 2). The 3D
 8 rotational view of insulative components in Figure 3f further indicates that the metallic grains serve
 9 as the filaments, not only “intersecting” the whole LCO layer, but also “interconnecting” with each
 10 other to form a percolated conductive cluster [51,52]. Meanwhile, from the deconvoluted view in
 11 Figure 3g, one can see that the metallic phase grains are interconnected in the lateral direction,
 12 building up a conductive network along the in-plane direction (between the source and drain
 13 electrodes), and thereby increasing the G_{SD} .
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 16 **Figure 4.** (a) G_{SD} of LCO-(104) NVRT as a function voltage plus number during the Long-term potentiation (LTP). The applied gate
 17 voltage pulse is 2 V with a width of 2 seconds. (b), (c) and (d) 3D distribution of conductive regions in LCO-(104) layer after 1st, 5th
 18 and 10th gate pulse cycles, respectively. The zero current regions were masked out in the 3D reconstruction maps.

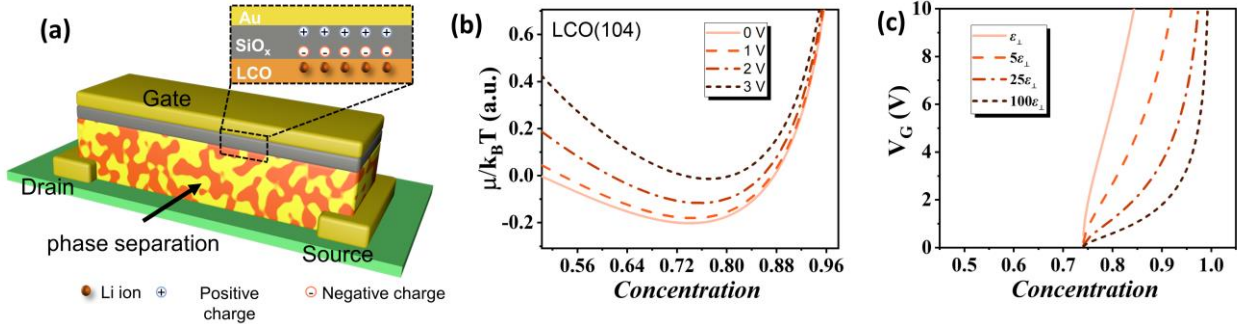
19 We further reconstructed the network of metallic LCO grains inside the LCO-(104) layer after a set
 20 of wide gate voltage pulses as shown in **Figure 4**. In Figure 4a, by applying a wide voltage pulse (2
 21 seconds), the G_{SD} as a function of the pulse number shows a tendency to saturate in the LTP curve.
 22 The LTP curves in Figures 1f-i were measured in the linear conductance modulation region far away
 23 from the device conductance reaching the saturation value ($G_{SD \max}$). Within this linear region, as
 24 shown in Figures 4b-d, one can find that the numbers of the metallic grains increased upon the gate
 25 voltage pulse cycles. Moreover, the statistical volume percentage of the conductive regions in these
 26 three states is about 21%, 67% and 82%, respectively. This indicates that after the device switching
 27 (suitable gate voltage/pulse was applied), the LCO layer nearly operating in the *post-percolation*
 28 region [11]. It is this *post-percolation* region that enables the device conductance (G_{SD}) to be
 29 changed continuously with good linearity as shown in Figures 1f-i. This conductance modulation

1 mechanism based on the percolation here was unambiguously supported by the 3D tomography c-
 2 AFM results, suggesting that the resistive switching mechanism in our LCO NVRT is distinct from
 3 the conventional memristors [53-56].

4 **2.4 Dynamics of the electrical field-induced MIT phase separation in the LCO layer**

5 As we demonstrated, the underlining fundamentals of the MIT switching in our NVRT are related to
 6 the lithium-ion diffusion [30], while it is unlikely that the suggested nano-battery mechanism [33,34]
 7 can guarantee the highly reversible switching processes since the huge volume expansion and solid-
 8 state-interphases caused by the lithiation of the SiO_x layer could be detrimental to the device
 9 reversibility [57]. Additionally, the formation of a percolation conductive network has been observed
 10 only inside the LCO layer according to the 3D nano-tomography conductive AFM measurement, as
 11 seen in Figures 3 and 4. Therefore, based on the optical observation, 3D c-AFM mapping and Raman
 12 spectra results, we believe that the switching mechanism of our NVRT should be triggered by an
 13 electric field-induced MIT phase separation [23,35,58] in the LCO channel layer, which involves
 14 short-term lithium-ion rearrangement in the LCO grain (inter-grain) or between the grain (intro-
 15 grain), rather than the massive redistribution by crossing the grain boundaries.

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18 **Figure 5.** (a) Schematic of LCO phase separating under an applied electrical field. (b) The reduced chemical potential of LCO-(104)
 19 as a function of the concentration under different gate voltage. (c) The critical phase separation voltage (V_{set}) of NVRTs with arbitrary
 20 LCO crystal orientation (different permittivity values: ϵ_{\perp} , $5\epsilon_{\perp}$, $25\epsilon_{\perp}$, $100\epsilon_{\perp}$) as a function of concentration. The simulation parameters
 21 can be found in the supplementary information.

22 The model of the device with a phase separation LCO channel layer is shown in **Figure 5a**. From the
 23 thermodynamic point of view, phase separation occurs in a homogenous binary system when the free
 24 energy of the system locates at the miscibility gap in which the spinodal and nuclear decomposition
 25 could both occur [59]. When a voltage V_G was applied to the gate electrode, the electrostatic
 26 potential ϕ was built within the LCO channel layer. The free energy of the LCO in the channel layer
 27 can be written as [23,58]

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$$F[c, \phi] = \int_v \left[G(c, \nabla c) - \frac{1}{2} \varepsilon(c) |\nabla \phi|^2 + \rho(c) \phi \right] dv \quad (1)$$

Where c is the normalized concentration of lithium-ion inside the channel layer, $c=0$ and $c=1$ represent the metallic and insulative LCO phases respectively. The actual concentration of $c=0$ and $c=1$ are equal to about $\text{Li}_{0.93-0.97}\text{CoO}_2$ and LiCoO_2 , respectively [18,60,61]. The first term, $G(c, \nabla c)$ function, represents the free energy density of the solid solution, and later two terms describe the total electrostatic energy, in which $\varepsilon(c)$ and $\rho(c)$ are the permittivity and charge carrier density, respectively. The lattice parameter changes between the metallic and insulative phases are less than 2% [62]. Moreover, it is assumed that the redistribution of lithium is a localized process (no filament formation), therefore the changes in mechanical free energy can be neglected here. According to the regular solution model and Cahn-Hilliard approximation [63]

$$G(c, \nabla c) = \frac{\Omega}{v} c(1 - c) + \frac{k_B T}{v} [c \ln c + (1 - c) \ln(1 - c)] + \frac{1}{2} \frac{\kappa}{v} |\nabla c|^2 \quad (2)$$

The first two terms represent the free energy density of the homogeneous solid solution, where Ω is the interaction coefficient between the species. The third term describes the volumetric phase boundary interaction, where κ is the gradient penalty. The initial LCO channel layer in the insulative state has a permittivity ($\varepsilon_i = \varepsilon(1)$) which is up to ten times of the dielectric constant of vacuum ε_0 [64], and is proportional to the lithium vacancy density inside the LCO lattice. At the same time, the permittivity of the metallic phase grows nonlinearly with the extraction of lithium ions and tends to be an infinite value (perfect conductor) after MIT transition ($\varepsilon_m = \varepsilon(0) \gg \varepsilon_i$). This characteristic of permittivity changes during the lithiation and delithiation process in lithium oxide-based phase separation materials can be described by using Clausius-Mossotti relation [23]: $\varepsilon(c) = \varepsilon_i \left(\frac{3}{c} - 2 \right)$. Therefore, from Equation (1), it can be found that the generic behaviour of the permittivity has significant effects on the system thermodynamic stability under the applied electric field. The total free energy is drastically lowered by the co-existence local area of the metallic LCO phase with higher permittivity $\varepsilon(c)$, especially at low lithium concentration region ($c \rightarrow 0$) where electrostatic energy is large. Namely, a higher electric field can lead to the spontaneous decomposition of the initial homogenous insulative LCO phase. Combining Equations (1) and (2), the diffusion electrochemical potential of lithium ions in the system can be expressed as

$$\mu = \frac{\delta F}{\delta c} = \Omega(1 - 2c) + k_B T \ln\left(\frac{c}{1-c}\right) + \kappa \nabla^2 c + \frac{3v}{2c^2} \varepsilon_i |\nabla \phi|^2 + v \rho'(c) \phi \quad (3)$$

Using this, we can now analyse the critical condition of system phase separation mediated resistive switching. Initially, the LCO layer is a single phase homogenous solid solution without phase boundaries, and there is no concentration gradient inside the channel layer ($\kappa \nabla^2 c = 0$), with the charge carrier density staying relatively constant before MIT transition ($v \rho'(c) \phi \approx 0$). When a pulse voltage with an amplitude higher than V_{set} is applied to the gate, the generated electric field between the gate and source electrodes may drive the system into the miscibility gap, in which the spinodal decomposition or nucleation and growth occurs from the local lithium deficient areas. The spinodal points of the initial homogenous system under the applied critical voltage V_{set} thus is given by

$$\frac{\partial \mu}{\partial c} = -2\Omega + \frac{k_B T}{c(1-c)} - \frac{3v}{c^3} \varepsilon_i |\nabla \phi|^2 = 0 \quad (4)$$

where $\nabla \phi = \frac{V_{set}}{L}$, L is the thickness of the LCO channel layer, thus the critical voltage

$$V_{set} = L \sqrt{\frac{c^3}{3v\varepsilon_i} \left[\frac{k_B T}{c(1-c)} - 2\Omega \right]} \quad (5)$$

According to Equation (3), Figure 5b shows the simulation plots of reduced chemical potential ($\frac{\mu}{k_B T}$) of LCO-(104) as a function of concentration (c) of the initial homogenous state (LiCoO_2 , $c=1$) under different applied voltages. Here, $\approx 50 \text{ nm}$, $\Omega \approx 2.6 k_B T$, $v \approx 1 \text{ nm}^3$, $\varepsilon_{i(104)} \approx 10\varepsilon_0$ [58, 64, 65]. We can see from Figure 5 that, with the increase of applied V_G value from 0 V to 3 V, the phase separation threshold concentration increases from about 0.74 to 0.79. This implies that the MIT phase separation can occur in the LCO channel layer under a smaller concentration fluctuation with the assistance of a localized lithium redistribution. Similar results of LCO-(003) can be found in Supplementary Figure S5. This is consistent with nanotomography 3D c-AFM results where redistribution of lithium ions in some LCO grains results in the formation of the metallic LCO phase. In this 3D map of physical properties, a homogeneous initial solid solution of lithium ions is driven to phase separate above a critical voltage (V_{set}), leading to the formation of electronically conducting local areas. These conductive areas connect to form a conductive network and increase the conductance between the source and drain electrodes, as sketched in Figure 5a. Meanwhile, the accumulation of opposite polarity carriers (electrons) on the LCO/SiO_x interface can also be

1 preserved by the dielectric SiO_x layer. This bulk effect, combined with the interfacial effect on the
2 LCO/SiO_x interface where the accumulated Li-ion were trapped in the SiO_x layer, retaining the
3 device at the high conductance state. In contrast, reversing the polarity (applying a positive gate
4 voltage) can drive the backflow of lithium-ions and destroy the phase separation ‘conductive
5 network’ with a time delay to allow for local rupturing of the percolation conductive network. In
6 summary, the positive/negative gate pulse voltage cycling leads to reversible creation and destruction
7 of the percolation cluster of the metallic phase with hysteresis due to spinodal decomposition of the
8 homogeneous state and nucleation/growth of new phases, meanwhile, the SiO_x serves as a trapping
9 layer to increase the retentivity of the NVRTs.

10
11 This model also can be used to explain the microscopic fundamentals of the switching threshold
12 voltage difference in LCO-(104) and LCO-(003) NVRTs. According to Equation (5), the critical
13 phase separation voltage (V_{set}) of four systems with the permittivity value of ϵ_{\perp} , $5\epsilon_{\perp}$, $25\epsilon_{\perp}$, $100\epsilon_{\perp}$
14 are plotted as a function of concentration in Figure 5c. ϵ_{\perp} is the permittivity of LCO crystal along the
15 *c*-axis. With the increase of LCO permittivity along the device’s out-of-plane direction, the critical
16 V_G for the LCO layer to reach the phase separation concentration is reduced. In other words, in the
17 system with high permittivity ($100\epsilon_{\perp}$), at the same V_G , negligible localized lithium redistribution
18 could trigger the insulative LCO ($c=1$) to enter the phase separation concentration region.
19 Considering that the in-plane (ϵ_{\parallel}) permittivity of LCO (001) crystal plane is higher than its out-of-
20 plane (ϵ_{\perp}) value [64], therefore, the channel layer permittivity between gate and source/drain
21 electrode in LCO-(104) NVRT (corresponding to the permittivity parallel to the Co-O octahedra
22 layer in LCO crystal lattice) is higher than the permittivity in LCO-(003) NVRT (corresponding to
23 the permittivity perpendicular to the Co-O octahedra plane in LCO crystal lattice). This difference is
24 even magnified at low-frequency region as discussed in SI, section 8. As a result, the smaller V_G
25 value was observed for driving the switching of LCO-(104) NVRT as shown in Figure 1f.

26 27 **2.5 Challenges and opportunities of Li-ion based three-terminal synaptic transistor**

28 After more than 10 years of effort put into the development of lithium-ion based memristor by the
29 worldwide scientific communities [5,22-23,28,33], significant advances have been achieved,
30 including optimized device architecture [32], switching performance [33], as well as fundamental
31 understanding [23, 35] of the resistive switching phenomena. Yet, some question remains elusive.

1 First of all, the role of oxygen vacancy on the channel layer's electrical conductivity is rarely
2 discussed in Li-ion based memristors. Oxygen vacancy plays a significant role in many metal oxides-
3 based memristors, especially the correlated oxides with MIT. For example, Jian et al. demonstrated
4 the synaptic action in three-terminal correlated oxide (SmNiO_3) devices [66]. Interestingly, they
5 found that the increase of oxygen vacancy leads to an increase in resistivity in the metallic state of
6 SmNiO_3 , while in the insulating regime, introducing oxygen vacancies modulates the resistivity of
7 SmNiO_3 channel layer non-monotonously. But in our case, the dependence of the conductance of
8 LCO channel layer on lithium concentration is relatively monotonous in both insulator and metallic
9 states, and the effect of oxygen vacancy can be negated in this lithium-ion migration induced MIT.
10 To be more specific, in the electrical field-polarized LCO thin film, the main charge carriers are
11 lithium-ions, and the migrations of oxygen vacancies are negligible compared to the lithium-ions
12 [45,67-68]. The activation energy of lithium-ion diffusion in side LCO is at the order of magnitude
13 of $\sim 10^{-2}$ - 10^{-1} eV [69-70], which is one order of magnitude smaller than that of oxygen vacancies.
14 Additionally, the mobility of lithium ions inside the LCO is higher than the mobility of oxygen
15 vacancies [67]. Such behaviour is explained as the oxygen ions are covalently bound with cobalt ions
16 and form the stable framework of edge-sharing CoO_6 octahedra layers [71], which guarantees the
17 crystal structure integrity during the reversible lithium-ion intercalation/deintercalation, enabling
18 LCO to be the most widely used and electrochemically stable lithium-ion battery cathode materials
19 [72]. Therefore, given the high mobility and low diffusion energy barrier of lithium-ions in LCO, we
20 believe that the applied electrical fields in our device mainly drive the lithium migration and
21 redistribution inside the LCO thin film. However, one should also notice that, at highly de-lithiated
22 states, the oxygen vacancies migration/formation energy barrier can be significantly reduced, even
23 causing oxygen evolution [73-75]. These oxygen vacancy migration-induced Co-O layer
24 rearrangements can cause irreversible phase transitions with the migration of oxygen vacancies being
25 detrimental to the LCO crystal structure integrity. Nevertheless, such conditions are avoided in our
26 memristor devices where the operation envelope does not reach such values.

27

28 Secondly, being a type of solid-state ionic-based memristor, the operation speed of LCO NVRTs
29 could be a bottleneck for their application for high-frequency neuromorphic computation. Although
30 the diffusion kinetics of lithium in solid is much better compared with other cations (such as Ag^+ and
31 Cu^+ in electrochemical metallization memories), the migrations of Lithium-ions in lithium transition
32 metal oxide often is accompanied by the phase transition with non-negligible energy barriers [76-78].
33 Recent reports on metal oxide-based memristors, such as TiO_2 [79], Ta_2O_5 [80], Nb_2O_5 [81] etc.,
34 show sub-microsecond/nanosecond operation speed with neuromorphic functionality, which is better

1 than the common electrochemical random-access memories that are limited by the slow ion kinetics
2 [82] on the order of milli-second write-read delays [22] and/or requiring long read pulses to ensure
3 accurate readout of a non-volatile change [83-84]. As discussed above, lithium-ion based NVRTs'
4 operation speed is intrinsically limited by the diffusion of Li^+ in the LCO, but it can be tackled by
5 shortening the characteristic diffusion length, represented by lithium-ion redistribution. This means
6 that, in our case, lithium-ion does not have to travel back and forth between two electrodes like a
7 "rocking chair" secondary battery, but it can redistribute between the percolation units inside the
8 resistive layer switches, switching between two slightly non-stoichiometric states (metallic $\text{Li}_{0.95}$ -
9 CoO_2 and insulative $\text{Li}_{0.5-0.95}\text{CoO}_2$) through inter- and intra-grain redistribution. Previous studies
10 suggested that improving the crystallinity of LCO material will increase the diffusivity by a factor of
11 10^3 and extend the operability to the μs regime [32,85]. Therefore, we believe that reducing the
12 characteristic diffusion length inside LCO, such as reducing the grain size or increasing the
13 dislocation density, can both reduce the Li^+ redistribution/diffusion time, resulting in a further
14 enhancement of the operation speed. This assumption has been validated by our preliminary results
15 measured on the device with a shorter channel layer distance of $10\ \mu\text{m}$. The scaled-down device has
16 shown a linear LTP behaviour under the stimulation plus with a width of $1\ \mu\text{s}$ (SI, Figure S8d).
17 Similar strategies were adopted by Onen et al. to achieve a nanosecond protonic programmable
18 memristor [86].

19

20 However, it is worth noticing that when compared with other types of memristor, the key advantages
21 of our NVRT memristor rely on the small energy consumption, absence of electroforming, process
22 tunability, and in particular, its high robustness. LCO-based memristor devices were found to
23 maintain conductance levels and hold the state of charge for several weeks after "write" operations
24 [33] as additional energy is needed to drive the lithium redistribution inside the percolation network
25 of LCO after the MIT phase separation. The high electrical resistivity of SiO_x layer ensures that the
26 resistance switching is non-volatile and shows a high level of robustness [30,33]. In some cases, Li
27 out-diffusion can contribute to self-discharge, a typical slow degradation process ($>1-2\%$ lithium
28 backflow per month) during the storage of lithium-ion batteries. But neural applications often target
29 training periods of less than one week after which the weight information can be read and stored
30 digitally if needed [32]. Therefore, we believe that our LCO NVRTs with good robustness can play a
31 significant role in the further development of neural applications.

32

33 **3. Conclusions**

1 In summary, we fabricated the three-terminal transistor with memristive behavior. In this transistor,
2 the key parameters, including the retention capability, thermal stability and switching voltage, can be
3 effectively adjusted by tuning the crystal orientation of the LCO channel layers between the source
4 and gate electrodes. The low switching voltage and high linearity were found in the NVRT device
5 with the LCO (104) crystal planes perpendicular to the ‘writing’ electrical field direction (parallel to
6 the device ‘reading’ electrical field direction). Significantly, the highly reversible trapping/releasing
7 effects of the SiO_x gate dielectric layer, rather than the interfacial solid-electrochemical reaction with
8 lithium-ions, guarantee the stability of long-term resistive switching cycles and low energy required
9 for the operation of such a neuromorphic device. The energy consumption per “write” (E_{write}) in our
10 LCO NVRTs is estimated at the order of magnitude of pico-joule which is comparable with the
11 recently reported values in three-terminal transistor [87], and it can be further reduced by scaling
12 down the device size (see energy consumption analysis in SI, section 9). Apart from the interfacial
13 SiO_x trapping effect, the electric field-induced phase separation, as well as the percolation of metallic
14 LCO phase within the channel layer, were observed in the three-terminal NVRT by *in situ* Raman
15 and 3D nano-tomography conductive AFM. The phase separation dynamic was numerically studied
16 by nonequilibrium thermodynamics analysis, in which we found that the dielectric permittivity of the
17 LCO channel layer plays a key role in determining the electrostatic energy term that can significantly
18 affect the phase separation threshold concentration. This phase separation mechanism of the LCO
19 switching layer was suggested as the synergy effect of the SiO_x interfacial trapping mechanism,
20 resulting in the robust resistive switching behavior of the LCO NVRTs. With further scaling down
21 the devices, we believe our LCO synaptic transistor based on the percolation theory with decent
22 linearity and tunable lithium ionic migration dynamic will be is a promising NVRT that overcomes
23 the shortcomings in the conventional ECMs and VCMs.

24

25 **4. Experimental Section**

26 **4.1 Thin-film deposition and device fabrication**

27 Polycrystalline LCO was deposited on silicon wafer substrates by using RF magnetron sputtering.
28 The substrates used were highly doped p-type Si (100) with and without the amorphous silicon oxide
29 layer. The epitaxial LCO films with (003) and (104) preferred orientations were grown on the (111)
30 and (100) crystal planes of SrTiO₃ (STO) single-crystal substrates, respectively. All LCO samples
31 used in this study were deposited by using magnetron sputtering (Flow rate: 25 sccm O₂ + 25 sccm
32 Argon; Areal power density: 30 W cm⁻²; Deposition pressure: 0.5 Pa). The three-terminal nonvolatile
33 redox transistors (NVRT) were fabricated using the bottom-up synthesis method. Two parallel Au
34 electrodes were patterned on the substrate before the LCO channel layer was deposited, serving as

1 source and drain electrodes. After the epitaxial LCO films were deposited on STO substrates, a ~80
2 nm amorphous SiO_x was deposited by reactive sputtering to cover the whole LCO channel area.
3 Lastly, a gold top electrode was patterned on top of the SiO_x dielectric layer as the gate electrode.
4 (The device section SEM can be found in Figure S18a. The initial conductivity between the drain
5 and source electrodes was determined by scanning the *I-V* curves between -1 V to 1 V, as seen in
6 Figure S18b. The pulse width for the measurements of the synaptic plasticity of NVRTs was
7 optimized by applying a set pulse of -3V with different pulse widths, as seen in Figure S19a, b. To
8 avoid the over-extraction of lithium ions, the device was measured at the linear region rather than the
9 non-linear region and saturation region as shown in Figure S19c.

11 **4.2 Characterization**

12 The surface morphology of thin films was captured by scanning electron microscopy (Hitachi SU-
13 8010). Resistive state characterizations and modifications of the as-prepared films and NVRT
14 devices were performed by conductive AFM (c-AFM), and the surface potential was measured by
15 Kelvin probe force microscopy (Bruker Dimension Icon). To check the reversibility of the local
16 resistive switching effects, a c-AFM image was recorded after tip ‘writing’ using a concentric square
17 method under the bias voltages of ± 4 V. For the *I-V* measurements, the bias was applied to the
18 bottom Si substrate, the tip was grounded. The platinum silicide tip ($k \sim 3$ N m⁻¹, $f_0 \sim 75$ kHz) was used
19 for c-AFM measurements. The Boron-doped polycrystalline diamond tip (All-in-one-DD, cantilever
20 D, $k \sim 100$ N m⁻¹, $f_0 \sim 450$ kHz) was used in the 3D nano-tomography c-AFM measurements. The
21 scratching force applied to the LCO is about 10 μ N, corresponding to about 1 nm penetration depth
22 per frame. The tip-milling depth was determined by measuring the step height in scratched surface
23 topography, as seen in Supplementary Figure S20. The obtained 3D c-AFM current map was stacked
24 using ImageJ software. All AFM measurements were performed in a glove box filled with Ar with
25 <1ppm O₂ and H₂O concentration. The electrical and memristive switching properties were
26 measured using the Keithley 4200 semiconductor parameter analyzer and an arbitrary function
27 generator (AFG31000, Tektronix), respectively. X-ray photoelectron spectroscopy (ESCALAB 250
28 Xi, USA) was used for the detection of surface elemental composition. The permittivity of LCO-
29 (003) and LCO-(104) thin films are measured by impedance spectroscopy, using an AC excitation
30 signal with an amplitude of 50 mV, in the frequency region between 10⁻¹-10⁷ Hz. As shown in Figure
31 S17a, the Nb-doped conductive STO bottom electrode [40] and 100-nm-thick Au top electrode are
32 used as two metallic blocking electrodes, forming a metal-insulator-metal stack structure. Raman
33 spectroscopy (HORIBA Jobin Yvon Evolution, 532 nm excitation) was employed to determine the
34 LCO phase transition during the switching. Optical video of NVRT during a switching was recorded

1 using a stereoscopic microscope (Motic SMZ-168). The cross-section of the LCO/SiO_x/Si stack was
2 prepared by focusing ion beam cutting (ZEISS CROSSBEAM 550). The nanostructure of the cross-
3 section structure was examined by TEM (FEI Talos F200).

4 5 **Credit authorship contribution statement**

6 **Weijian Zhang:** Investigation, Formal analysis, Writing – original draft, Writing - review &
7 editing, Validation. **Yue Chen:** Formal analysis, Writing – original draft, Writing - review &
8 editing, Validation, supervision. **Chenjie Xu, Chun Lin, Jianming Tao, Yingbin Lin, Jiabin Li:**
9 Characterization. **Oleg V. Kolosov, Zhigao Huang:** Conceptualization, Supervision, Project
10 administration, Funding acquisition, Writing - review & editing.

11 12 **Conflict of Interest**

13 The authors declare no conflict of interest.

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