

1 **Degassing of volcanic extrusives on Mercury: Potential contributions to transient**
2 **atmospheres and buried polar deposits**

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21 **Abstract**

22

23 The surface of Mercury is dominated by extensive, widespread lava plains that formed early in
24 its history. The emplacement of these lavas was accompanied by the release of magmatic
25 volatiles, the bulk of which were lost to space via thermal escape and/or photodissociation. Here
26 we consider the fate of these erupted volatiles by quantifying the volumes of erupted volcanic
27 plains and estimating the associated masses of erupted volatiles. The concentrations and
28 speciation of volatiles in Mercury's magmas are not known with certainty at this time, so we
29 model a wide range of cases, based on existing experimental data and speciation models, at 3–7
30 $\log fO_2$ units below conditions determined by the iron-wüstite buffer. Cases range from relatively
31 low gas content scenarios (total exsolved gas mass of 9×10^{15} kg) to high gas content scenarios
32 (total exsolved gas = 5×10^{19} kg). We estimate that the average duration of a transient volcanic
33 atmosphere resulting from a single eruption would be between ~250 and ~210,000 years,
34 depending on the volume, degassed volatile content, and eruption rate of an individual eruption,
35 as well as the fO_2 conditions of the planet's interior. If a dense transient atmosphere was ever
36 surface-bound long enough for the released volatiles to be transported to and cold-trapped at
37 Mercury's polar regions, those trapped volatiles are predicted to be well-mixed with the regolith,
38 and at least 16 m beneath the surface given regolith gardening rates. These volatiles would have
39 a composition and age distinctly different from those of the H₂O-ice deposits observed at the
40 poles of Mercury today.

41 **Main text**

42

43 **1 Introduction**

44

45 Discovering that Mercury is volatile-bearing is one of the most exciting findings of the
46 MErcury Surface, Space ENvironment, GEochemistry, and Ranging (MESSENGER) mission
47 given the planet's heliocentric distance. For example, Mercury has abundant and widely
48 distributed pyroclastic deposits, indicative of volatile-rich eruptions (e.g., Head et al., 2009;
49 Kerber et al., 2009; Thomas et al., 2014a, 2014b; Goudge et al., 2014; Jozwiak et al., 2018;
50 Pajola et al., 2021). The size of these deposits suggests that they formed from an eruptive process
51 that was even more volatile-rich than the process that formed lunar pyroclastic deposits (Kerber
52 et al., 2009). Mercury also has intriguing hollows suggestive of sublimation, implying that
53 abundant volatiles are present beneath the surface (Blewett et al., 2013). Additionally,
54 MESSENGER X-Ray Spectrometer (XRS) and Gamma-Ray Spectrometer (GRS) measurements
55 indicate the widespread and abundant presence of K, S, Na, and Cl, and that the planet is
56 volatile-rich (e.g., Nittler et al. 2011; Peplowski et al. 2012, 2014; Evans et al. 2015).

57 Magmatism is a primary mechanism by which volatiles are transferred from the interiors
58 of planets to the surface (e.g., Moore, 1970; Greeley, 1987). For example, laboratory analyses of
59 Apollo samples revealed the presence of volatiles (H-, C-, S-, Cl-, and F-bearing compounds)
60 trapped in primitive magmas (e.g., Saal et al., 2008), suggesting that lunar volcanism has
61 effectively transferred volatiles from the interior to the surface of the Moon. As shown by the
62 lunar volcanic record (e.g., Head and Wilson, 2017), basaltic volcanic eruptions can be of two
63 types: effusive and explosive. Effusive eruptions involve the generation of mantle partial melts,

64 their ascent toward the surface in magma-filled cracks (dikes) driven by buoyancy and
65 overpressurization, and their eruption to the surface to form lava flows, accompanied by
66 degassing at the vent (e.g., Wilson and Head, 2018). Needham and Kring (2017) estimated that
67 $\sim 10^{16}$ kg of CO and S and $\sim 10^{14}$ kg of H₂O were released during the effusive formation of lunar
68 lava plains (the maria), with the bulk of volatiles being released during peak mare emplacement
69 at ~ 3.5 Gyr. The mare-erupted volatiles may have even been present in sufficient masses to
70 produce a transient atmosphere (Needham and Kring, 2017), although the duration of and
71 intervals between mare-forming eruptions have an important control on the density and lifetime
72 of any volcanically-derived atmosphere (Head et al., 2020). Interestingly, the production of a
73 substantially dense atmosphere would aid in the transport of volatiles to cold-trapping regions,
74 and volcanically-derived volatiles have been predicted to be cold-trapped at the lunar poles today
75 (e.g., Arnold, 1979; Crotts and Hummels, 2009; Needham and Kring, 2017).

76 Like the Moon, Mercury has been extensively resurfaced by large expanses of lava plains
77 (Fig. 1) (Head et al., 2009, 2011; Denevi et al., 2013; Byrne et al., 2016, 2018). If effusive
78 volcanism released substantial volatiles on the Moon during the production of the maria, it is
79 possible that substantial volatiles were also released on Mercury during the production of
80 volcanic plains, albeit with different chemical species and abundances (Nittler et al., 2011;
81 Zolotov, 2011). Given the more widespread presence of pyroclastic deposits (Kerber et al., 2009;
82 Thomas et al., 2014a, 2014b; Goudge et al., 2014; Jozwiak et al., 2018), and its generally higher
83 volatile contents relative to the Moon, Mercury may be even more likely to have had a transient
84 volcanically-derived atmosphere than the Moon. Here we seek to understand the magnitude and
85 potential fate of erupted volatiles. We analyze the volume of volcanic plains on Mercury and
86 estimate the amount of released gases, specifically for volatiles predicted to be indigenous to the

87 planet. We then estimate the typical eruption frequency and associated released magma volumes
88 and volatile masses. Finally, we discuss the potential fate of erupted volatiles and their
89 relationship to volatiles observed in high-latitude cold traps on Mercury (e.g., Lawrence et al.,
90 2013). The pre-eruptive volatile contents of Mercury's lavas are not well known, and existing
91 experimental data do not constrain speciation well at mercurian igneous conditions (low oxygen
92 fugacity, high S). Therefore, we examine a wide range of concentration and speciation cases in
93 order to provide bounds on the amount of volatiles released during effusive eruptions.

94

95 **2 Methods**

96

97 **2.1 Volume of erupted lavas on Mercury**

98 Figure 1 shows the distribution of volcanic plains on Mercury. Smooth plains (SPs) are
99 widely distributed across the planet and were originally suggested to be volcanic due to their
100 tendency to embay other features and pond in topographic lows (e.g., Murray et al., 1974; Strom
101 et al., 1975). Observations of these plains with MESSENGER later revealed a variety of volcanic
102 landforms and morphologies (Head et al., 2009, 2011; Byrne et al., 2016; Denevi et al., 2013),
103 and the SPs are interpreted to be from effusive, flood-mode eruptions (Head et al., 2009, 2011;
104 Byrne et al., 2016, 2018). Denevi et al. (2013) found multiple lines of evidence (e.g., flooding
105 and embayment relationships, color properties, and relationships with volcanic vents) indicating
106 a volcanic origin for >65% of the SPs. As a conservative estimate, we assume in this study the
107 total surface area of volcanic SPs (VSPs) to be 65% of the total mapped SPs surface area.

108 Figure 1 also delineates intercrater plains (ICPs), the most spatially dominant unit on
109 Mercury. The ICPs are more heavily cratered than the SPs, suggesting they are relatively older

110 (Whitten et al., 2014). They were originally suggested to be volcanic because they occupy large
111 volumes and because there is no obvious impact basin from which they may have been sourced
112 as ejecta (e.g., Murray et al., 1974; Strom et al., 1975). The color properties of ICPs and SPs are
113 similar (Murchie et al., 2015) and many of the contacts between these units exist as gradational
114 boundaries (Whitten et al., 2014). These observations have led to the interpretation that ICPs are
115 volcanic, emplaced in multiple, large-volume effusive flows (similar to the SPs), and have since
116 been modified by impact cratering (Denevi et al., 2013; Whitten et al., 2014; Byrne et al., 2018).
117 However, the origin of some ICPs is still debated and so we treat two different scenarios for
118 estimating the total volume of effused lavas: first considering only VSPs and then also including
119 the ICPs. In the models where ICPs are included, we assume that an additional ~40% of the
120 planet's surface is covered by volcanically-derived ICPs.

121 For these two scenarios, we estimate the volume of plains by multiplying the surface area
122 of each plains deposit by its estimated thickness, assuming VSPs are ~25% and volcanic ICPs
123 are 40% of the planet's total surface area. Previous analyses indicate that the majority of volcanic
124 plains on Mercury are between 0.5 km and 4 km thick (Byrne et al., 2018 and references
125 therein). For plains that do not have previously resolved thicknesses, we estimate their
126 thicknesses for three cases: (1) low volume (assuming a plains thickness, T , of 0.1 km), (2)
127 intermediate volume ($T=1$ km), and (3) high volume ($T=4$ km).

128 In this analysis, we do not include the effect of explosively-formed pyroclastic deposits,
129 which are relatively small and extend well into the post-regional plains part of Mercury's history
130 (Thomas et al., 2014a, 2014b; Goudge et al., 2014; Jozwiak et al., 2018). Instead, we focus on
131 the more widespread, and volumetrically-important, effusive volcanic plains.

132

133 **2.2 Mercury's magmatic volatile content**

134

135 We aim to provide upper and lower bounds for the amount of gas released on Mercury
136 during extrusive plains-forming volcanism by exploring a wide parameter space of degassed
137 contents and oxygen fugacities. In contrast to the Moon and Mars, we do not currently have any
138 known samples of Mercury, which could help in the determination of its volatile content. In lieu
139 of samples, XRS and GRS measurements of Mercury's surface and near-surface can provide
140 some insight. These measurements - indicating abundances of K (Peplowski et al., 2012), C
141 (Murchie et al., 2015), S (Nittler et al., 2011), Na (Peplowski et al., 2014), and Cl (Evans et al.,
142 2015) - have been used widely in experiments and petrologic modeling to infer the melting
143 conditions and mantle source compositions for surface lavas (e.g., Namur et al., 2016). The
144 measurements of elevated S abundances (up to 4 wt. % at the surface; Nittler et al., 2011) also
145 allow for the calculation of the planet's oxygen fugacity (fO_2), which is estimated to be between
146 IW-3 and IW-7 (where IW-3 is 3 log units below the iron-wüstite oxygen buffer), indicating that
147 Mercury is the most reduced terrestrial planet (Zolotov, 2011; Zolotov et al., 2013; Namur et al.,
148 2016; McCubbin et al., 2017).

149 Our primary approach to estimating volcanic degassing rates uses experimentally-
150 measured solubilities of volatiles at reducing conditions and at pressures relevant to mantle
151 magma-generation regions (1–3 GPa). Thus, we are assuming that the melts are sourced from the
152 mantle (90–270 km), and not crustal melting or direct impact heating. We use the melting
153 pressures (rather than eruptive pressures), as this sets an upper limit to the amount of volatiles
154 that can be mobilized by melting and delivered to the surface. Currently, there are no published
155 experimental data that meet all of the desired criteria: data derived from experiments that vary

156 fO_2 values from IW-3 to IW-7, reproduce surface to mantle pressures appropriate to Mercury,
157 include all volatiles (particularly S), and reproduce a major elemental composition similar to that
158 of NSP and ICP lavas. Therefore, values in Table 1 are gathered from a range of experimental
159 studies. These studies sometimes vary in magma composition and experimental conditions, and
160 so are unfortunately not self-consistent (e.g., some values are from Fe-bearing melts, others are
161 Fe-free; some melts have S, others do not). In some cases, the values had to be extrapolated to
162 IW-7. The high-pressure solubilities set the maximum concentration of volatiles in magmas
163 delivered to the surface, equivalent to assuming that the melts were volatile-saturated during
164 melting and that there was no degassing during transport, as typically occurs on Earth. Both
165 assumptions are not likely to be true, and so the actual concentrations in the pre-eruptive magmas
166 are likely to be lower. We then estimate how much of those volatiles are actually degassed to the
167 atmosphere, versus remaining in the lava as unexsolved volatiles, being trapped in vesicles, or
168 being precipitated as solids. How much of the volatiles actually degasses depends on numerous
169 factors, including how quickly the magmas ascend, how much they are oxidized (or reduced) by
170 crustal interaction, how pressure affects solubility, and what solid phases would precipitate from
171 the magma and/or condense from the volcanic gas. Since none of these factors are well
172 constrained, we make very simple assumptions about the fraction of gas released to estimate
173 high- and low-gas cases. These estimates are meant to place upper and lower bounds on
174 degassing, but given the current state of knowledge, are unlikely to be accurate as to the exact
175 composition or speciation of the degassed volatiles.

176 To supplement this approach, we also calculate degassing scenarios for specific cases of
177 volatile speciation presented in Zolotov (2011) (Supporting Information). For 45/48 of these
178 cases, the overall amounts of degassed volatiles fall within the range of our high- and low-gas

179 models, though the details of the gas composition differ (Tables S1–S3). The major conclusions
180 of our work do not change when using these different models.

181 At Mercury’s reducing conditions, S has high concentrations and affects the partitioning
182 and speciation of other volatiles in the melts (Zolotov, 2011; Zolotov et al., 2013; Armstrong et
183 al., 2015; Namur et al., 2016; Anzures et al., 2020). Highly reducing conditions promote high S
184 solubility via bonding with Ca, Mg, and Na in the melt structure, and this solubility has been
185 measured in various experimental studies (e.g., McCoy et al., 1999; Namur et al., 2016; Anzures
186 et al., 2020). At the high pressures (up to 4 GPa) and temperatures (1200–1750 °C) predicted for
187 mercurian mantle conditions, S solubility in a sulfide-saturated experimental melt increases from
188 ~1 wt. % to ~7 wt. % as fO_2 decreases from IW-3 to IW-7 (Namur et al., 2016). At present, how
189 much of this S will degas from the magma is not known. A significant factor in this uncertainty
190 is how much the lavas are oxidized (or reduced) as they transit the crust (Zolotov, 2011).
191 Oxidation will greatly lower S solubility, and thus promote abundant S degassing. The high S
192 that remains in the lavas, as measured by XRS, suggests that a significant amount of the
193 dissolved S is trapped in sulfides, indicating limited (or lack of) crustal oxidation of the magma.
194 We consider two cases: Model A, a low-gas scenario in which only 1% of the S content degasses
195 (the rest forming sulfides), and Model B, a high-gas scenario in which 10% of the S content
196 degasses. Given the uncertainties about Mercury’s interior, our intention is to explore a broad
197 parameter space with different models. Table 1 displays the various S contents explored here,
198 and S content in the melts are assumed to be the experimentally-measured sulfur concentrations
199 at sulfide saturation (SCSS). Note that in reality, volatile abundances are not simply determined
200 by solubility limits, but also depend on bulk abundances. Our goal here is provide firm upper and
201 lower bounds on the problem, and the solubility limits are a good upper bound.

202 For Mercury's interior, a major consideration is how elevated S abundances influence the
203 melt, given that S saturation affects the bonding environment of other species, particularly in a
204 low-O system. Measurements of sulfide-saturated experimental melts indicate that there is an
205 increase in Cl solubility with decreasing fO_2 (e.g., Zolotov, 2011; Evans et al., 2015; Anzures et
206 al., 2020). From these experiments, we estimate that Cl may be present in mercurian melts at
207 abundances between 10 ppm and 800 ppm for fO_2 values between IW-3 and IW-7, respectively
208 (Table 1). Again, it is not yet understood how much of any specific volatile species is released
209 during eruptions on Mercury, so we estimate degassed contents for Cl (and all other non-S
210 volatiles) as 10% for our low-gas Model A and 100% for our high-gas Model B.

211 C solubility has also been measured in various experimental studies. For example, Ardia
212 et al. (2013) measured dissolved C in synthetic basaltic melts at pressures up to 3 GPa,
213 temperatures between 1400 and 1450 °C, and fO_2 between IW-3 and IW-5. More recently, Dalou
214 et al. (2019) measured dissolved C in reduced basaltic glasses at pressures up to 3 GPa and
215 temperatures between 1400 and 1600 °C. Both studies provide similar estimates for dissolved C
216 at IW-3 (100 ppm), which we use here (Table 1), although they do not explore lower fO_2
217 conditions. We estimate C at IW-7 as 10 ppm (Table 1) using the results of Armstrong et al.
218 (2015), who experimentally studied (1.2 GPa, 1400 °C) the solubility and speciation of dissolved
219 volatiles in mafic melts specifically for C-O-H-N species. Notably, the experiments of
220 Armstrong et al. (2015) included the presence of H_2O , which is not predicted to be present in
221 Mercury's interior (Nittler et al., 2011; Zolotov, 2011). The values used in our models (Table 1)
222 reflect a decrease in C solubility with decreasing fO_2 , but we note that recent experiments on
223 sulfide-saturated systems suggest that C solubility increases with decreasing fO_2 (Anzures et al.,
224 2020) and future work can help refine our models presented here.

225 Armstrong et al. (2015) also predicted dissolved N- and H-bearing melt species to be
226 stable at high pressure and reducing conditions. N solubility was measured by Libourel et al.
227 (2003) in S- and Fe-free basaltic melts at a range of fO_2 values between IW-1.3 and IW-8.3. We
228 estimate the maximal possible N content for IW-3 and IW-7 from these experiments (Table 1).
229 Maximal H contents at IW-3 are estimated from Dalou et al. (2019), and extrapolated down to
230 IW-7 from Ardia et al. (2013) (Table 1).

231 Finally, we use estimates of C to estimate O content; assuming that all O in gas is present
232 as CO or COS (Zolotov, 2011), we calculate the elemental abundance of O from the estimated
233 moles of C released (Table 1).

234 We next estimate the total mass of volatiles delivered to the surface during the formation
235 of volcanic plains on Mercury. We use the terms “delivered to the surface” and “released” to
236 describe all of the volatiles that reach the surface, including both the volatiles that were
237 outgassed to the atmosphere during initial dike penetration to the surface and
238 hawaiian/strombolian pyroclastic activity and those that remain trapped in the solidified magma
239 (e.g., vesicles, etc.) and diffuse out over much longer time scales (e.g., Wilson and Head, 2017,
240 2018). With our estimates of species released during mercurian eruptions (Table 1), we calculate
241 the total mass of released volatiles, G , for each volcanic plains deposit from

$$242 \quad G = V \times \rho_m \times n \quad (\text{Eq. 1}),$$

243 where V is the total erupted volume, ρ_m is the bulk density of volcanic deposits on Mercury
244 ($\sim 3014 \text{ kg m}^{-3}$; Padovan et al., 2015), and n is the estimated mass fraction of volatiles released
245 from magmas delivered to the surface (Table 1).

246 We stress that Table 1 represents our estimates of the possible volatiles that were released
247 during plains-forming eruptions on Mercury, but the actual volatile species and their degassed

248 content remain unknown. We cannot emphasize enough that our approach is to bound the
249 problem, and then present the different possibilities within these bounds. We examine a wide
250 parameter space of degassed contents and oxygen fugacities in order to derive reasonable
251 estimates for the amount of gas released on Mercury during plains-forming eruptions, and
252 demonstrate how sensitive the models are to the gas content of the magmas. Future *in situ*
253 analyses, as well as new experimental studies, are essential for more accurately describing the
254 volatile speciation and content of mercurian lavas.

255

256 **3 Results**

257

258 **3.1 Total volume of erupted lavas**

259

260 The total volume of erupted lavas estimated for Mercury's VSPs ranges between $\sim 1.1 \times$
261 10^7 and 4.6×10^7 km³ depending on the estimated thicknesses of the plains (Fig. 2), equating to a
262 global equivalent layer (GEL) of lava that is 0.15–0.61 km thick. The total amount of erupted
263 lavas is likely to have been higher given that not all volcanic deposits are preserved at the surface
264 today (Whitten et al., 2014). Assuming ICPs are also volcanic flows (Murray et al., 1974; Strom
265 et al., 1975; Denevi et al., 2013; Whitten et al., 2014; Byrne et al., 2018), then the estimated
266 volume of erupted lavas increases by up to an order of magnitude; the total erupted VSP and ICP
267 volume is estimated to be $\sim 1.4 \times 10^7$ – 1.7×10^8 km³ (~ 0.19 – 2.3 km GEL) (Fig. 2).

268 Our conservative estimates of the total volume of erupted VSPs on Mercury are an order
269 of magnitude higher than that most recently estimated for the Moon by Needham and Kring
270 ($\sim 8.9 \times 10^6$ km³; 0.23 km GEL), and similar to the $\sim 10^7$ km³ (0.26 km GEL) estimated

271 previously (Head and Wilson, 1992; Evans et al., 2016). The total volume of erupted lavas on
272 Mercury is up to 2 orders of magnitude higher than that on the Moon if the ICPs are indeed
273 volcanic. Overall, SP volcanism on Mercury appears to have largely begun to wane prior to ~ 3.5
274 Ga (Byrne et al., 2016; see Supporting Information), when mare volcanism may have been most
275 active on the Moon (Hiesinger et al., 2011).

276 Uncertainties in the erupted volume stem from uncertainties associated with the estimates
277 of volcanic plains thicknesses. While the thicknesses of some individual plains units have been
278 resolved from stratigraphy (Byrne et al., 2018 and references therein), the thicknesses of others
279 have not been analyzed in detail. Individual plains units are not predicted to be uniform in
280 thickness; they vary in thickness as a function of eruption conditions, magmatic flux, viscosity,
281 and geologic setting (Wilson and Head, 2008, 2017; Head et al., 2009, 2011; Denevi et al., 2013;
282 Whitten and Head, 2013). To account for these uncertainties, as discussed in Section 2.1, we
283 have computed the volume for three different volume scenarios.

284

285 **3.2 Mass of erupted volatiles**

286

287 We calculate the overall mass of degassed volatiles on Mercury (Table 2; Fig. 3) using
288 estimates of total erupted lavas for the low-, intermediate-, and high-volume cases (Fig. 2). First
289 we consider only the volatiles released during VSP-forming eruptions. Assuming an upper fO_2 of
290 IW-3 in the intermediate-volume case, we estimate that $\sim 2.0 \times 10^{16}$ kg of volatiles were released
291 during the formation of VSPs in a low-gas scenario (Model A), and $\sim 2.0 \times 10^{17}$ kg in a high-gas
292 scenario (Model B). Assuming a lower fO_2 of IW-7 in the intermediate-volume case, we estimate
293 that $\sim 7.9 \times 10^{17}$ kg of volatiles were released during the formation of SPs in Model A, and $\sim 7.9 \times$

294 10^{18} kg in Model B. Our estimates for S-, C-, H-, N-, Cl-, and O-bearing species are shown in
295 Table 2.

296 If the ICPs are indeed volcanic, then the estimated masses of released volatiles would be
297 even greater. For example, for the intermediate-volume case and an fO_2 of IW-3, $\sim 4.5 \times 10^{16}$ kg
298 volatiles were released in Model A and $\sim 4.5 \times 10^{17}$ kg in Model B when including the
299 contributions from both VSP-forming and ICP-forming eruptions. When assuming a lower fO_2 of
300 IW-7, these estimates increase to $\sim 1.8 \times 10^{18}$ kg for Model A and $\sim 1.8 \times 10^{19}$ kg for Model B.

301 We also complete these calculations for specific mercurian degassing models presented
302 by Zolotov (2011) in the Supporting Information. We find that the major conclusions presented
303 here do not change when considering these specific models, given that our models are much
304 more sensitive to the total amount of erupted gas as opposed to the specific species.

305 The difference in the abundance of released volatiles between Mercury and the Moon is
306 largely a function of the total volume of erupted lavas, which is estimated to be greater on
307 Mercury (Section 3.1). Overall, the large volumes of erupted lavas on the surface imply that the
308 VSP-forming eruptions released at least $\sim 9.3 \times 10^{15}$ kg of volatiles over time, representing a
309 conservative estimate under IW-3 conditions in Model A. While Model A represents a
310 conservative low-gas scenario (Table 1), the mass of erupted volatiles would be lower for more
311 undersaturated magmas.

312

313 **4 The potential for volcanically-derived transient atmospheres**

314

315 Although abundant volatiles are predicted to have been released during the formation of
316 lava plains on Mercury (Fig. 3) and the Moon (Wilson and Head, 2018), the fate of such volatiles

317 is an important open question. It has been proposed that large amounts of released volatiles on
318 the Moon could have produced a transient atmosphere that aided in the transport of volatiles to
319 cold-trapping regions (Needham and Kring, 2017). Critically, the possibility of such an
320 atmosphere is dependent not only on volatile abundance and speciation, but also on degassing
321 dynamics, eruption lifetimes, atmospheric gas dynamics, and atmospheric loss rates.

322 The Moon, for example, was very volcanically active between ~4 and 3.5 Ga on geologic
323 timescales, but individual eruptions occurred on relatively short (~100 day) timescales (Wilson
324 and Head, 2017, 2018), implying that there were significant periods of inactivity between
325 eruptions. Head et al. (2020) estimated this repose time between average lunar eruptions to be
326 ~13,000–40,000 years.

327 Here, we perform similar calculations for Mercury to solve for the average time interval
328 between average eruptions, τ_i , from

329
$$\tau_i = \frac{\tau_T}{V_T/V_e} \quad (\text{Eq. 2}),$$

330 where τ_T is the timespan of volcanic activity (estimated to be ~1 Gyr; Byrne et al., 2018), V_T is
331 the total erupted volume of plains (estimated to be $\sim 10^7$ km³ for VSPs and $\sim 10^8$ km³ for VSPs +
332 ICPs in Section 3.1), and V_e is the average volume of a single eruption (estimated to be ~200–
333 400 km³ from the heights and lateral extents of lobate flow fronts; Wilson and Head, 2008). If
334 we assume that volcanic activity was relatively constant over τ_T , then τ_i is estimated to be
335 between ~20,000 and 40,000 years for VSP-forming eruptions, but between ~2,000 and 4,000
336 years when also considering ICP-forming eruptions. As with the Moon, it is possible that the
337 frequency of eruptions was higher at the start of τ_T and then decreased through time (see
338 Supporting Information), given the thermal and magmatic evolution of the planet (e.g., Wilson
339 and Head, 2008; 2017).

340 Released volatiles must remain in the atmosphere for sufficient durations in order to
341 migrate and become cold-trapped before they are photochemically destroyed or lost to space
342 (Section 5). In order to determine the average duration of volcanically-derived transient
343 atmospheres on Mercury, we first calculate the average eruption duration, τ_e , from

$$344 \quad \tau_e = \frac{V_e}{F_l} \quad (\text{Eq. 3}),$$

345 where V_e the volume of erupted lava and F_l is the flux of erupted lava. For typical eruptions on
346 Mercury, V_e is between 200 and 400 km³ and F_l is between 10³ and 10⁷ m³ s⁻¹ (Wilson and Head,
347 2008); thus, τ_e is estimated to be between ~2,300 and 4,600 days for low-flux eruptions, and < 1
348 day for extremely high-flux eruptions (Table 3).

349 From τ_e , we calculate the flux of erupted gas, F_g , using

$$350 \quad F_g = \frac{G}{\tau_e} \quad (\text{Eq. 4}),$$

351 where G is the mass of gas released for an average eruption. While we estimate that the total
352 mass of erupted volatiles on Mercury may be between ~10¹⁵ and 10¹⁹ kg (Section 3.2), it will of
353 course be substantially less for individual eruptions. Wilson and Head (2008) estimated that the
354 average volume of a single eruption is between ~200 and 400 km³, which implies that the total
355 gas released, G , for average mercurian eruptions is ~1.6 × 10¹¹–3.3 × 10¹² kg assuming an $f\text{O}_2$ of
356 IW-3 and ~6.5 × 10¹²–1.3 × 10¹⁴ kg assuming an $f\text{O}_2$ of IW-7. With these values, F_g is estimated
357 to be ~830–8.3 × 10⁷ kg s⁻¹ for IW-3, and ~33,000–3.3 × 10⁹ kg s⁻¹ for IW-7 (Table 3).

358 We next estimate the scale height, H , of a volcanically-derived atmosphere to be 15 km
359 from

$$360 \quad H = \frac{Q \times T}{m \times g} \quad (\text{Eq. 5}),$$

361 where Q is the universal gas constant ($8.314 \text{ J mol}^{-1} \text{ K}^{-1}$), T is Mercury's mean surface
 362 temperature ($\sim 440 \text{ K}$), m is the mean molecular mass of erupted volatiles ($\sim 64.1 \text{ kg kmol}^{-1}$
 363 assuming an S_2 atmosphere), and g is the gravitational acceleration at Mercury's surface (3.7 m
 364 s^{-2}). The scale height is dependent on temperature, and thus varies with location (latitude or time-
 365 of-day). For example, H is between ~ 4 and 25 km depending on whether the eruption occurred
 366 on the nightside or dayside, respectively, due to the large differences in diurnal surface
 367 temperature ($\Delta T \approx 600 \text{ K}$). The temperature may also vary during the eruption lifetime due to the
 368 greenhouse effect, or through geologic time due to changes in solar luminosity. Note that
 369 because of Mercury's greater surface gravity and a predicted S-based atmosphere (as opposed to
 370 a CO-H₂O dominated lunar atmosphere), H is less than the estimated H for a lunar volcanically-
 371 derived atmosphere (38 km ; Head et al., 2020), which would also vary depending on the time
 372 and location of the eruption ($\Delta T \approx 300 \text{ K}$).

373 We use H (estimated from the mean surface temperature) to solve for the surface density
 374 of a volcanically-derived atmosphere, ρ_s , using

$$375 \quad \rho_s = \frac{G}{4 \pi R^2 \times H} \quad (\text{Eq. 6}),$$

376 where R is the planet's radius (2440 km). For IW-3, ρ_s is estimated to be $\sim 1.4 \times 10^{-7}$ – 2.9×10^{-6}
 377 kg m^{-3} and for IW-7, ρ_s is estimated to be $\sim 5.6 \times 10^{-6}$ – $1.1 \times 10^{-4} \text{ kg m}^{-3}$ (Table 3).

378 The pressure of a collisionally-supported atmosphere, P_s , is calculated using

$$379 \quad P_s = \rho_s \times g \times H \quad (\text{Eq. 7}).$$

380 We estimate P_s to be ~ 0.01 – 0.2 Pa for IW-3 and ~ 0.3 – 6 Pa for IW-7 (Table 3), although this
 381 likely represents a maximum local surface pressure that quickly decays in time and space.

382 Finally, the flux of gas released during individual eruptions can be compared with the
383 loss rate of gases from the atmosphere. We estimate the total duration of a volcanically-derived
384 atmosphere, τ_d , from

$$385 \quad \tau_d = \frac{G}{F_{esc}} \quad (\text{Eq. 8}).$$

386 (Note, see Houghton (2002) for more details on Eqs. 5–8). To date, we know little about the rate
387 of atmospheric escape, F_{esc} , on Mercury, although various estimates have been made for the
388 Moon. For example, Vondrak (1974) estimated a thermal escape of 60 kg s^{-1} , and this value was
389 adopted by both Needham and Kring (2017) and Head et al. (2020) when evaluating lifetimes of
390 potential volcanically-derived transient atmospheres on the Moon. However, the thermal escape
391 flux exponentially depends on molecular mass such that larger molecules escape more slowly
392 (e.g., Tucker et al., 2021), and the molecular mass of the atmosphere was not considered by these
393 groups. For a heavier S-based atmosphere and due to the higher gravity on Mercury, the escape
394 rate would be smaller on Mercury than on the Moon. Tucker et al. (2021) found that thermal
395 escape processes are less important in a collisional atmosphere, and sputtering and
396 photodissociation dominate, calculating a mass loss rate of 17 kg s^{-1} for a CO collisional
397 atmosphere. These processes may be stronger at Mercury due to its proximity to the Sun. In
398 absence of Mercury-specific studies, these lunar-specific values provide a first-order
399 approximation for our analysis, and we approximate F_{esc} as 20 kg s^{-1} . Solving for τ_d implies that
400 a typical volcanically-derived atmosphere would decay within $\sim 250\text{--}210,000$ years (Table 3).
401 Note that using an F_{esc} of 60 kg s^{-1} would decrease these estimates to $\sim 85\text{--}69,000$ years.

402 Above we estimated the average time between eruptions on Mercury, τ_i , to be $\sim 20,000\text{--}$
403 $40,000$ years, although possibly as short as $\sim 2,000\text{--}4,000$ years if the ICPs (as volcanic origin)
404 are included. Comparisons between τ_d and τ_i indicate the following:

- 405 1. If VSPs are the only volcanic plains on Mercury, then a transient volcanic atmosphere
406 would likely have dissipated prior to a subsequent eruption (Fig. 4). It is possible that
407 transient volcanic atmospheres co-existed in low $f\text{O}_2$ (IW-7) and high-gas (Model B)
408 conditions given that the predicted lifetime of volcanically-derived atmospheres would
409 have been longer (Fig. 4).
- 410 2. In the more likely case that ICPs, like the VSPs, are volcanic, the majority of our models
411 suggest that the average duration of transient, volcanic atmospheres would have exceeded
412 the average duration between individual eruptions, suggesting that the frequency of
413 eruptions may have resulted in the combination of denser, longer-lived atmospheres (Fig.
414 4). Only for the low-gas, high $f\text{O}_2$ scenarios would this not be the case (Table 3).

415 These implications are depicted in Figure 4, where τ_d and τ_i are compared for low-volume
416 scenarios ($V_e=200 \text{ km}^3$).

417 The average surface pressure and density of transient atmospheres vary in each case and
418 are highest for longer-lived atmospheres (Table 3). These properties are expected to decrease
419 through time as an atmosphere decays, although may increase if individual eruptions overlap in
420 time (denoted by the less transparent portions of plots in Figure 4). With the generation of a
421 sufficiently dense atmosphere, the migration of volatiles is no longer governed by ballistic
422 hopping, but instead by the collisions between molecules (Stewart et al., 2011; Prem et al.,
423 2015). While such an atmosphere remains gravitationally bound, its lower layers are shielded
424 from photodestruction (the primary loss mechanism for volatiles on airless bodies), allowing
425 more time for molecules to reach permanent cold traps (Stewart et al., 2011; Prem et al., 2015).
426 Determining the ultimate fate of volatiles in transiently dense atmospheres on nominally airless
427 bodies requires detailed modeling that depends on eruption dynamics, atmospheric loss rates, gas

428 dynamics, and compositional evolution in an atmosphere that is evolving on geologically rapid
429 timescales (e.g., Prem et al., 2015, 2019). Such modeling will be an important next step in
430 assessing how volcanic eruptions altered Mercury's atmospheric density and composition, and
431 how the collisional dynamics of transiently dense atmospheres affected the transport and
432 retention of interior volatiles on Mercury.

433

434 **5 Buried volatiles at the poles of Mercury?**

435

436 Needham and Kring (2017) point out that only a fraction of the predicted water released
437 from mare formation ($\sim 10^{14}$ kg) is needed to account for all of the remotely sensed water at the
438 lunar poles ($\sim 10^{11}$ kg), and if mare volcanism did deliver ice to the lunar poles, it is likely to be
439 buried in the subsurface. We find that Mercury released even greater volumes of volcanic
440 material over time than the Moon (Fig. 2), and therefore greater amounts of volatiles were likely
441 released (Fig. 3). Is it possible that some fraction of erupted volatiles was deposited in polar cold
442 traps on Mercury?

443 Interestingly, K, Na, and Cl appear to have higher concentrations in Mercury's northern
444 terrain, and it is possible that these volatiles may be released from warmer, equatorial terrains
445 and lost to space or redistributed to cooler terrains (Peplowski et al., 2012; Nittler et al., 2018).
446 However, not all volatiles on Mercury appear to be spatially correlated with temperature,
447 including S, which is concentrated at mid-equatorial latitudes (Weider et al., 2015; Nittler et al.,
448 2018). The northern enhancements of K, Na, and Cl may instead be related to compositionally
449 distinct magmas, as they show a general correlation with the large Northern SPs unit (Peplowski
450 et al., 2015; Evans et al., 2015). However, if temperature-induced mobilization of volatiles has

451 occurred on Mercury, then the northern concentrations of K, Na, or Cl may be examples of
452 volatiles migrating to the poles, similar to what could have happened with outgassed species
453 early on in mercurian history. A critical test of the competing origin hypotheses for the northern
454 volatile anomalies will be BepiColombo (Benkhoff et al., 2010) measuring the abundance of K,
455 Na, and Cl at the south polar region of Mercury, where there is no large volcanic plains unit
456 comparable to the one seen at the north pole, but there is a similar thermal environment that
457 would allow for the cold-trapping of volatiles.

458 In addition to K, Na, and Cl, there is an increase in H at Mercury's north polar region,
459 where the detection of enhanced neutron suppression (Lawrence et al., 2013) and reflectance
460 measurements (Neumann et al., 2013; Deutsch et al., 2017) are consistent with the presence of
461 nearly pure H₂O deposits cold-trapped at the poles. These H-rich deposits have highly reflective
462 radar properties that are also indicative of a nearly pure H₂O composition, with <5% silicates by
463 volume (Butler et al., 1993). The ice deposits have not been covered by regolith gardening
464 processes (Butler et al., 1993; Crider and Killen, 2005) and have distinct reflectance properties
465 (Chabot et al., 2016), sharp geologic contacts (Chabot et al., 2016), and crater spatial densities
466 (Deutsch et al., 2019) suggestive of geologically young ages <200 Myr (Crider and Killen, 2005;
467 Lawrence et al., 2013; Deutsch et al., 2019).

468 The composition and age of these surface ice deposits indicate that they are not derived
469 from volcanism. At mantle pressures and Mercury's extremely reducing conditions, H₂O is not
470 predicted to be present in magmas (e.g., Nittler et al., 2011; Zolotov, 2011) and the very young
471 surface ages of ice deposits (Deutsch et al., 2019) are inconsistent with the timing of Mercury's
472 volcanic activity. The H₂O ices observed at the surface of the poles are more likely to have been

473 delivered by an external, cometary impactor (e.g., Butler et al., 1993; Chabot et al., 2016; Ernst
474 et al., 2018; Deutsch et al., 2019).

475 However, given our analysis here, it is important to consider the possibility that volatiles
476 other than H₂O were released from Mercury's interior and contributed some fraction of materials
477 to polar cold traps. These volatiles would be distinctly different from the H₂O polar deposits
478 observed today, raising the question: Where would they be and how could they be identified?

479 Any volatiles that are deposited on the surface of Mercury (originating in the interior or
480 externally delivered) must face the space weathering environment, including strong irradiation
481 and thermal stresses (e.g., McCord and Clark, 1979; Domingue et al., 2014) and high-velocity
482 impactors (e.g., Borin et al., 2009; Domingue et al., 2014). While the elevated elemental
483 abundances of volatiles on Mercury (Nittler et al., 2011; Zolotov, 2011; Peplowski et al., 2011,
484 2012; Evans et al., 2012) suggest that perhaps the regolith is effective at trapping volatiles, the
485 high temperatures of non-polar surfaces are a major limiting factor governing volatile
486 sequestration and retention. If volatiles do succumb to sequestration in the polar region, they can
487 subsequently be destroyed through impact vaporization and melting, but also preserved by ejecta.

488 If any individual, volcanically-derived atmosphere was sufficiently dense and long-
489 lasting, then perhaps some fraction of volatiles migrated to and became cold-trapped at the poles
490 before the atmosphere completely attenuated. We find that such atmospheres were probably
491 uncommon, but may have been produced under certain extreme conditions (high eruption
492 volume, high lava flux, low oxygen fugacity; Table 3). The location of an eruption also has some
493 control over the distribution of migrated volatiles (Prem et al., 2015, 2019). While the Northern
494 SPs (Fig. 1) coincide with today's polar cold traps, these cold traps largely post-date the
495 formation of the SPs, so perhaps older south polar cold traps would provide a better record of

496 volcanically-derived volatiles. If we assume that a layer of volcanically-derived volatiles was
497 ever cold-trapped at Mercury's poles during one of these more extreme volcanic events, then we
498 can use a regolith gardening rate to predict the depth at which they would be concentrated. For
499 example, if volatiles were delivered 3.8 Ga (before major plains-forming volcanism began to
500 wane), then the regolith gardening rate of $\sim 0.43 \text{ cm Myr}^{-1}$ derived by Crider and Killen (2005)
501 suggests that the peak concentration of remnant volatiles would be at $\sim 16 \text{ m}$ in the subsurface
502 today. We consider this to be a minimum estimate, given that the regolith gardening rate derived
503 by Crider and Killen (2005) is for the recent past ($< 100 \text{ Ma}$) and the volcanism we analyze here
504 was most active before $\sim 3.5 \text{ Ga}$, when impact rates were much higher (e.g., Le Feuvre and
505 Wieczorek, 2011). The volatiles would be well-mixed with the regolith and have a highly
506 heterogeneous spatial distribution due to the stochastic impact gardening process (e.g., Crider
507 and Killen, 2005).

508 The detection of volcanically-derived volatiles (distinct from impact-derived volatiles in
509 composition) would provide important new insight into Mercury's volcanic history, interior
510 geochemical conditions, and exospheric/atmospheric evolution. Indigenous volatiles well-mixed
511 with the regolith at least 16 m below the surface would not have been detectable by
512 MESSENGER, nor would it be with BepiColombo, given the sensing depths of each spacecraft's
513 spectrometers. But future analyses of young crater ejecta might provide insight into the presence
514 of buried volatiles. BepiColombo (Benkhoff et al., 2010) can also help refine many of the
515 uncertainties associated with our current models. For example, analyses of plains morphometry
516 can improve estimates of the volume of erupted lavas, and studying the morphology of the ICPs
517 would be helpful in determining the extent to which they are volcanic. Higher-resolution images
518 could also improve model production functions and thus the emplacement ages of plains units.

519 Furthermore, more detailed information on the numerous pyroclastic deposits and associated
520 vents could increase our knowledge about whether they represent shallow concentrations of
521 volatiles prior to eruption (Jozwiak et al., 2018), and how they compare to the effusive volcanic
522 deposits documented here. Additional elemental measurements (particularly of S) can help refine
523 geochemical models used to estimate the oxygen fugacity of Mercury's interior. Ultimately,
524 sample analyses of Mercury's volcanic plains are needed to directly measure the speciation and
525 volatile content of lavas ([https://science.nasa.gov/science-red/s3fs-](https://science.nasa.gov/science-red/s3fs-public/atoms/files/Mercury%20Lander.pdf)
526 [public/atoms/files/Mercury%20Lander.pdf](https://science.nasa.gov/science-red/s3fs-public/atoms/files/Mercury%20Lander.pdf)).

527

528 **6 Conclusion**

529

530 Here we provide the first comprehensive estimates of the amounts of predicted volatiles
531 (S, C, H, N, Cl, and O) released on Mercury via extrusive volcanic plains-forming eruptions
532 using a variety of fO_2 conditions, degassing scenarios, and erupted volume estimates. We
533 estimate that $\sim 1.1\text{--}4.6 \times 10^7 \text{ km}^3$ of lava erupted to form volcanic smooth plains (estimated here
534 to be $\sim 25\%$ of the planet's total surface area), which is greater than the $\sim 8.9 \times 10^6 \text{--}1.0 \times 10^7$
535 km^3 of lava predicted for the lunar maria (Head and Wilson, 1992; Evans et al., 2016; Needham
536 and Kring, 2017). A critical open question is the origin of Mercury's intercrater plains, which are
537 the most spatially dominant geologic unit on the planet (Whitten et al., 2014). If an additional
538 40% of the surface area of Mercury is also volcanic (Murray et al., 1974; Strom et al., 1975;
539 Denevi et al., 2013; Whitten et al., 2014; Byrne et al., 2018), then potentially up to 2 orders of
540 magnitude more lavas were erupted to form volcanic plains on Mercury than erupted to form the
541 lunar maria, over a similar length of time. The large volumes of erupted lavas on Mercury imply

542 that large amounts of volatiles were released during the eruptive phases, which may have
543 resulted in the transient production of enhanced atmospheric pressures with lifetimes between
544 ~250 and ~210,000 years.

545 It is possible that these atmospheric lifetimes were greater than the time of repose
546 between average mercurian eruptions (~20,000–40,000 years for SPs-forming eruptions or
547 2,000–4,000 years when also including ICPs-forming eruptions), creating dense, collisional
548 transient atmospheres. Such atmospheres would enable more volatiles to migrate across the
549 planet before they are photodestroyed and lost to space, increasing the probability of the erupted
550 volatiles being cold-trapped (Stewart et al., 2011; Prem et al., 2015). For volatiles that did
551 become cold-trapped at Mercury’s poles, we estimate that they would be well-mixed with the
552 regolith and at least ~16 m below the subsurface given regolith gardening rates (Crider and
553 Killen, 2005). Such sequestered volatiles would be distinctly different from the polar H₂O-ice
554 deposits observed on Mercury’s surface today, both in composition (Butler et al., 1993;
555 Lawrence et al., 2013) and in age (Crider and Killen, 2005; Lawrence et al., 2013; Deutsch et al.,
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557

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559

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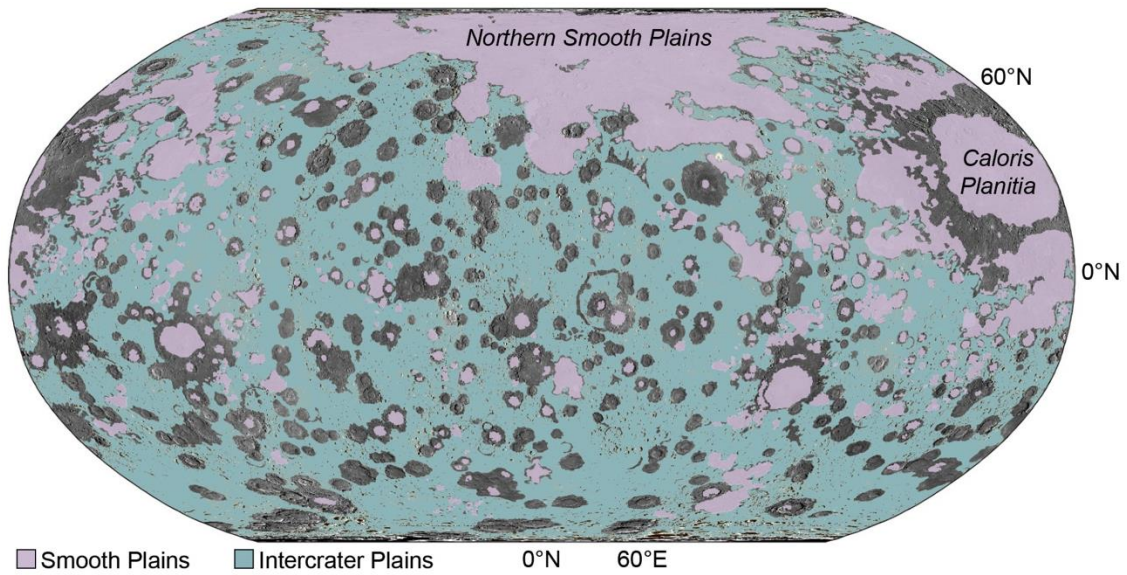
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810 **Figures**

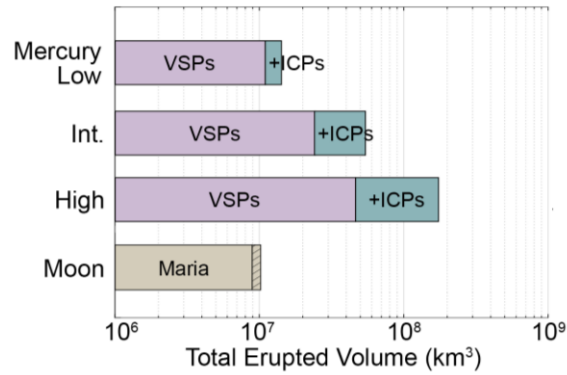


811

812 **Fig. 1.** The global distribution of smooth plains and intercrater plains on Mercury in Robinson

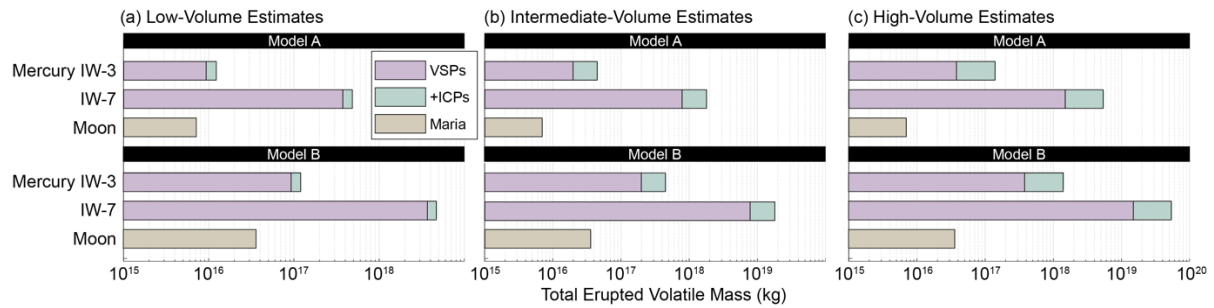
813 projection (modified from Byrne et al., 2018).

814



815

816 **Fig. 2.** Estimates of the total volume of erupted lavas on Mercury using low (0.1 km),
817 intermediate (1 km), and high (4 km) estimates for previously unresolved plains thicknesses.
818 These are compared with the predicted volume of erupted lavas on the Moon using the data from
819 Needham and Kring (2017) as well as Head and Wilson (1992) and Evans et al. (2016) in
820 hatching. Note that the volume is shown on a log scale.



821

822

Fig. 3. Estimates of the total mass of volatiles released on Mercury from eruptions forming

823

volcanic smooth plains (VSPs) and intercrater plains (ICPs), compared with the total mass of

824

volatiles released during the formation of the maria estimated by Needham and Kring (2017).

825

For fO_2 values of IW-3 and IW-7, estimates are computed for low-gas (Model A) and high-gas

826

(Model B) scenarios. Recall, Model A estimates S abundances to be 1% SCSS and C-, H-, O-, N-

827

, and Cl-abundances from 10% degassing amounts, while Model B estimates 10% of S and 100%

828

of other volatiles degassed. We use a range of thicknesses (0.1, 1, and 4 km) for plains that lack

829

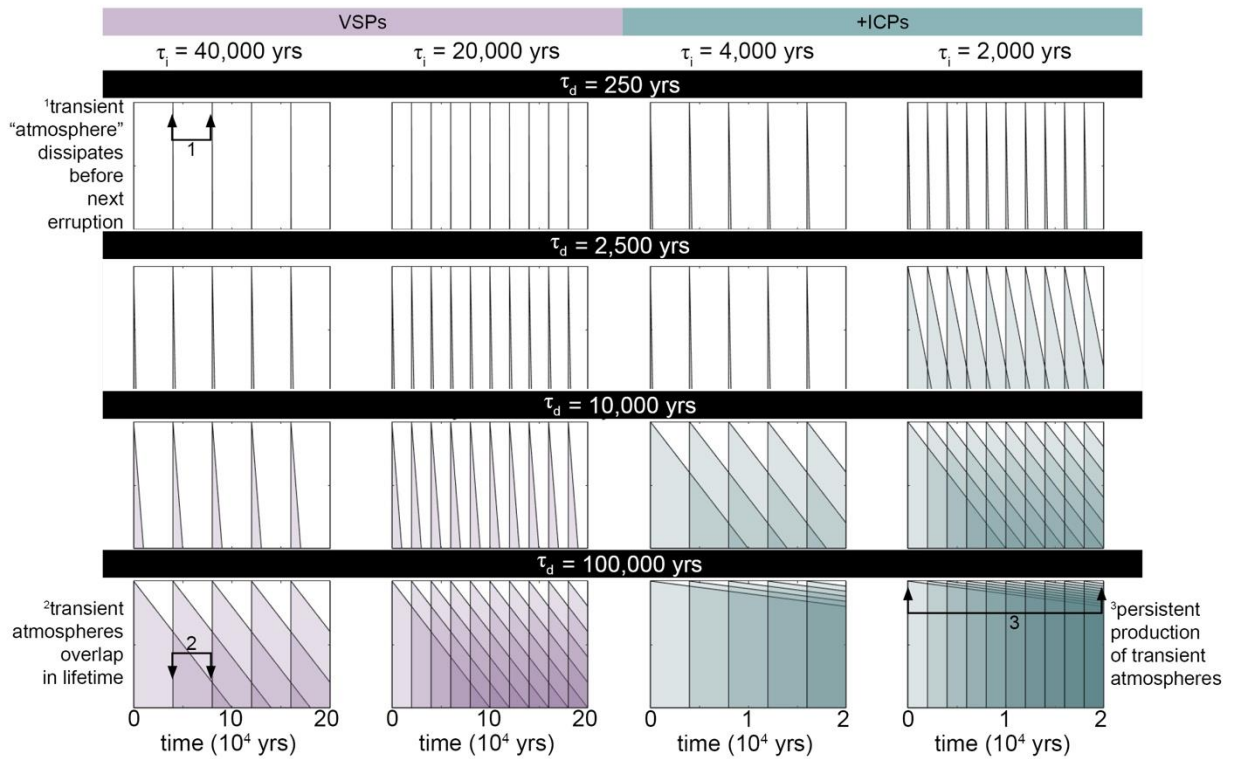
previous estimates to derive (a) low-volume, (b) intermediate-volume, and (c) high-volume

830

estimates. Note that the lunar estimates do not vary from (a) to (c), but do vary between low- and

831

high-gas scenarios.



832

833 **Fig. 4.** Estimates of the total duration of volcanically-derived transient atmospheres from various

834 model runs for low-volume scenarios ($V_e=200 \text{ km}^3$; Table 3), where each triangular peak

835 represents the formation of a volcanically-derived transient atmosphere, and the width of the

836 triangle represents the duration of such an atmosphere. Although the individual eruptions are

837 schematically represented as triangles, the decay of an eruption is not predicted to be linear. As

838 the estimated time interval between average eruptions (τ_i) decreases across the columns, the

839 intermittency spacing between eruptions decreases; eruptions are occurring every 40,000 years in

840 the leftmost column and as frequently as every 2,000 years in the rightmost column. As the

841 estimated duration of an average transient volcanically-derived atmosphere (τ_d) increases down

842 the rows, the duration of a transient atmosphere increases; τ_d ranges from 250 to 2,500 years at

843 IW-3 under degassing Models A and B, respectively, and from 10,000 to 100,000 years at IW-7

844 under Models A and B. Note that the x-axis is longer (200,000 years) in the first two columns
845 than in the last two columns (20,000 years).

846 **Tables**

847

		Estimated Volatiles Delivered to the Surface							
		Mass, <i>n</i> (ppm)				Amount (wt. %)			
		IW-3		IW-7		IW-3		IW-7	
Element	Potential Degassing Species	Model A	Model B	Model A	Model B	Model A	Model B	Model A	Model B
S	S ₂ , S ₃ , CS ₂ , S ₂ Cl, COS, SCl ₂	100	1,000	700	7,000	36	36	6	6
C	CO, CS ₂ , COS	10	100	1	10	3.6	3.6	0.01	0.01
H	H ₂ S, H ₂ , HCl	100	1,000	50	500	36	36	0	0
O	COS	13	133	1.3	13	4.9	4.9	0.01	0.01
N	N ₂	50	500	10,000	100,000	18	18	92	92
Cl	Cl ₂ , S ₂ Cl, SCl ₂ , HCl, NaCl, (NaCl) ₂ , KCl	1	10	80	800	0.4	0.4	0.7	0.7

848

849 **Table 1.** Predicted volatile species for mercurian magmas. S abundances are estimated to be 1%
850 of the estimated sulfur concentration at sulfide saturation (SCSS) for Model A and 10% of the
851 estimated SCSS for Model B. Abundances of other elements are estimated from degassing
852 amounts of 10% and 100% for Models A and B, respectively. Potential degassing species are
853 discussed in detail by Zolotov (2011), Evans et al. (2015), Armstrong et al., (2015) and Dalou et
854 al. (2019).

Mass of Released Volatiles (kg)														
S		C		H		N		Cl		O		Sum		
Degassing Model														
Model A (Low Gas)	Model B (High Gas)	Model A	Model B	Model A	Model B	Model A	Model B	Model A	Model B	Model A	Model B	Model A	Model B	
IW-3														
Volume Scenario	<i>Volcanic Smooth Plains Only</i>													
Low	3.4E+15	3.4E+16	3.4E+14	3.4E+15	3.4E+15	3.4E+16	1.7E+15	1.7E+16	3.4E+13	3.4E+14	4.4E+14	4.5E+15	9.3E+15	9.3E+16
Intermediate	7.3E+15	7.3E+16	7.3E+14	7.3E+15	7.3E+15	7.3E+16	3.6E+15	3.6E+16	7.3E+13	7.3E+14	9.5E+14	9.7E+15	2.0E+16	2.0E+17
High	1.4E+16	1.4E+17	1.4E+15	1.4E+16	1.4E+16	1.4E+17	7.0E+15	7.0E+16	1.4E+14	1.4E+15	1.8E+15	1.9E+16	3.8E+16	3.8E+17
<i>Volcanic Smooth Plains and Inter crater Plains</i>														
Low	4.3E+15	4.3E+16	4.3E+14	4.3E+15	4.3E+15	4.3E+16	2.1E+15	2.1E+16	4.3E+13	4.3E+14	5.6E+14	5.7E+15	1.2E+16	1.2E+17
Intermediate	1.6E+16	1.6E+17	1.6E+15	1.6E+16	1.6E+16	1.6E+17	8.1E+15	8.1E+16	1.6E+14	1.6E+15	2.1E+15	2.2E+16	4.5E+16	4.5E+17
High	5.0E+16	5.0E+17	5.0E+15	5.0E+16	5.0E+16	5.0E+17	2.5E+16	2.5E+17	5.0E+14	5.0E+15	6.5E+15	6.6E+16	1.4E+17	1.4E+18
IW-7														
<i>Volcanic Smooth Plains Only</i>														
Low	2.4E+16	2.4E+17	3.4E+13	3.4E+14	1.7E+15	1.7E+16	3.4E+17	3.4E+18	2.7E+15	2.7E+16	4.4E+13	4.4E+14	3.7E+17	3.7E+18
Intermediate	5.1E+16	5.1E+17	7.3E+13	7.3E+14	3.6E+15	3.6E+16	7.3E+17	7.3E+18	5.8E+15	5.8E+16	9.5E+13	9.5E+14	7.9E+17	7.9E+18
High	9.8E+16	9.8E+17	1.4E+14	1.4E+15	7.0E+15	7.0E+16	1.4E+18	1.4E+19	1.1E+16	1.1E+17	1.8E+14	1.8E+15	1.5E+18	1.5E+19
<i>Volcanic Smooth Plains and Inter crater Plains</i>														
Low	3.0E+16	3.0E+17	4.3E+13	4.3E+14	2.1E+15	2.1E+16	4.3E+17	4.3E+18	3.4E+15	3.4E+16	5.6E+13	5.6E+14	4.7E+17	4.7E+18
Intermediate	1.1E+17	1.1E+18	1.6E+14	1.6E+15	8.1E+15	8.1E+16	1.6E+18	1.6E+19	1.3E+16	1.3E+17	2.1E+14	2.1E+15	1.8E+18	1.8E+19
High	3.5E+17	3.5E+18	5.0E+14	5.0E+15	2.5E+16	2.5E+17	5.0E+18	5.0E+19	4.0E+16	4.0E+17	6.5E+14	6.5E+15	5.4E+18	5.4E+19

856

857 **Table 2.** Mass of erupted volatiles released during VSP- and ICP-forming eruptions estimated for fO_2 values of IW-3 and IW-7 and

858 different degassing models. Model A represents a low-gas model where S abundances are estimated to be 1% of the estimated SCSS

859 and abundances of other elements are estimated from degassing amounts of 10%. Model B is a high-gas model where 10% of S and

860 100% of other volatiles are estimated to have degassed. For each case, we present the low-, intermediate-, and high-volume estimates
861 corresponding to different estimates of plains thicknesses (0.1, 1, and 4 km, respectively, for plains with previously unresolved
862 thicknesses).

	V_e (km ³)	fO_2	Degassing Model	G (kg)	F_l (m ³ s ⁻¹)	τ_e (days)	F_g (kg s ⁻¹)	ρ_s (kg m ⁻³)	P_s (Pa)	τ_d (years)
<i>Low-Flux Scenario</i>										
Low-Volume Scenario	200	IW-3	A	1.6×10^{11}	10^3	2,300	800	1.4×10^{-7}	0.008	250
			B	1.6×10^{12}			8,000	1.4×10^{-6}	0.08	2,500
		IW-7	A	6.5×10^{12}			33,000	5.6×10^{-6}	0.32	10,000
			B	6.5×10^{13}			330,000	5.6×10^{-5}	3.2	100,000
High-Volume Scenario	400	IW-3	A	3.3×10^{11}		4,600	830	2.9×10^{-7}	0.016	520
			B	3.3×10^{12}			8,300	2.9×10^{-6}	0.16	5,200
		IW-7	A	1.3×10^{13}			33,000	1.1×10^{-5}	0.64	21,000
			B	1.3×10^{14}			330,000	1.1×10^{-4}	6.4	210,000
<i>High-Flux Scenario</i>										
Low-Volume Scenario	200	IW-3	A	1.6×10^{11}	10^7	0.23	8.0×10^6	1.4×10^{-7}	0.008	250
			B	1.6×10^{12}			8.0×10^7	1.4×10^{-6}	0.08	2,500
		IW-7	A	6.5×10^{12}			3.3×10^8	5.6×10^{-6}	0.32	10,000
			B	6.5×10^{13}			3.3×10^9	5.6×10^{-5}	3.2	100,000
High-Volume Scenario	400	IW-3	A	3.3×10^{11}		0.46	8.3×10^6	2.9×10^{-7}	0.016	520
			B	3.3×10^{12}			8.3×10^7	2.9×10^{-6}	0.16	5,200
		IW-7	A	1.3×10^{13}			3.3×10^8	1.1×10^{-5}	0.64	21,000
			B	1.3×10^{14}			3.3×10^9	1.1×10^{-4}	6.4	210,000

864

865 **Table 3.** Predicted values to describe mercurian eruptions and resulting volcanically-derived
866 atmospheres, including average volume of a single eruption (V_e), oxygen fugacity (fO_2), mass of
867 total erupted gas (G), eruption rate (F_l), eruption duration (τ_e), gas release rate (F_g), transient
868 atmospheric surface density (ρ_s), transient atmospheric surface pressure (P_s), and transient
869 atmosphere duration (τ_d).