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## The occurrence of the REE-vanadate wakefieldite in the rhyolitic Joe Lott Tuff

--Manuscript Draft--

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# **1 The occurrence of the REE-vanadate wakefieldite in the rhyolitic Joe Lott Tuff,**

## **2 Utah, USA**

3

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10 Running head: Wakefieldite in rhyolitic ignimbrite

11

## ABSTRACT

The high-silica rhyolitic Joe Lott Tuff was erupted at  $19.2 \pm 0.4$  Ma from the Mount Belknap caldera, SW Utah. Certain units in the tuff contain two species of wakefieldite, the Nd- and Y-dominant types. They occur in disseminated streaks and patches in association with rhodochrosite, calcite, Fe oxide, cerite, and a Mn silicate (caryopilit?) thought to have been deposited from hydrothermal fluids. The wakefieldites contain the highest levels of As ( $\leq 15.34$  wt.%  $\text{As}_2\text{O}_5$ ) and P ( $\leq 5.7$  wt.%  $\text{P}_2\text{O}_5$ ) yet recorded in the mineral, pointing to significant solid solution towards chernovite and xenotime. Thorium levels are also unusually high ( $\leq 14.2$  wt.%). The source of the hydrothermal fluid(s) is unknown but may be related to uranium mineralization in the region, in that As, V and U are commonly associated in such deposits.

**KEYWORDS:** Joe Lott Tuff, wakefieldite-(Nd), wakefieldite-(Y), As, P and Th enrichment

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25

26

27 **Introduction**

28 Four species of the *REE*-vanadate wakefieldite are known: wakefieldite-(Ce) (Deliens and  
29 Piret, 1977, 1986), wakefieldite-(La) (Witzke *et al.*, 2008), wakefieldite-(Nd) (Moriyami *et*  
30 *al.*, 2010), and wakefieldite-(Y) (Miles *et al.*, 1971). They occur in a large number of  
31 parageneses, including Fe-Mn deposits, granitic pegmatites and silicified wood, and in a wide  
32 range of mineral associations. All occurrences of which we are aware, and their host rocks,  
33 are listed in the Supplementary Material. The total number is 23, although some reports are  
34 not accompanied by confirmatory X-ray diffraction or electron microprobe analytical data.  
35 Published analyses have shown that there is considerable compositional variation in  
36 wakefieldite, with significant degrees of solid solution between each species and with  
37 chernovite ( $\text{YAsO}_4$ ) and xenotime ( $\text{YPO}_4$ ). Calcium, Si and Th can also be significant  
38 components.

39 This paper describes previously unrecorded wakefieldite-(Nd) and wakefieldite-(Y) from a  
40 high-silica rhyolitic ash-flow deposit, the Joe Lott Tuff, Utah, USA. The new data have  
41 allowed us to comment on compositional variation in the phase and to propose the major  
42 substitution schemes. It is suggested that the wakefieldite crystallized from carbonate-rich  
43 fluids possibly associated with local vanadium-uranium mineralization.

44

45 **Joe Lott Tuff**

46 The Joe Lott Tuff Member of the Mount Belknap Volcanics is a rhyolitic ash flow tuff sheet  
47 associated with the collapse of the Mount Belknap caldera in west-central Utah (Fig. 1;

48 Cunningham and Steven, 1979; Budding *et al.*, 1987). Erupted at  $19.2 \pm 0.4$  Ma, it has a  
2

49 volume of 150 km<sup>3</sup>. The tuff is a composite sheet, changing laterally from a single cooling  
50 unit near source to four distinct cooling units distally (Fig. 2). The Lower Unit is up to 64 m  
51 thick and has a basal vitrophyre. Initial collapse of the caldera accompanied eruption of the  
52 Lower Unit. The unit is followed upward by a Middle Unit up to 43 m thick, a 26 m thick  
53 Pink Unit, and an Upper Unit 31 m thick. The poorly welded, pumice-rich Pink Unit  
54 comprises two ash-flow tuffs (17 m and 9 m thick) separated by a fall layer 0.5 m thick (Fig.  
55 3).

56 The major variability in the tuff is in the degree of welding and the abundance of  
57 phenocryst phases. The most densely welded, eutaxitic, rocks are at the base of the Lower  
58 Unit (the basal vitrophyre); the degree of welding increases upwards in both the Lower and  
59 Middle Units (Budding *et al.*, 1987). With the exception of the basal vitrophyre, the  
60 groundmass of all samples is largely devitrified.

61

## 62 **Occurrence of wakefieldite**

63 Wakefieldite has been studied in two samples; M831, from the upper part of the Middle Unit,  
64 and JLT4.1, from the upper ash-flow layer in the Pink Unit. M831 is poorly welded, with  
65 pumices and shards in a devitrified matrix rich in lithophysae up to 0.7 mm across. It contains  
66 about 1 modal % phenocrysts of quartz, sanidine, plagioclase, augite, FeTi oxides, apatite,  
67 zircon and monazite (Budding *et al.*, 1987). JLT4.1 is more densely welded and lithophysae-  
68 bearing, with the main phenocrysts being alkali feldspar, quartz and magnetite.

69 The wakefieldite occurs in disseminated veins and patches, associated with rhodochrosite,  
70 calcite, Fe oxides, cerite, a Mn silicate (caryopilitic?), monazite and quartz (Fig. 4). The  
71 assemblage is similar to that of the type wakefieldite-(Nd) from the Arose stratiform deposit,

72 Japan, reported by Moriyama *et al.* (2010) as hematite, caryopilitite, calcite and rhodochrosite.  
73 In the Joe Lott tuff, the wakefieldite is relatively abundant: in JLT4.1, over 100 crystals have  
74 been identified in a single thin section. Crystals are invariably small; the majority are  $\leq 5$   $\mu\text{m}$   
75 across, although a few are up to 10  $\mu\text{m}$ . Representative occurrences of wakefieldite-(Nd) are  
76 shown in Fig. 5, in two cases being associated with rhodochrosite and in two associated with  
77 magnetite phenocrysts. The crystals show several forms, including platy, prismatic and  
78 rounded. Contacts with neighbouring minerals are normally very sharp.

79

## 80 **Analytical methods**

81 Accessory phases were initially identified by SEM, using a Zeiss Sigma<sup>TM</sup>/VP FE (field  
82 emission) – SEM equipped with new generation SDD-type two EDS (XFlash 6/10<sup>TM</sup>)  
83 detectors produced by Bruker. An acceleration voltage of 30 kV and a 120  $\mu\text{m}$  aperture were  
84 used. Mineral compositions were determined by electron microprobe, using a Cameca Five  
85 FE microprobe equipped with five wavelength dispersive spectrometers and large crystals.  
86 The analytical conditions were: accelerating voltage 15 kV and probe current 40 nA, with  
87 counting times of 20 s on peak and 10 s on each of two background positions. The standards,  
88 crystals and X-ray lines used, and approximate detection limits are given in Table 1. The  
89 ‘PAP’ $\phi(\rho Z)$  program of Pouchou and Pichoir (1991) was used for corrections.

90

## 91 **Mineral compositions**

92 *Composition of wakefieldite-(Nd) and wakefieldite-(Y)*

93 Sample JLT4.1 contains both wakefieldite-(Nd) and wakefieldite-(Y); M831 has only  
94 wakefieldite-(Y) (Fig. 6; Table 1). After Nd, the most abundant *REE* in wakefieldite-(Nd) is  
95 La, followed by Y. Total *REE+Y* are in the range 0.82-0.90 apfu. The chondrite-normalized  
96 *REE* patterns (Fig. 7a) show strong negative Ce anomalies ( $\text{Ce}/\text{Ce}^*$  0.08-0.18), peaks at Pr,  
97 then a steady decrease to the *HREE* with large negative Eu anomalies ( $\text{Eu}/\text{Eu}^*$  0.06-0.30).  
98 The minor troughs at Ho in some patterns may be an artefact since the levels of the element  
99 are close to the detection limits. Inter-crystal variations are exemplified by  $[\text{La}/\text{Ce}]_{\text{CN}}$  (4-12)  
100 and  $[\text{La}/\text{Yb}]_{\text{CN}}$  (6-24). The dominant cations replacing the *REE* are Ca ( $\leq 0.05$  apfu) and Th  
101 (0.05-0.09 apfu). Arsenic (0.03-0.11 apfu) is the main substituent for V (0.77-0.89 apfu), with  
102 lesser amounts of P (0.02-0.04 apfu) and Si ( $\leq 0.04$  apfu).

103 Compared to the Nd-dominant variety, wakefieldite-(Y) in M831 has higher Th (0.05-0.15  
104 apfu), As (0.05-0.37 apfu), P (0.04-0.11 apfu) and Si ( $\leq 0.20$  apfu), and lower *REE+Y* (0.69-  
105 0.75 apfu), and V (0.36-0.76 apfu). The crystals show, with one exception, positive Ce  
106 anomalies ( $\text{Ce}/\text{Ce}^*$  1.4-2.2), peak at Sm and have negative Eu anomalies ( $\text{Eu}/\text{Eu}^*$  0.13-0.26)  
107 (Fig. 7b). The wakefieldite-(Y) contains  $\leq 0.18$  wt%  $\text{SO}_3$  ( $\leq 0.01$  apfu). The only other report  
108 of anions in wakefieldite of which we are aware is of  $\text{SO}_3$  in fossilized wood from the Czech  
109 Republic (0.14 wt.%; Matysova *et al.*, 2016), although Khoury *et al.* (2015) refer to the  
110 occurrence in marbles in central Jordan of a Ca-rich, U-and S-bearing analogue of  
111 wakefieldite-(Ce)  $[(\text{Ce},\text{Ca},\text{U})(\text{VO}_4)(\text{SO}_4)]$ . The wakefieldite-(Y) from sample JLT4.1 is  
112 different to that in sample M831 in having lower abundances of As, P, Th and V (Fig. 8). The  
113 patterns in chondrite-normalized *REE* plots in JLT4.1 are broadly similar to those for  
114 wakefieldite-(Nd) in the same rock (Figs. 7a, 7b).

115

116 **Substitution mechanisms**

117 In this section, the new and published analyses are used to look for generally applicable  
118 substitution schemes. Among the *REE*, the main substitutions are  $\text{Y}_1\text{HREE}_{-1}$  (Fig. 9a) and  
119  $\text{LREE}_1(\text{Y}, \text{HREE})_{-1}$  (Fig. 9b). Calcium may be incorporated into wakefieldite by the  
120 substitution  $\text{Ca}^{2+} + \text{Th}^{4+} = 2(\text{REE} + \text{Y})^{3+}$ , as is observed in *inter alia* the monazite group but  
121 the overall correlation is not strong (Fig. 10a;  $r^2 = 0.32$ ) because the various suites follow  
122 separate subtrends.

123 The major substituent for V in the Joe Lott samples is As ( $\text{As}_1\text{V}_{-1}$ ), with As levels up to  
124 0.37 apfu (15.34 wt.%  $\text{As}_2\text{O}_5$ ) in wakefieldite-(Y) (Fig. 8a). These are the highest values yet  
125 recorded in wakefieldite. In their study of *LREE*- and Y-arsenates from a Fe-Mn deposit in the  
126 Maritime Alps, Miyawaki and Nakai (1996) found up to 30 mol% of  $\text{LREEVO}_4$ , broadly  
127 similar in amount to the entry of As into wakefieldite. Phosphorus also substitutes for V ( $\text{P}_1\text{V}_{-1}$ )  
128 in significant amounts in the Joe Lott minerals ( $\leq 0.11$  apfu; 5.7 wt.%  $\text{P}_2\text{O}_5$ ), again the  
129 highest values yet recorded in wakefieldite (Fig. 8b). It is still uncertain, however, whether  
130 there is a complete  $\text{YVO}_4 - \text{YPO}_4$  solid solution (Kolitsch and Holtstam, 2004; Hetherington  
131 *et al.*, 2008). Silicon is present at levels  $\leq 0.19$  apfu (4.1 wt.%  $\text{SiO}_2$ ), although some high  
132 values may be due to beam contamination by neighbouring quartz grains. The mechanism of  
133 incorporation of Si is uncertain: one possibility is similar to the thorite exchange in xenotime,  
134  $\text{Y}^{3+} + \text{P}^{5+} = \text{Th}^{4+} + \text{Si}^{4+}$  (Förster, 2006), as inferred by Miles *et al.* (1971) for the type  
135 wakefieldite-(Y); however, the correlation in the Joe Lott tuff data is weak ( $r^2 = 0.34$ ) (Fig.  
136 10b).

137 As noted above, in their study of *LREE*- and Y-arsenates Miyawaki and Nakai (1996)  
138 found up to 30 mol.% of  $\text{LREEVO}_4$ , in *LREE*- and Y-arsenates. They suggested that the  
139 presence of large  $\text{AsO}_4$  tetrahedra could enable Y-arsenates to accept the larger *LREE* ions. In

140 the Joe Lott case, however, the situation is reversed: the highest As contents are accompanied  
141 by higher *HREE*+Y and lower *LREE* (Figs. 11a, b).

142

143 **Formation conditions**

144 The wakefieldite-(Nd) and wakefieldite-(Y) in the Joe Lott Tuff are found in veins and  
145 patches associated with rhodochrosite, calcite, cerite, monazite, quartz, Fe-oxide and  
146 caryopilite (?), strongly suggesting that they are of hydrothermal origin. This is consistent  
147 with the negative Ce anomalies in wakefieldite-(Nd): the mineral was formed in fluids  
148 depleted in Ce by oxidation of Ce<sup>3+</sup>, with the Ce then entering cerite (c.f. Witzke *et al.*, 2008).

149 We have no independent evidence of the T-P conditions under which they crystallized.  
150 However, Bakker and Elburg (2006) found that wakefieldite- (Ce) in diopside-titanite veins in  
151 Arkaroola, Flinders Range, South Australia, was formed by remobilization of *LREE* and Y  
152 from titanite and/or the granitic host rock by a hydrothermal fluids of fairly pure H<sub>2</sub>O at T  
153 <200°C and P <50 MPa.

154 The large number of parageneses in which wakefieldite has been found is reflected in the  
155 many mechanisms proposed for its formation. Miles *et al.* (1971) proposed that the type  
156 wakefieldite-(Y) is a secondary mineral, possibly derived by leaching of Y-bearing hellandite  
157 [(Ca,REE)<sub>4</sub>Y<sub>2</sub>Al<sub>2</sub>(Si<sub>4</sub>B<sub>4</sub>O<sub>22</sub>)(OH<sub>2</sub>)]. The type wakefieldite-(Ce) was formed, along with  
158 vanadinite [Pb<sub>5</sub>(VO<sub>4</sub>)<sub>3</sub>Cl], in an oxidation zone in a silicified limestone (Deliens and Piret,  
159 1977). Howard *et al.* (1995) reported wakefieldite-(Ce) occurring with Sr-rich zeolites and  
160 fluorite, suggesting that the *REE* and V were carried by hydrothermal solutions during the last  
161 stages of formation of the zeolites. Wakefieldite-(Ce) occurs with roscoelite, a vanadium  
162 mica, in reduction spots in Devonian sandstones in Banffshire, Scotland. The roscoelite is

163 thought to have formed by a reaction involving a change in redox potential of the groundwater  
164 and the release of V from V-rich Fe-Ti oxides in the sandstone (van Panhuys-Sigler *et al.*,  
165 1996). Moriyami *et al.* (2010) proposed that the type wakefieldite-(Nd) was formed during  
166 prehnite-pumpellyite facies metamorphism by recrystallization and hydration of Fe-and Mn-  
167 hydroxide. A solid solution of wakefieldite-(Ce) and wakefieldite-(Y) was formed in silicified  
168 plant issue of Lower Palaeozoic age from the Studenec area, Czech Republic, as a secondary  
169 mineral during post-depositional diagenesis (Matysova *et al.*, 2016).

170 The association of wakefieldite with carbonates in the Joe Lott Tuff strongly suggests that  
171 the mineral was deposited from CO<sub>2</sub>-rich hydrothermal fluids but this study has provided no  
172 evidence of the source of the inferred fluids. However, the Mount Belknap Volcanics (23-14  
173 Ma) formed above a western and eastern source area spanning the central part of the  
174 Marysvale volcanic field (Fig. 1). Intrusions in the source area resulted in hydrothermally  
175 altered rocks and deposits mostly of uranium and molybdenum (Rowley *et al.*, 1994). The  
176 uranium mines in the Marysvale region were operated by the Vanadium Corporation of  
177 America. While the mineralization described in this paper is clearly of a different type, it may  
178 also have been related to fluids released by these intrusions.

179

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249

250 **Figure captions**

251 Fig. 1. Locality map of the Mount Belknap Caldera in southwestern Utah, USA, showing the  
252 distribution of the Joe Lott Tuff and the location of samples (JLT4.1 and M831) used in this  
253 study.

254 Fig. 2. Stratigraphic relationships in the Joe Lott Tuff Member, the underlying Bullion  
255 Canyon Volcanics and the overlying crystal-rich member of the Mount Belknap Volcanics  
256 (after Budding *et al.*, 1987, fig. 4). The approximate positions of samples M831 and JLT4.1  
257 are shown.

258 Fig. 3. Pink Unit, exposed near the junction of State Road 4 and Interstate 89. Two ash-flows  
259 are separated by a thin fall layer (white; arrowed).

260 Fig. 4. False colour back-scattered electron (BSE) image of a thin section of JLT4.1. The dark  
261 cores of the rounded lithophysae (green) are composed of caryopilite (?) and silica. Yellow  
262 areas – calcite; red – rhodochrosite, blue – quartz and alkali feldspar. The pink crystal is  
263 titanomagnetite.

264 Fig. 5. BSE images of wakefieldite-(Nd) (Wf). **(a)** In rhodochrosite (Rds) associated with  
265 magnetite phenocryst (Mag). **(b)** Associated with magnetite phenocryst, which also has  
266 inclusions of cerite (Cer) and ilmenite (Ilm). Afs is an alkali feldspar phenocryst. **(c)**  
267 Subhedral crystal embedded in rhodochrosite. Rds is italicised to show the textural difference  
268 to that in Fig. 5a; the darker Rds to the right is more calcic. Qtz - quartz. **(d)** As inclusion in  
269 magnetite phenocryst. The pale rim marked X is an unidentified Mn, Pb, Al, Ca silicate.

270 Fig. 6. Ce-Nd-Y (apfu) plot for the Joe Lott Tuff and comparative suites. Data sources: Joe  
271 Lott Tuff – Supplementary Table 1; fossilized wood – Matyssova *et al.* (2016); Arose –  
272 Moriyami *et al.* (2010); Arkaroola – Bakker and Elburg (2006); Tifernine - Baudracco-Gritti  
273 *et al.* (1987). Wakefieldite-(La) is not plotted.

274 Fig. 7. Chondrite-normalized REE plots for (a) wakefieldite-(Nd) and (b) wakefieldite-(Y) in  
275 the Joe Lott Tuff. Data source: Supplementary Table 1, analysis numbers 1, 4, 9, 13, 14, 16,  
276 22, 17. Normalizing factors from Sun and McDonough (1989).

277 Fig. 8. (a) V-As and (b) V-P plots for Joe Lott Tuff and comparative suites. Data sources as in  
278 Figure 6.

279 Fig. 9. (a) Y – HREE and (b) (Y+HREE) – LREE plots for Joe Lott Tuff and comparative  
280 suites. Data sources as in Fig. 6, plus Glücksstern – Witze *et al.* (2008).

281 Fig. 10. (a)  $\text{Ca}^{2+} + \text{Th}^{4+} = 2(\text{REE} + \text{Y})^{3+}$  and (b)  $\text{Y}^{3+} + \text{P}^{5+} = \text{Th}^{4+} + \text{Si}^{4+}$  as possible  
282 substitution schemes in wakefieldite.

283 Fig. 11. Arsenic plotted against (a) (HREE + Y) and (b) LREE for Joe Lott Tuff and  
284 comparative suites.

Cover letter

The paper reports a very unusual occurrence of the rare mineral wakefieldite in a rhyolitic ignimbrite. New EPMA analyses extend the compositional range of the mineral with especially high values of As and P. The occurrence may have been formed from CO<sub>2</sub> rich metasomatizing fluids.

Figure 1

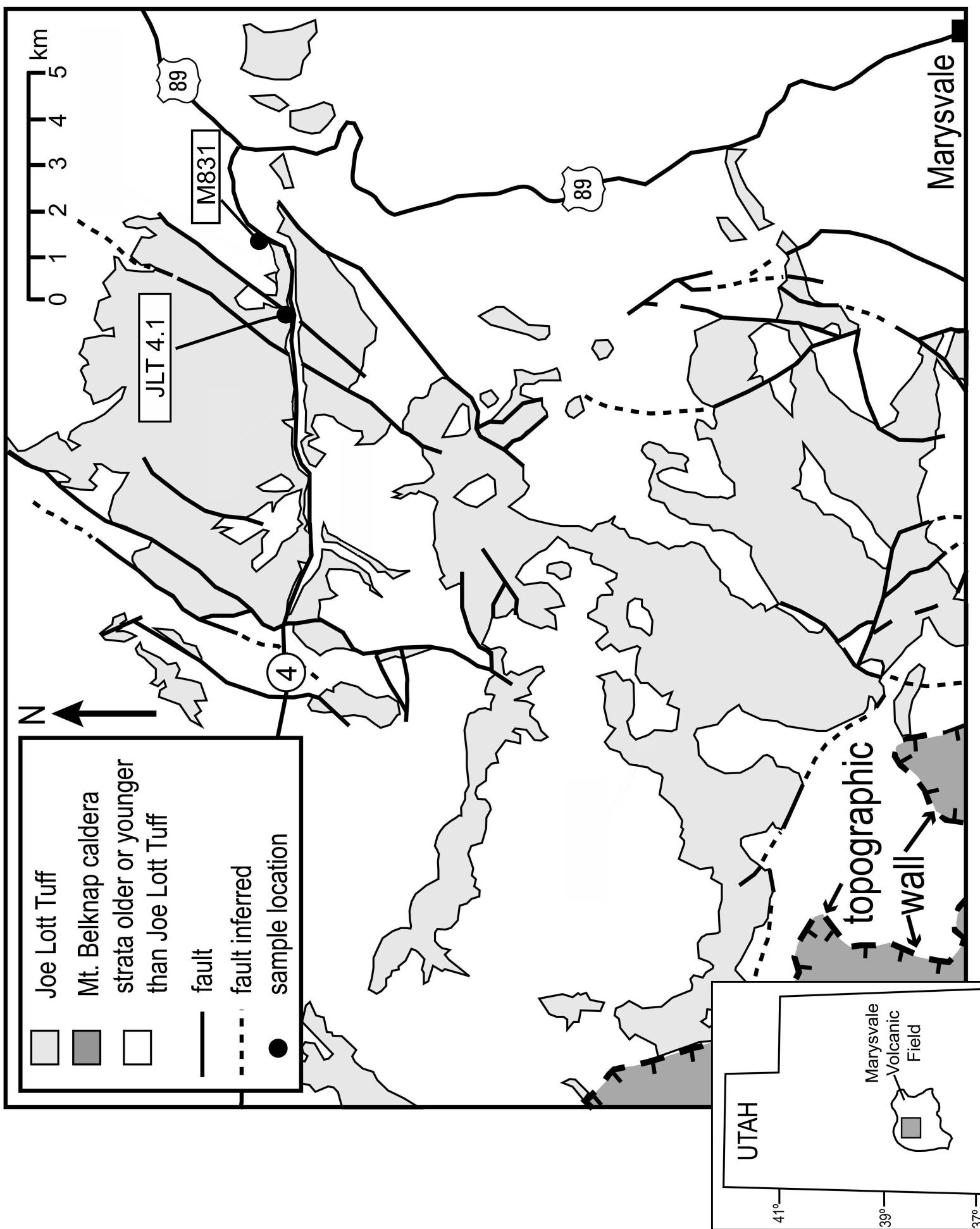
[Click here to access/download;Figure;Fig.1.eps](#)

Figure 2

[Click here to access/download;Figure;Fig.2.eps](#)

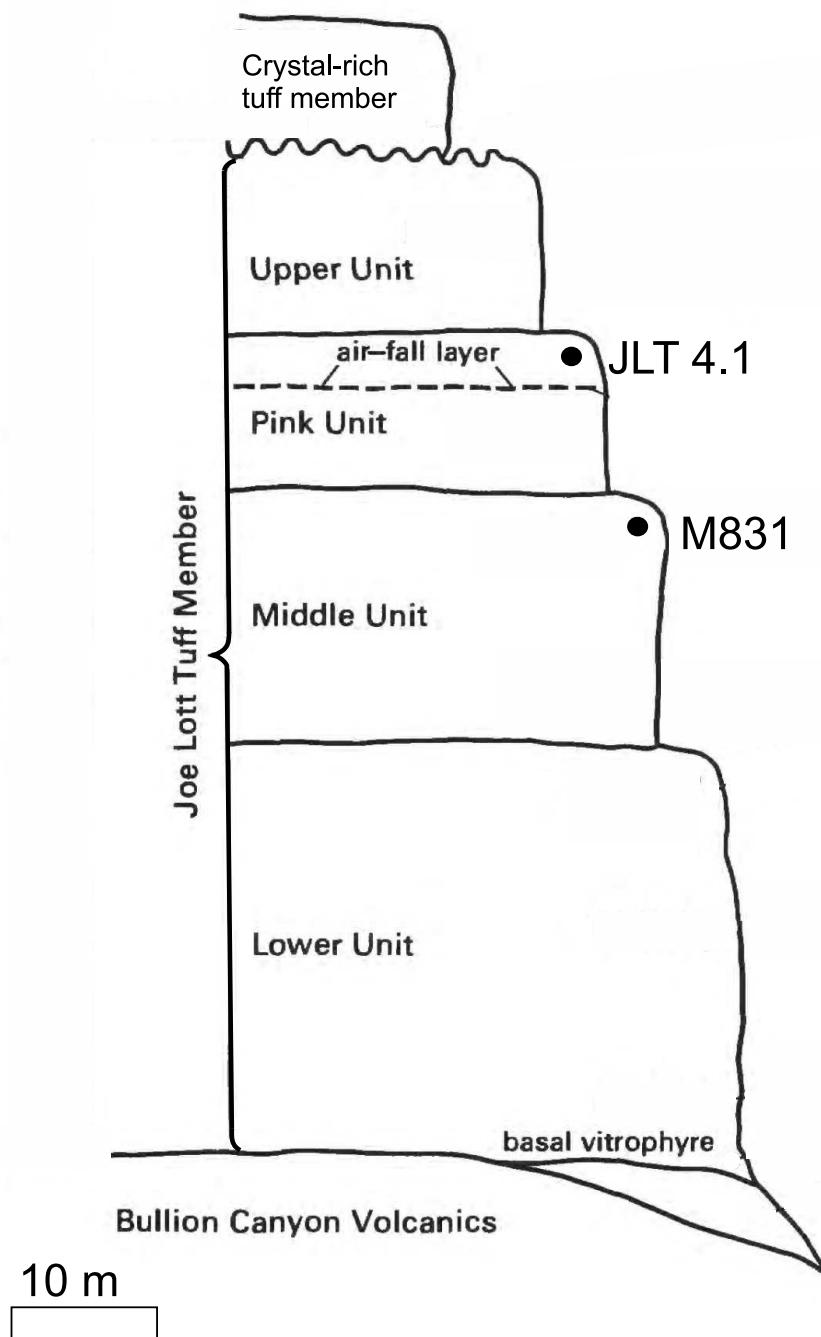


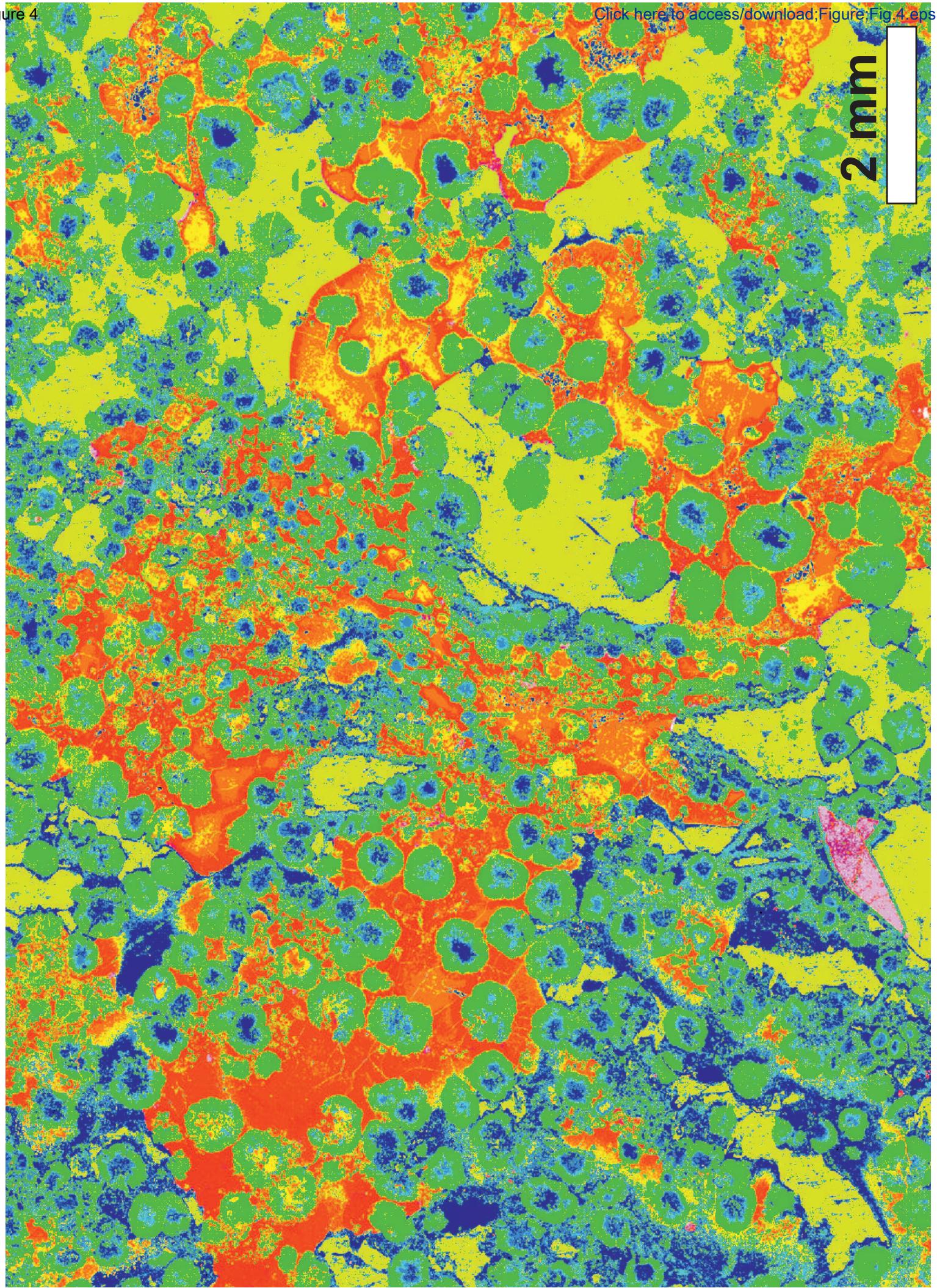
Figure 3

[Click here to access/download;Figure;Fig.3.eps](#) 



Figure 4.

[Click here to access/download/](#)Figure/Fig.4.eps



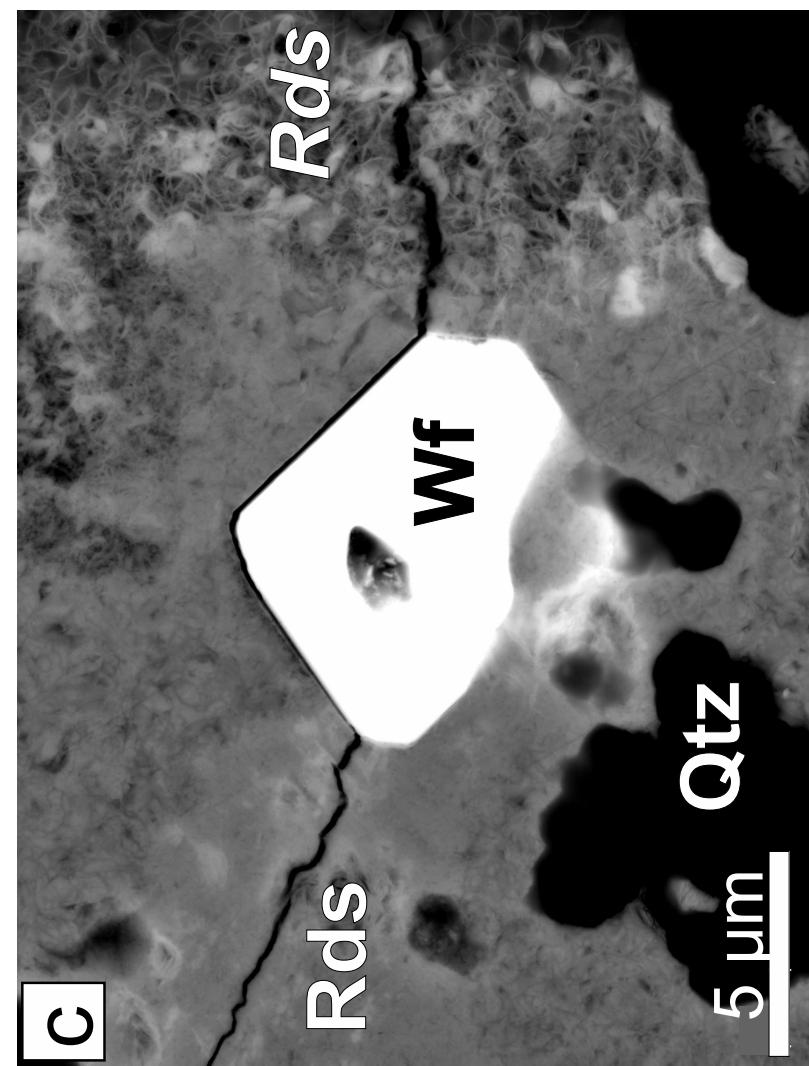
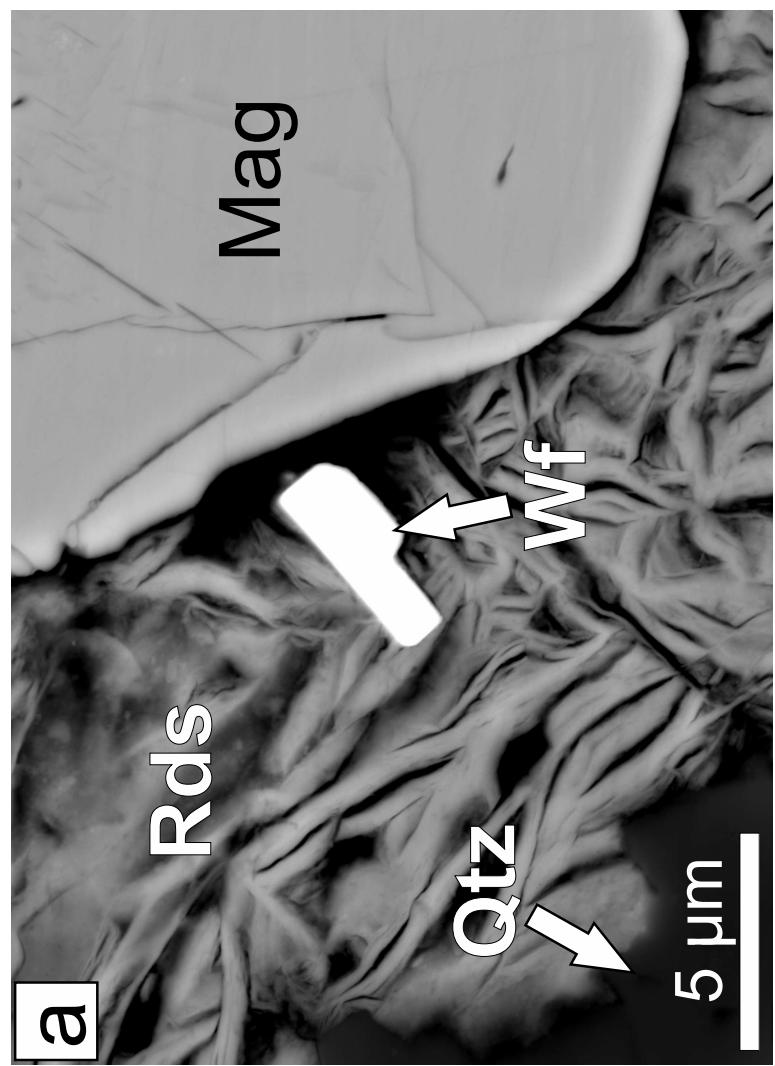
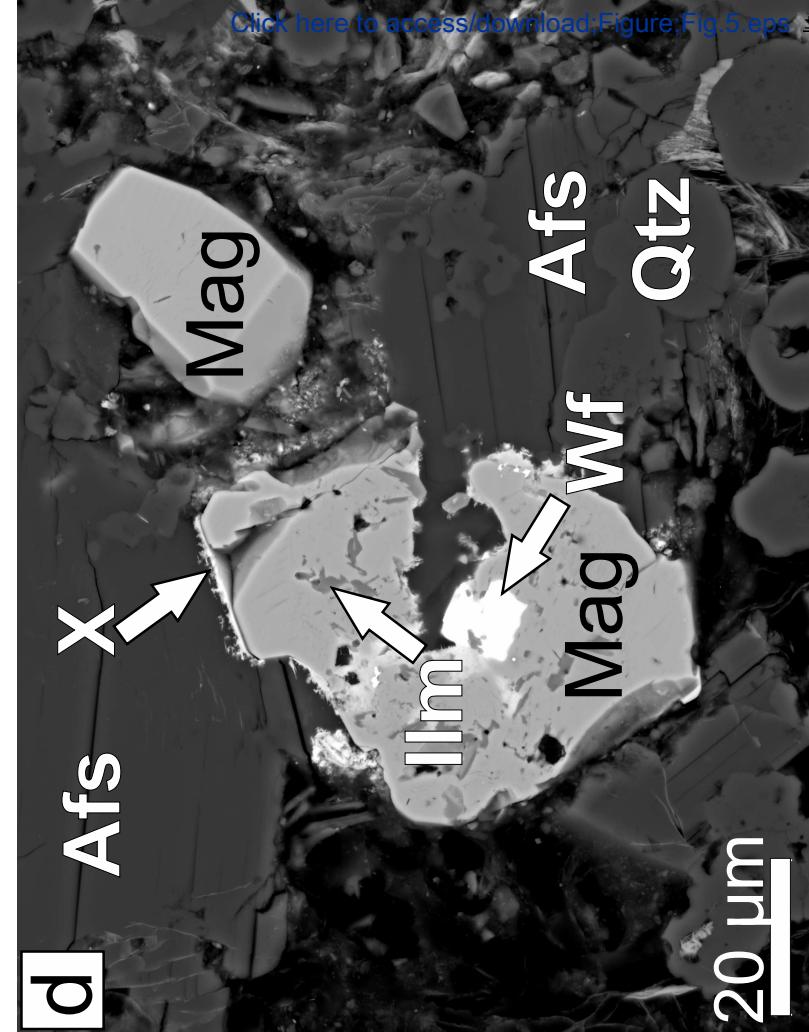
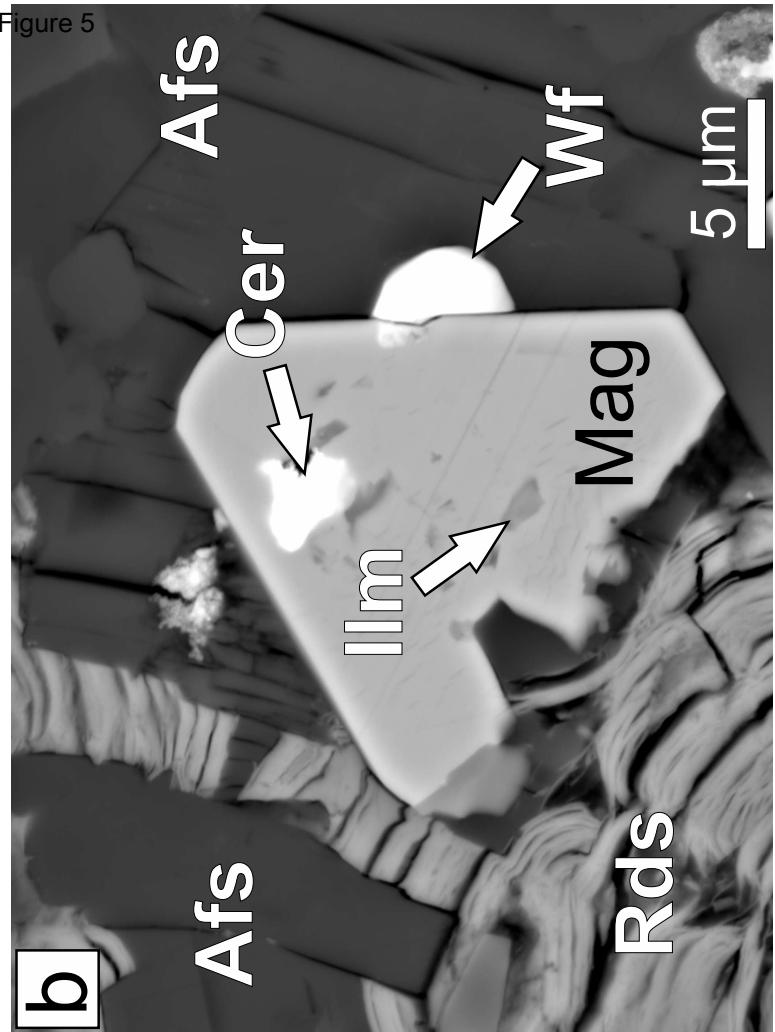


Figure 6

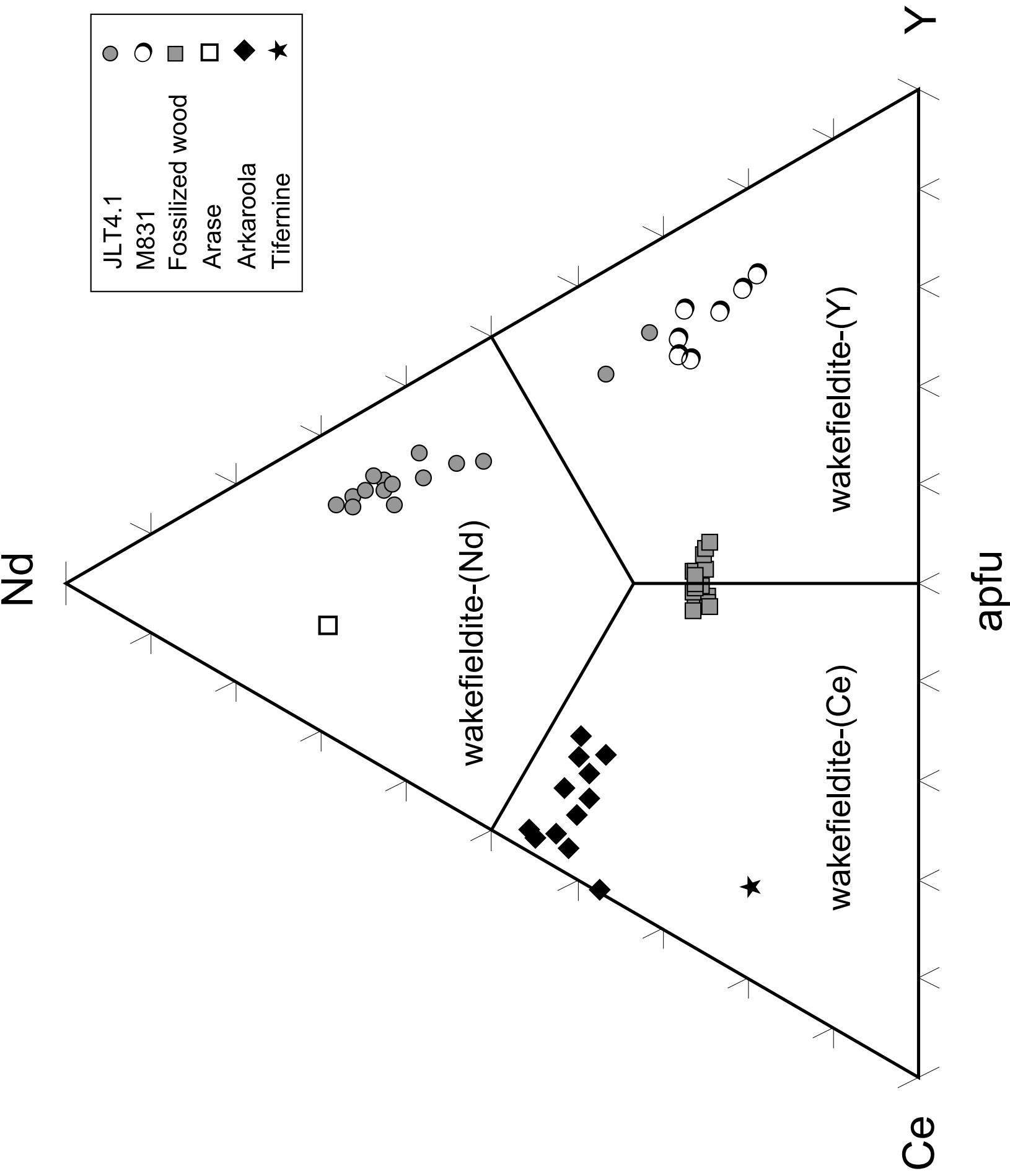
[Click here to access/download/FIGURE;Fig.6.eps](#)

Figure 7

Click here to access/download; Figure; Fig.7AB.eps

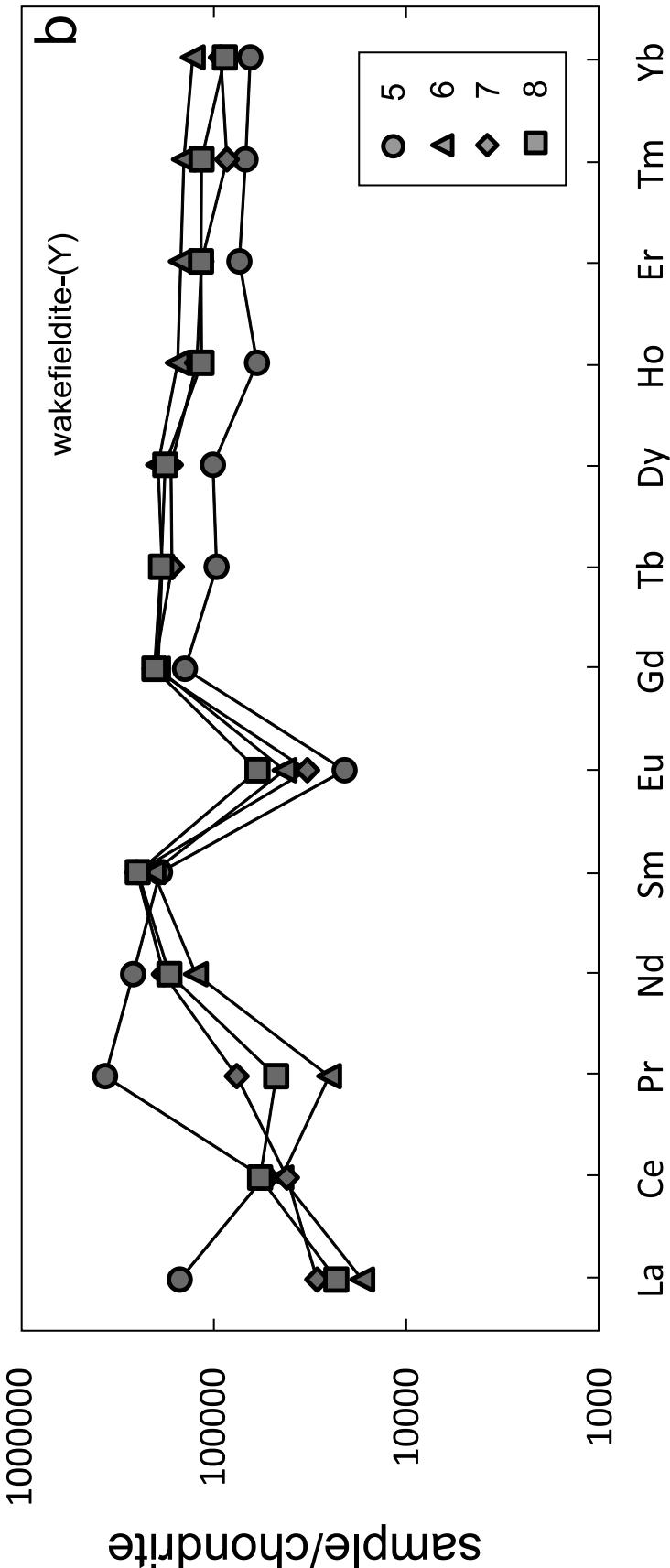
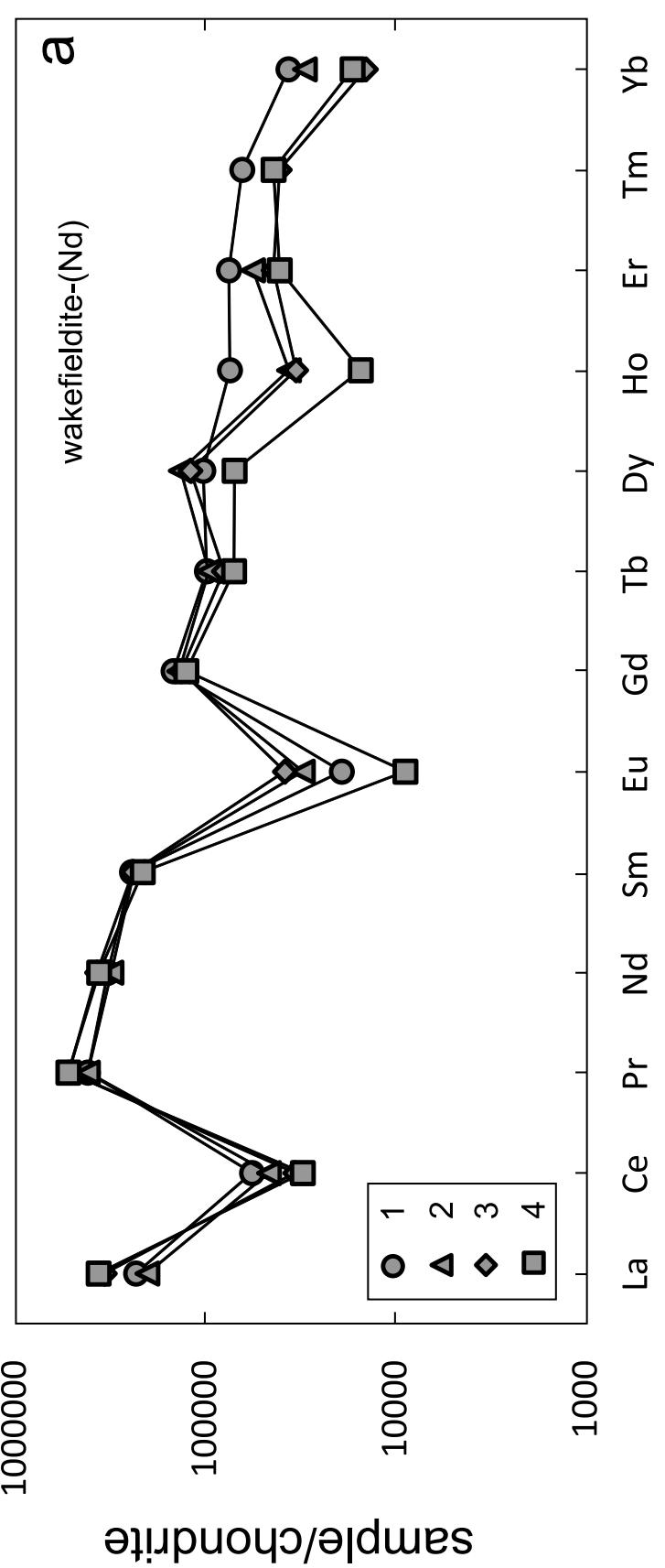
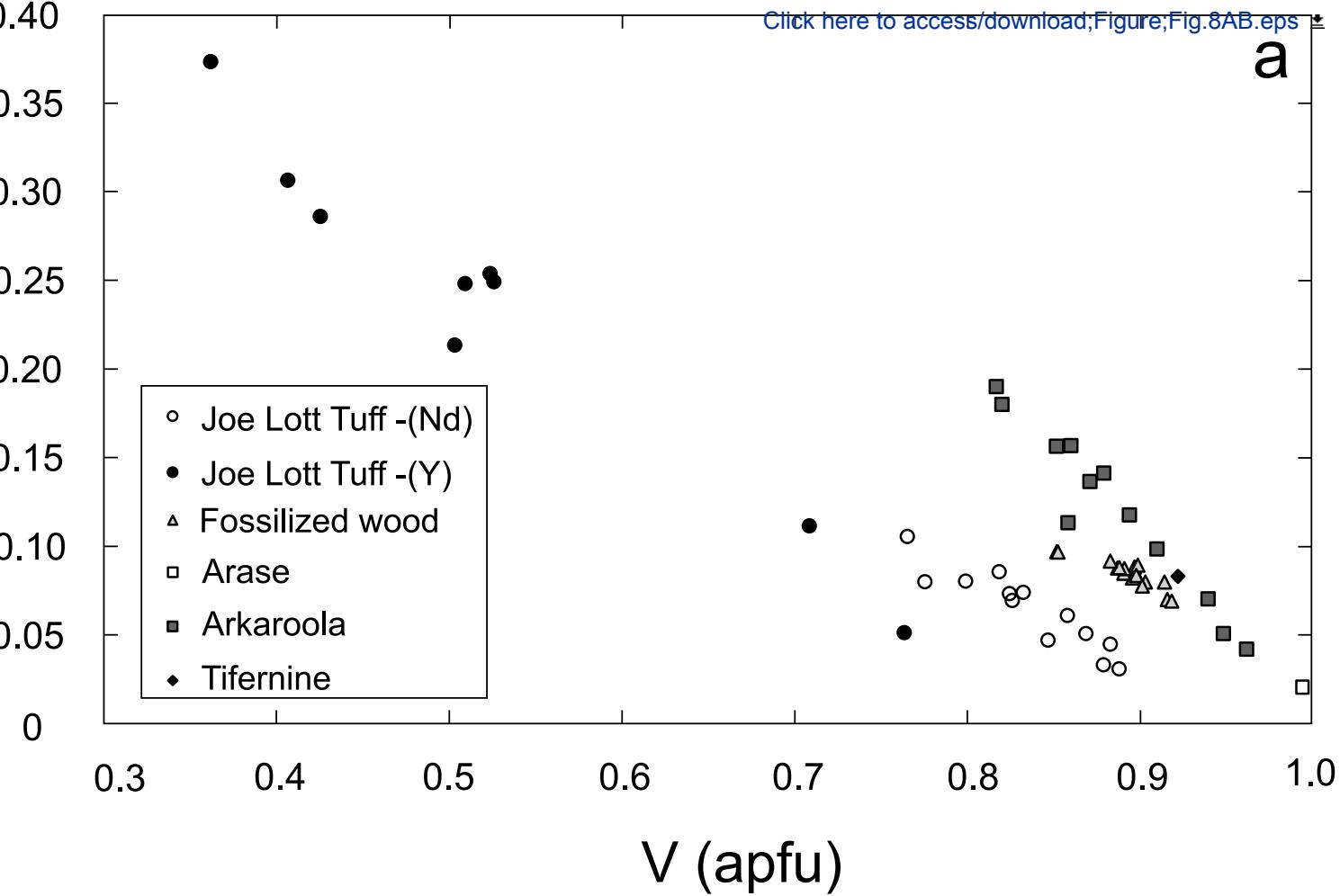


Figure 8 Click here to access/download; Figure; Fig.8AB.eps

As (apfu)

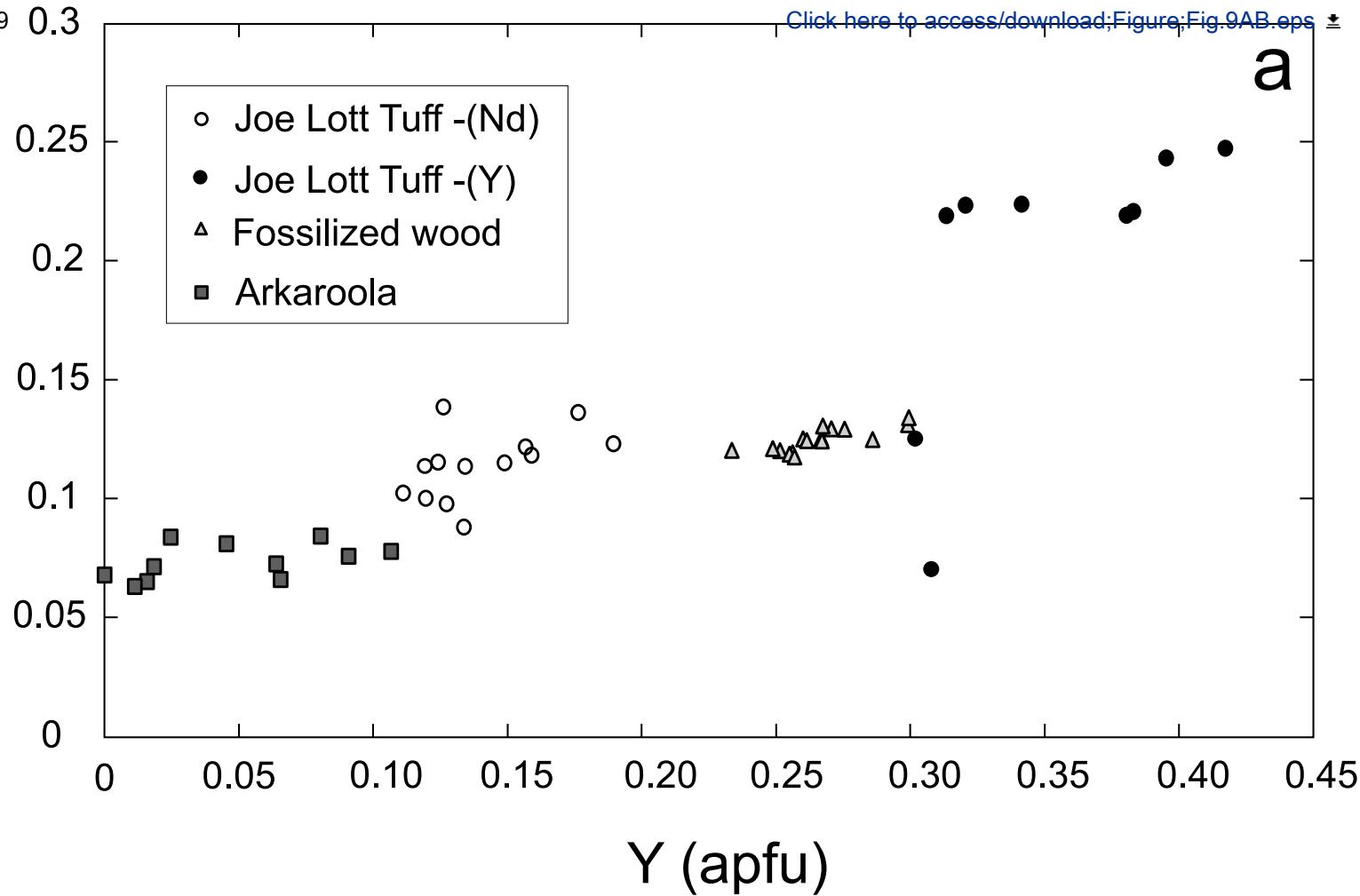
**a**

P (apfu)

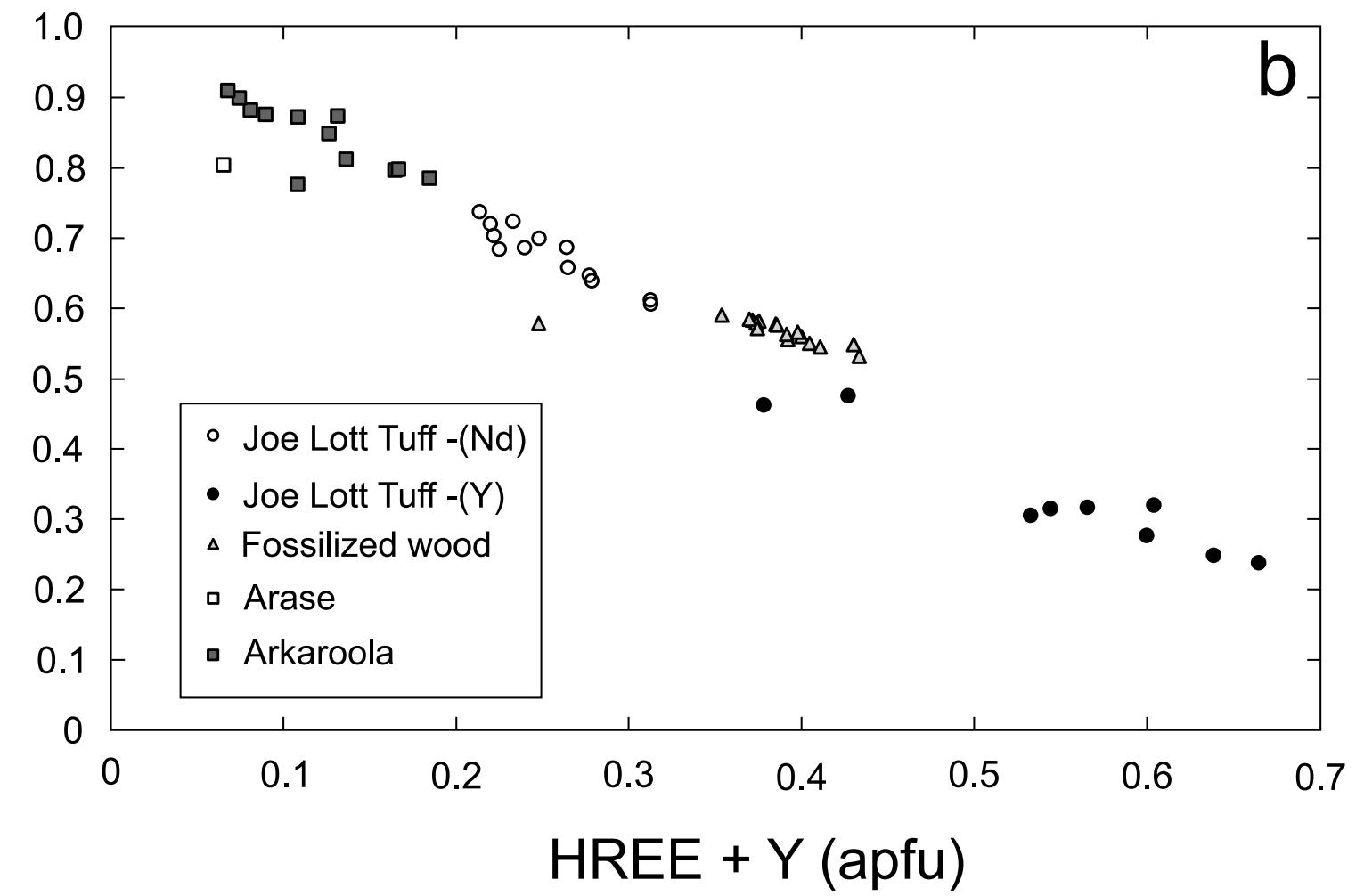
**b**

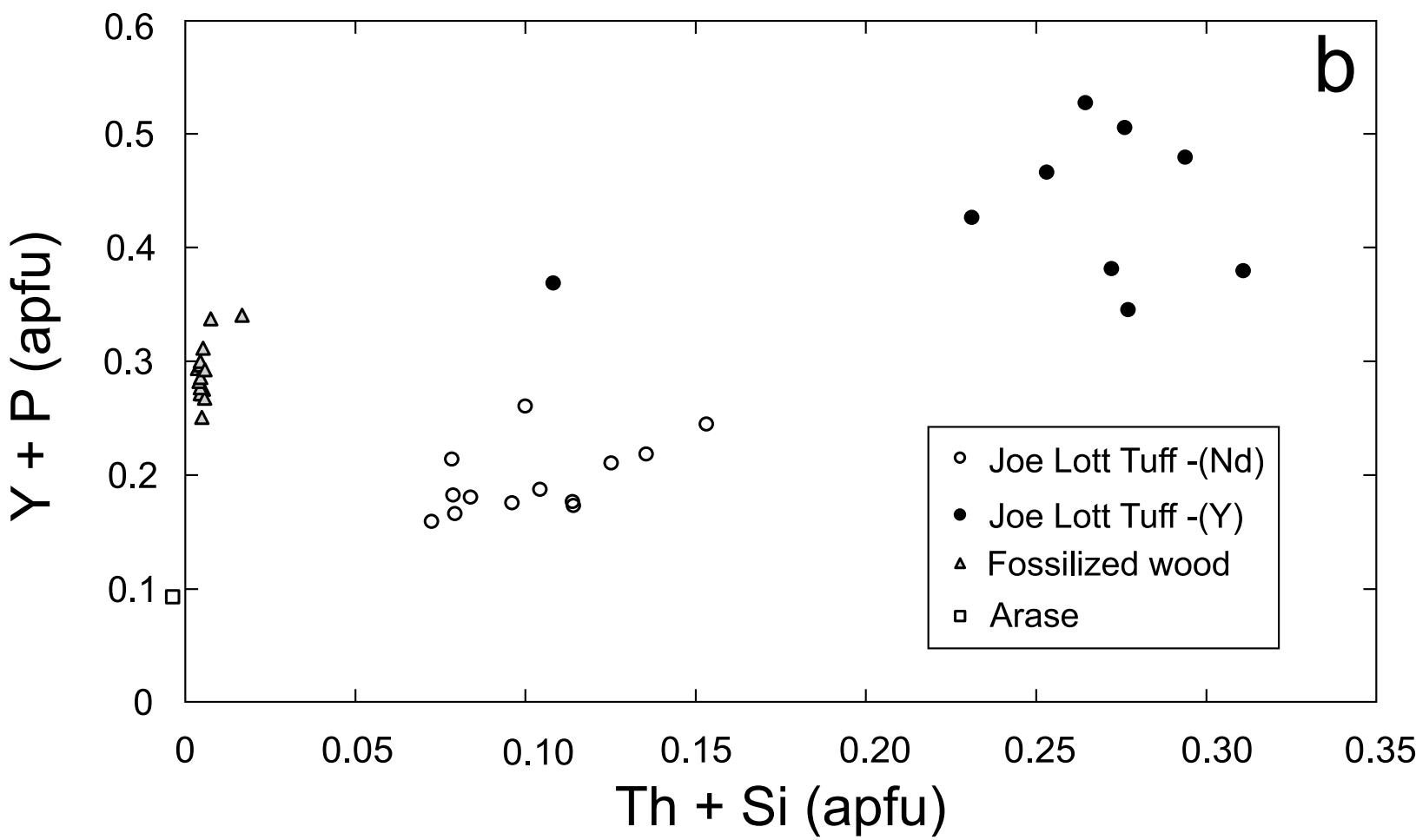
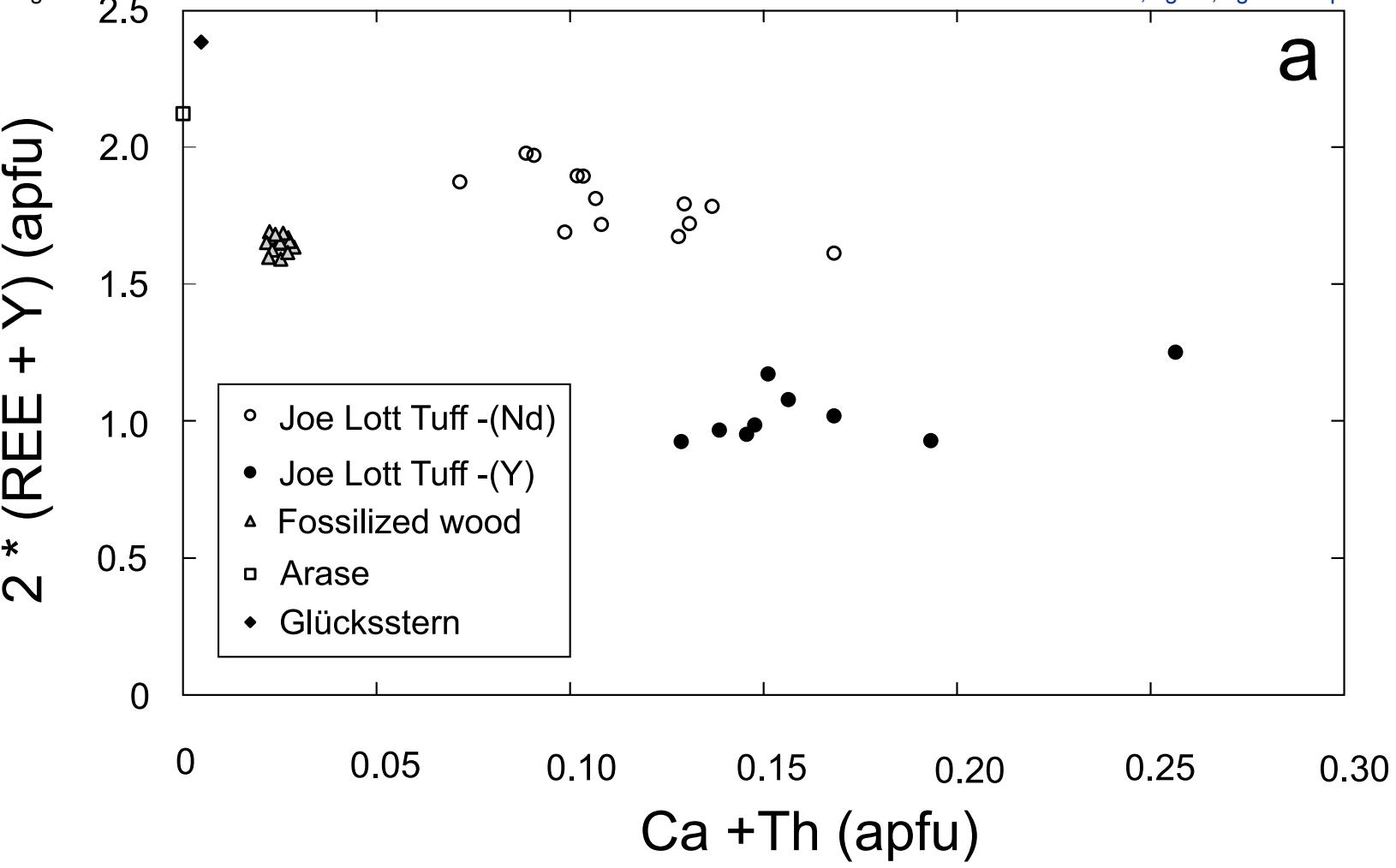
Figure 9 Click here to access/download; Figure; Fig.9AB.eps

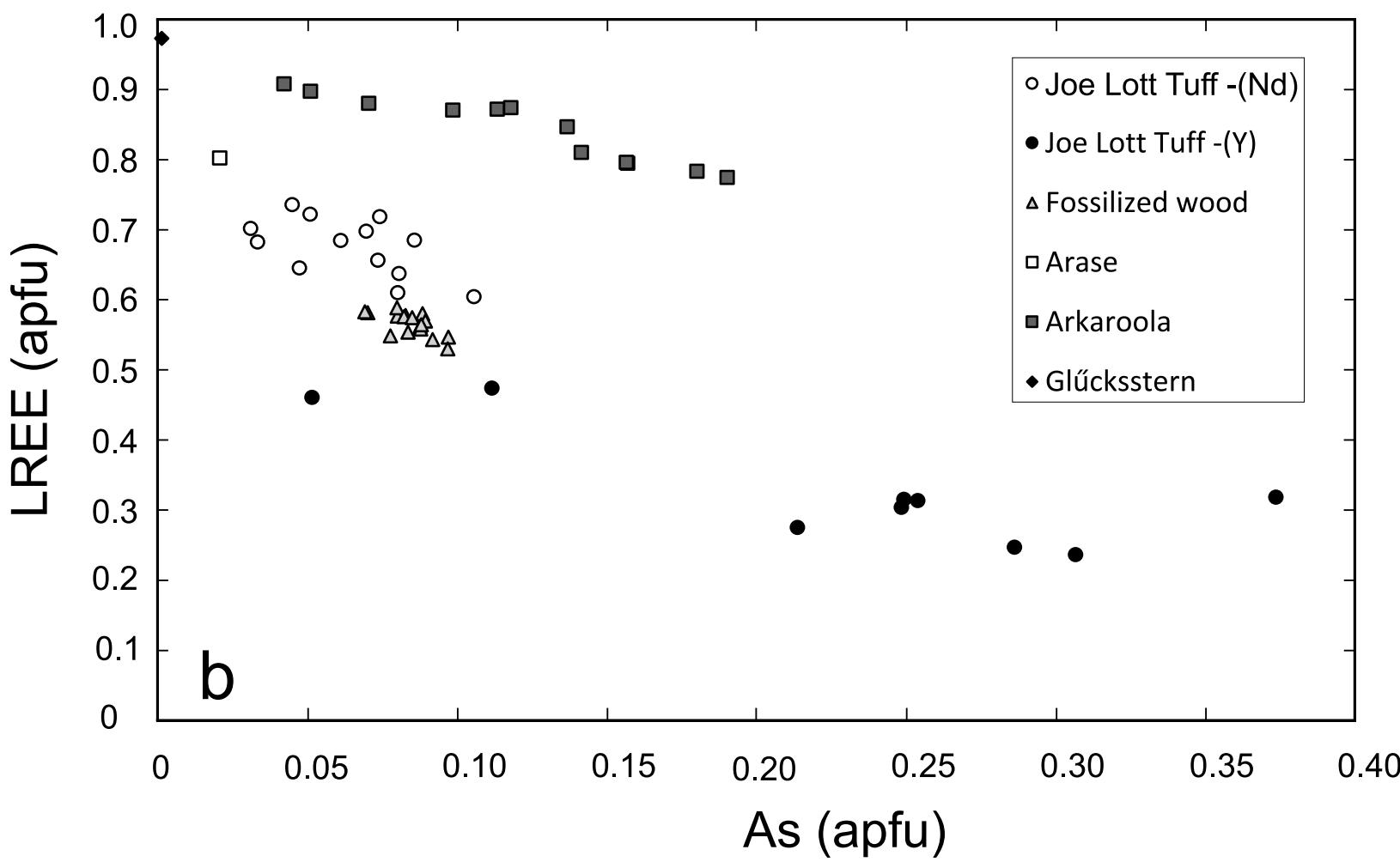
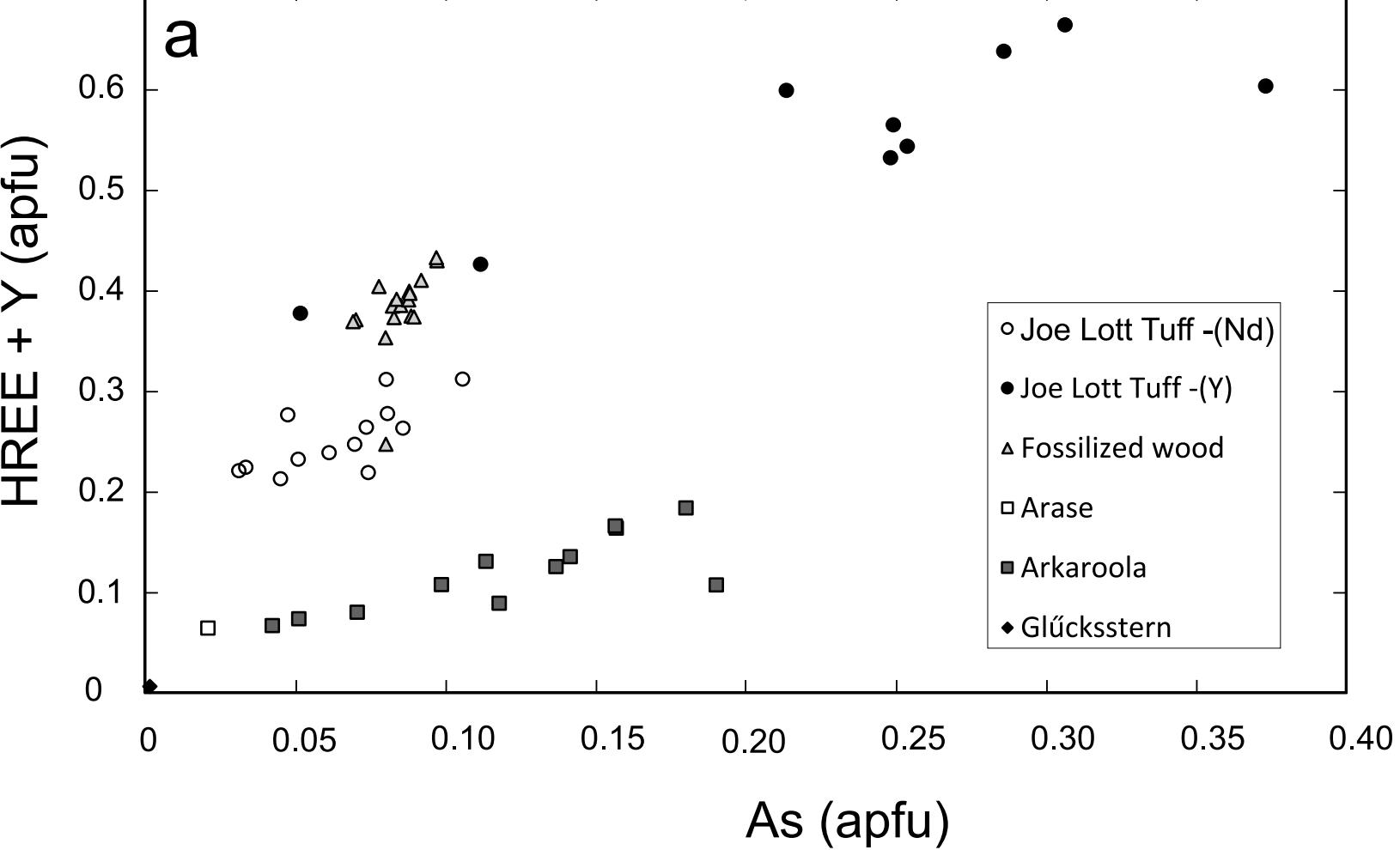
HREE (apfu)



LREE (apfu)







## Sheet1

Table 1. EPMA method summary

Element/ x-ray line	Crystal	Peak Overlap Correction	Standard	Approx. Det. Limit (ppm)	Peak/Backg. Time (s)	Approx. Std Dev. wt%
Al K $\alpha$	TAP		Al <sub>2</sub> O <sub>3</sub>	200	30/30	0.02
As L $\alpha$	TAP		ZnAs <sub>2</sub>	760	30/30	0.10
Bi M $\alpha$	LPET		Bi (metal)	740	30/30	0.07
Ca K $\alpha$	LPET		Wollastonite	160	20/20	0.02
Ce L $\alpha$	LPET		CePO <sub>4</sub>	500	30/30	0.06
Dy L $\alpha$	LLIF		DyPO <sub>4</sub>	1460	20/40	0.19
Er L $\alpha$	LLIF	Tb L $\beta$ <sub>4</sub>	ErPO <sub>4</sub>	1450	20/40	0.14
Eu L $\alpha$	LLIF	Pr L $\beta$ <sub>2</sub> , Nd L $\beta$ <sub>3</sub>	EuPO <sub>4</sub>	120	30/60	0.01
Gd L $\alpha$	LLIF	La L $\gamma$ <sub>2</sub> , Ce L $\gamma$	GdPO <sub>4</sub>	1280	20/40	0.15
Ho L $\alpha$	LLIF	Gd L $\beta$	HoPO <sub>4</sub>	160	30/60	0.02
Lu M $\alpha$	LTAP	Yb M $\beta$	LuPO <sub>4</sub>	120	110/180	0.13
L $\alpha$ L $\alpha$	LPET		LaPO <sub>4</sub>	470	30/30	0.01
Mg K $\alpha$	TAP		MgO	240	20/20	0.02
Nb L $\alpha$	LPET		Nb (metal)	420	60/60	0.04
Nd L $\alpha$	LLIF	Ce L $\beta$	NdPO <sub>4</sub>	1200	20/40	0.40
P K $\alpha$	TAP	Y L $\beta$	CePO <sub>4</sub>	260	20/20	0.04
Pb M $\alpha$	LPET	Y L $\gamma$ <sub>3</sub>	Crocoite	200	140/280	0.01
Pr L $\alpha$	LPET	La L $\beta$ , V K $\alpha$	PrPO <sub>4</sub>	290	30/60	0.06
S K $\alpha$	LPET		FeS <sub>2</sub>	180	20/20	0.01
Si K $\alpha$	TAP		SiO <sub>2</sub>	220	20/20	0.03
Sm L $\alpha$	LLIF	Ce L $\beta$ <sub>2</sub>	SmPO <sub>4</sub>	1260	20/40	0.18
Sr L $\alpha$	TAP		SrSO <sub>4</sub>	460	30/60	0.04
Tb L $\alpha$	LLIF		TbPO <sub>4</sub>	1150	30/60	0.10
Th M $\alpha$	LPET		ThO <sub>2</sub>	480	60/120	0.12
Ti K $\alpha$	LLIF		TiO <sub>2</sub>	420	20/40	0.03
Tm L $\alpha$	LLIF	Sm L $\gamma$ , Dy L $\beta$ <sub>4</sub>	TmPO <sub>4</sub>	270	30/60	0.02
U M $\alpha$	LPET	Th M $\beta$	UO <sub>2</sub>	400	80/80	0.02
V K $\alpha$	LLIF		V (metal)	500	20/40	0.31
Y L $\alpha$	TAP		YPO <sub>4</sub>	450	60/60	0.09
Yb M $\alpha$	LTAP	Dy M $\gamma$	YbPO <sub>4</sub>	400	120/120	0.03
Zr L $\alpha$	LPET		Zr (metal)	600	30/30	0.05

Table 2. Representative compositions of wakefieldite in the Joe Lott Tuff

Sample	Wakefieldite-(Nd)				Wakefieldite-(Y)			
	JLT4.1				JLT4.1		M831	
	1	2	3	4	5	6	7	8
<b>wt.%</b>								
CaO	1.04	1.28	0.99	1.92	4.80	0.61	0.68	0.84
SrO	0.10	bd	bd	0.07	bd	-	-	-
FeO*	-	-	-	-	-	0.05	0.43	bd
Y <sub>2</sub> O <sub>3</sub>	5.93	5.85	5.59	6.13	13.91	17.11	13.12	12.58
La <sub>2</sub> O <sub>3</sub>	7.82	10.62	8.74	10.39	4.22	0.47	0.64	0.70
Ce <sub>2</sub> O <sub>3</sub>	3.60	2.42	2.47	2.23	3.96	3.20	4.14	4.34
Pr <sub>2</sub> O <sub>3</sub>	5.79	5.66	5.43	6.01	4.16	0.28	0.53	0.50
Nd <sub>2</sub> O <sub>3</sub>	20.13	19.05	20.27	20.27	14.54	6.82	9.38	8.58
Sm <sub>2</sub> O <sub>3</sub>	4.63	3.59	4.61	3.87	3.43	3.71	4.46	4.07
Eu <sub>2</sub> O <sub>3</sub>	0.11	0.19	0.16	0.06	0.14	0.29	0.40	0.36
Gd <sub>2</sub> O <sub>3</sub>	3.23	2.86	3.38	3.04	3.37	4.67	4.89	4.55
Tb <sub>2</sub> O <sub>3</sub>	0.39	0.31	0.32	0.31	0.42	0.81	0.82	0.79
Dy <sub>2</sub> O <sub>3</sub>	3.23	2.97	3.42	2.08	2.97	5.75	5.27	5.00
Ho <sub>2</sub> O <sub>3</sub>	0.18	0.27	0.30	0.10	0.39	1.01	0.76	0.84
Er <sub>2</sub> O <sub>3</sub>	0.93	0.87	0.98	0.78	1.40	2.84	2.23	2.31
Tm <sub>2</sub> O <sub>3</sub>	0.11	0.17	bd	0.13	0.20	0.42	0.34	0.24
Yb <sub>2</sub> O <sub>3</sub>	0.38	0.29	0.34	0.33	1.26	2.52	1.71	1.69
SiO <sub>2</sub>	1.40	1.20	1.04	0.88	1.51	3.21	3.00	3.42
TiO <sub>2</sub>	bd	0.12	bd	0.11	bd	-	-	-
Nb <sub>2</sub> O <sub>5</sub>	bd	0.10	bd	bd	-	-	-	-
ThO <sub>2</sub>	5.62	8.04	6.56	5.61	5.04	11.29	12.89	14.18
UO <sub>2</sub>	bd	bd	bd	bd	bd	bd	bd	bd
PbO	bd	0.04	bd	bd	0.05	-	-	-
P <sub>2</sub> O <sub>5</sub>	0.72	0.54	0.68	0.62	1.95	5.73	3.15	3.36
V <sub>2</sub> O <sub>5</sub>	29.38	32.54	31.11	32.78	26.30	13.43	17.26	16.46
As <sub>2</sub> O <sub>5</sub>	3.13	1.56	2.80	1.45	5.24	12.80	10.58	10.15
SO <sub>3</sub>	bd	bd	bd	bd	bd	0.10	0.05	0.06
Total	97.85	100.54	99.19	99.17	99.26	97.12	96.73	95.02
<b>Formulae on the basis of 4 oxygens</b>								
Ca	0.047	0.056	0.044	0.084	0.210	0.030	0.033	0.042
Fe <sup>2+</sup>	-	-	-	-	-	0.002	0.017	0.000
Y	0.134	0.127	0.124	0.134	0.302	0.417	0.321	0.313
La	0.177	0.231	0.194	0.227	0.092	0.008	0.011	0.012
Ce	0.056	0.036	0.038	0.033	0.059	0.054	0.070	0.074
Pr	0.090	0.084	0.083	0.090	0.062	0.005	0.009	0.009
Nd	0.306	0.278	0.302	0.297	0.212	0.112	0.154	0.143
Sm	0.068	0.051	0.066	0.055	0.048	0.059	0.071	0.066
Eu	0.002	0.003	0.002	0.001	0.002	0.005	0.006	0.006

Gd	0.046	0.039	0.047	0.041	0.046	0.071	0.074	0.071
Tb	0.005	0.004	0.004	0.004	0.006	0.012	0.012	0.012
Dy	0.044	0.039	0.046	0.027	0.039	0.085	0.078	0.075
Er	0.013	0.012	0.014	0.011	0.019	0.044	0.035	0.037
Yb	0.005	0.004	0.004	0.004	0.016	0.035	0.024	0.024
Ti	0.000	0.002	0.000	0.001	0.000	-	-	-
Th	0.054	0.075	0.062	0.052	0.047	0.118	0.135	0.151
U	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Sum	1.048	1.039	1.031	1.060	1.158	1.055	1.049	1.036
V	0.826	0.879	0.858	0.888	0.708	0.407	0.524	0.509
Si	0.060	0.049	0.043	0.036	0.062	0.147	0.138	0.160
As	0.070	0.033	0.061	0.031	0.112	0.307	0.254	0.248
P	0.026	0.019	0.024	0.022	0.067	0.111	0.061	0.067
Sum	0.981	0.980	0.986	0.976	0.949	0.971	0.976	0.984
S	0.000	0.000	0.000	0.000	0.000	0.003	0.002	0.002
$\Sigma$ cations	2.03	2.02	2.02	2.04	2.11	2.03	2.03	2.02

FeO\*, all Fe as Fe<sup>2+</sup>. bd, below detection. Dash - not determined





## Supplementary Table 1

Supplementary Table 1. Compositions of wakefieldite in Joe Lott Tuff

Sample	Wakefieldite-(Nd)							
	JLT 4.1	1	2	3	4	5	6	7
wt.%								
P <sub>2</sub> O <sub>5</sub>	1.28	0.98	0.72	1.11	0.54	0.70	0.59	0.62
V <sub>2</sub> O <sub>5</sub>	28.37	28.87	29.38	26.54	32.54	30.85	32.05	28.11
As <sub>2</sub> O <sub>5</sub>	4.95	3.68	3.13	3.47	1.56	2.18	2.06	3.17
Nb <sub>2</sub> O <sub>5</sub>	bd	0.09	bd	0.13	0.10	0.10	bd	0.09
SiO <sub>2</sub>	1.34	2.12	1.40	1.71	1.20	1.60	0.77	1.29
TiO <sub>2</sub>	bd	0.03	bd	0.68	0.12	0.10	0.05	0.85
ThO <sub>2</sub>	5.91	5.92	5.62	8.67	8.04	7.24	5.27	6.56
UO <sub>2</sub>	bd	bd	bd	bd	bd	bd	bd	bd
SO <sub>3</sub>	bd	bd	bd	bd	bd	bd	bd	bd
Y <sub>2</sub> O <sub>3</sub>	8.72	7.03	5.93	7.49	5.85	7.19	5.01	5.34
La <sub>2</sub> O <sub>3</sub>	6.58	6.83	7.82	5.75	10.62	8.77	9.80	7.81
Ce <sub>2</sub> O <sub>3</sub>	4.15	3.29	3.60	3.38	2.42	2.29	2.41	2.58
Pr <sub>2</sub> O <sub>3</sub>	4.74	5.13	5.79	4.74	5.66	5.10	6.02	4.81
Nd <sub>2</sub> O <sub>3</sub>	18.19	19.34	20.13	17.41	19.05	18.63	21.66	17.89
Sm <sub>2</sub> O <sub>3</sub>	4.43	4.51	4.63	4.32	3.59	4.18	4.55	4.18
Eu <sub>2</sub> O <sub>3</sub>	0.13	0.18	0.11	0.21	0.19	0.27	0.20	0.34
Gd <sub>2</sub> O <sub>3</sub>	3.56	3.58	3.23	3.33	2.86	3.29	3.26	3.07
Tb <sub>2</sub> O <sub>3</sub>	0.43	0.33	0.39	0.42	0.31	0.38	0.34	0.37
Dy <sub>2</sub> O <sub>3</sub>	3.05	3.41	3.23	4.01	2.97	3.43	2.82	4.79
Ho <sub>2</sub> O <sub>3</sub>	0.49	0.14	0.18	0.24	0.27	0.15	0.06	0.26
Er <sub>2</sub> O <sub>3</sub>	1.45	1.00	0.93	1.08	0.87	1.09	0.79	0.93
Tm <sub>2</sub> O <sub>3</sub>	0.19	0.15	0.11	bd	0.17	0.27	bd	bd
Yb <sub>2</sub> O <sub>3</sub>	0.72	0.57	0.38	0.59	0.29	0.51	0.28	0.41
CaO	2.59	0.94	1.04	0.86	1.28	0.89	0.91	1.33
FeO*	-	-	-	-	-	-	-	-
SrO	0.07	0.08	0.10	bd	bd	0.08	0.11	bd
PbO	0.05	bd	bd	bd	0.04	0.04	bd	bd
Total	101.39	98.20	97.85	96.14	100.54	99.33	99.01	94.80

Formulae on the basis of 4 oxygens

Ca	0.113	0.042	0.047	0.041	0.056	0.040	0.041	0.063
Fe <sup>2+</sup>	-	-	-	-	-	-	-	-
Y	0.189	0.157	0.134	0.176	0.127	0.159	0.111	0.126
La	0.143	0.152	0.177	0.135	0.231	0.194	0.217	0.184
Ce	0.062	0.050	0.056	0.055	0.036	0.035	0.037	0.042
Pr	0.071	0.078	0.090	0.076	0.084	0.077	0.091	0.078
Nd	0.265	0.289	0.306	0.275	0.278	0.276	0.322	0.284
Sm	0.062	0.065	0.068	0.066	0.051	0.060	0.065	0.064
Eu	0.002	0.003	0.002	0.003	0.003	0.004	0.003	0.005

1	Gd	0.048	0.050	0.046	0.049	0.039	0.045	0.045	0.045
2	Tb	0.006	0.005	0.005	0.006	0.004	0.005	0.005	0.005
3	Dy	0.040	0.046	0.044	0.057	0.039	0.046	0.038	0.068
4	Er	0.020	0.014	0.013	0.016	0.012	0.015	0.011	0.014
5	Yb	0.009	0.007	0.005	0.008	0.004	0.006	0.004	0.006
6	Ti	0.000	0.001	0.000	0.023	0.001	0.028	0.000	0.003
7	Th	0.055	0.056	0.054	0.087	0.075	0.068	0.050	0.066
8	U	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
9	Sum	1.086	1.016	1.048	1.074	1.040	1.059	1.040	1.054
10	V	0.765	0.799	0.826	0.775	0.879	0.847	0.883	0.824
11	Si	0.055	0.089	0.060	0.076	0.049	0.066	0.032	0.057
12	As	0.106	0.081	0.070	0.080	0.033	0.047	0.045	0.074
13	P	0.044	0.035	0.026	0.042	0.019	0.025	0.021	0.023
14	Sum	0.970	1.003	0.981	0.973	0.980	0.985	0.980	0.978
15	S	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
16	$\Sigma$ cations	2.06	2.02	2.03	2.02	2.02	2.02	2.02	2.03

FeO\*, all Fe as Fe<sup>2+</sup>. bd, below detection. Dash - not determined

Wakefieldite-(Y)							
					JLT 4.1	M831	
9	10	11	12	13	14	15	16
0.55	0.68	0.94	1.01	0.62	1.95	1.16	5.73
31.34	31.11	29.72	27.93	32.78	26.3	30.01	13.43
2.32	2.8	3.35	3.70	1.45	5.24	2.56	12.80
bd	bd	bd	0.33	bd	-	-	-
0.95	1.04	1.01	0.94	0.88	1.51	5.21	3.21
bd	bd	bd	0.68	0.11	bd	bd	-
5.14	6.56	5.26	4.59	5.61	5.04	8.76	11.29
bd	bd	bd	bd	bd	bd	bd	bd
bd	bd	bd	bd	bd	bd	bd	0.10
5.34	5.59	5.30	6.31	6.13	13.91	15.02	17.11
9.66	8.74	8.16	7.15	10.39	4.22	7.99	0.47
2.45	2.47	2.82	2.86	2.23	3.96	3.21	3.20
5.96	5.43	5.98	5.45	6.01	4.16	4.30	0.28
20.78	20.27	21.50	20.02	20.27	14.54	11.96	6.82
4.36	4.61	4.86	4.14	3.87	3.43	2.05	3.71
0.26	0.16	0.16	0.12	0.06	0.14	bd	0.29
3.25	3.38	3.32	3.21	3.04	3.37	1.92	4.67
0.35	0.32	0.32	0.37	0.31	0.42	0.21	0.81
3.56	3.42	2.34	2.69	2.08	2.97	1.57	5.75
0.22	0.30	0.10	0.28	0.10	0.39	0.23	1.01
0.84	0.98	0.84	1.08	0.78	1.40	0.99	2.84
0.12	bd	0.11	0.18	0.13	0.20	0.19	0.42
0.28	0.34	0.39	0.57	0.33	1.26	0.91	2.52
0.88	0.99	1.16	0.53	1.92	4.80	1.26	0.61
-	-	-	-	-	-	-	0.05
bd	bd	0.09	0.07	0.07	bd	0.08	-
bd	bd	bd	bd	bd	0.05	0.07	-
98.61	99.19	97.73	94.21	99.17	99.26	99.66	97.12
0.040	0.044	0.053	0.025	0.084	0.210	0.052	0.030
-	-	-	-	-	-	-	0.002
0.119	0.124	0.120	0.149	0.134	0.302	0.308	0.417
0.216	0.194	0.184	0.169	0.227	0.092	0.164	0.008
0.038	0.038	0.044	0.046	0.033	0.059	0.045	0.054
0.091	0.083	0.092	0.088	0.090	0.062	0.060	0.005
0.311	0.302	0.325	0.317	0.297	0.212	0.164	0.112
0.063	0.066	0.071	0.063	0.055	0.048	0.027	0.059
0.004	0.002	0.002	0.002	0.001	0.002	0.000	0.005

1								
2								
3	0.045	0.047	0.047	0.047	0.041	0.046	0.025	0.071
4	0.005	0.004	0.004	0.005	0.004	0.006	0.003	0.012
5	0.048	0.046	0.032	0.038	0.027	0.039	0.019	0.085
6	0.012	0.014	0.012	0.016	0.011	0.019	0.013	0.044
7	0.004	0.004	0.005	0.008	0.004	0.016	0.011	0.035
8	0.001	0.001	0.001	0.023	0.003	0.000	0.000	-
9								
10	0.049	0.062	0.051	0.046	0.052	0.047	0.077	0.118
11	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
12	1.045	1.032	1.043	1.044	1.063	1.158	0.968	1.055
13								
14	0.869	0.858	0.832	0.818	0.888	0.708	0.763	0.407
15	0.040	0.043	0.043	0.042	0.036	0.062	0.201	0.147
16	0.051	0.061	0.074	0.086	0.031	0.112	0.052	0.307
17	0.020	0.024	0.034	0.038	0.022	0.067	0.038	0.111
18	0.979	0.986	0.983	0.984	0.976	0.949	1.053	0.971
19								
20								
21	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.003
22								
23								
24	2.02	2.02	2.02	2.00	2.04	2.11	2.02	2.03

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	17	18	19	20	21	22
10	3.15	3.36	5.72	5.17	4.36	4.27
11	17.26	16.46	13.99	16.69	17.12	11.76
12	10.58	10.15	11.9	8.96	10.26	15.34
13	-	-	-	-	-	-
14	3.00	3.42	3.68	4.06	2.46	2.86
15	-	-	-	-	-	-
16	12.89	14.18	10.22	10.49	11.06	11.34
17	bd	bd	bd	bd	bd	bd
18	0.05	0.06	0.10	0.06	0.08	0.18
19	13.12	12.58	16.14	15.67	13.80	15.45
20	0.64	0.70	0.46	0.50	0.68	0.81
21	4.14	4.34	3.31	3.75	3.66	2.99
22	0.53	0.50	0.33	0.49	0.60	0.85
23	9.38	8.58	7.18	8.33	9.60	10.02
24	4.46	4.07	3.80	3.86	4.49	4.52
25	0.40	0.36	0.32	0.34	0.35	0.22
26	4.89	4.55	4.94	4.34	4.85	4.89
27	0.82	0.79	0.80	0.70	0.77	0.72
28	5.27	5.00	5.70	5.25	5.27	4.92
29	0.76	0.84	0.99	0.79	0.77	0.80
30	2.23	2.31	2.63	2.41	2.20	2.21
31	0.34	0.24	0.43	0.30	0.36	0.25
32	1.71	1.69	2.16	2.05	1.68	1.80
33	0.68	0.84	0.64	0.75	0.79	0.62
34	0.43	bd	0.07	bd	bd	bd
35	-	-	-	-	-	-
36	-	-	-	-	-	-
37	96.73	95.02	95.51	94.96	95.21	96.82
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50	0.033	0.042	0.032	0.037	0.039	0.031
51	0.017	0.000	0.003	0.000	0.000	0.000
52	0.321	0.313	0.395	0.380	0.341	0.383
53	0.011	0.012	0.008	0.008	0.012	0.014
54	0.070	0.074	0.056	0.063	0.062	0.051
55	0.009	0.009	0.006	0.008	0.010	0.014
56	0.154	0.143	0.118	0.136	0.159	0.167
57	0.071	0.066	0.060	0.061	0.072	0.073
58	0.006	0.006	0.005	0.005	0.006	0.003
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1	0.074	0.071	0.075	0.066	0.075	0.075
2	0.012	0.012	0.012	0.010	0.012	0.011
3	0.078	0.075	0.084	0.077	0.079	0.074
4	0.035	0.037	0.041	0.038	0.035	0.035
5	0.024	0.024	0.030	0.029	0.024	0.026
6	-	-	-	-	-	-
7	0.135	0.151	0.107	0.109	0.117	0.120
8	0.000	0.000	0.000	0.000	0.000	0.000
9	1.049	1.036	1.032	1.026	1.043	1.077
10	0.524	0.509	0.425	0.503	0.526	0.362
11	0.138	0.160	0.169	0.185	0.114	0.133
12	0.254	0.248	0.286	0.214	0.249	0.374
13	0.061	0.067	0.111	0.100	0.086	0.084
14	0.976	0.984	0.992	1.002	0.975	0.953
15	0.002	0.002	0.003	0.002	0.003	0.006
16	2.03	2.02	2.02	2.03	2.02	2.03

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## Wakefieldite occurrences

### **Canada**

Wakefieldite-(Ce), veins in porphyritic trachyte, Osoyoos Mining Division, Yellow Lake,

British Columbia (Howard *et al.*, 1995)

Wakefieldite-(Y), in zoned granite pegmatite, near Wakefield Lake, Quebec (Miles *et al.*, 1971)

### **USA**

Wakefieldite-(Y), Huron River uranium prospect, Baraga County, Michigan (Carlson *et al.*, 2007)

Wakefieldite-(Y), lining voids in euxenite-(Y), White Tank Mountains, Maricopa County, Arizona (Anthony *et al.*, 1995)

Wakefieldite-(Nd), wakefieldite-(Y), in Joe Lott rhyolitic ignimbrite, Utah (this paper)

Wakefieldite-(Y), as rims on chernovite-(Y), Taylor Creek Tin District, Sierra County, New Mexico (Foord & Hlava, 1991)

### **Scotland**

Wakefieldite-(Ce) in reduction spots in sandstone, Gamrie Bay, Banffshire (Van Panhuys-Sigler *et al.*, 2018)

### **Germany**

Wakefieldite-(La), in hydrothermal barytes veins, Glücksstern Mine, Friedrichroda, Thuringia (Witzke *et al.*, 2008).

1 Wakefieldite-(Ce) and wakefieldite-(Nd), manganese mine, Ilfeld, Harz (Gröbner *et al.*, 2011)  
2

3 Wakefieldite-(La), Bellerberg volcano, Mayen-Koblenz, Rhineland-Palatinate (Blass, 2010).  
4

5  
6 Reported in mindat.org.: accessed 25/12/2018  
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8  
9 Wakefieldite-(Ce), Clara Mine, Rankach Valley, Baden-Württenberg (Kolitsch *et al.*, 2005)  
10

11  
12 **Austria**  
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14  
15 Wakefieldite-(Ce), wakefieldite-(Y), Obernberger Tribulaun, North Tyrol (Kolitsch *et al.*,  
16  
17 2018)  
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20  
21 **Czech Republic**  
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23  
24 Wakefieldite-(Ce,Y), in silicified wood near Studenec (Matysova *et al.*, 2016)  
25

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27  
28 *Probable* wakefieldite-(Ce), in teschenite sill, Krmelin, Frýdek-Místek, Moravia-Silesia  
29  
30 (Matýsek, 2013)  
31

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33  
34 **Italy**  
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36  
37 Wakefieldite-(La), Cerchiara Mine, La Spezia Province, Liguria (reported in mindat.org;  
38  
39 accessed 19/1/2018)  
40

41  
42 Wakefieldite-(Ce), Pianciano, Roman Volcanic Province. Fluor-ore-carbonatite units  
43  
44 interbedded with silicate volcanics (Stoppa *et al.*, 2016).  
45

46  
47 Wakefieldite-(Y), on quartz-calcite matrix, Borgata Oberti, Montaldo di Mondovi (Cadoni *et*  
48  
49 *al.*, 2011)  
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52  
53 **Morocco**  
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55  
56 Wakefieldite-(Ce), quartzose dyke?, Tifernine plateau (Baudracco-Gritti *et al.*, 1987)  
57

1           ***Egypt***  
2  
3       Wakefieldite-(Y), in silver-gold mineralization associated with uranium bearing minerals and  
4  
5       base metal sulphide, El Sheik Soliman area, south Sinai (Sallam *et al.*, 2014).  
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9           ***Democratic Republic of the Congo***  
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12       Wakefieldite-(Ce), Kobokobo pegmatite, Sud-Kivu (Deliens & Piret, 1977, 1986)  
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16           ***Namibia***  
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19       Wakefieldite-(Y), mineralized quartz vein, Garnsberg region (Bezing *et al.*, 2014)  
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22           ***Jordan***  
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25       Wakefieldite-(Ce), marbles, Tulul Al Hamman area (Khoury *et al.*, 2015)  
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29           ***Japan***  
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32       Wakefieldite-(Nd), Arase stratiform ferromanganese deposit, Kochi Prefecture (Moriyama *et*  
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34       *al.*, 2010).  
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37           ***Australia***  
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40       Wakefieldite-(Ce), diopside-titanite veins, Arkaroola, Flinders Range, South Australia  
41  
42       (Bakker & Elburg, 2006),  
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