

1 **Core Ideas**

- 2 • Wheat biomass increased by 33% when poor quality compost and Fe-WTR were
- 3 combined
- 4 • Combined amendments and control outperformed single amendments of compost and
- 5 WTR
- 6 • Bioavailable metals measured after the pot trial did not increase
- 7 • Co-application of compost with WTR may counter the negative effects of P sorption

8 **Better Together: Water Treatment Residual and Poor-Quality Compost**

9 **Improves Sandy Soil Fertility**

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17

18 **Abstract**

19 Water treatment residual (WTR) is an under-utilized clean water industry byproduct, generally
20 disposed to landfill. This study assesses the benefits and risks of ferric-WTR as a soil
21 amendment or co-amendment for plant growth in a nutrient-poor sandy soil. A 12-week pot
22 trial tested the efficacy of WTR and a locally available, low-quality, municipal compost as
23 single (1, 5, 12.5% dry mass) and co-amended treatments (1:1 WTR:compost ratio, at 2%, 10%
24 and 25%) on wheat growth in a sandy soil. The low total N content of the compost and low
25 WTR P and K contents resulted in significantly lower (up to 50% lower; $p < 0.05$) plant biomass

26 in single amendments compared to the control, while the highest co-amendment produced
27 significantly higher plant biomass (33% higher; $p < 0.05$) than the control. This positive co-
28 amendment effect on plant growth is attributed to balanced nutrient provision, with P and K
29 from the compost and N from the WTR. Foliar micronutrient and Al levels showed no toxic
30 accumulation, and co-amended foliar Mn levels increased from near deficient (20 mg/kg) to
31 sufficient (50 mg/kg). Total WTR metals were well below maximum land application
32 concentrations (USDA). Trace element bioavailability remained the same (Ni, Cu, Hg) or
33 significantly decreased (B, Al, Cr, Mn, Fe, Zn, As, Cd; $p < 0.05$) during the pot trial. These
34 results suggest, within this context, that a WTR-compost co-amendment is a promising soil
35 improvement technology for increasing crop yields in sandy soils.

36 Keywords: Fe-WTR, Waste Recycling, drinking water purification, Arenosol

37

38 **1. Introduction**

39 Water treatment residual (WTR) is a global byproduct of drinking water treatment which
40 purifies raw water to produce drinking water for municipalities. Basibuyuk and Kalat (2004)
41 reported that several million tons of WTR are produced in Europe every year, with production
42 estimated to double within the next decade. In Africa, WTR production is also set to increase
43 due to a growing population requiring increasing access to clean drinking water. WTR is most
44 commonly disposed in landfill, both globally (Basta et al., 2000) and within South Africa
45 (Herselman, 2013). Alternative uses of this waste byproduct are of global interest to water
46 companies, many of which are looking towards zero waste strategies to reduce costs, and
47 contribute to the United Nations Sustainable Development Goals (SDG 12, Responsible
48 Production and Consumption; UN, 2016).

49

50 WTR consists of flocculating agents (ferric and aluminium oxyhydroxides), de-watering agents
51 (polyelectrolytes), activated carbon, and flocculated material from the catchment dams,
52 including clay particles, microbes and dissolved organic matter (Matilainen et al., 2010). Given
53 the soil-like composition of WTR, land application is an important potential disposal option.
54 The implications of land application have been well researched (Ippolito et al., 2011). A major
55 limitation encountered with land application of WTR is the high P-fixation capacity of the Fe
56 and Al-oxyhydroxides (Elliot and Dempsey, 1991; Ippolito et al., 2003; Norris and Titshall,
57 2012). Addition of WTR to soils results in yield loss and P-deficiency symptoms in maize
58 (Rengasamy et al., 1980), lettuce (Elliott and Singer, 1988) and sorghum-Sudan grass (Heil
59 and Barbarick, 1989). Another problematic factor is the high concentrations of bioavailable Al
60 and Mn in WTR (Ippolito et al., 2011; Novak et al., 2007; Titshall and Hughes, 2005), which
61 may result in phytotoxic conditions.

62

63 Compost is commonly used to improve both chemical (fertility and phytotoxicity) and physical
64 (aggregation and water holding capacity) properties of soils. It is well-established that compost
65 addition can reduce the P sorption capacity of Al and Fe oxides in soils (Havlin et al., 2005),
66 yet the use of WTR and compost as a co-amendment is not well-researched. Hsu and Hseu
67 (2011) looked at the co-addition of a good quality (C:N ratio = 20, total N = 3.9%) compost
68 with Al-WTR. In contrast to the above-mentioned studies, they observed an increase in the
69 growth of Bahia grass with Al-WTR added as a single amendment. Co-addition of the compost
70 improved growth but not significantly. Compost increased plant available P in co-amended
71 treatments, although plant tissue P was not significantly affected. In many small-scale farming
72 systems in Africa, compost quality is often poor, with high C:N ratios and typically low total
73 N contents (Vanlauwe and Giller, 2006). Our research findings showcase the first use of a
74 ferric-WTR and poor quality compost co-amendment as a cost-effective soil improvement

75 technology to improve crop productivity through balanced nutrient provision, in sandy soils
76 from Southern Africa.

77 Sandy soils are ubiquitous throughout Africa, where despite their low fertility and low water
78 holding capacity, they support crop production in small-scale dryland systems. Dryland
79 farming in sandy soil has a high risk of crop failure due water stress, which is exacerbated in
80 nutrient-deprived plants (Steynberg et al., 1989). Infertile soils affect both plant growth and
81 human nutrition. For example, communities solely subsisting on crops grown in sandy soils in
82 Maputuland, South Africa, had elevated incidences of dwarfism and endemic osteoarthritis due
83 to nutrient deficiencies (Ceruti et al., 2003).

84

85 The Cape Flats region, just outside Cape Town, has nutrient poor, sandy soils of aeolian origin.
86 The area is predominantly occupied by low-income communities and hosts the largest informal
87 settlement in the Western Cape (Statistics South Africa, 2016). Residential urban agriculture
88 is uncommon, mainly due to lack of space, but also due to the nutrient poor soils and restricted
89 access to irrigation water. However, in a community where unemployment levels are over
90 30% (Western Cape Government, 2017), backyard vegetable gardens can provide fresh
91 produce to supplement the common maize staple. Thus, any improvement to the soils in terms
92 of increased water holding capacity and nutrient provision, could stimulate backyard
93 gardening, impacting community health and wellbeing. The Faure raw water treatment works
94 is the main supplier of potable water to the City of Cape Town, producing approximately 14
95 000 tons Fe-WTR per year (personal communication, City of Cape Town Municipality, 2018),
96 and lies physically close to the Cape Flats region. Currently Faure WTR is transported
97 approximately 50 km to a local landfill site. Therefore, if the Faure WTR could be used to
98 improve the Cape Flats soil it would be beneficial for both the municipality and the local
99 inhabitants. In this study we focus on the contamination risks and plant response of WTR

100 amendments and compare the effect of WTR and a typical low quality compost both added
101 separately and as co-amendments on plant yield, bioavailable metals and plant nutrient levels
102 in typical sandy Cape flats soil.

103 **2. Materials and methods**

104 **2.1 Sample collection and characterization**

105 Water Treatment Residual was obtained from Faure water treatment works, outside Cape
106 Town. The main storage dam, and the only reservoir supplying Faure at the time of sampling,
107 is the Theewaterskloof dam. The plant uses $\text{Fe}_2(\text{SO}_4)_3$, $\text{Ca}(\text{OH})_2$, a chemical coagulant
108 (Praestol 2540, a copolymer of acrylamide and sodium acrylate) and varying amounts of
109 activated carbon for odour control (Titshall and Hughes, 2005). The resulting WTR is a mixture
110 of ferric hydroxides, reservoir sediments, flocculated organic acids, coagulant and activated
111 carbon. Samples of WTR were collected on three dates - 28 February, 9 May and 15 May 2017.
112 During this period the Western Cape was experiencing a severe drought, and turbidity and
113 odour levels were elevated due to the increased microbial blooms. This increased coagulant
114 and activated carbon use during water purification. The three individual samples were air-
115 dried (30°C, 1 month) before being crushed to pass through a 2 mm sieve. The three individual
116 samples were chemically analysed to assess elemental variation, before being thoroughly
117 combined for re-analysis and subsequent application in incubations, chemical analyses, and
118 plant trials.

119

120 The commercially available compost used in this study is made from municipal green waste
121 (chipped garden refuse) and was used and analysed without sieving. The total C and N content
122 of the compost was analysed on a milled subsample.

123

124 The sandy soil was collected from a fallow field outside Brackenfell (Western Cape). The
125 Quartzipsamment soils of this region are typical acid variants of the Cape Flats sands. These
126 sands are windblown marine deposits, that have been leached of all carbonates, have an
127 inherently low nutrient status and are mildly acidic (Schloms et al., 1983). The top 30 cm of
128 soil was collected, air-dried and passed through a 2 mm sieve before analysis. Details of the
129 basic characterization methods and statistical analysis are provided in the Supporting
130 Information.

131 *2.1.1 Trace element content and availability*

132 Trace elements (TE) were measured in i) aqua regia (USEPA method 3015a), and ii) NH_4NO_3
133 (representing bioavailable fraction) following the DIN 19730 procedure (Herselman, 2013).
134 Extracts, prepared in triplicate, were analysed for metals using ICP-MS with an Agilent 8800
135 QQQ ICP-MS.

136 **2.2 Pre-Trial Incubation Analyses**

137 Incubation profiles of pH, EC, Mn and P were assessed, to inform application rates. Six
138 application levels (0, 10, 25, 50, 75 and 100%) of (a) WTR and (b) a 1:1 WTR-Compost
139 mixture were added on a dry weight % basis to the soil. Each air-dried sample (50 g) was wet
140 to field water capacity, covered in parafilm to prevent moisture loss and incubated at room
141 temperature ($\pm 25^\circ\text{C}$) in duplicate for two weeks. Samples were regularly weighed to confirm
142 moisture retention. Samples were analyzed post-incubation for pH, EC, Mn and P as described
143 in Supplementary Materials.

144 **2.3 Pot Trials**

145 Pot trials were set up to assess the impact of increasing application rates of WTR, compost and
146 the WTR-Compost (WTR-Comp) co-amendment on wheat growth and elemental
147 accumulation in nutrient-poor sandy soils. The application rates used were 0 (control), 1, 5
148 and 12.5% (w/w) for the single compost or WTR treatments and 0, 2, 10 and 25% (w/w) for

149 the 1:1 WTR-Comp co-amendment. All treatments were prepared in triplicate. Pots (5 L) were
150 packed to a bulk density of 1500 kg/m³. Six wheat seeds (*Triticum aestivum L.*) per pot were
151 planted and thinned to 3 plants per pot after germination. Pots were weighed and watered twice
152 a week, maintaining field water capacity. Greenhouse pot placement was randomized and
153 randomly re-organized twice during the 3-month trial. Pots were fertilized using the wheat
154 recommendation of the Fertilizer Society of South Africa (FSSA, 2007) for Western Cape
155 sandy soils (N = 130, P = 50, K = 75, Ca = 40, Mg = 13 and S = 40 kg/ha). The 500 mL fertilizer
156 concentrate was added as three applications over the 3 month trial period.

157 **2.4 Post-Trial Analyses**

158 After 3 months of growth, the pot trial was terminated. The above-ground plant material was
159 harvested by cutting the plant at soil level. Roots were weighed after soil material was removed.
160 Plant material was oven-dried (60°C) overnight and weighed per pot. Total macro- and
161 micronutrients of the dried above-ground plant material were determined using the Kjeldahl
162 method (N), and acid digestion and ICP-MS (P, Ca, Mg, K, Na, Fe, B, Zn, Mn, Cu and Al;
163 Elsenberg Plant Laboratory).

164

165 Soil from the pots was sieved (2 mm) to remove roots and air-dried. The NH₄NO₃ extractable
166 metals (see Section 2.1), were measured on the pre- and post-trial soil mixtures.

167

168 **3 Results and Discussion**

169 **3.1 WTR, Compost and Soil Characterization**

170 The properties of the sandy soil, WTR and compost are given in Supplemental Table S1. The
171 soil is mildly acid (pH_{water} = 6.5), with very low EC (64 μS/cm), total C (0.6%) and N (0.04%).
172 The P level in the soil (52 mg/kg) is above the 33 mg/kg recommended for most crops
173 (Mehlich, 1985). Bray II K levels in the soil are extremely low (9 mg/kg), falling well below

174 the recommended 50 mg/kg for winter wheat production (FSSA, 2007). The WTR has a neutral
175 pH in water (6.8) and low EC (319 $\mu\text{S}/\text{cm}$). The total C is 17%, which includes flocculated
176 dissolved organic C and the added activated carbon. The total N content of the WTR is 0.35%,
177 which is in the typical range for South African WTRs (0.02 – 0.52%), but lower than reported
178 for Faure WTR in 2005 (0.52%; Titshall and Hughes, (2005)). Thus, the severe drought had
179 not significantly increased the total N content of the WTR. The mineral N content (165 mg/kg)
180 of the WTR falls within the range of typical WTRs in South Africa (Titshall and Hughes, 2005)
181 and those reviewed by Ippolito et al. (2011). The Mehlich III P concentration in the WTR is
182 within the lower region of the range reported by Dayton and Basta (2001), between 1.6 and
183 54.4 mg/kg.

184

185 The compost has a slightly alkaline pH_{water} (7.8), very high EC (5410 $\mu\text{S}/\text{cm}$) and a relatively
186 low total C content (9.6%) for a compost. Despite an acceptable C:N ratio (25), the total N
187 content of the compost (0.38%) falls well below the 1% threshold recommended in composts
188 intended for fertilizer use (Barker, 1997). The mineral N content (7 mg/kg) of the compost is
189 also very low, falling short of that required to support crop growth (50-200 mg/kg; (Mulvaney,
190 1996)). On the other hand, the compost has ample plant available K and P (145 and 2944
191 mg/kg, respectively).

192 The aqua regia metal concentrations of the three Faure WTR samples collected at different
193 times are shown in Supplemental Table S2. Iron is the dominant metal (14-19%), with
194 substantial Al concentrations (5.3- 7.7%). Manganese is variable (0.05 – 0.29%) but lower than
195 the values reported by Titshall and Hughes (2005) for Faure WTR in 2005 (0.7 and 1.8%). The
196 source of Mn in the Faure WTR is anticipated to be from impurities in the ferric sulphate or
197 lime used during the water treatment process (Titshall and Hughes, 2005). The lower Mn values
198 measured in this study suggests that purer sources or lower quantities of these additives are

199 currently being used. With the exception of Mn, Zn and Ni, which were higher in summer
200 (February), the trace elements in the WTR do not differ substantially between sampling dates.
201 The metal concentrations of all samples are well below both the United States Environmental
202 Protection Agency (USEPA, 2000) and the more conservative South African guidelines
203 (Herselman, 2013) for the maximum allowable limits for land application.

204

205 The bioavailable metals (NH_4NO_3 extract) for the soil, composite WTR and compost are given
206 in Table 1. Prior to WTR land application in South Africa, receiving soils must be analysed for
207 bioavailable metals to assess the soils' suitability for receiving waste (Herselman, 2013). The
208 Cape Flats sand has metal concentrations far below the maximum limit permitted for soils that
209 will receive WTR (Herselman, 2013). The pure WTR had slightly elevated bioavailable Mn
210 concentrations (17 mg/kg). There are no plant micronutrient thresholds for NH_4NO_3 extracts,
211 so Mehlich III extracts of the soil, compost and WTR were analyzed. The Mn concentrations
212 in the Mehlich III extracts were 2.2 (± 0.2), 22.9 (± 0.7) and 124.0 (± 2.6) mg/kg for the soil,
213 compost and WTR, respectively. The available Mn in the soil is well below the critical
214 minimum level required for crop growth (10 mg/kg; (Sims and Johnson, 1991)) and Mn
215 deficiencies could be expected. There are no clear guidelines for phytotoxic Mn levels in soils,
216 but application of the WTR in the Cape Flats sand up to rates of 10% (w/w) would bring the
217 Mn concentrations close to the minimum critical level. The compost contained a very high
218 bioavailable As concentration (141 $\mu\text{g}/\text{kg}$), which may be due to pesticide residues in
219 municipal green waste or inclusion of treated wood in the composted material (Adriano, 2001).
220 This compost was selected for its low C and N content. The elevated As was an unexpected
221 property of the widely used compost, and although it adds an interesting aspect to the study,
222 the emphasis is on metals in the WTR, rather than metals in an inherently variable compost
223 stream.

3.2 Pre-Trial incubation studies

224
225 Prior to the pot trial design, 14-day incubations were performed at field water capacity with i)
226 WTR and ii) a 1:1 WTR-Compost co-amendment added to the sandy soil, at 6 application rates
227 between 0 and 100% (dry w/w). The results of the incubation studies (Figure 1) provide insights
228 into the effects extreme loadings of WTR and WTR-Compost co-amendments might have on
229 important soil parameters. Both WTR and the co-amendment increased pH (Figure 1a), which
230 would benefit acid soils, although increasing the pH above 7.5 is undesirable as it can result
231 in trace element deficiencies (Havlin et al., 2005). The higher pH readings in the incubation
232 studies, compared to the initial characterization (Supplemental Table S1) is assigned to longer
233 equilibrium times during the incubation. At higher loadings the 1:1 WTR-Compost co-
234 amendment exceeded 500 $\mu\text{S}/\text{cm}$ (Figure 1b) which is considered the critical EC level (in a 1:5
235 water extract) where plant growth is affected negatively (Sonmez et al., 2008). The compost is
236 likely to be the main contributor to salinity with an $\text{EC} > 5000 \mu\text{S}/\text{cm}$ (Supplemental Table
237 S1). To keep EC within tolerable levels, the 1:1 WTR-Compost co-amendment loadings should
238 be below 25%. The high P-sorption potential of the WTR is evident from the incubations
239 (Figure 1c) and increases with WTR application rate in the single amendment. However,
240 compost co-addition increases plant-available P suggesting that the compost might alleviate
241 this limitation to a degree. Bioavailable Mn concentrations increase linearly with increasing
242 loading rates (Figure 1d). These incubation results suggest that maximal application rates
243 should be kept below 25% WTR to prevent phytotoxic Mn conditions developing in the soil.
244 Based on these incubation studies the maximum WTR application rate was set at 12.5% and
245 the WTR-Comp co-amendment was set at 25%.

246

247 Figure 1 Pre-trial incubations, investigating the effect of increasing application rates of WTR and a 1:1 WTR-
248 Compost (WTR:Comp) co-amendment on (a) pH, (b) EC (c) P (Mehlich III) and d) Mn (Mehlich III). Average
249 of duplicate incubations shown.

3.3 Pot Trial: Post-Harvest Plant Physiology and Chemistry

250
251 The above- and below-ground biomass of the treatments are shown in Figure 2a and b,
252 respectively. The WTR-Comp co-amendment resulted in significantly higher (up to 33%;
253 $p < 0.05$) above-ground biomass than the control at the two highest application rates (10 and
254 25% WTR-Comp). The individual compost and WTR treatments had a significant negative
255 effect on above-ground biomass (up to 50% lower), with biomass concomitantly decreasing
256 with increasing amendment rates (Figure 2a). The below-ground biomass for the highest
257 amendment loadings showed a similar pattern, significantly lower root biomass in the single
258 amendments than the control, while the co-amended treatment showed no significant difference
259 to the control (Figure 2b).

260

261 Figure 2 The effect of single WTR (W) and compost (C) amendments, and co-amendments (WC), on plant
262 growth parameters, (a) total above-ground biomass and (b) root biomass. Bars that do not differ significantly
263 ($p < 0.05$) contain the same letter.

264 At the end of the trial, plants in all treatments except for the 12.5% WTR started to show N –
265 deficiency symptoms through older leaf yellowing and senescence despite fertilizer
266 application. Plants in the 12.5% WTR treatment did not show deficiency symptoms, due to the
267 fact that this treatment was significantly stunted (Figure 2a) and thus utilized less of the applied
268 N, which was confirmed by the leaf N-levels (Figure 3a). Although plants from all treatments
269 were well below the critical N-level (3%) for wheat (Plank and Donohue, 2000), the 12.5%
270 WTR treatment had the highest N weight percent, followed by the 5% WTR treatment. The
271 highest co-amendment (25% WTR-Comp) showed significantly higher (30%; $p < 0.05$) leaf N-
272 levels than the control, despite these plants being 33% larger. The compost amended treatments
273 all showed similar leaf N-levels to the control, although plants in the higher loadings were
274 severely stunted (Figure 2a).

275

276 Figure 3 Foliar macronutrient contents of harvested wheat plants as a weight percentage a)-c) and as absolute
277 accumulation in grams d)-f) for WTR (W), compost (C) and WTR+Compost (WC). Critical macronutrient
278 levels for wheat (Plank and Donohue, 2000) shown by red lines. Bars that do not differ significantly ($p < 0.05$)
279 contain the same letter.

280 The leaf P-levels showed the opposite trend to the N-levels, with the two lowest single WTR
281 amendments having significantly lower leaf P-levels than the control while all single
282 amendment compost treatments had significantly higher P-levels than the control (Figure 3b).
283 The two highest co-amended treatments did not show a significant difference to the control in
284 terms of P content. All treatments were above the 0.15% critical level for P in wheat (Plank
285 and Donohue, 2000), except for the two lowest WTR treatments. The slightly higher P content
286 of the 12.5% WTR treatment is attributed to smaller plant size. Potassium levels are generally
287 below the critical level of 2% (Plank and Donohue, 2000) but all treatments significantly
288 increased the K level compared to the control (Figure 3c).

289

290 The poor plant response to the compost is not surprising, considering the low total and mineral
291 N content of this material (Supplemental Table S1). The fact that the compost treatments
292 performed worse than the control suggests that N-immobilization is taking place in these
293 treatments. This is also illustrated by the total grams of N taken up by the plants (Figure 3d),
294 which shows the plants in the compost treatment assimilated the lowest amount of nitrogen
295 into their leaves. In contrast, the two highest co-amendments took up significantly more N
296 than the control or the single WTR treatments. The same trend is observed with the absolute
297 amount of P in the leaves (Figure 3e). While the single compost treatments showed the highest
298 weight % P (Figure 3b), the co-amendment treatments showed higher absolute P-levels,
299 because the biomass of these plants was greater. The same was true for K accumulation (Figure
300 3f).

301

302 When interpreting these growth response results in light of the nutrient contents in the compost
303 and WTR it is clear that both amendments are providing different macronutrients, with the
304 WTR adding mineral N while the compost contributes P and K. Although total provision of
305 nutrients by the co-amendment is likely to be the main cause of improved growth, there is also
306 the potential for the organic matter from the compost to sorb to the WTR surface and prevent
307 the fixation of added P to the oxide surfaces (Havlin et al., 2005).

308

309 The foliar micronutrient and Al levels of the wheat plants are given in Figure 4. Foliar Mn in
310 the control is at the lowest critical limit for wheat growth (Figure 4b). Addition of compost
311 with WTR at 25% had the largest effect on foliar Mn, raising the concentration to sufficiency
312 levels (20-150 mg/kg). This increase was significantly greater than addition of WTR alone,
313 indicating a synergistic effect on plant uptake of Mn in the co-amendment. Possible reasons
314 for this synergy include lowering of the redox potential in the soil and addition of Mn-
315 associated microbiomes, which may aid in Mn mobilization in the rhizosphere (Rengel, 2015).
316 Manganese is often flagged as a possible problematic metal in WTR (Novak et al., 2007;
317 Titshall and Hughes, 2005). The incubation experiments also indicate that Mn phytotoxicity
318 might be an issue at higher loadings (Figure 1d). The foliar analysis shows that even at the
319 highest levels of WTR application (12.5%), the foliar Mn concentrations were only at sufficient
320 levels and far below the toxicity threshold (380 mg/kg) for small grains (Keisling et al., 1984).
321 Thus, for nutrient poor soils, such as the Cape Flats sands, WTR-Compost co-amendments
322 could constitute an important source of Mn plant nutrition although careful monitoring under
323 field conditions would be required if repeated WTR additions were made to such a sandy soil.

324

325 Figure 4 Foliar micronutrient and Al concentrations provided with critical values for wheat production (Plank
326 and Donohue, 2000) and Al toxicity threshold (Pais and Benton Jones, 1997) for WTR (W), compost (C) and
327 WTR+Compost (WC). Bars that do not differ significantly ($p < 0.05$) contain the same letter.

328 Aluminium constitutes up to 7.7% of the WTR used in this study (Supplemental Table 2), thus
329 Al toxicity in plants was considered a potential risk when applying the material to an acid soil.
330 Only treatments with the highest loading of WTR (12.5% WTR and 25% WTR-Comp) showed
331 a significant increase in foliar concentrations (Figure 4f) and these were well below (less than
332 half) the Al toxicity level for crops (Pais and Benton Jones, 1997).

333 **3.4 Pot Trial: Bioavailable trace elements**

334 Bioavailable TE were measured before and after the pot trial on selected treatments (Table 2).
335 Before the trial B, Mn, Fe, Ni, Cu and As concentrations were significantly higher ($p < 0.05$)
336 in the 25% WTR-Comp treatment while Al, Zn and Cd concentrations were significantly lower
337 ($p < 0.05$) compared to the Before-trial control. Before-trial WTR and WTR-Comp treatments
338 had significantly lower Pb, while compost on its own had significantly higher Pb ($p < 0.05$) than
339 the Before-trial control. The higher TE bioavailability before the trial is attributed to the higher
340 TE content of the amendments (Table 1), while the lower Al, Zn and Cd availability is most
341 likely due to an increased pH in the soil system (Figure 1a).

342 When adding a waste to a soil, it is important to consider any mobilizing effects plant growth
343 might have on the bioavailability of metals, thus before and after trial comparisons were made.
344 The TE either showed no change or significantly decreased in bioavailability after the trial
345 (Table 2). For all the compost- and WTR-treated soils, extractable Mn concentrations were
346 significantly lower after the pot trial. Importantly, phytotoxic Al was not mobilized and either
347 showed little change or decreased during the trial.

348 Plant available As levels were elevated in the compost (Table 1) and, for the 12.5% compost
349 treatment, levels were beyond the threshold for soils to receive additional WTR (Herselman,
350 2013), both before and after the pot trial. Pre-trial As concentrations (11.7 $\mu\text{g}/\text{kg}$) in the 25%
351 WTR-Comp were significantly lower ($p < 0.05$) than in the pre-trial 12.5% compost treatment
352 (29.3 $\mu\text{g}/\text{kg}$). This is attributed to the capacity of WTR to strongly chemisorb As (McCann et

353 al., 2018; Sarkar et al., 2007) and suggests WTR addition to an As-rich compost could reduce
354 bioavailable As content.

355 With the exception of As in the compost treatment, the bioavailable TE measured after the
356 trial were substantially below the maximum extractable threshold for receiving soils (Table 2).
357 In addition, all TE concentrations are far below the soil screening guidelines for the protection
358 of water sources (Table 2) thus the risks of trace metal contamination of ground- and surface
359 water, even at very high WTR application rates, measured under pot trial conditions, appears
360 low. The maximum rates applied in this trial are unrealistically high (375 tons WTR + 375 tons
361 compost/ha), but suggest multiple applications of WTR at lower rates would keep TE levels
362 within guideline levels. Further work is required to establish responsible application rates for
363 WTR-Compost amendments. In addition, elevated As in the compost, highlights the
364 importance of screening the compost used as a co-amendment for known contaminants.

365 **3.5 Implications for WTR-Compost co-amendments**

366 In African small-scale farming systems, organic residues are often available but are of poor
367 quality with high C:N ratios and/or low total N (Vanlauwe and Giller, 2006). The compost used
368 in this study was of extremely poor quality, with low total C and N contents, high salinity and
369 unacceptably high As levels. Addition of WTR to this compost provided mineral N, increased
370 certain deficient trace elements and decreased the bioavailable As content, creating a more
371 favorable growth medium than compost on its own. The compost, in turn, provided K and
372 countered or reduced P-sorption tendencies of the WTR.

373

374 The mildly acidic sandy soils used in this study are ubiquitous in Africa (Jones et al., 2013)
375 and communities relying solely on these soils for food are at greater risk of malnutrition due to
376 insufficient soil micronutrients (Ceruti et al., 2003). Our results suggest that WTR-Compost
377 co-amendments show potential to improve crop productivity where the two materials are

378 abundantly available and within the context of considering transport costs versus economic and
379 social benefits of improved soil function.

380

381 The potential risks associated with land application of wastes are contamination of soil and
382 groundwater resources (Pritchard et al., 2010). Sandy soils lack clays and sesquioxides, which
383 sequester contaminants and often buffer the soil and underlying groundwater against
384 contamination. Sandy soils, especially the acid variants, are considered high risk for land
385 application of wastes (Pritchard et al., 2010). Provided that the sandy soils are well drained to
386 maintain aerobic conditions, addition of sesquioxide-rich WTR could in fact increase the buffer
387 capacity of such soils. The results obtained here suggest that even at extreme loadings (375 ton
388 WTR/ha), contamination risks from heavy metals are low, although these need to be verified
389 under field conditions using multiple WTR applications.

390 Wheat was used in this study as an indicator crop, however, in subsistence agriculture leafy
391 greens are frequently grown to supplement the maize staple. Leaf nutrition and metal uptake
392 in edible leaves needs to be determined in assessing the safety of WTR land application. In
393 addition, there are other potential toxicity risks of WTR land application, which are seldom
394 addressed. These include microbial contamination from polluted water sources, phyto-uptake
395 and toxicity of micropollutants (e.g. pharmaceuticals, pesticides, plasticizers), as well as the
396 toxicity of the chemical additives used in coagulation and flocculation. All of these risks should
397 be investigated before large-scale land applications of WTR are permitted on susceptible sandy
398 soils.

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406

407 **Supplementary Material** includes description of basic characterization methods, statistical
408 methods and characterization data for materials used

409 **References**

410 Adriano D.C. (2001) Trace elements in Terrestrial Environments Springer, New York, NY.

411 Barker A.V. (1997) Composition and Uses of Compost, Agricultural Uses of By-Products
412 and Wastes, American Chemical Society. pp. 140-162.

413 Basibuyuk M., Kalat D.G. (2004) The use of waterworks sludge for the treatment of
414 vegetable oil refinery industry wastewater. Environ. Technol. 25:373-380.

415 Ceruti P., Pooley J., Fey M.V. (2003) Soil nutrient deficiencies (Mseleni Joint Disease) and
416 dwarfism in Maputaland, South Africa, in: H. C. W. Skinner and A. R. Berger (Eds.),
417 *In Geology and Health: Closing the Gap*, Oxford University Press, New York. pp.
418 151-154.

419 Dayton E.A., Basta N.T. (2001) Characterization of drinking water treatment residuals for
420 use as a soil substitute. Water Environ Res 73:52-7.

421 DEA. (2010) Framework for the management of contaminated land. , May 2010.

422 Elliot H.A., Dempsey B.A. (1991) Agronomic effects of land application of water treatment
423 sludges. J. Am. Water Works Assoc. 84:126.

424 Elliott H.A., Singer L.M. (1988) Effect of water treatment sludge on growth and elemental
425 composition of tomato (*Lycopersicon esculentum*) shoots. Commun Soil Sci Plan
426 19:345-354. DOI: 10.1080/00103628809367943.

427 FSSA. (2007) FSSA fertilizer handbook, Fertilizer Society of South Africa.

- 428 Havlin J.L., Beaton J.D., Tisdale S.L., Nelson W.L. (2005) Soil fertility and Fertilizers: an
429 introduction to nutrient management. 7th ed. Pearson Prentice Hall, Upper Saddle
430 River, N.J.
- 431 Heil D.M., Barbarick K.A. (1989) Water Treatment Sludge Influence on the Growth of
432 Sorghum-Sudangrass. *J. Environ Qual.* 18:292-298. DOI:
433 10.2134/jeq1989.00472425001800030008x.
- 434 Herselman J.E. (2013) Guideline for the utilisation and disposal of water treatment residue.
435 Water Research commission No. 559/13.
- 436 Hsu W.M., Hseu Z.Y. (2011) Rehabilitation of a Sandy Soil With Aluminum-Water
437 Treatment Residual. *Soil Sci* 176:691-698.
- 438 Ippolito J.A., Barbarick K.A., Elliott H.A. (2011) Drinking water treatment residuals: a
439 review of recent uses. *J Environ Qual* 40:1-12.
- 440 Ippolito J.A., Barbarick K.A., Heil D.M., Chandler J.P., Redente E.F. (2003) Phosphorus
441 retention mechanisms of a water treatment residual. *J Environ Qual* 32:1857-64.
- 442 Jones A., Breuning-Madsen H., Brossard M., Dampha A., Deckers J., Dewitte O., Gallali T.,
443 Hallett S., Jones R., Kilasara M., Le Roux P., Michéli E., Montanarella L., Spaargaren
444 O., Thiombiano L., Van Ranst E., Yemefack M., Zougmore R. (2013) Soil Atlas of
445 Africa. European Commission, Publications Office of the European Union,
446 Luxembourg. 176 pp. ISBN 978-92-79-26715-.
- 447 Keisling T.C., Thompson L.F., Slabaugh W.R. (1984) Visual symptoms and tissue
448 manganese concentrations associated with manganese toxicity in wheat. *Commun*
449 *Soil Sci Plan* 15:537-540. DOI: 10.1080/00103628409367495.
- 450 Matilainen A., Vepsäläinen M., Sillanpää M. (2010) Natural organic matter removal by
451 coagulation during drinking water treatment: a review. *Adv Colloid Interfac* 159:189-
452 197.

- 453 McCann C.M., Peacock C.L., Hudson-Edwards K.A., Shrimpton T., Gray N.D., Johnson
454 K.L. (2018) In situ arsenic oxidation and sorption by a Fe-Mn binary oxide waste in
455 soil. *J Hazard Mater* 342:724-731. DOI: 10.1016/j.jhazmat.2017.08.066.
- 456 Mehlich A. (1985) Mehlich 3 soil test extractant: A modification of Mehlich 2 extractant.
457 *Commun. Soil Sci. Plant Anal.* 15:1409.
- 458 Mulvaney R.L. (1996) Inorganic Nitrogen in: D. L. Sparks (Ed.), *In Methods of Soil*
459 *Analysis*, Soil Sci. Soc. Am., Madison, Wis.
- 460 Norris N., Titshall L.W. (2012) THE distribution of inherent phosphorus in fifteen water
461 treatment residues from South Africa. *Water SA* 38:715-720.
- 462 Novak J.M., Szogi A.A., Watts D.W., Busscher W.J. (2007) Water treatment residuals
463 amended soils release Mn, Na, S, AND C. *Soil. Sci.* 172:992-1000. DOI:
464 10.1097/ss.0b013e3181586b9a.
- 465 Pais I., Benton Jones J. (1997) *The Handbook of Trace Elements* St Lucie Boca Raton,
466 Florida.
- 467 Plank C.O., Donohue S.J. (2000) Small Grain-Barley, Oats, Rye, Wheat, in: C. R. Cambell
468 (Ed.), *Reference sufficiency ranges for plant analysis in the southern region of the*
469 *United States*, Southern Cooperative Series Bulletin no. 394.
- 470 Pritchard D.L., Penney N., McLaughlin M.J., Rigby H., Schwarz K. (2010) Land application
471 of sewage sludge (biosolids) in Australia: risks to the environment and food crops.
472 *Water Sci Technol* 62:48-57. DOI: 10.2166/wst.2010.274.
- 473 Rengasamy P., Oades J.M., Hancock T.W. (1980) Improvement of soil structure and plant
474 growth by addition of alum sludge. *Commun Soil Sci Plan* 11:533-545. DOI:
475 10.1080/00103628009367061.
- 476 Rengel Z. (2015) Availability of Mn, Zn and Fe in the rhizosphere. *J Soil Sci Plant Nut*
477 15:397-409.

- 478 Sarkar D., Makris K.C., Vandanapu V., Datta R. (2007) Arsenic immobilization in soils
479 amended with drinking-water treatment residuals. *Environ Pollut* 146:414-419. DOI:
480 <https://doi.org/10.1016/j.envpol.2006.06.035>.
- 481 Schloms B., Ellis F., Lambrechts J.J.N. (1983) Soils of the coastal platform, in: H. Deacon, et
482 al. (Eds.), *Fynbos palaeoecology: A preliminary synthesis*, South African National
483 Scientific Programmes Report No 75, South African National Scientific Programmes
484 Report No 75.
- 485 Sims J.T., Johnson C.V. (1991) Micronutrient soil tests, in: J. J. Mortvedt, et al. (Eds.),
486 *Micronutrients in Agriculture*, Soil Science Society of America, Madison, WI.
- 487 Sonmez S., Buyuktas D., Okturen F., Citak S. (2008) Assessment of different soil to water
488 ratios (1:1, 1:2.5, 1:5) in soil salinity studies. *Geoderma* 144:361-369. DOI:
489 <https://doi.org/10.1016/j.geoderma.2007.12.005>.
- 490 Statistics South Africa. (2016) Provincial profile: Western Cape, Statistics South Africa.
491 Report 03-01-07.
- 492 Steynberg R.E., Nel P.C., Hammes P.S. (1989) Drought sensitivity of maize (*Zea mays* L.) in
493 relation to soil fertility and water stress during different growth stages. *SA J Plant Soil*
494 6:83-85.
- 495 Titshall L.W., Hughes J.C. (2005) Characterisation of some South African water treatment
496 residues and implications for land application. *Water SA* 31:299-307.
- 497 USEPA. (2000) Maximum concentration permitted for Land Application: Biosolids
498 Technology fact sheet Land Application of Biosolids.
- 499 Vanlauwe B., Giller K.E. (2006) Popular myths around soil fertility management in sub-
500 Saharan Africa. *Ag Ecosyst Environ* 116:34-46. DOI:
501 <https://doi.org/10.1016/j.agee.2006.03.016>.

502 Western Cape Government. (2017) City of Cape Town: Socio-Economic Profile, Western
503 Cape Government.

504

505 **Figure Captions**

506 Figure 5 Pre-trial incubations, investigating the effect of increasing application rates of WTR and a 1:1 WTR-
507 Compost (WTR:Comp) co-amendment on (a) pH, (b) EC (c) P (Mehlich III) and d) Mn (Mehlich III). Average
508 of duplicate incubations shown.

509 Figure 6 The effect of single WTR (W) and compost (C) amendments, and co-amendments (WC), on plant
510 growth parameters, (a) total above-ground biomass and (b) root biomass. Bars that do not differ significantly
511 ($p < 0.05$) contain the same letter.

512 Figure 7 Foliar macronutrient contents of harvested wheat plants as a weight percentage a)-c) and as absolute
513 accumulation in grams d)-f) for WTR (W), compost (C) and WTR+Compost (WC). Critical macronutrient
514 levels for wheat (Plank and Donohue, 2000) shown by red lines. Bars that do not differ significantly ($p < 0.05$)
515 contain the same letter.

516 Figure 8 Foliar micronutrient and Al concentrations provided with critical values for wheat production (Plank
517 and Donohue, 2000) and Al toxicity threshold (Pais and Benton Jones, 1997) for WTR (W), compost (C) and
518 WTR+Compost (WC). Bars that do not differ significantly ($p < 0.05$) contain the same letter.

519

520 **Tables**

521 Table 1 Bioavailable trace element concentrations ($\mu\text{g}/\text{kg}$) in the pot trial materials, together
 522 with threshold limits for metal concentrations in the soil where WTR will be applied
 523 (Herselman, 2013)

Element	Receiving soil limit	Soil	WTR	Compost
B		31.5	188.7	659.0
Al		208.7	60.3	2473.1
Mn		194	17000	343
Fe		126.8	130.8	1534.2
Ni	1200	3.3	94.7	19.5
Cu	1200	9.8	363.6	113.9
Zn	5000	57.6	100.0	96.3
As	14	1.7	30.1	141.3
Cd	100	0.2	0.5	0.3
Hg	7	0.03	<0.05	0.06
Pb	3500	1.0	1.4	5.1

524

525 Table 2 Trace element concentrations in 1M NH_4NO_3 extracts of selected soil treatments
 526 analyzed before and after the wheat pot trial.

Element	Receiving soil limit ^a $\mu\text{g}/\text{kg}$	Soil Screening values ^b mg/kg	$\mu\text{g}/\text{kg}$							
			Control		12.5% Compost		12.5% WTR		25% WTR+Comp	
			Before	After	Before	After	Before	After	Before	After
B			31.5	32.5	215.3	176.7	26.0	26.6	152.7*	100.8
Al			208.7	191.6	146.4*	111.6	28.3	23.9	67.5*	44.3
Mn			194.3	173.8	283.2*	101.9	3931.6*	404.8	2292.6*	473.9
Fe			126.8	110.3	371.8*	309.2	52.7	32.5	252.2*	133.2
Ni	1200	91	3.3	3.4	5.8*	4.7	8.4	7.6	9.9	9.4
Cu	1200	200	9.8	20.1	60.2	57.9	29.5	29.6	37.2	34.3
Zn	5000	3700	57.6	51.3	44.6*	33.0	19.4*	10.2	23.2*	13.6
As	14	5.8	1.7	1.9	29.3	17.4	4.8*	1.7	11.7*	4.1
Cd	100	7.5	0.16	0.18	0.13*	0.08	0.08	0.04	0.09	0.06
Hg	7	1	0.03	0.03	0.04	0.05	0.02	0.01	0.03	0.02
Pb	3500	20	1.02	1.29	1.89*	1.24	0.11	0.11	0.14	0.13

527 ^a. According to (Herselman, 2013); ^b South African Soil screening values for the protection of water sources using a dilution
 528 factor of 20 (DEA, 2010)

529 *Marks significance between Before and After concentrations at a 95% confidence limit, for comparisons made between
 530 treatments p values are provided in the text.

531