

1 Quantifying the helium and hydrocarbon accumulation processes using noble 2 gases in the North Qaidam Basin, China

3 Wen Zhang ^{a,b,c}, Yuhong Li ^{a,*}, Fenghua Zhao ^d, Wei Han ^a, Junlin Zhou ^a, Greg Holland ^e, Zheng
4 Zhou ^{b,**}

5 a. Xi'an Center, China Geological Survey, 710054, China

6 b. Lancaster Environment Centre, Lancaster University, LA1 4YQ, UK

7 c. Institute of Geology, Chinese Academy of Geological Sciences, Beijing 100037, China

8 d. College of Geoscience and Surveying Engineering, China University of Mining and Technology,
9 100083, Beijing, China

10 e. School of Earth and Environmental Sciences, The University of Manchester, M13 9PL, UK

11
12 **Abstract:** Limited reserves and insecure resource supply have led to global shortage crises in
13 helium, a vital gas for cryogenic engineering and other countless industrial manufacturing
14 processes. Despite the attention drawn by global supply disruptions, the helium accumulation
15 mechanism in natural gas fields remains poorly understood. Noble gases are excellent tracers for
16 studying migration and accumulation processes of fluids in the subsurface and can be used to
17 investigate the influence of subsurface fluids on helium accumulation. We present noble gas
18 isotope and abundance data as well as major gas compositional data from 10 producing wells in
19 three gas fields in the North Qaidam Basin, China. Helium is more concentrated in the Mabei and
20 Dongping gas fields ($2.06 - 48.4 \times 10^{-4} \text{ cm}^3 \text{ STP/cm}^3$) than in the Niudong gas field ($1.15 - 1.42 \times 10^{-4}$

* Correspondence to: No. 438, Youyi East Road, Beilin District, Xian, China.

** Corresponding author.

E-mail address: wenzhangcn@outlook.com (Wen Zhang); lyuhong@mail.cgs.gov.cn (Yuhong Li);
z.zhou4@lancaster.ac.uk (Zheng Zhou);

21 $^4\text{ cm}^3\text{ STP/cm}^3$). The helium is mainly radiogenic, with $^3\text{He}/^4\text{He}$ ratios of 0.01-0.05 Ra, where Ra
22 is the atmospheric value of $^3\text{He}/^4\text{He}$, and lacks significant contribution from the mantle (0.03-
23 0.67%). The noble gases derived from air-saturated water (^{20}Ne , ^{36}Ar , ^{84}Kr and ^{130}Xe) can be
24 explained by an oil-modified groundwater-exsolution model with excess heavy noble gases. The
25 calculated $V_{\text{oil}}/V_{\text{water}}$ and $V_{\text{gas}}/V_{\text{water}}$ ratios indicate that the Mabei region is the most oil-rich area
26 and the Dongping region has the driest natural gas, which is consistent with the geological context.
27 These ratios further support the fractionation models. The strong linear relationship between ^4He
28 and ^{20}Ne ($R^2=0.98$) suggested that ^4He was dissolved into groundwater before migrating into the
29 oil or gas phase. The initial ^4He concentrations in groundwater can accumulate within 0.31-2.78
30 Myr assuming a ^4He flux from the entire crustal section. According to the fractionation model,
31 helium in groundwater partitions into the gas phase when contacting hydrocarbons. The different
32 volume ratios among oil, gas and water during the equilibration process cause much greater
33 variability in the helium concentrations in the gas phase (e.g., 6.08×10^{-4} to $2.01\times 10^{-3}\text{ cm}^3\text{ STP/cm}^3$
34 in Mabei) than those in the groundwater phase (e.g. 9.18×10^{-3} to $1.39\times 10^{-2}\text{ cm}^3\text{ }^4\text{He STP/cm}^3\text{ H}_2\text{O}$
35 in Mabei). Hydrocarbons play a critical role in helium accumulation and dilution. Helium-rich
36 natural gas fields are characterized by old groundwater systems and moderate hydrocarbon
37 abundance. This study has succeeded in quantitatively assessing the helium accumulation process
38 in natural gas fields in the North Qaidam Basin and revealed that both groundwater and
39 hydrocarbon phases control the helium accumulation in the subsurface environment. This outcome
40 has broad implications for the prediction of hydrocarbon and helium as resources.

41

42 **Keywords:** Noble gases; Hydrocarbon charge; Helium accumulation; North Qaidam Basin

43

44 **1. Introduction**

45 Helium is an important by-product of natural gas processing and liquefied natural gas (LNG)
46 production. It is widely used in cryogenic engineering equipment, particularly the nuclear
47 magnetic resonance imaging in medical diagnostics. It also has huge applications in space
48 programmes, electronics, optical fibre manufacturing and scientific research (Cai et al., 2010;
49 Nuttall et al., 2012, Boreham et al., 2018). The world has experienced shortages of helium twice,
50 in 2006-2007 and 2013 (Ballentine, 2017; Bare et al., 2016). The imminent depletion of Bureau
51 of Land Management (BLM) helium reserves in the U.S.A. and any future unstable helium supply
52 from LNG production in Qatar are very likely to cause another “helium crisis”. Therefore, it is
53 urgent to better understand the helium accumulation mechanisms in natural gas fields.

54 ^4He is produced by the radiogenic decay of ^{235}U , ^{238}U and ^{232}Th in minerals (Ballentine and
55 Burnard, 2002). The release and diffusion of helium have been studied in many U- and Th-rich
56 minerals, such as apatite, zircon, monazite, and titanite, and it has been demonstrated that, at
57 temperatures below the closure temperature of the host mineral, helium is released from these
58 minerals over geological timescales (Cherniak and Pyle, 2008; Cherniak and Watson, 2011; Farley,
59 2000; Reiners et al., 2002). After release, helium is transported into subsurface fluid system, where
60 it is influenced by different fluids, such as oil, gas and groundwater. Previous studies suggested
61 that there was a relationship between ^4He accumulation and groundwater movement (Ballentine
62 and Lollar, 2002; Brown, 2010; Tolstikhin et al., 2017). However, despite their importance, the
63 influences of hydrocarbons and groundwater on helium accumulation remain poorly constrained.

64 Noble gases are excellent tracers for a variety of subsurface fluid processes in hydrothermal,
65 volcanic and petroleum systems (Ballentine et al., 2002; Barry et al., 2018; Cao et al., 2018; Darrah
66 et al., 2015; Gilfillan et al., 2008; Holland et al., 2013). Due to their inert natures, noble gases are

67 rarely influenced by chemical reactions and biological processes (Ozima and Podosek, 2002).
68 Noble gases in subsurface fluid systems can be classified as atmospheric, mantle or radiogenic
69 components, each of which has distinct isotopic signatures (Ballentine et al., 2002). Air-derived
70 noble gases (ADGs, e.g., ^{20}Ne and ^{36}Ar) dissolved into groundwater in the recharge area, are
71 transported to the subsurface and are fractionated by the solubility differences in different phases.
72 This fractionation allows an investigation into the nature of the different phases, the relative
73 volumes involved and the migration processes that occur in the subsurface (Ballentine et al., 1991;
74 Barry et al., 2016; Zartman et al., 1961; Zhou et al., 2005). In addition, ADGs have been applied
75 in groundwater protection studies to identify the source of fugitive gas contamination in an aquifer
76 (Darrah et al., 2014; Wen et al., 2017) and to estimate the influence of enhanced oil recovery on
77 water resources (Barry et al., 2018). Radiogenic noble gases (e.g., ^4He , $^{21}\text{Ne}^*$, and $^{40}\text{Ar}^*$) have
78 also been used to constrain the residence time of gas reservoirs and groundwaters in the subsurface
79 (Holland et al., 2013; Pinti et al., 2011; Zhou and Ballentine, 2006; Zwahlen et al., 2017).

80 Since noble gases are a powerful tool to trace groundwater movement and hydrocarbon
81 charging in the subsurface, which have major impacts on helium accumulation, the noble gas data
82 from natural gas fields in the North Qaidam Basin can provide an excellent opportunity to address
83 a number of key issues regarding helium and hydrocarbon accumulation processes in gas fields,
84 such as the presence of oil and gas phases, the extent of phase equilibrium in open or closed
85 systems, the volume ratios of different phases involved in the system (oil, gas and water), and the
86 quantification of the influence of groundwater and hydrocarbons on helium enrichment and
87 dilution.

88

89

90 **2. Geological background**

91 The Qaidam Basin, which has an area of approximately 121,000 km², is located in the
92 northern part of the Qinghai-Tibet Plateau in China. Several gas and oil fields (e.g., Dongping,
93 Niudong, Lenghu, Pingtai, Mabei, Nanbaxian, and Yuka) have been discovered in the North
94 Qaidam Basin in the last 60 years (Fig. 1a). The predominant hydrocarbon source rocks in this
95 area are Lower-Middle Jurassic strata, which are mainly classified as Type II-III and are composed
96 of dark or grey-black mudstones, carbonaceous mudstones and coal (Li et al., 2016; Ma et al.,
97 2015; Tian et al., 2017). Unlike the Niudong field, no hydrocarbon source rocks are present
98 beneath the Mabei and Dongping gas fields (Fig. 1b), suggesting that the natural gas in the Mabei
99 and Dongping fields migrated from adjacent depressions. The carbon isotope values of methane
100 suggest that the maturity of the hydrocarbon source rocks in the North Qaidam Basin varies with
101 vitrinite reflectance (R_o), which ranges from 0.85% to 4.76% (Tian et al., 2017). The gas reservoirs
102 are mainly in the Xiaganchaigou Formation (E₃) in Mabei and in Jurassic strata (J₁) and the
103 Ganchaigou Formation (E₃) in Niudong. Furthermore, the majority of the gas in Dongping is
104 distributed in bedrock, which mainly consists of Neoproterozoic and early Palaeozoic granite and
105 metamorphic rocks (Ma et al., 2015). Oil was mainly generated in the Miocene period (N₁), and
106 gas was mainly generated after the Pliocene (N₂) (Luo et al., 2013).

107

108 **3. Methodology**

109 **3.1 Sample collection**

110 Gas samples (n=10) were collected in high-pressure stainless-steel cylinders from producing
111 wellheads in three gas fields in the North Qaidam Basin: the Dongping gas field (n=3), the Mabei
112 gas field (n=4), and the Niudong gas field (n=3). The high-pressure cylinders were flushed with

113 the produced gas for at least 5 min to evacuate air in the connecting pipes and all dead space prior
114 to sample collection. Once samples were shipped to the laboratory in Xi'an Center, China
115 Geological Survey, gas samples were decanted from the cylinders and transferred to refrigeration-
116 grade copper tubes with an external diameter of 10 mm. A two-stage pressure regulator was used
117 to step down the pressure from the well stream to 1-2 bars as the gas flowed into the copper tubes.
118 The copper tubes were flushed with the produced gas for 10 min before being sealed with stainless-
119 steel clamps (Zhang et al., 2019). The remaining gas in the cylinders was shipped to the Key
120 Laboratory of Petroleum Resource Research, Chinese Academy of Sciences, Lanzhou, for major
121 gas composition analysis and a complete suite of hydrocarbon gas composition tests. A MAT271
122 mass spectrometer (MS) and an on-line continuous flow gas chromatograph (GC) were used for
123 these compositional analyses, as described in detail in Zhang et al. (2018). The analysis errors
124 were less than 1 vol% for CH₄, CO₂ and N₂ and less than 10 vol% for other gases. The major gas
125 component results of two samples (Mabei 1 and Dongping 3) were not acquired due to shipping
126 loss.

127 **3.2 Analytical techniques**

128 Noble gas analyses were conducted in the subsurface fluid isotope geochemistry laboratory
129 at the Lancaster Environment Centre, Lancaster University. Before gas was released from the
130 copper tube to the high-vacuum prep line, a quadrupole MS (Hiden Analytical HAL-201) was
131 used to detect any possible leak and eliminate air contamination during testing. First, samples were
132 expanded into a calibrated volume, the pressure of which was detected using an MSK manometer.
133 Then, the samples were expanded into the rest prep line for the purification and separation
134 processes. Specifically, samples were expanded onto a Ti-sponge getter held at 800°C and then
135 cooled to room temperature for 20 mins to remove reactive gases and hydrocarbons, followed by

136 further cleaning using a hot getter (SAEC GP50) held at 250°C for 15 mins. Noble gases were
137 separated by a charcoal cold finger refrigerated by liquid nitrogen (trapping Ar, Kr and Xe) and a
138 Janis cryogenic trap (trapping He and Ne).

139 All noble gases (He, Ne, Ar, Kr and Xe) were sequentially released from the cold finger and
140 cryogenic trap. Two cold getters held at room temperature (SAES GP50) were used to further
141 remove hydrogen prior to expanding the gas into the NGX noble gas mass spectrometer (IsotopX).
142 Due to the overlap of release temperature, Ar and Kr were released simultaneously but measured
143 separately by releasing the gas stored in different segregated lines into the NGX. The concentration
144 and all isotopic ratios of the five noble gases were determined. Doubly charged $^{40}\text{Ar}^{++}$ and CO_2^{++}
145 were measured to correct ^{20}Ne and ^{22}Ne following the methods of Niedermann et al. (1993).

146 Blanks were measured once per week following the same procedures as used for the samples.
147 The average blank corrections for ^4He , ^{20}Ne , ^{40}Ar , ^{84}Kr and ^{130}Xe were 0.29 vol.%, 0.05 vol.%,
148 1.28 vol.%, 0.06 vol.% and 0.57 vol.%, respectively. An air standard was analyzed each day to
149 ensure the stability and reproducibility of the system.

150

151 **4. Results**

152 **4.1 Major gas compositions**

153 Major gas compositions from 8 wells are shown in Table 1. The produced gases primarily
154 consist of methane (C_1), ranging from 76.64 vol% to 95.18 vol%, with minor longer-chain
155 hydrocarbons (C_2 and C_3). Gas dryness ($\text{C}_1/\text{C}_{1-5}$) (Lorant et al., 1998) is an indicator of
156 hydrocarbon maturity because longer-chain hydrocarbons tend to crack into short-chain molecules
157 at a higher temperature or over longer time periods (Behar et al., 1997). Gas dryness was calculated
158 to be 0.97 in Dongping, 0.89-0.91 in Niudong, and 0.85-0.96 in Mabei, suggesting that the

159 Dongping gas field has the driest and most mature gas. In addition, $\delta^{13}\text{C}_{\text{C1}}$ can also reflect the
160 hydrocarbon maturity because bonds with ^{12}C have a higher crack propensity than those with ^{13}C ,
161 endowing the ^{12}C isotope with a tendency to partition into hydrocarbons formed in the early stage
162 of hydrocarbon generation. The gas dryness is consistent with the $\delta^{13}\text{C}_{\text{C1}}$ values in previous studies,
163 ranging from -19.8‰ to -28.5‰ in Dongping, -35.3‰ to -28.4‰ in Mabei and -35.8‰ to -30.9‰
164 in Niudong (Tian et al., 2018; Zhou et al., 2016). The gas samples also include minor proportions
165 of N_2 (1.15-8.87%), CO_2 (0.10-0.53%) and O_2 (0.02-0.19%).

166 **4.2 Noble gases**

167 He, Ne, Ar, Kr and Xe concentrations and isotopic ratios are given in Table 2.

168 Helium is more concentrated in Mabei and Dongping than in the Niudong gas field. The ^4He
169 concentration ranges from 6.08 to $20.14 \times 10^{-4} \text{ cm}^3/\text{cm}^3$ at standard temperature and pressure (STP)
170 in the Mabei gas field, from 2.06 to $48.4 \times 10^{-4} \text{ cm}^3 \text{ STP}/\text{cm}^3$ in the Dongping gas field, and from
171 1.15 to $1.42 \times 10^{-4} \text{ cm}^3 \text{ STP}/\text{cm}^3$ in the Niudong gas field (Table 2). Helium isotopic ratios are
172 0.024-0.049 Ra in Mabei and 0.010-0.017 Ra in Niudong and Dongping, where Ra is the helium
173 isotopic ratio in the atmosphere ($\text{Ra}=1.4 \times 10^{-6}$, Mamyrin and Tolstikhin, 1984). Using a simple 2-
174 endmember mixing model between sub-continental lithospheric mantle-like (6.1 Ra, Gautheron
175 and Moreira, 2002) and upper crustal (0.008 Ra, Ballentine and Burnard, 2002) endmembers, the
176 mantle contributions of helium are only 0.27-0.67% in Mabei and 0.03-0.15% in Dongping and
177 Niudong, showing that helium in the three gas fields is mainly radiogenic.

178 The ^{20}Ne concentrations in the three gas fields range from 3.45×10^{-9} to $6.62 \times 10^{-8} \text{ cm}^3$
179 STP/cm^3 . Measured $^{20}\text{Ne}/^{22}\text{Ne}$ ratios vary between 8.75 ± 0.14 and 9.96 ± 0.16 , showing minor
180 deviations from the atmospheric value of 9.8. $^{21}\text{Ne}/^{22}\text{Ne}$ ratios vary from 0.0328 ± 0.0009 to
181 0.0504 ± 0.0012 , slightly higher than the value of air (0.029). Neon isotope ratios can be explained

182 by mass fractionation and nucleogenic crustal ^{21}Ne and ^{22}Ne addition (Fig. 2). Nucleogenic ^{21}Ne
183 and ^{22}Ne are mainly generated by ^{17}O , ^{18}O (α , n) ^{20}Ne , ^{21}Ne ; ^{19}F (α , n) ^{22}Na (β^+) ^{22}Ne ; and ^{19}F (α ,
184 p) ^{22}Ne in the crust, and the typical crustal endmember shown in fig.2 has a $^{21}\text{Ne}/^{22}\text{Ne}$ value of
185 approximately 0.47 at the $^{20}\text{Ne}/^{22}\text{Ne}$ zero intercept (Kennedy et al., 1990, Ballentine and Burnard,
186 2002). Mabei and Dongping samples accumulate more nucleogenic ^{21}Ne than do Niudong samples.

187 The samples have ^{36}Ar concentrations ranging from 1.89 to $21.56 \times 10^{-8} \text{ cm}^3 \text{ STP/cm}^3$ and
188 significantly different $^{40}\text{Ar}/^{36}\text{Ar}$ ratios. The $^{40}\text{Ar}/^{36}\text{Ar}$ ratios, ranging from 1726 ± 36 to 2019 ± 45 in
189 Mabei, 883 ± 16 to 2862 ± 75 in Dongping, and 393 ± 6 to 432 ± 8 in Niudong, differ markedly from
190 the atmospheric $^{40}\text{Ar}/^{36}\text{Ar}$ ratio of 298.56 ± 0.31 (Lee et al., 2006). Due to the little mantle
191 contribution indicated by the $^3\text{He}/^4\text{He}$ ratios in Section 4.2.1, the excess ^{40}Ar was likely generated
192 by the radiogenic decay of potassium in the crust.

193 The ^{84}Kr concentrations range from 0.89 to $8.57 \times 10^{-9} \text{ cm}^3 \text{ STP/cm}^3$. The measured Kr
194 isotopic ratios are indistinguishable from the atmospheric ratio. The $^{86}\text{Kr}/^{84}\text{Kr}$ ratio ranges between
195 0.301 ± 0.0053 and 0.310 ± 0.0058 , similar to that of air (0.3035, Aregbe et al., 1996). Furthermore,
196 the ^{130}Xe concentrations range from 2.08 to $12.00 \times 10^{-11} \text{ cm}^3 \text{ STP/cm}^3$, and the Xe isotopic ratios
197 are also similar to those of air. Specifically, the $^{132}\text{Xe}/^{130}\text{Xe}$ ratios of 6.41 ± 0.298 to 6.74 ± 0.364 are
198 similar to the atmospheric value (6.61, Pepin, 2000).

199

200 **5. Discussion**

201 **5.1 Fractionation processes of atmosphere-derived noble gases**

202 The noble gases, such as ^{20}Ne , ^{36}Ar , ^{84}Kr and ^{130}Xe , are predominantly derived from air,
203 without significant contributions from radiogenic, nucleogenic or fissiogenic sources (Ballentine
204 and Burnard, 2002). These gases originally dissolved into surface water under recharge conditions

205 and moved into the subsurface during aquifer recharge (Ballentine et al., 2002). The initial noble
206 gas solubility in water increases with mass ($\text{Ne} < \text{Ar} < \text{Kr} < \text{Xe}$) and is controlled by the recharge
207 conditions, such as temperature, salinity, and recharge elevation. When groundwater encounters
208 hydrocarbons (e.g. during the hydrocarbon filling process), each of these elements is partitioned
209 into the oil or gas, and this process is controlled by Henry's law (Barry et al., 2017; Wen et al.,
210 2017; Zhou et al., 2005). The characteristics of the atmospheric noble gases in these fluids are
211 assumed to be conservative due to their inert chemical properties and limited subsurface sources.
212 Therefore, atmosphere-derived noble gases (^{20}Ne , ^{36}Ar , ^{84}Kr and ^{130}Xe) and their ratios are used
213 to model the history of multi-component (i.e., water, gas and oil) interactions within conventional
214 and unconventional hydrocarbon systems (Ballentine et al., 1991; Barry et al., 2018; Byrne et al.,
215 2018; Zhou et al., 2005).

216 We assume a temperature of 10°C, a salinity of 0M NaCl, and an elevation of sea level as the
217 recharge conditions for the three gas fields. The reservoir conditions of the Mabei, Dongping and
218 Niudong gas fields differ significantly due to the various depths of the reservoirs, which are
219 approximately 1000m, 3000m and 2000m, respectively (one exception is the Dongping 3 well
220 with a depth of ~2000 m) (Zhou et al., 2016). The temperatures and pressures were calculated
221 based on an average geothermal gradient of 3.4°C/100 m and a pressure gradient of 13.3 bar/100m.
222 Additionally, the groundwater salinity is assumed to be 2.02 M NaCl equivalent (~118 g/L on
223 average). The oil density varies from 0.78-0.83 g/cm³ and is assumed to be 0.80 g/cm³ for
224 simplification. The geothermal gradient, pressure gradient, groundwater salinity and oil density
225 were measured by the Exploration and Development Research Institute, PetroChina Qinghai Oil
226 Field Company. Henry's constants of the noble gases in water and oil were calculated from
227 empirical equations under the above parameters and corrected with fugacity coefficients and

228 activity coefficients (Ballentine et al., 2002, Dymond and Smith, 1980, Smith and Kennedy, 1983).
229 Henry's constants used in the following models are listed in Table 3.

230 **5.1.1 Single-stage phase fractionation (gas-water equilibrium)**

231 A single-stage phase equilibrium postulates that noble gases in the gas phase are directly
232 degassed from groundwater during gas-groundwater equilibrium. Ne is more prone to degassing
233 from groundwater than is Ar due to the lower solubility of Ne in water, meaning that the gas phase
234 in equilibrium with water has a higher $^{20}\text{Ne}/^{36}\text{Ar}$ ratio than that in the water phase. Regardless of
235 whether the system is closed or open, the highest $^{20}\text{Ne}/^{36}\text{Ar}$ ratio occurs in the first bubble
236 equilibrated with groundwater under reservoir temperature and pressure (RTP) conditions, and
237 these values are 0.310, 0.223 and 0.250 in the Mabei, Dongping and Niudong gas fields,
238 respectively, except for the Dongping 3 well (0.250). The $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in the gas phase decreases
239 with an increase in the volume ratio of gas to water, which approaches 0 in an open system and
240 becomes infinitely close to the value in the original air-saturated water (ASW) in a closed system
241 (0.143). We observed that the measured $^{20}\text{Ne}/^{36}\text{Ar}$ ratios in Niudong gas (0.120-0.133) were lower
242 than the minimum ratio of 0.143 in the gas phase in a closed system, which can only be explained
243 by gas-water fractionation in an open system.

244 The influence of the variation in recharge conditions on the models have been considered. 1)
245 When the salinity of recharge water is higher than the assumed value in this study (0 M NaCl), the
246 initial $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in ASW and the lowest $^{20}\text{Ne}/^{36}\text{Ar}$ ratio equilibrated into gas phase in a closed
247 system are higher than 0.143, which is higher than part of the measured $^{20}\text{Ne}/^{36}\text{Ar}$ ratio (0.120-
248 0.291), therefore, only the open system fractionation model could explain the data. 2) As the
249 recharge temperature decreased gradually, a recharge temperature of 0 °C would result in a
250 minimum $^{20}\text{Ne}/^{36}\text{Ar}$ ratio partitioning into the gas phase in a closed system of 0.122, which could

251 explain the measured data ($0.120 \pm 0.002 - 0.291 \pm 0.006$) within errors. However, at the recharge
 252 temperature of 0°C , the highest $^{20}\text{Ne}/^{36}\text{Ar}$ ratio that could occur in the gas phases in Mabei both
 253 in a closed and an open system decreases to 0.250, which could not explain the measured values
 254 in this area (0.210-0.291). Therefore, a recharge temperature of 0°C could not be used in the
 255 modelling. Based on the considerations above, a recharge temperature of 10°C was used together
 256 with the Rayleigh fractionation law to model the gas-water equilibrium process in an open system
 257 at the Qaidam Basin (Ballentine et al., 2002 and references therein):

$$258 \quad \left(^{20}\text{Ne}/^{36}\text{Ar}\right)_{\text{water}} = \left(^{20}\text{Ne}/^{36}\text{Ar}\right)_{\text{ASW}} f^{\alpha-1} \quad (1)$$

$$259 \quad \alpha = \frac{K_{\text{Ne}(\text{water})}^d}{K_{\text{Ar}(\text{water})}^d} \quad (2)$$

260 where $\left(^{20}\text{Ne}/^{36}\text{Ar}\right)_{\text{water}}$ is the $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in the groundwater phase after equilibrium;
 261 $\left(^{20}\text{Ne}/^{36}\text{Ar}\right)_{\text{ASW}}$ is the $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in the original air-saturated water (ASW); f is the
 262 fraction of ^{36}Ar remaining in the groundwater phase, and α is the fractionation coefficient for the
 263 gas/groundwater system. $K_{\text{Ne}(\text{water})}^d$ and $K_{\text{Ar}(\text{water})}^d$ are dimensionless Henry's constants of Ne
 264 and Ar in groundwater. As an infinitesimally small amount of gas equilibrates with groundwater,
 265 the $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in the gas phase after equilibrium $\left(^{20}\text{Ne}/^{36}\text{Ar}\right)_{\text{gas}}$ is given by:

$$266 \quad \left(^{20}\text{Ne}/^{36}\text{Ar}\right)_{\text{gas}} = \left(^{20}\text{Ne}/^{36}\text{Ar}\right)_{\text{water}} \alpha \quad (3)$$

267 Based on the measured $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in the gas phase, the ^{36}Ar remaining in the
 268 groundwater phase (f) can be calculated by Equations (1)-(3). Then, the ^{36}Ar concentration in the
 269 gas phase $\left(^{36}\text{Ar}\right)_{\text{gas}}$ (m^3/m^3) can be calculated as follows:

$$270 \quad \left(^{36}\text{Ar}\right)_{\text{gas}} = \left(^{36}\text{Ar}\right)_{\text{ASW}} f K_{\text{Ar}(\text{water})}^d / Z \quad (4)$$

271 where $\left(^{36}\text{Ar}\right)_{\text{ASW}}$ is the ^{36}Ar concentration in ASW ($\text{m}^3 \text{STP}/\text{m}^3$); the other parameters are
 272 the same as those in Equations (1)-(3). Z is the compression factor that converts noble gas
 273 concentrations from STP (273.15 K at 1 bar) to RTP and is given by Barry et al. (2016):

274 $Z = 273.15P_R/T_R$ (5)

275 where P_R is the reservoir pressure in atm and T_R is the reservoir temperature. Based on the
276 measured $^{20}\text{Ne}/^{36}\text{Ar}$ ratio, the calculated ^{36}Ar concentrations in the gas phase of Mabei range from
277 5.56 to $7.35 \times 10^{-7} \text{ cm}^3 \text{ STP/cm}^3$, which is 6.9-9.2 times the measured ^{36}Ar concentrations in natural
278 gas. Similarly, the calculated ^{36}Ar concentrations in the gas phase are 2.8-11.0 and 1.6-3.3 times
279 the measured ^{36}Ar concentrations in the Dongping and Niudong gas fields, respectively. This
280 suggests that the gas-groundwater fractionation in an open system cannot explain the measured
281 ^{20}Ne and ^{36}Ar concentrations. Therefore, there is likely another process that can dilute the air-
282 derived noble gases (ADGs) before or after gas-water equilibration. In addition, the $^{20}\text{Ne}/^{36}\text{Ar}$ ratio
283 in the Dongping 3 well (0.287) is higher than the modelling maximum value (0.250) in the gas
284 phase at Dongping RTP, indicating that single-stage gas-groundwater fractionation cannot explain
285 the measured ADGs characteristics.

286 **5.1.2 Two-stage phase fractionation (oil-modified groundwater-exsolution)**

287 Based on the theory of hydrocarbon generation, all types of hydrocarbon source rocks,
288 including the type II-III in the North Qaidam Basin, can produce both oil and gas phases (Yang et
289 al., 1985). Furthermore, long-chain hydrocarbons (oil) are generated in an earlier stage than
290 methane (Tissot and Welte, 1984). Because the Mabei and Niudong gas fields are distributed in
291 oil and gas areas and Mabei is near the Nanbaxian oil field (Fig. 1a), we sought to characterize the
292 noble gas in the natural gas by oil-groundwater fractionation followed by gas-groundwater
293 equilibrium, similar to the oil-modified groundwater exsolution (OMG-E) model in Zhou et al.
294 (2012) and the double distillation model in Battani et al. (2000). In this work, we follow the OMG-
295 E model, which suggests that an oil phase is equilibrated with the groundwater during the early
296 stage of hydrocarbon generation, this process is then followed by subsequent equilibrium between

297 gas and the fractionated groundwater.

298 Involvement of the oil phase can explain two discrepancies in the single-stage gas-
299 groundwater equilibration. 1) The measured ^{20}Ne and ^{36}Ar concentrations are lower than the
300 predicted concentrations. Because noble gases are more soluble in oil than in water, the
301 equilibration between an oil phase and groundwater results in the transfer of noble gases from the
302 groundwater to the oil. Therefore, the process reduces the noble gas concentrations in the
303 groundwater before gas-groundwater fractionation, resulting in less noble gas degassing into the
304 gas phase in the gas-groundwater equilibration process. 2) The measured $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in the
305 Dongping 3 sample is higher than the predicted maximum value. Relative to Ne, Ar is more soluble
306 in oil than in water (Ballentine et al., 2002); therefore, the equilibration between oil phase and
307 groundwater phase increases the $^{20}\text{Ne}/^{36}\text{Ar}$ ratio in the water phase and increases the $^{20}\text{Ne}/^{36}\text{Ar}$
308 ratio fractionation limit in the gas phase in contact with the water phase.

309 Since hydrocarbon source rocks are distributed in depressions rather than under the reservoirs
310 in the Mabei and Dongping areas, the gas and oil must migrate laterally and vertically for several
311 kilometres from adjacent depressions (Fig. 1). The long migration distance supported our model
312 that the equilibrium among gas, oil and water proceeds in an open system from initial stage for
313 both oil-water and gas-water systems. The two stages of Rayleigh oil-water fractionation and
314 Rayleigh gas-water fractionation in an open system are described by the following equations):

$$315 \quad (^{20}\text{Ne}/^{36}\text{Ar})_{gas} = (^{20}\text{Ne}/^{36}\text{Ar})_{ASW} f_1^{(\alpha_1-1)} f_2^{(\alpha_2-1)} \alpha_2 \quad (6)$$

$$316 \quad (^{36}\text{Ar})_{gas} = (^{36}\text{Ar})_{ASW} f_1 f_2 K_{Ar(water)}^d / Z \quad (7)$$

317 where $(^{20}\text{Ne}/^{36}\text{Ar})_{gas}$ and $(^{20}\text{Ne}/^{36}\text{Ar})_{ASW}$ are the $^{20}\text{Ne}/^{36}\text{Ar}$ ratios in the gas phase and
318 original groundwater (ASW), respectively; f_1 and f_2 are the fractions of ^{36}Ar remaining in the
319 groundwater phase after oil-water and after gas-water equilibrium, respectively; and Z is the

320 compression factor given by Equation (5). α_1 and α_2 are the Ne-Ar fractionation coefficients for
 321 the oil/water and gas/water systems, given by:

$$322 \alpha_1 = \frac{K_{Ne(water)}^m K_{Ar(oil)}^m}{K_{Ne(oil)}^m K_{Ar(water)}^m} \quad (8); \quad \alpha_2 = \frac{K_{Ne(water)}^m}{K_{Ar(water)}^m} \quad (9)$$

323 where the superscript m means that Henry's constant (K) is expressed in the unit of atm
 324 kg/mol. The modelling results (Fig. 3) suggest that the data from the Mabei and Niudong gas fields
 325 show a similar trend to that of the modelled line and fit the gas-groundwater fractionation in an
 326 open system well.

327 However, the data from Dongping are not consistent with the model line in an open system
 328 (not shown in Fig.3); therefore, we also modelled the situation considering gas-water fractionation
 329 in a closed system to figure out the best explanation for data in Dongping area. We still assume
 330 that the oil-water fractionation proceeds in an open system due to the long-distance migration of
 331 hydrocarbon. Oil-water fractionation in an open system and gas-water fractionation in a closed
 332 system are described by the following equations:

$$333 \left({}^{20}Ne / {}^{36}Ar \right)_{gas} = \left({}^{20}Ne / {}^{36}Ar \right)_{ASW} f_1^{(\alpha_1 - 1)} \frac{\frac{V_{gas}}{V_{water}} + \frac{1}{K_{Ar(water)}^d}}{\frac{V_{gas}}{V_{water}} + \frac{1}{K_{Ne(water)}^d}} \quad (10)$$

$$334 \left({}^{36}Ar \right)_{gas} = \left({}^{36}Ar \right)_{ASW} f_1 \left(\frac{V_{gas}}{V_{water}} + \frac{1}{K_{Ar(water)}^d} \right)^{-1} / Z \quad (11)$$

335 where V_{gas}/V_{water} is the volume ratio of gas to water under RTP and all other parameters are
 336 the same as those in the above equations. Solving Equations (10) and (11), we can obtain the f_1
 337 and V_{gas}/V_{water} values. Since the Dongping 3 well has a reservoir depth of approximately 2000 m,
 338 which is different from the other two wells in this area (3000 m), the Dongping 3 well is modelled
 339 separately. The modelling results show that the data from Dongping are well explained by the
 340 model of oil-water fractionation in an open system and gas-groundwater fractionation in a closed
 341 system (Fig. 3).

342 Although Fig. 3 provides a good explanation for the measured data, we also investigated
 343 other modelling situations to test whether the measured Mabei and Niudong data could fit the
 344 fractionation model that was used to explain the Dongping data. Our results showed that this model
 345 is less consistent with the Mabei data compared to the model line in an open system. It also cannot
 346 explain the Niudong data since the data are distributed below the boundary line (the gas-water
 347 equilibrium line when the V_{oil}/V_{water} ratio equals 0). Therefore, we use the model of oil-water and
 348 gas-water fractionation both in an open system to explain the Mabei and Niudong data and the
 349 model of oil-water fractionation model in an open system and gas-water in a closed system to
 350 interpret the measured Dongping values.

351 5.1.3 Volume ratio of oil, gas and groundwater

352 The relative oil, gas and groundwater amounts in the system can be quantified based on the
 353 models discussed in previous section. For oil-water and gas-water Rayleigh fractionation model
 354 in an open system, step-wise oil-water and gas-water equilibration can approximate Rayleigh
 355 fractionation if the volume ratios V_{oil}/V_{water} and V_{gas}/V_{water} are low in each stage (Zhou et al., 2005).
 356 Equations for V_{oil}/V_{water} and V_{gas}/V_{water} in a single equilibration process between oil and water
 357 phases at STP are given by Battani et al. (2000) and Ballentine et al. (2002), respectively:

$$358 \frac{V_{oil}}{V_{water}} = \frac{\rho_{water} K_{Ar}^m(oil)}{\rho_{oil} K_{Ar}^m(water)} \left(\frac{1-f_1}{f_1} \right) \quad (12)$$

$$359 \frac{V_{gas}}{V_{water}} = \frac{Z(1-f_2)}{f_2 K_{Ar}^d(water)} \quad (13)$$

360 where V_{oil} , V_{gas} , and V_{water} are the volumes of oil, gas and groundwater, respectively, in the
 361 oil-gas-water system under RTP and ρ_{oil} and ρ_{water} are the densities of oil and water, which are 0.80
 362 g/cm³ and 1 g/cm³, respectively. The other parameters are the same as those in the equations above.
 363 We assume that a small volume of oil (gas) equilibrates with the groundwater in each individual
 364 stage ($V_{oil}/V_{water}=V_{gas}/V_{water}=3 \times 10^{-5}$, Zhou et al., 2005); then, the fraction of ³⁶Ar remaining in the

365 groundwater phase in each step ($f_{1single}$ and $f_{2single}$) can be calculated by Equations 12 and 13.
366 Furthermore, the numbers of individual stages of oil/water (n_1) and gas/water (n_2) equilibration
367 can be calculated by $f_1=f_{1single}^{n_1}$ and $f_2=f_{2single}^{n_2}$. The sum of the volume ratios of oil (gas) to
368 water defines the total volume ratio of oil (gas) to water involved in the oil-water and gas-water
369 fractionation processes.

370 For the model of oil-water fractionation in an open system and gas-water fractionation in a
371 closed system, the V_{gas}/V_{water} ratio at RTP can be obtained by solving Eqs. (10) and (11), and the
372 V_{gas}/V_{water} ratio at STP can further be obtained through multiplying by the compression factor Z .
373 The V_{oil}/V_{water} ratio in this model can also be calculated by Eq. (12). The results are listed in Table
374 4.

375 The ratios among the gas, oil and groundwater in the OMG-E model agree well with the
376 geological context, which is independent evidence for the credibility of the model. The modelled
377 V_{gas}/V_{water} ratio of the Dongping 3 well (0.71) is lower than the values of other wells in the three
378 gas fields (1.09-39.15), consistent with the lower gas production and higher water production in
379 the Dongping 3 area (Li et al., 2014). In addition, the model results reveal that the V_{gas}/V_{oil} ratios
380 range from 75.97 to 151.48 in Niudong and from 292.97 to 1081.28 in Dongping (except for
381 Dongping 3=17.20), while the ratios vary between 9.93 and 13.36 in Mabei. These ratios agree
382 well with the oil reservoirs discovered in the Mabei gas field (Fig. 1b) and the lower gas dryness
383 of Mabei (0.76-0.98) compared to those of Niudong (0.89-0.90) and Dongping (0.95-0.99) (Tian
384 et al., 2018; Zhou et al., 2016 and this study). These results suggest that Mabei is the most oil-rich
385 area among these three gas fields. Although some gas samples in Mabei are characterized by high
386 coefficients of gas dryness (up to 0.98), these findings are not contradictory to the modelled low
387 V_{gas}/V_{oil} ratio in Mabei because pockets of oil could fail to be preserved if oil migration occurred

388 earlier than the development of the trapping structure.

389 Assuming the groundwater quantities in the three gas fields are similar, the V_{oil}/V_{water} and
390 V_{gas}/V_{water} ratios can provide a means to evaluate the production of gas and oil in the gas fields.
391 Except for that of Dongping 3, the V_{oil}/V_{water} ratios indicate that the generated oil quantity in Mabei
392 is approximately 2.9 and 3.9 times of those in the Niudong and Dongping gas fields, respectively.
393 Similarly, the V_{gas}/V_{water} ratios show that the generated gas amount in Mabei is only 36.2% and
394 5.7% of those in Niudong and Dongping. Therefore, the Dongping area is the most favourable area
395 for gas accumulation, especially the No. 5 block, where the Dongping 171 well is located.

396 **5.1.4 Heavy noble gas enrichments**

397 The heavy noble gases (Kr and Xe) in the samples are characterized by air-like isotopic
398 compositions, lacking the accumulation of fissiogenic Kr and Xe. Based on the OMG-E model
399 used in this study, the calculated Kr and Xe concentrations in the gas phase partitioned from ASW
400 are only 16-67% and 0.14-30% of the measured values. It is uncertain whether the released Kr and
401 Xe equilibrated with deep meteoric formation waters before partitioning into the oil and gas phase,
402 which complicates the explanation of the measured Kr and Xe data (Barry et al., 2018). Therefore,
403 the solubility-controlled fractionation model cannot reasonably explain the measured data.
404 Because heavy noble gases are preferentially absorbed on organic sediments and siliceous fossils
405 relative to He, Ne and Ar (Podosek et al., 1981, Matsuda and Nagao, 1986), when hydrocarbons
406 are generated, heavy noble gases are released and migrate into the fluid system, providing an
407 additional contribution to the heavy noble gas concentrations other than that provided by ASW
408 (Barry et al., 2016; Torgersen and Kennedy, 1999; Zhou et al., 2005). Excess heavy noble gas
409 enrichment has been demonstrated to be a common phenomenon in coalbed methane, gas fields,
410 oil fields, shale gas systems, and CO₂ gas fields (Barry et al., 2018; Gilfillan et al., 2008; Wen et

411 al., 2017; Zhou et al., 2005).

412 **5.2 Helium accumulation process**

413 Commercial helium is often exploited from CH₄ gas fields; however, helium concentrations
414 vary in a big range among gas fields globally (Ballentine and Lollar, 2002; Boreham et al., 2018
415 and references therein). This fact gives rise to an interesting question: what are the critical
416 processes controlling helium accumulation in natural gas fields? In this work, we aim to
417 understand this question based on the example of the gas fields in the North Qaidam Basin.

418 **5.2.1 Relationship between helium and groundwater**

419 The ³He/⁴He ratios (0.01-0.05 Ra) in the North Qaidam Basin indicate that helium in the gas
420 fields is mainly derived from the radiogenic decay of uranium and thorium in the crust. ²⁰Ne in
421 natural gas is derived from ASW due to lack of other sources in the crust and mantle (Ballentine
422 and Burnard, 2002). Therefore, the relationship between ⁴He and ²⁰Ne can reflect the influence of
423 groundwater on helium accumulation. As shown in Fig. 4, the linear relationship between ⁴He and
424 ²⁰Ne indicates that ⁴He was transported together with ²⁰Ne by groundwater before migrating into
425 the oil or gas phase. Because He and Ne have similar solubilities in oil and water, solubility-
426 controlled fractionation has little effect on the ⁴He/²⁰Ne ratio. Therefore, the ⁴He/²⁰Ne ratio is only
427 controlled by the ratio between quantity of the involved groundwater to the amount of helium
428 production. A similar relationship between ⁴He and ²⁰Ne is also shown in the Hugoton-Panhandle
429 gas field (Ballentine and Lollar, 2002), southwestern Kansas (Danabalan, 2017), and the Weihe
430 Basin, China (Zhang et al., 2019).

431 In addition, taking the Mabei gas field as an example, the solubility of He under reservoir
432 conditions is approximately 0.724 cm³STP/g, which is much higher than the calculated initial ⁴He
433 concentrations in groundwater (9.18-13.88×10⁻³ cm³STP/cm³ H₂O, see Section 5.2.2). This

434 suggests that helium should have dissolved in groundwater before equilibrating with the oil and
 435 gas phases.

436 **5.2.2 Initial ⁴He concentrations in the groundwater and dilution of helium by CH₄ in the** 437 **reservoir**

438 The initial concentration of ⁴He (as well as ²¹Ne* and ⁴⁰Ar*) in groundwater before oil-gas-
 439 water interaction can be estimated based on the measured ⁴He (or ²¹Ne*, ⁴⁰Ar*) concentration in
 440 the gas phase and the fractionation process. Taking ⁴He as an example, the initial ⁴He concentration
 441 in groundwater in the Mabei and Niudong gas fields (Rayleigh oil-water fractionation and
 442 Rayleigh gas-water fractionation) is given by:

$$443 \quad ({}^4\text{He})_{\text{initial}} = \frac{({}^4\text{He})_{\text{gas}} Z}{K_{\text{He}(\text{water})}^d f_1^{\alpha'_1} f_2^{\alpha'_2}} \quad (14)$$

$$444 \quad \alpha'_1 = \frac{K_{\text{He}(\text{water})}^m K_{\text{Ar}(\text{oil})}^m}{K_{\text{He}(\text{oil})}^m K_{\text{Ar}(\text{water})}^m}; \quad (15) \quad \alpha'_2 = \frac{K_{\text{He}(\text{water})}^m}{K_{\text{Ar}(\text{water})}^m} \quad (16)$$

445 The initial ⁴He concentration in the groundwater in the Dongping gas field (open oil-water
 446 fractionation and closed gas-water fractionation) is given by:

$$447 \quad ({}^4\text{He})_{\text{initial}} = \frac{({}^4\text{He})_{\text{gas}} Z}{f_1^{\alpha'_1}} \left(\frac{V_{\text{gas}}}{V_{\text{water}}} + \frac{1}{K_{\text{He}(\text{water})}^d} \right) \quad (17)$$

448 where the parameters in Equations (14)-(17) are the same as those in Equations (6), (7), (12),
 449 and (13). α'_1 and α'_2 are the He-Ar fractionation coefficients for the oil/water and gas/water
 450 systems, respectively. The results show that He was more concentrated in the groundwater of
 451 Mabei and Dongping ($0.92\text{-}1.40 \times 10^{-2} \text{ cm}^3 \text{ }^4\text{He STP/cm}^3 \text{ H}_2\text{O}$) than in Niudong ($2.07\text{-}3.21 \times 10^{-3}$
 452 $\text{ cm}^3 \text{ }^4\text{He STP/cm}^3 \text{ H}_2\text{O}$) (Table 4). These values agree with the higher ⁴⁰Ar/³⁶Ar ratios in the Mabei
 453 and Dongping fields (883-2862) than those in the Niudong field (393-432). ²¹Ne* and ⁴⁰Ar* have
 454 similar patterns as ⁴He, which are discussed in Section 5.3. The reason for the lower initial ⁴He
 455 concentration in the groundwater in Niudong is discussed in Section 5.2.3. Here, we focus on the

456 variation in the ^4He concentrations in the gas and groundwater in Mabei and Dongping, which can
457 be measured by the coefficient of variation (C.V.). This parameter is defined as follows:

$$458 \text{ C. V.} = \frac{\sigma}{\mu} \quad (18)$$

459 where σ is the standard deviation and μ is the mean value of ^4He concentrations in gas or
460 groundwater phase. Results suggest that the C.V.s of ^4He contents in gas phases in Mabei and
461 Dongping are 0.39 and 1.10 respectively, higher than those of initial ^4He concentrations in
462 groundwater in the two gas fields (C.V.=0.17 and 0.27, respectively). It suggests, despite the large
463 variation in the measured ^4He concentrations in the gas phase in Mabei and Dongping, the
464 estimated initial ^4He concentrations in groundwater vary to a less extent. The ^4He variation in the
465 gas phase in Mabei and Dongping can be explained by different CH_4 volumes present in the system,
466 i.e., CH_4 reduces the ^4He concentrations in the gas phase. The major gases (e.g., CH_4) have a
467 significant dilution effect on ^4He and other noble gases partitioned into the gas phase in a similar
468 fashion to the dilution effect of CH_4 on ^{36}Ar in the shale gas system (Byrne et al., 2018).

469 5.2.3 Accumulation of He in groundwater

470 Helium is mainly produced by the radiogenic decay of ^{235}U , ^{238}U and ^{232}Th in rocks.
471 Subsequently, several mechanisms control the release of helium from rock and minerals (e.g.,
472 recoil, diffusive loss, fracturing and mineral transformation). Thus, the He accumulation in
473 groundwater is contributed by in situ production and an external flux (Barry et al., 2017; Torgersen
474 and Clarke, 1985). The in situ accumulation and external flux of ^4He in groundwater are given by
475 Torgersen (1980) and Zhou and Ballentine (2006) respectively:

$$476 \left[^4\text{He} \right]_{\text{in situ production}}^{\text{groundwater}} = \frac{\rho \Lambda J_4 (1-\phi)}{\phi} t \quad (19)$$

$$477 \left[^4\text{He} \right]_{\text{external flux}}^{\text{groundwater}} = \frac{J_4 \rho H}{\phi h} t \quad (20)$$

478 where ρ is the density of porous rock in g/cm^3 ; A is the ^4He transfer efficiency from the rock
479 matrix to the groundwater, which is assumed to be 1; ϕ is the rock porosity; and t is the
480 groundwater residence time in years. Additionally, H is the average thickness of the helium source
481 rock in km; h is the aquifer thickness in km; and J_4 is the radioactive production of ^4He in
482 groundwater in $\text{cm}^3 \text{ STP } ^4\text{He}/\text{g}_{\text{rock}} \text{ year}$, given by Craig and Lupton (1976):

$$483 \quad J_4 = 0.2355 \times 10^{-12} [U]\{1 + 0.123([Th]/[U] - 4)\} \quad (21)$$

484 [U] and [Th] are the U and Th concentrations, respectively, in rocks in ppm. Based on
485 Equations (19) and (21), we assume that ^4He in groundwater is accumulated between the
486 deposition of the Eocene layers (56 Ma) and the beginning of the hydrocarbon filling process (23
487 Ma); then, the maximum in situ ^4He production in groundwater is calculated to be $7.80\text{-}16.4 \times 10^{-4}$
488 $\text{cm}^3 \text{ } ^4\text{He STP}/\text{cm}^3 \text{ H}_2\text{O}$. The value is one order of magnitude lower than the calculated initial ^4He
489 concentration in groundwater; therefore, the contribution of an external ^4He flux from the crust
490 should be considered in this case.

491 An external ^4He flux from the entire crust (upper crust and lower crust) is used to calculate
492 the time needed for groundwater to accumulate the observed concentration of helium. We choose
493 the average U and Th concentrations of acidic rock (3.5 and 18 ppm, respectively, Vinogradov and
494 Ryabchikov, 1962) as the values of the upper crust because of the granitic-metamorphic basement.
495 The U and Th concentrations of the lower crust are taken to be 0.28 and 1.07ppm respectively
496 (Zhou and Ballentine, 2006). Results show that the ^4He accumulation period in Mabei is the
497 longest (1.84-2.78Ma), followed by Dongping (1.07-2.11Ma) and Niudong (0.31-0.35Ma). This
498 suggests that the groundwater in Niudong is younger than that in Mabei and Dongping, which is
499 why the calculated initial ^4He concentrations in groundwater in the former are lower than those in
500 the latter.

501 **5.2.4 Insignificant contribution of ⁴He to the gas reservoirs after their formation**

502 The ⁴He contribution to the gas reservoir after gas filling can be derived from the external
503 flux and in situ production. Since groundwater is widespread in the crust and ⁴He in the gas
504 reservoir has a close relationship with groundwater (see Section 5.2.1), the external ⁴He flux from
505 the deep crust should dissolve into groundwater in the first stage. Because it is less possible for
506 the gas reservoir to equilibrate with large-scale groundwater after hydrocarbon filling, in which
507 some ⁴He is partitioned into the gas reservoir from the groundwater phase, the external ⁴He flux
508 contribution after gas filling can be ignored. The in-situ production of ⁴He (cm³ STP/cm³) after
509 the formation of reservoir can be calculated using the following equation after Byrne et al. (2018):

$$510 \quad [{}^4\text{He}]_{in-situ}^{gas} = \frac{\rho \Delta J_4 (1-\phi) t}{\phi Z} \quad (22)$$

511 where Z is the compression factor of CH₄ under reservoir pressure and temperature conditions
512 given by Equation 5 and accounts for the dilution of produced ⁴He by pressurized CH₄. The other
513 parameters are the same as those in Equations (19)-(21). Results show that the accumulated ⁴He
514 concentration in the gas reservoir after the hydrocarbon filling (~2.58 Myr in Dongping and
515 Niudong area, Cao et al., 2013) is only (1.70-6.18)×10⁻⁷ cm³ /cm³, which is 3-4 orders of
516 magnitude less than the measured ⁴He concentration in the gas phase. This observation suggests
517 that the contribution of in situ ⁴He production to the reservoir after hydrocarbon filling is negligible.
518 The relationship between ⁴He and ²⁰Ne (Fig. 4) indicates that the majority of ⁴He is migrated from
519 groundwater to the gas phase along with ²⁰Ne rather than directly from the rocks to the gas phase,
520 which further demonstrates that there was little ⁴He contribution to the gas reservoir after
521 hydrocarbon filling.

522 **5.2.5 Other crust-derived noble gases**

523 In addition to ^4He , ^{21}Ne and ^{40}Ar also have significant contributions from the crust. The
524 explicit crust-derived components for Ne and Ar ($^{21}\text{Ne}^*$ and $^{40}\text{Ar}^*$) are corrected for atmospheric
525 contributions by assuming a pure atmospheric source for ^{20}Ne and ^{36}Ar . The in situ production
526 ratios of $^4\text{He}/^{40}\text{Ar}^*$ and $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ are calculated to be 7.5 and 3.19×10^{-7} , respectively, on the
527 basis of average U, Th, and K concentrations of 3.5 ppm, 18 ppm, and 3.34%, respectively, in
528 acidic rock (Vinogradov and Ryabchikov, 1962) and an average crustal $^4\text{He}/^{21}\text{Ne}^*$ ratio of
529 2.33×10^7 (Ballentine and Burnard, 2002). As He and Ne have similar solubilities in water, the
530 linear relationship between the measured $^4\text{He}/^{40}\text{Ar}^*$ and $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ values (Fig. 6a) indicates that
531 the scattered $^4\text{He}/^{40}\text{Ar}$ ratios are controlled by solubility-dependent fractionation. The crust
532 derived noble gas concentrations in the initial groundwater are calculated based on the OMG-E
533 model. Results (Fig. 6b) show that there are two sources for ^4He , $^{21}\text{Ne}^*$ and $^{40}\text{Ar}^*$: the first source
534 features low $^4\text{He}/^{40}\text{Ar}^*$ ratios (4.16-6.15) and low $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ ratios ($1.30\text{-}2.21 \times 10^{-7}$), which are
535 slightly lower than the in situ production ratios; the second source has high $^4\text{He}/^{40}\text{Ar}^*$ ratios (14.4-
536 17.9) and low $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ ratios ($4.31\text{-}6.90 \times 10^{-7}$). Although the $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ ratios in the second
537 group have large variations, the values are consistent within error. This suggests that the scattered
538 $^4\text{He}/^{40}\text{Ar}^*$ and $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ values in Fig. 6a are likely evolved from the two sources in Fig. 6b.

539 A relationship between ^4He and $^{21}\text{Ne}^*$ (the slopes of the fitting lines in Fig. 6) is expected
540 because ^4He is produced by the radiogenic decay of U and Th and Ne is generated by nucleogenic
541 routes, which are controlled by the amount of α -particles, which in turn is controlled by U and Th
542 concentrations. The different $^4\text{He}/^{40}\text{Ar}^*$ ratios may be caused by the local elemental ratio between
543 U, Th and K or the preferential release of He compared to Ar (Ballentine and Burnard, 2002).

544 **5.3 He accumulation processes in the North Qaidam Basin**

545 Based on the findings in previous sections, we summarized the helium accumulation
546 processes in the North Qaidam Basin in the following 5 steps (Fig. 5).

547 (1) Helium in the North Qaidam Basin is mainly derived from crust and generated by
548 radiogenic decay of U and Th, which are suggested by the $^3\text{He}/^4\text{He}$ ratios of 0.011-0.049 Ra.

549 (2) Majority of the produced helium is released and preserved in groundwater. The close
550 relationship between ^4He and ^{20}Ne suggests that the majority of ^4He in gas reservoir is exsolved
551 from groundwater to the gas phase along with ^{20}Ne rather than directly migrated from the rocks to
552 the gas phase.

553 (3) Helium concentrations in the groundwater increase with the residence time of the water.
554 The initial ^4He contents in the groundwater in Mabei and Dongping ($2.07\text{-}3.21 \times 10^{-3} \text{ cm}^3 \text{ STP/cm}^3$)
555 are much higher than those in Niudong ($0.92\text{-}1.40 \times 10^{-2} \text{ cm}^3 \text{ STP/cm}^3$), which takes 1.07-2.78Ma
556 and 0.31-0.35Ma, respectively to accumulate when assuming a ^4He flux from the entire crust. The
557 younger groundwater in Niudong gas field than those in Mabei and Dongping gas fields cause the
558 lower He concentrations in the former than the latter.

559 (4) Helium in the groundwater is degassed by oil and gas and subsequently transported into
560 reservoir along with the hydrocarbons. In the North Qaidam Basin, oil and gas phases
561 subsequently contact with groundwater and partition helium in groundwater into the hydrocarbon
562 phases (oil-modified groundwater-exsolution model). In this process, different volumes of oil, gas
563 and groundwater were involved and caused various ^4He concentrations in gas phases (C.V. = 0.39
564 and 1.10 in Mabei and Dongping) compared to those in initial groundwater (C.V. = 0.17 and 0.27
565 in Mabei and Dongping). Dongping 171 produces a higher gas quantity ($V_{\text{gas}}/V_{\text{water}}=39.15$) and a
566 lower ^4He concentration ($(^4\text{He})_{\text{gas}}=0.021\%$) compared to Dongping 3 ($V_{\text{gas}}/V_{\text{water}} = 0.71$ and
567 $(^4\text{He})_{\text{gas}} = 0.484\%$), suggesting that, although the existence of gas phase is a necessary condition

568 to partition helium out of groundwater, a large amount of hydrocarbons can dilute the helium
569 concentration in the gas phase, which reduces the commercial value of the reservoir for producing
570 helium gas.

571 (5) There is insignificant helium contributed to the gas reservoir after the reservoir formation
572 due to the absence of large-scale gas-water contact after hydrocarbon filling.

573 In summary, a gas field associated with old groundwater systems and filled by moderate
574 amount of hydrocarbon gases are more likely to accumulate high concentrations and amounts of
575 helium for exploration purposes.

576

577 **6. Conclusions**

578 We present high-precision noble gas (He, Ne, Ar, Kr and Xe) isotope and abundance data as
579 well as major gas compositional data from 10 producing wells in the Mabei, Dongping and
580 Niudong gas fields in the North Qaidam Basin, China. These data are used to quantifiably
581 investigate helium accumulation mechanism and hydrocarbon charging processes in a water-oil-
582 gas system. Results can provide insights and guidance in exploring helium as a strategic resource.

583 $^3\text{He}/^4\text{He}$ ratios show that there is little contribution from the mantle source, indicating that
584 the majority of noble gases in the gas reservoirs are derived from ASW and the crust. The ASW-
585 derived noble gases (^{20}Ne , ^{36}Ar , ^{84}Kr and ^{130}Xe) can be accounted for by the OMG-E model with
586 excess heavy noble gases. The data from Mabei and Niudong can be interpreted as open-system
587 fractionation between oil and water followed by gas and water fractionation. The Dongping data
588 can be explained by equilibration in an open oil/water system followed by a closed gas/water
589 system. The calculated $V_{\text{oil}}/V_{\text{water}}$ ratios and $V_{\text{gas}}/V_{\text{water}}$ ratios suggest that Mabei is the most oil-
590 rich area and Dongping has the driest natural gas, which is consistent with the geological context

591 and supports our fractionation model.

592 The linear relationship between ^4He and ^{20}Ne indicates that ^4He is dissolved into the
593 groundwater and mixed with ^{20}Ne before in contact with the oil or gas phase. The estimated initial
594 ^4He concentrations in the groundwater are less variable (C.V.=0.39 and 1.10 for Mabei and
595 Dongping, respectively) than the measured ^4He concentrations in the gas phase (C.V.=0.17 and
596 0.27, respectively), indicating that CH_4 has a significant dilution effect on ^4He and other noble
597 gases partitioned into the gas phase. The groundwater residence times are calculated to be 0.31-
598 2.78 Myr assuming an external ^4He flux derived from the entire crust. The groundwater in Niudong
599 is younger than that in Mabei and Dongping, which is consistent with the lower He concentrations
600 in the Niudong gas field than those in the Mabei and Dongping gas fields. The calculation based
601 on He production suggests that little ^4He was contributed to the gas reservoir after filling of the
602 hydrocarbons in the reservoir. Based on the OMG-E model, the $^4\text{He}/^{40}\text{Ar}^*$ and $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ ratios
603 in the gas phase are derived from two sources, which are ultimately controlled by the in situ
604 production ratio.

605 The helium accumulation process can be summarised into 5 steps. 1) Helium is generated by
606 radiogenic decay of U and Th in the crust. 2) Majority of helium is released and dissolved into
607 groundwater present in rock fractures. 3) Groundwater with a long residence time can accumulate
608 high amount of He. 4) Helium in the groundwater partitions into oil and gas and is subsequently
609 transported into the hydrocarbon reservoir, where hydrocarbons have a dilution effect on the
610 helium concentration in the gas phase. 5) There is insignificant contribution of He to the reservoir
611 after the formation of gas reservoir. It is more likely to succeed in helium exploration at gas fields
612 associated with old groundwater systems and moderate hydrocarbon reserve.

613

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621

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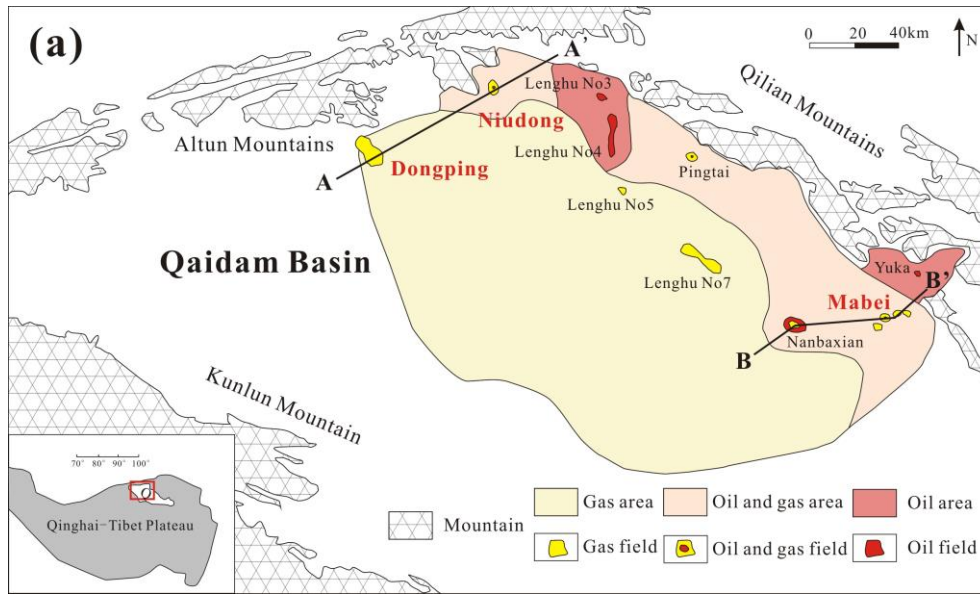
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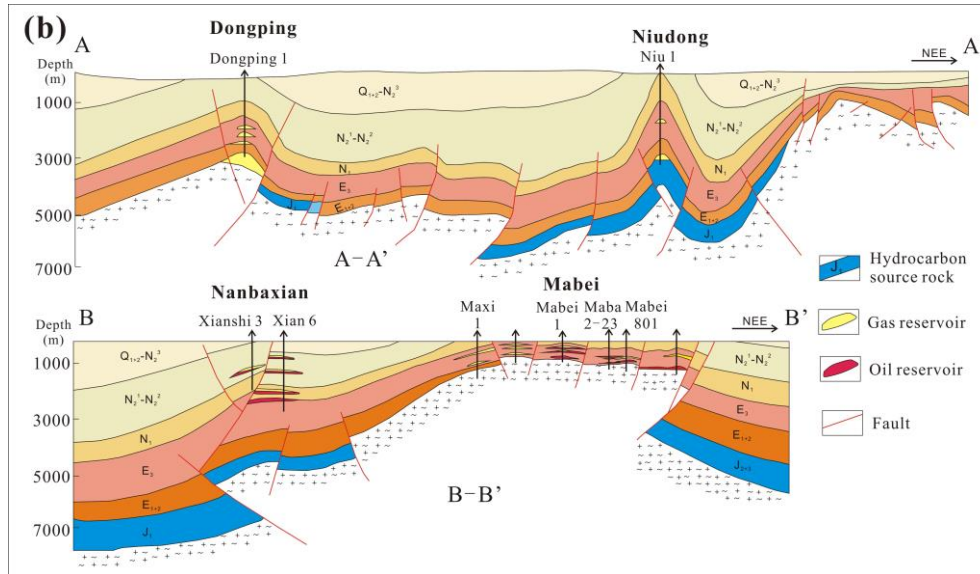
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1 **Figures**

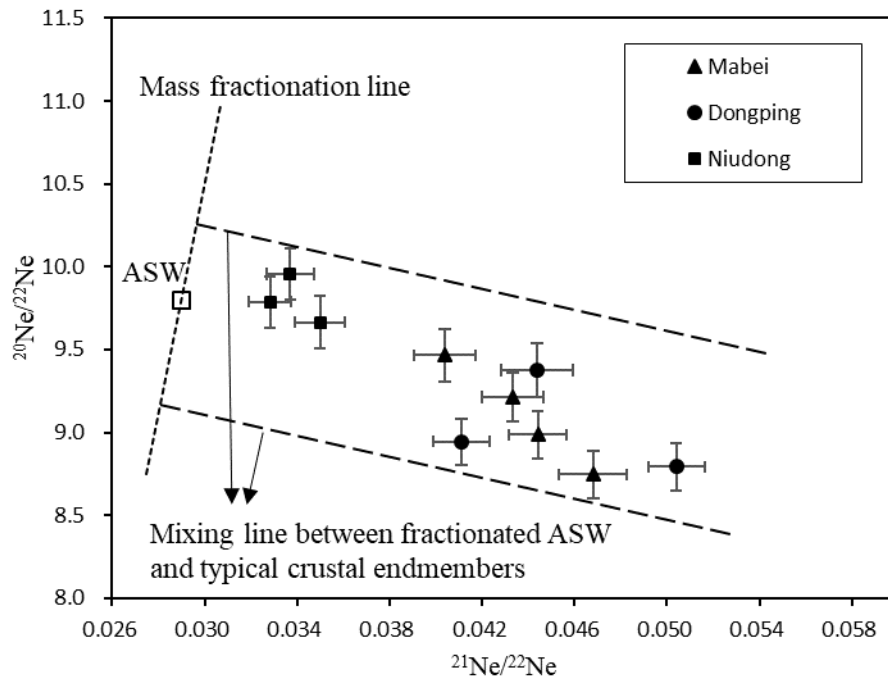


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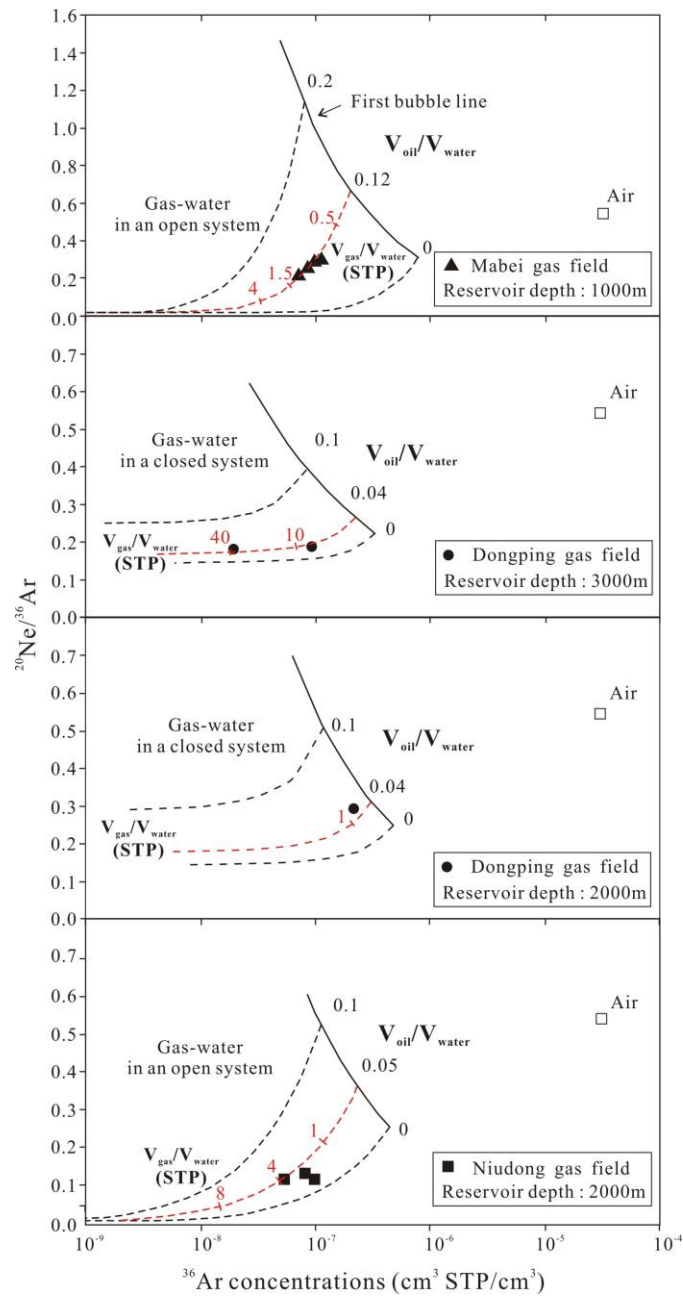
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4 **Fig. 1.** (a) Geological map of the North Qaidam Basin. Gas samples were collected from the Mabei,
 5 Dongping and Niudong gas fields, which are shown in bold red font. (b) Geological cross sections
 6 of the Dongping-Niudong gas field (A-A') and the Nanbaxian-Mabei oil and gas field (B-B'). The
 7 sampled production wells in Mabei field are labelled in the B-B' cross section. Since the sample
 8 locations are close to each other in both the Dongping and Niudong gas fields, only representative
 9 wells (Dongping 1 and Niu 1) are shown in the A-A' section.

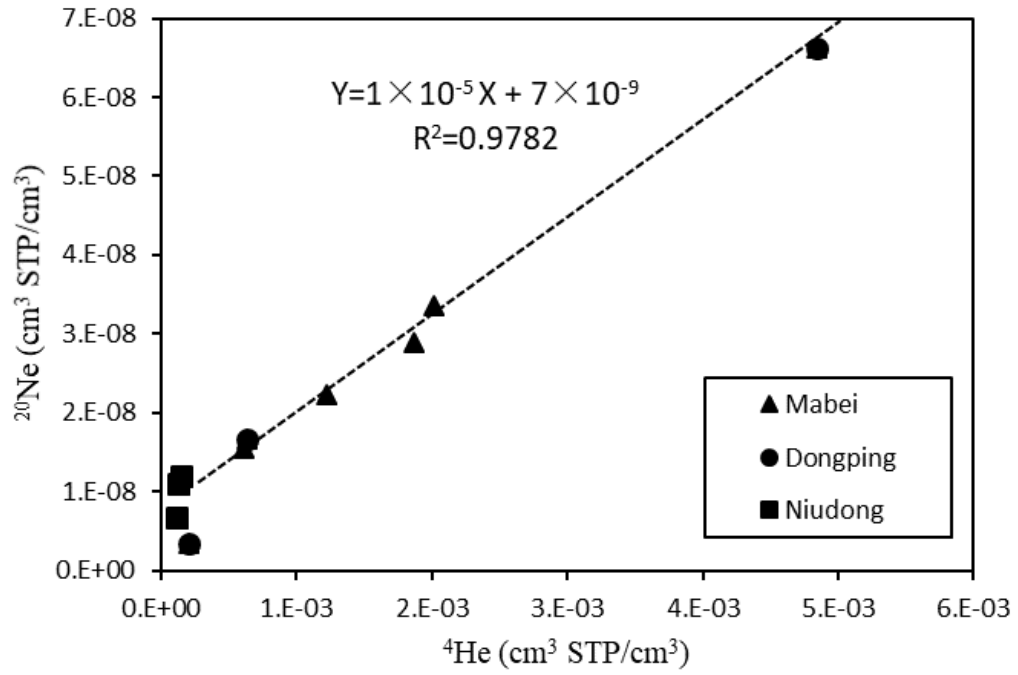


10
 11 **Fig. 2.** Ne isotopic ratios ($^{20}\text{Ne}/^{22}\text{Ne}$ vs. $^{21}\text{Ne}/^{22}\text{Ne}$) in produced gases from the Mabei, Dongping
 12 and Niudong gas fields. The data can be explained by mixing between mass-fractionated air-
 13 saturated water (ASW) and typical crustal endmembers.

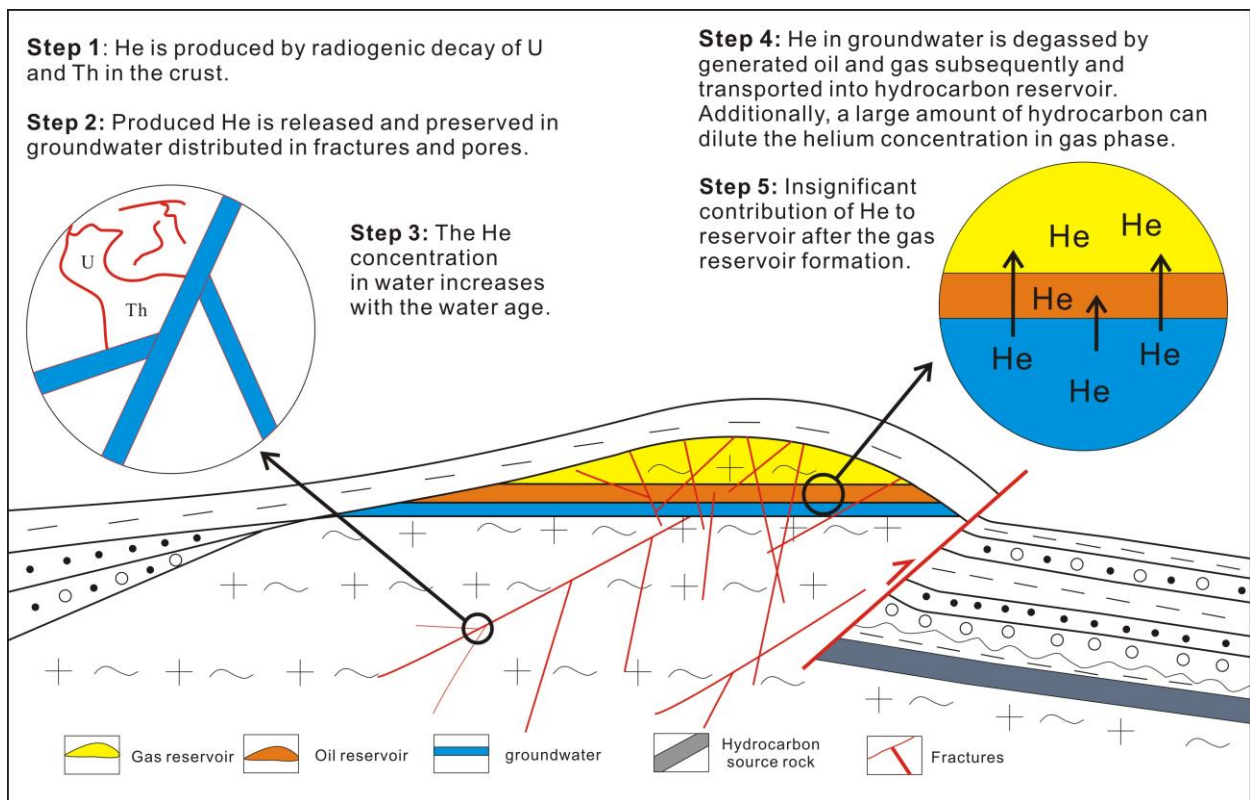
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 16 **Fig. 3** Two-stage phase fractionation modelling results. Black solid line represents noble gas
 17 characteristics of the first bubble from groundwater that has equilibrated with oil before gas-water
 18 equilibrium. Dashed line indicates the noble gas features in the gas phase with different gas/water
 19 ratios, in which the red dashed line suggests the approximate model result. The $V_{\text{oil}}/V_{\text{water}}$ ratio
 20 and $V_{\text{gas}}/V_{\text{water}}$ ratio are labelled near the lines. Sample Dongping 3 was modelled separately in the
 21 third plot since the gas reservoir is located at a depth of $\sim 2000\text{m}$, shallower than those associated
 22 with other wells in the Dongping area ($\sim 3000\text{m}$ in depth).



23
 24 **Fig. 4.** The linear relationship between ^4He and ^{20}Ne concentrations ($\text{cm}^3 \text{ STP/cm}^3$) in the Mabei,
 25 Dongping and Niudong gas fields.
 26

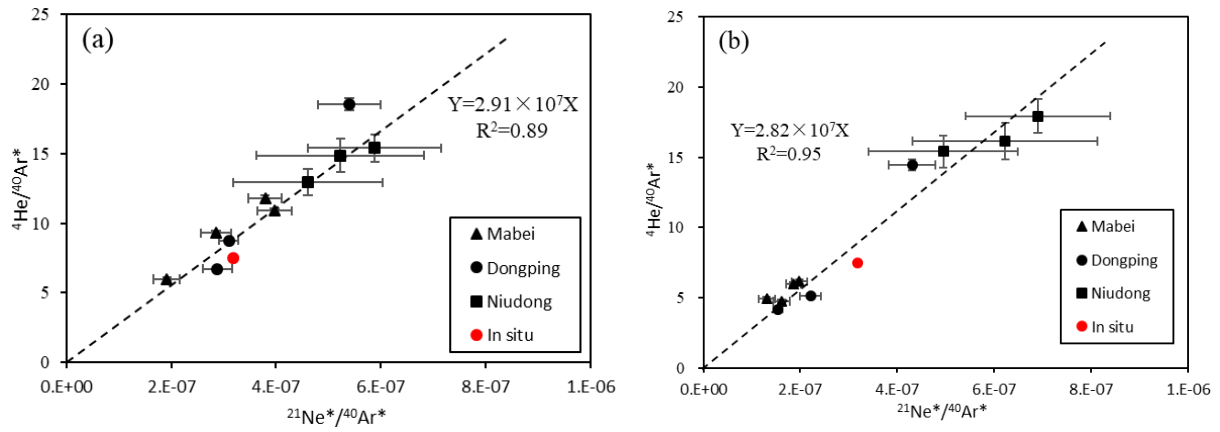


27

28 **Fig. 5.** Cartoon illustrating the helium accumulation processes in the North Qaidam Basin,

29 China, which can be divided into 5 steps. See detailed discussion in Section 5.3.

30



31
 32 **Fig. 6.** (a) Measured ${}^4\text{He}/{}^{40}\text{Ar}^*$ vs. ${}^{21}\text{Ne}^*/{}^{40}\text{Ar}^*$ in the gas phase. (b) Calculated ${}^4\text{He}/{}^{40}\text{Ar}^*$ vs.
 33 ${}^{21}\text{Ne}^*/{}^{40}\text{Ar}^*$ in the initial groundwater. The red dot represents the in-situ production ratio of
 34 ${}^4\text{He}/{}^{40}\text{Ar}^*$ and ${}^{21}\text{Ne}^*/{}^{40}\text{Ar}^*$ being 7.5 and 2.81×10^{-7} , respectively. Characteristics of the measured
 35 crust-derived noble gases in gas phase as shown in plot (a) are controlled by the oil-modified
 36 groundwater-exsolution fractionation process on the base of the two kinds of sources for ${}^4\text{He}$,
 37 ${}^{21}\text{Ne}^*$ and ${}^{40}\text{Ar}^*$ in groundwater and rocks shown in plot (b).

1 **Table 1 Major gas composition data (in volume fraction, vol. %) for the three gas fields in the North Qaidam**
 2 **Basin.**

Sample	C ₁	C ₂	C ₃	N ₂	CO ₂	O ₂	Gas Dryness
<i>Mabei gas field</i>							
Mabei 801	76.64	8.50	2.98	8.87	0.53	0.05	0.85
Maxi 1	87.63	2.76	0.39	8.54	0.30	0.03	0.96
Maba 2-23	78.83	8.59	2.93	7.41	0.10	0.02	0.85
<i>Dongping gas field</i>							
Dongping 171	95.18	2.48	0.39	1.15	0.10	0.05	0.97
Dongping 1	89.99	1.96	0.31	6.87	0.21	0.03	0.97
<i>Niudong gas field</i>							
Niu 1	89.04	6.30	1.69	1.41	0.19	0.13	0.91
Niu 1-2-10	86.97	7.60	2.12	1.54	0.25	0.16	0.89
Niu 1-2-11	86.47	7.47	2.25	2.08	0.25	0.19	0.89

3

4 **Table 2 Noble gas (helium, neon, argon, krypton, and xenon) isotope systematics in gases from the North Qaidam Basin^a.**

Sample	⁴ He×10 ⁻⁴	²⁰ Ne×10 ⁻⁸	³⁶ Ar×10 ⁻⁸	⁸⁴ Kr×10 ⁻⁹	¹³⁰ Xe×10 ⁻¹¹	³ He/ ⁴ He (R/Ra) ^b	²⁰ Ne/ ²² Ne	²¹ Ne/ ²² Ne	⁴⁰ Ar/ ³⁶ Ar	⁸⁶ Kr/ ⁸⁴ Kr	¹³² Xe/ ¹³⁰ Xe
	cm ³ STP/cm ³										
Air	0.05	1645	3142	650	366	1	9.80	0.029	298.56	0.305	6.61
Mabei Gas Field											
Mabei 801	18.6±0.2	2.90±0.03	9.92±0.20	3.43±0.04	4.13±0.20	0.0489±0.0010	8.75±0.14	0.0468±0.0015	2019±45	0.310±0.006	6.74±0.36
Mabei 1	12.2±0.1	2.23±0.03	8.53±0.16	3.38±0.04	5.58±0.20	0.0356±0.0014	9.21±0.15	0.0433±0.0013	1835±38	0.307±0.005	6.62±0.27
Maxi 1	6.08±0.06	1.54±0.02	7.11±0.13	3.61±0.04	5.18±0.20	0.0242±0.0010	9.47±0.16	0.0404±0.0013	1726±36	0.309±0.005	6.73±0.28
Maba 2-23	20.1±0.2	3.36±0.04	11.12±0.18	3.64±0.04	4.70±0.21	0.0372±0.0009	8.99±0.14	0.0444±0.0013	1817±35	0.301±0.006	6.84±0.33
Dongping Gas Field											
Dongping 171	2.06±0.021	0.35±0.01	1.89±0.02	0.89±0.01	2.08±0.09	0.0101±0.0003	9.38±0.16	0.0444±0.0015	883±16	0.302±0.005	6.52±0.30
Dongping 1	6.37±0.064	1.67±0.02	8.98±0.12	3.44±0.04	4.49±0.20	0.0102±0.0005	8.94±0.14	0.0411±0.0012	1350±22	0.301±0.005	6.60±0.32
Dongping 3	48.4±0.5	6.62±0.08	21.56±0.52	8.57±0.10	12.00±0.52	0.0174±0.0004	8.79±0.15	0.0504±0.0012	2862±75	0.303±0.006	6.41±0.30
Niudong Gas Field											
Niu 1	1.21±0.012	1.10±0.01	8.19±0.11	3.49±0.04	5.70±0.27	0.0148±0.0005	9.79±0.16	0.0328±0.0009	410±7	0.304±0.006	6.76±0.35
Niu 1-2-10	1.15±0.012	0.68±0.01	5.45±0.08	2.85±0.03	3.99±0.18	0.0154±0.0006	9.67±0.16	0.0350±0.0011	432±8	0.305±0.005	6.58±0.32
Niu 1-2-11	1.42±0.014	1.20±0.01	9.78±0.13	4.83±0.05	7.57±0.35	0.0146±0.0005	9.96±0.16	0.0337±0.0010	393±6	0.304±0.005	6.42±0.32

5 ^a 1σ errors are shown in the table after the plus/minus signs.

6 ^b ³He/⁴He ratios (R) are normalized to the air value Ra = 1.4×10⁻⁶.

7

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9 **Table 3 Henry's constants of noble gases in water and light oil (API=34) at modeling conditions^a.**

Depth (m)	Temperature (°C)	Solvent	Henry's constants					
			Helium	Neon	Argon	Krypton	Xenon	
<i>Recharge</i>								
0	10	Salinity: 0 M NaCl Water (dimensionless)	105	85	23	12	7	
<i>Reservoir (Mabei)</i>								
1000	44	Salinity: 2.02 M NaCl Water (dimensionless)	158	148	68	43	32	
		Salinity: 2.02 M NaCl Water (atm kg/mol)	4112	3857	1777	1129	842	
		API: 34 Oil (atm kg/mol)	700	590	119	40	12	
<i>Reservoir (Dongping 1 and 171)</i>								
3000	112	Salinity: 2.02 M NaCl Water (dimensionless)	110	112	72	53	48	
		Salinity: 2.02 M NaCl Water (atm kg/mol)	3488	3553	2274	1671	1506	
		API: 34 Oil (atm kg/mol)	392	358	139	49	27	
<i>Reservoir (Dongping 3 and Niudong)</i>								
2000	78	Salinity: 2.02 M NaCl Water (dimensionless)	134	130	74	51	42	
		Salinity: 2.02 M NaCl Water (atm kg/mol)	3853	3754	2144	1478	1220	
		API: 34 Oil (atm kg/mol)	524	460	128	44	18	

10 ^a Henry's constants of noble gases in water and oil are calculated from empirical equations from Ballentine et al. (2002) and corrected with fugacity coefficients and activity coefficients (Dymond and
11 Smith, 1980, Smith and Kennedy, 1983).

12 **Table 4 Volume ratios of oil to groundwater and gas to groundwater, and He concentrations in the initial groundwater.**

Sample	$^{20}\text{Ne}/^{36}\text{Ar}$	f_1	f_2	$V_{\text{oil}}/V_{\text{water}}$	$V_{\text{gas}}/V_{\text{water}}$ STP	$V_{\text{gas}}/V_{\text{oil}}$	Initial concentration in groundwater ^a (cm^3 STP/g)		
							^4He ($\times 10^{-3}$)	$^{21}\text{Ne}^*$ ($\times 10^{-11}$)	$^{40}\text{Ar}^*$ ($\times 10^{-4}$)
<i>Mabei gas field</i>									
Mabei 801	0.286 ± 0.006	0.24	0.47	0.12	1.21	10.06	13.9 ± 0.1	44.7 ± 3.6	22.6 ± 0.6
Mabei 1	0.252 ± 0.005	0.27	0.45	0.11	1.28	11.71	9.52 ± 0.10	32.5 ± 3.2	20.1 ± 0.5
Maxi 1	0.210 ± 0.005	0.22	0.35	0.13	1.69	13.36	9.18 ± 0.09	24.4 ± 3.2	18.7 ± 0.5
Maba 2-23	0.291 ± 0.006	0.27	0.51	0.11	1.09	9.93	11.9 ± 0.1	37.0 ± 3.1	19.9 ± 0.4
<i>Dongping gas field</i>									
Dongping171	0.179 ± 0.003	0.62	-	0.04	39.15	1081.28	11.1 ± 0.1	33.1 ± 3.6	7.69 ± 0.2
Dongping1	0.185 ± 0.003	0.74	-	0.02	6.81	292.97	7.05 ± 0.07	30.6 ± 2.9	13.8 ± 0.3
Dongping3	0.287 ± 0.008	0.58	-	0.04	0.71	17.20	14.0 ± 0.2	51.8 ± 3.0	33.6 ± 0.9
<i>Niudong gas field</i>									
Niu 1	0.133 ± 0.002	0.55	0.29	0.04	3.49	77.71	2.30 ± 0.02	7.39 ± 2.2	1.49 ± 0.1
Niu 1-2-10	0.122 ± 0.002	0.49	0.24	0.05	4.02	75.97	3.21 ± 0.03	12.3 ± 2.5	1.77 ± 0.1
Niu 1-2-11	0.120 ± 0.002	0.75	0.31	0.02	3.28	151.48	2.07 ± 0.02	7.97 ± 2.4	2.27 ± 0.1

13 ^a $^{21}\text{Ne}^*$ and $^{40}\text{Ar}^*$ represent the crust-derived components of ^{21}Ne and ^{40}Ar calculated by $^{21}\text{Ne}^* = ^{21}\text{Ne} - 0.00298 \times ^{20}\text{Ne}$ and $^{40}\text{Ar}^* = ^{40}\text{Ar} - 298.56 \times ^{36}\text{Ar}$.