Experimental Investigation of the Temperature Distribution in a Microwave-Induced Plasma Reactor

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Abstract: It is urgent to reduce CO_2 emissions to mitigate the impacts of climate change. The development of advanced conversion technologies integrated with plasma torches provides a path for the optimisation of clean energy recovery from biomass and wastes, thus substituting fossil fuels utilization. This article presents the temperature characterisation within a laboratory-scale microwave-induced plasma reactor operated with air, H₂O and CO₂ as the plasma working gases. The benefits associated with the plasma torch are highlighted and include rapid responses of the plasma and the temperature profile within the reactor to changing operating conditions. The average temperature near the side wall in the laboratory-scale reactor is proportional to the applied microwave power, ranging from 550°C at 2 kW to 850°C at 5 kW, while significantly higher temperatures are locally present within the plasma plume. The described system demonstrates promising conditions that are ideal for effective energy recovery from biomass and wastes into clean fuel gas.

Keywords: Microwave-induced plasma torch, air, steam, carbon dioxide, advanced conversion technologies

Abbreviation:

ACT: Advanced Conversion Technologies DC: Direct Current IPCC: Intergovernmental Panel on Climate Change MFC: Mass Flow Controller MIP: Microwave-Induced Plasma SLPM: Standard Litre Per Minute SOFC: Solid Oxide Fuel Cell

1 1. Introduction

2 1.1. Context and Motivation

3 The global surface temperature has abnormally increased by approximately 1°C during the last century, causing 4 modification of the climate system, a phenomenon commonly referred to as climate change [1]. According to the 5 Intergovernmental Panel on Climate Change (IPCC), this phenomena is "increasing the likelihood of severe, pervasive 6 and irreversible impacts for people and ecosystems", driven by sea-level rise and increases in the frequency and magnitude 7 of extreme events (heat waves, droughts, floods, cyclones and wildfires) [2]. To limit the risks related to climate change, 8 189 Parties ratified the Paris Agreement in 2015, which aims to keep the global temperature rise below 2°C above pre-9 industrial levels [3]. The IPCC estimates this would equate to a global CO2 emissions reduction of 25% from 2010 levels 10 by 2030 and reaching net zero around 2070 [4]. This will require major modifications for society, especially in the energy 11 sector, where fossil fuel consumption is currently at its highest and represents more than 80% of the primary energy 12 sources [5]. To meet this challenge, renewable energy generation must be rapidly implemented from wind and solar 13 resources, but also from sustainably grown biomass. At the same time, waste generation has been ever increasing globally 14 with implications for the environment and human health [6]. In fact, approximately 70% of waste worldwide is either 15 landfilled or dumped [6], both of which are recognised as the least favoured waste management practices according to 16 the waste hierarchy described in the European Directive 2008/98/EC on waste [7]. This work investigates the development 17 of advanced conversion technologies (ACTs) integrated with microwave-induced plasma (MIP) torches, which has the 18 potential to both efficiently generate renewable energy and fuels from biomass as well as to provide a solution for 19 sustainable solid waste management.

20 *1.2. Plasma conversion technologies*

21 Gasification and pyrolysis are ACTs that have the potential to reduce CO₂ emissions associated with energy generation 22 from biomass and waste compared to that of incineration [8]. Gasification is a thermal treatment with partial oxidation 23 from a controlled oxygen supply operating at elevated temperatures in the range 800-1,600°C, and produces a synthetic 24 gas (syngas) [9]. The syngas generated is rich in CO and H_2 , and can be used in a wide range of applications, such as 25 electricity generation, as a precursor in the chemical industry or for the production of liquid and gaseous fuels [10]. In 26 contrast, pyrolysis is the thermal dissociation of a solid feedstock in the absence of oxygen in the temperature range of 27 400-800°C [9]. The products of pyrolysis are char, pyrolysis oil, and pyrolysis gas, which can all be used for energy 28 generation with yields varying according to the operating conditions [11].

29 The implementation of plasma torches in ACT systems presents an opportunity to enhance biomass and waste 30 conversion while reducing harmful gaseous emissions [12]. In fact, plasma, sometimes referred as "the fourth state of 31 matter", provides exceptional treatment conditions such as extremely high temperatures and very high concentrations of 32 energetic and chemically active species (electrons, ions, excited species and photons) [13]. Plasma gasification 33 technologies are generally based on DC (Direct Current) plasma torches, however such systems suffer from high 34 operational costs, generally associated with the short lifespan of expensive electrodes in an oxidising environment 35 (<100 h) [14]. This issue can be overcome using MIP torches, capable of generating pure H₂O plasma without degradation 36 of any components [15].

37 *1.3. Review of MIP reactors*

38 Tang et al. [16] defined the different plasma reactor configurations including the plasma fixed bed reactor, the plasma 39 moving bed reactor and the plasma entrained-flow bed reactor. The plasma fixed bed operates in batch sequence whereas 40 the plasma moving-bed reactor is fitted with a feedstock inlet enabling continuous operation [16]. In the plasma entrained-41 flow reactor, the powdered feedstock is injected within the plasma flame [16].

42 Several studies investigated the potential of integrated MIP torches within ACT processes using different reactor 43 configurations. The pyrolysis of small biomass and waste samples <10 g has been described using Ar [17-19] and N₂ [20-44 22] as plasma working gases. However, the use of inert gases causes high dilution leading to the generation of syngas 45 with an extremely low calorific value. It is therefore preferred to apply chemically reactive plasma gases to the feedstock 46 such as air and H₂O. Most of the experimental MIP gasification studies have been undertaken with entrained-flow bed 47 reactors where powdered coal [15, 23-25] or biomass [26, 27] is introduced within the high temperature plasma plume. 48 However, Ho et al. [28] identified that the main challenge to improve MIP gasification performance using entrained-flow 49 bed reactors is the fuel retention time. In fact, in the entrained-flow configuration, the feedstock is rapidly pushed away

from the reaction zone by the high plasma velocity, thus limiting the conversion efficiency. In addition, the entrainedflow configuration requires pre-processing and size reduction of the feedstock, itself an energy intensive process.

The development of MIP fixed and moving bed reactors are alternative technologies that have received far less attention than entrained-flow reactors. Sanlisoy and Carpinlioglu [29, 30] recently developed a tubular MIP fixed bed reactor and investigated the gasification of a variety of fuels in an air plasma. Uhm et al. [31, 32] described the only MIP moving bed reactor available in the literature. However, it is a demonstration-scale reactor fitted with two plasma torches with an operational specification of up to 75 kW, and operates at a lower frequency of 915 MHz compared to the usual 2.45 GHz used at laboratory-scale. The reactor achieved complete coal conversion in steam plasma while generating a syngas rich in H₂ and CO with volumetric concentration of 40% and 32% respectively [31, 32].

This study presents the temperature distribution measured within a laboratory-scale MIP moving-bed reactor, which to the best of the authors' knowledge, has not been previously described in the literature. Air, CO₂ and H₂O are three gases of interest in ACTs, and their use as plasma working gases are investigated here. The effect of applied microwave power and working gas flow rates on the temperature distribution are described.

63 2. Materials and Methods

64 The experimental rig can be divided in three main sections: the microwave plasma source and its components, the 65 plasma working gas supply and the plasma reactor. A block diagram of the experimental set-up is presented in Fig. 1.

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69 2.1. The microwave plasma source

70 Plasma is generated in the MIP torch, which is initially a Downstream plasma source (Sairem SAS, Neyron, France). 71 Microwaves are generated using a Sairem GMP G4 microwave generator composed of a high-voltage supply where the 72 microwaves can be controlled. The microwave generator is connected to a magnetron head equipped with a circulator. 73 The magnetron generates microwaves at a frequency of 2.45 GHz up to 6 kW of forward power. The microwaves then 74 propagate through WR340 waveguides equipped with a 3-stub tuner and a sliding short-circuit enabling the minimisation 75 of the reflected microwave power to less than 1% of the forward microwave power whilst maintaining a stable plasma. 76 Therefore, almost all the power in the microwaves is transmitted to the plasma and the overall efficiency of the system is 77 almost equivalent to the conversion efficiency from electricity to microwaves that is in the range 60-70%. The plasma 78 working gas flows through a 35 cm long quartz tube with internal and external diameter of 25.6 and 30 mm respectively. 79 The plasma is manually ignited by inserting a tungsten rod. The plasma is maintained by absorbing the microwave power 80 within the downstream plasma source. The plasma plume propagates downstream and exits the quartz tube at its bottom 81 tip. Fig. 2 presents an open-air plasma plume generated with the downstream plasma source.



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Fig. 2. Picture of the microwave-induced plasma torch in operation with air.

84 2.2. Working gas supply

The microwave plasma source can generate a stable plasma using air, H_2O and CO_2 , as well as mixtures of these gases. Fig. 3 shows the plasma plume obtained for the three different plasma working gases studied. The flow rates of air and CO₂ are regulated using mass flow controllers (MFC) (Alicat Scientific, Tucson AZ, USA) operating in the range 0 to 100 SLPM (standard litre per minute). A gas cylinder supplies the CO₂ whereas air comes from the building compressed air lines. A precision steam generator (Cellkraft, Stockholm, Sweden) provides H₂O at flow rates in the range 10 to 50 g/min and at temperatures up to 200°C.



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Fig. 3. Pictures of plasma plumes for different plasma working gases.

93 2.3. The plasma reactor

94 A plasma reactor has been designed and built for integration with the MIP torch. It is schematically presented in Fig. 4. 95 The reactor has an internal diameter and length of approximately 15 cm and 35 cm, respectively. It is internally insulated 96 with a 3 cm layer of fire cement and externally using thermal webbing tape. The plasma reactor is equipped with a 97 feedstock inlet, a safety line and a syngas outlet. The feedstock inlet is a 2-inch pipe inclined by 45° enabling the feedstock 98 to be dropped into the reactor while the plasma torch is in operation. The feedstock can therefore be continuously inserted 99 in the reactor in a moving bed configuration. Nevertheless, the feedstock inlet is not used in this study and is sealed off 100 with a blind flange. The safety line is fitted with a rupture disc rated at 0.3 bar to manage the risk of overpressure in the 101 reactor. The syngas outlet is a 2-inch pipe allowing the gas to exit near the top of the reactor. The top of the reactor is 102 directly connected to the bottom of the downstream source. The plasma is released in the reactor at the bottom of the 103 quartz tube.

Four type K thermocouples are located within the plasma reactor to investigate temperature distribution. The thermocouples are named T1, T2, T3 and T4 and are sited at 31 cm, 17 cm, 10 cm and 3 cm from the internal top of the reactor, respectively. The thermocouple T4 measures the temperature above where the plasma is released as the quartz tube extends about 4 cm from the top of the reactor. The average temperature in the reactor (TA) is calculated as the average of the four thermocouples T1, T2, T3 and T4. For the results presented in Section 3.1 to 3.4, the thermocouples protrude approximately 2 cm into the reactor beyond the refractory lining and measure the temperature close to the side wall. In Section 3.5, the thermocouples are repositioned to investigate the radial temperature gradient in the reactor. The distance between the tip of the thermocouple and the refractory lining is varied between 2 and 5 cm. The temperatures presented in Section 3.5 are recorded by repeating a the following protocol: the reactor is first warmed-up using an applied microwave power of 5 kW for a period of 60 minutes, and then the temperature is recorded after 20 minutes of operation in the specific conditions of flow rate and applied microwave power.





Fig. 4. Schematic representation of the plasma reactor.

118 3. Results and Discussion

119 *3.1. Temperatures with air plasma*

120 This section presents the temperature distribution in the plasma reactor when operating the plasma torch with air. 121 Fig. 5 depicts the temperature evolution over time when operating the plasma torch at constant conditions of 70 SLPM 122 air and 5 kW applied microwave power. The graph indicates a very quick temperature increase during the first minutes 123 of operation with a rate exceeding 150°C per minute. In fact, within the four first minutes, the average temperature (TA) 124 rose from ambient 20°C to 650°C. After 10 minutes of operation, the temperature profile evolves to a linear increase with 125 a decreasing slope over time. TA increases from 750°C after 20 minutes to 890°C after almost 2 hours of operation. 126 Nevertheless, the temperature increase rate diminishes below 1°C per minute during the last 10 minutes but does not 127 completely stabilise after 2 hours. The long temperature increase time is attributed to the large heat capacity of the fire 128 cement layer thus taking time to warm up. Conversely, the heat stored in the fire cement refractory layer will be released 129 when cooling down the inside of the reactor. The fire cement layer therefore has a great influence on the temperature 130 measured within the reactor as it slows down heating and cooling rates, which also protects the metallic parts of the 131 reactor from thermal shock. The experimental duration is limited to approximatively 2 hours because of the limited 132 capacity of the cooling system in place for the magnetron operation.

Fig. 5 also provides information on the temperature distribution within the reactor. It is important to bear in mind that the temperatures T1, T2, T3 and T4 are measured close to the side wall, and thus provide a minimum on a cylindrical plane as higher temperatures are expected along the axis of the cylinder in line with the plasma plume. The highest temperatures are recorded by the thermocouples T1 and T2 at the bottom of the reactor. After 2 hours of operation, T1 and T2 are up to 950°C, representing the minimum temperature in the lower half of the reactor. However, lower temperatures are measured by the thermocouple T3, despite the fact that it is located closer to where the plasma plume is

139 released, compared to T1 and T2. This observation can be attributed to the cone shape (horizontal expansion when moving

down vertically) of the heat released by the plasma plume. Furthermore, measured temperatures by thermocouple T3 are

141 generally in accordance with the average temperature calculated within the reactor. The lowest value is recorded by the

142 thermocouple T4 at the top of the reactor, reaching up to 800°C.





151 152 Fig. 5. Temperatures in reactor with 70 SLPM air plasma at 5 kW applied microwave power.

Fig. 6 presents the evolution of the temperature in the reactor when varying the applied microwave power at a constant air flow rate. The microwave power is modified with 20 minutes intervals in the range 2 to 5 kW. It is evident that the temperature measured is proportional to the applied microwave power. In fact, a 1 kW increase in applied microwave power engenders a quick increase of TA in the range 80-100°C within less than 5 minutes. In addition, the temperature variations are similar at the different locations in the reactor and the microwave power does not affect the temperature distribution under the specific operating conditions presented in Fig. 6.



Fig. 6. Temperatures in reactor with 70 SLPM air plasma and varying applied microwave power.

153 The effects of the air flow rate are also investigated with variations in the range of 40 to 80 SLPM at a constant applied 154 microwave power (Fig. 7). A fixed microwave power of 3 kW has been chosen as it enables plasma operation in a wider 155 range of air flow rates. The maximum air flow rate is limited to 80 SLPM reflecting the maximum the compressed air 156 supply could provide. The air supply was limited to 40 SLPM as low flow rate to power ratio can lead to the degradation 157 of the quartz tube. Fig. 7 shows how the flow rate affects the temperature distribution. It is particularly visible at the 158 lowest flow rate of 40 SLPM, with a sudden increase of temperature at T1, with decreases observed at both T2 and T3. 159 This can be attributed to two main phenomena associated with the flow reduction: 1) a thinner plasma plume reduces the 160 horizontal spreading of the heat; 2) lower turbulent flow pattern and air mixing in the reactor. Moreover, the temperature 161 T4 is not significantly influenced by the change of air flow rate highlighting low turbulence in the reactor, especially in

the upper section. Generally, an increase of 10 SLPM of the flow rate results in a decrease of the average temperature in

the range 10-20°C, which can be explained by a heat dilution effect. However, this is not evident at the lowest flow rate

164 of 40 SLPM, because the large temperature disparities in the reactor makes TA more uncertain.





Fig. 7. Temperatures in reactor varying air flow rate at 3 kW microwave power.

167 *3.2. Temperatures with* H_2O *plasma*

168 The main advantage of a microwave plasma torch is that it can be operated using carrier gases such as steam, which 169 would otherwise corrode the electrodes of DC systems. This section of the manuscript presents the temperatures recorded 170 in the reactor when operating with pure H₂O plasma.

171 Fig. 8 presents the temperature distribution in the reactor by varying the applied microwave power when operating 172 H₂O plasma at a constant flow rate of 50 g/min. The lowest microwave power is 3 kW as it was empirically determined 173 that a minimum of 2.5 kW was required to sustain a pure H₂O plasma. Similarly, as for air, the temperatures are directly 174 proportional to the applied microwave power. Nevertheless, lower temperature increases in the range 50-70°C per 1 kW 175 microwave power are observed, whereas the increase was in the range 80-100°C when using air. Furthermore, contrary 176 to air plasma, the applied microwave power affects the temperature distribution when operating with H_2O . Whereas T1 and T2 have very close values at 5 kW, a decrease of the microwave power results in a more distinct reduction of the 177 178 temperature at T1 than T2. In fact, at 3 kW, the temperature at T1 is comparable with the temperature recorded at T3. 179 This is attributed to a reduction of the length of the H₂O plasma plume at low power, thus shifting the highest temperature 180 region from the bottom to the middle of the reactor.





Fig. 8. Temperatures in reactor with 50 g/min H₂O and varying applied microwave power.

The influence of the H₂O flow rate is presented in Fig. 9 with variation between 30 and 50 g/min at constant microwave power. This was undertaken at microwave power of 3 kW because operation at higher applied microwave power and low flow rate of 30 g/min would damage the quartz tube. As can be seen in Fig. 9, the variation of H₂O flow rate does not significantly affect the average temperature in the reactor. However, similarly as for air, lower H₂O flow rates result in an increase in temperature measured at the bottom of the reactor. Thereby, the highest temperatures were measured by T2 at flow rate of 50 g/min whereas the hottest region moves down to T1 at a flow rate of 30 g/min.







191 *3.3. Temperatures with CO₂ plasma*

The third gas of interest in this study is CO_2 because of its reforming properties that are of interests for biomass and wastes conversion to syngas. Moreover, CO_2 is one of the main greenhouse gases and new utilisations of the gas could reduce fossil fuel derived emissions, thus mitigating climate change. The plasma torch is successfully operated with CO_2 at considerably lower flow rates than air. In fact, as presented in Fig. 10, a stable CO_2 plasma can be generated at a flow rate of 25 SLPM for an applied microwave power in the range 2-5 kW. Operation at higher flow rates of up to 100 SLPM is possible but could not be sustained for long periods due to the limited capacity of the gas supply.

As presented in Fig. 10, the temperature T1 is substantially higher than T2 at high microwave powers of 4 and 5 kW
 reaching up to 1,000°C. This phenomenon is typical for low flow rates as described previously for air and steam plasmas.
 However, at a lower microwave power of 2 kW, the temperature at T1 drops lower than the temperatures observed at T2

and T3. It confirms the tendency, as seen for H_2O , that low microwave power causes a reduction of the size of the plasma plume limiting the heat propagation to the bottom of the reactor, which is further exacerbated at low flow rates. The average temperature was also found to be directly proportional to the applied microwave power, with a rapid temperature increase in the range 50-100°C per kW. In addition, the temperature T1 is the most affected with temperature variation in the range 100-150°C per 1 kW change, highlighting non-homogeneous temperature distribution in the reactor, especially at low flow rates.





Fig. 10. Temperatures in reactor with 25 SLPM CO₂ and varying applied microwave power.

209 3.4. Comparison of plasma gases

210 Fig. 11 presents the average temperature in the reactor when varying the applied microwave power for the three plasma 211 working gases. The temperatures are taken after more than one hour of plasma operation and 20 minutes at the specific 212 applied microwave power and flow rate, and uses the results presented in Fig. 6, 8 and 10. Whereas it was previously 213 demonstrated that the measured temperatures are partially related to the gas flow rate and length of the experiment, Fig. 11 214 shows the main driver of the temperature is the applied microwave power. In fact, the average temperatures are very close 215 for the three plasma working gases with maximum differences of 50°C at microwave powers above 3 kW. At a lower 216 microwave power of 2 kW, higher temperature differences are noted, explained by thermal stratification in the reactor 217 causing uncertainties in the calculated averages. Generally, the average temperatures close to the side wall of the reactor 218 are approximately 550, 650, 750 and 850°C at an applied microwave power of 2, 3, 4 and 5 kW, respectively.





222 The major drawback of plasma torches is their high energy consumption, which can have a detrimental impact on the 223 overall process efficiency [12]. Hence, it is essential to optimise the operating conditions according to feedstock 224 characteristics. In fact, the plasma ACT process can be applied to a wide range of wastes including biomass wastes, refuse 225 derived fuel (such as mixtures of plastics, paper, wood and dried organic materials), hazardous wastes, used tyres, paper mill wastes, medical wastes or sewage sludge wastes [14, 33]. The performance and optimal operating conditions depend 226 227 on the waste specific properties such as elemental composition, lower heating value (LHV), ash content, volatile matter 228 content, contaminants, bulk density and size [34]. These properties influence the temperature required for effective solid 229 conversion.

230 The experimental results highlight the direct relationship between the applied microwave power and temperature in 231 the reactor. It provides control of the processing environment, thus enabling real-time optimisation of the system. The 232 temperature is also affected by the occurring chemical reactions driven by the nature of the plasma working gas. The use 233 of air increases the temperature and thus conversion through exothermic oxidation reactions, but reduces the calorific 234 value of the generated gas because of N₂ dilution as well as fuel oxidation into CO₂ and H₂O [34]. H₂O and CO₂ plasma 235 treatments enhance reforming reactions and generates syngas with a higher heating value [34]. However, reforming 236 reactions are endothermic and would rely on the heat provided by the plasma, thus requiring higher energy consumption 237 of the plasma torch.

238 Numerical models are useful tools to theoretically optimise the operating conditions of plasma ACT processes and 239 can be used to identify a balance between feedstock conversion and electricity consumption by operating with a mixture 240 of plasma working gases. For example, Ismail et al. [35] determined that an equivalence ratio (ratio of oxygen in reactor to the stoichiometric amount of oxygen required for full feedstock oxidation) of 0.3 and a steam to fuel ratio of 0.5 were 241 242 the most favourable conditions for high-quality syngas generation from municipal solid wastes in a plasma fixed-bed 243 gasification reactor. In contrast, this work provides a first characterisation of our MIP reactor that will enable the 244 conversion efficiency to be experimentally studied at laboratory scale by varying the operating conditions for different feedstocks. 245

246 *3.5. Radial temperature gradients*

247 To investigate the radial temperature gradient within the reactor, the experiments were repeated, and temperature 248 measurements were made at distances between 2-5 cm away from the side wall of the reactor. Fig. 12 presents the 249 temperature recorded for different positions of the thermocouples with 70 SLPM air and an applied microwave power of 5 kW. As expected, higher temperatures were measured when moving the thermocouples towards the centre of the reactor. 250 251 In fact, the temperatures recorded at T1 show a relatively linear increase, ranging between 10-20°C per 1 cm displacement 252 to the centre of the reactor. A similar trend is observed at T2 and T3 whereby the thermocouples were positioned between 253 2 and 4 cm from the side wall. However, the temperature drastically increased at T2 and T3 when moving the 254 thermocouples 5 cm from the side wall towards the centre of the reactor. At this position, the thermocouples are closest 255 to the plasma plume, with the highest temperatures measured exceeding 1050°C at an applied microwave power of 5 kW. Similar trends were achieved at T3 with the highest temperatures recorded when operating the plasma torch with 70 SLPM 256 257 air and applied microwave powers of 2, 3 and 4 kW.

258 To avoid damage to the thermocouples at T3, given the expected exponential increase in temperature towards the 259 centre of the plasma plume, a maximum measurement distance of 5 cm from the reactor wall was instated. In fact, the 260 temperature distribution is characterised by extremely high temperatures, as a result of the plasma plume, at the top centre 261 of the reactor. The temperature was found to decrease when moving away from the plasma plume as described. Whilst 262 the temperature around the plasma plume can be measured using thermocouples, the temperature within the plasma is 263 usually estimated by comparing optical emission spectroscopy measurements with simulated spectra. A previous study 264 from the authors showed that the temperature in H_2O and CO_2 plasmas ranged from approximatively 6,000°C where the 265 plasma is generated to 2,300°C 14 cm downstream of the plasma plume [36]. Similar temperatures are expected in air plasma [37]. Moreover, it was demonstrated that the temperature in the centre of the plasma was not significantly 266 267 influenced by the applied microwave power [36]. In this study, it was demonstrated that an increase in applied microwave 268 power resulted in a greater proportion of working gas being ionised, resulting in an enlargement of the plasma plume 269 diameter within the quartz tube. The temperatures recorded in the reactor are therefore proportional to the applied

270 microwave power mainly because of the impact it has on the ratio of hot to cold gasses in the quartz tube as opposed to a

change in plasma temperature.



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Fig. 12. Temperature in reactor with 70 SLPM air and 5 kW applied microwave power varying the distance of the thermocouples to the refractory lining between 2 and 5 cm.

Fig. 13 presents the temperatures recorded in the reactor, across the different thermocouple positions, with 40 SLPM air at 3 kW applied microwave power. As depicted in Fig. 12, highest temperatures were recorded close to the centre of the reactor. Nevertheless, even when the thermocouples were positioned 5 cm from the side wall, the highest temperatures were recorded at T1 and not T3 as experienced with higher air flow rate. This confirms that lower plasma gas flow rates tend to reduce the horizontal heat expansion along the plasma plume resulting in higher temperatures at the bottom of the reactor.



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Fig. 13. Temperature in reactor with 40 SLPM air and 3 kW applied microwave power varying the distance of the thermocouples to the refractory lining between 2 and 5 cm.

284 *3.6. Benefits of the plasma reactor*

The experimental study shows that temperatures ranging from 500°C to 1,050°C are achieved within the presented plasma reactor. The system is thus able to generate appropriate temperatures above 600°C that are required for efficient conversion of solid carbon and light hydrocarbons [34]. The solid fuel can be injected in the reactor through the feedstock inlet and sit at the bottom of the reactor, where some of the highest temperatures were recorded. The reactor configuration thus enables long fuel retention time that should enhance the conversion efficiency [28]. Nevertheless, one of the main issues with ACT is the presence of tars contaminating the generated gas [14]. Tars are a complex mixture of condensable hydrocarbons causing plugging and fouling of pipes and downstream equipment [38]. Because the generated fuel gas 292 must pass close to the plasma plume to exit the reactor at the top, substantial tars reforming is expected due to the plasma's 293 exceptional properties. First, the plasma's extremely high temperatures will enhance endothermic steam and dry 294 reforming reactions as well as the thermal cracking reactions of tars. In addition, plasma is composed of very chemically 295 reactive species including electrons, ions, excited molecules and photons. These species are able to stimulate chain 296 processes in a way that cannot be accomplished in conventional chemistry [13]. This effect is usually designated as plasma 297 catalysis as it enhances chemical reaction rates in a similar manner to the presence of a catalyst [13]. A previous study by 298 the authors on the same system demonstrates that active plasma species (e.g. O, OH and H radicals in H₂O plasma) are 299 still present 14 cm below the plasma plume [36]. The dissociation of H₂O in plasma is dominated by H, O, OH, H₂ and 300 O₂ species, whereas CO₂ mainly dissociates into CO, C, O, C₂ and O₂ [39, 40]. The benefits of H₂O and CO₂ dissociation 301 in MIP is that it provides oxygen for full and partial-oxidation reactions that favour fuel conversion, whilst generating 302 additional H₂ and CO which serves to further enrich the product syngas [39, 40]. Several experimental studies have shown 303 high tar conversion for syngas cleaning using MIP torches [41]. However, those experiments investigate the application of MIP reforming in a second step process for gas cleaning following conventional ACT. Therefore, tar conversion might 304 305 be significantly lowered when the MIP torch is used for both gasification and gas cleaning in the same reactor. The MIP 306 moving bed reactor described in this study will enable the experimental investigation of the quantity and composition of 307 tars present in syngas by varying the feedstock composition and operating conditions.

The MIP-ACT process can be completed with a syngas cleaning step to meet the downstream requirements for syngas quality. Tars are conventionally removed at low temperatures using wet scrubbing processes, which presents drawbacks in relation to a reduction in efficiency due to syngas cooling or the additional costs incurred to treat and dispose of the tar contaminants [42]. Another approach is the *in-situ* decomposition of tars into valuable CO and H₂ that will serve to enrich the generated syngas. This can be achieved at elevated temperatures using catalysts, however their costs and rapid deactivation limit their use [42], or through virtue of internal cracking of the tars within the reactor itself.

314 *3.7. Scale-up and deployment*

315 The commercialisation of ACT technologies can be limited by the expensive gas cleaning required to meet downstream requirements [34]. The use of a microwave plasma torch should improve the quality of the gas generated, 316 317 thus reducing the size and cost of the gas cleaning step. Processes involving MIP torches have been widely investigated 318 in laboratories but can also be scaled-up for industrial applications [28]. The first possibility is to integrate a multitude of 319 plasma torches into one large reactor, which provides the advantage of improved heat distribution within the reactor. 320 Moreover, MIP torches with capacity up to 100 kW are commercially available at a lower frequency of 915 MHz whereas 321 systems at 2.45 GHz are generally limited to 15 kW [43]. One of the main advantages of higher capacity microwave generators is their improved efficiency. Whereas the conversion efficiency from electricity to microwaves is circa 70% 322 323 at frequency of 2.45 GHz, it is considerably improved at lower frequency of 915 MHz and can reach up to 88%, therefore 324 positively affecting the overall efficiency of the proposed MIP-ACT process when scaling-up [44]. Uhm et al. described 325 the high efficiency of a coal gasification process using two MIP torches of 75 kW operating at frequency of 915 MHz, 326 and capable of generating 500 kW of syngas [31, 32]. In addition, high power magnetrons between 0.5 and 1 MW could 327 potentially be developed for large scale applications at low frequency of 433.96 MHz [43].

328 The gas generated from industrial microwave plasma ACT systems could be used in various downstream applications, 329 including liquid fuel production, chemicals production or electricity generation, which all enable substitution of fossil 330 fuels and could mitigate against climate change [14]. Electricity generation can be undertaken using gas engines, gas 331 turbines or solid oxide fuel cells (SOFC). The emerging SOFC systems have high electrical efficiency up to 65% and 332 could engender a technology shift, by making integrated MIP-ACT-SOFC systems more efficient, especially at relatively 333 small scales below 1 MW [45]. In addition, such system enable a multitude of integration options. Song and Chun 334 demonstrate that up to 65% of CO₂ could be converted to combustible CO in a MIP torch and in the presence of biomass 335 char [46]. Therefore, parts of the exhaust of a downstream electricity generation device, rich in carbon dioxide, could be 336 recycled back into the plasma torch, thus reducing the CO₂ emissions per unit of energy generated. Perna et al. [47] 337 affirmed that an advanced power plant combining plasma gasification and SOFC could achieve high electrical efficiencies 338 in the range 35-45%, substantially higher than the average 20% of conventional incineration power plant with a steam 339 turbine. The plasma ACT system has the potential to recover energy from wastes with high efficiency and could be 340 deployed where the wastes are generated, thus providing clean energy for on-site use. At the same time, pollution and

costs associated with waste transportation can be avoided. Our research suggests that, as part of the transition to a more
 sustainable circular economy that is demanded by the European Commission [48], the use of small scale, compact,
 decentralised systems should be considered.

344 4. Conclusion

345 This work presents the temperature measurement in a MIP moving-bed reactor at laboratory scale designed for waste 346 and biomass energy recovery. Beside the extreme temperatures within the plasma plume, high temperatures are recorded with values approaching 1,000°C at the bottom of the reactor. The study shows here that the average temperature near the 347 side wall within the laboratory scale system is proportional to the applied microwave power and varies from 550°C at 348 2 kW to 850°C at 5 kW. While the temperatures recorded are of the same order for the three plasma working gases studied, 349 350 high flow rates and applied microwave power results in more homogeneous temperature distribution within the reactor, 351 which should be improved through further design. Our experimental results suggest that control of operating parameters 352 on a short temporal resolution provides further advantages when using an MIP torch for the direct management of the 353 processing environment. This study provides important results for the preparation of biomass and wastes plasma ACT 354 experiments that will enable the efficiency of the process and the quality of the syngas generated to be empirically 355 determined. Ideal conditions are described for effective conversion of organic solid into clean combustible gas, which 356 could balance the high energy consumption of the plasma torch. The proposed plasma ACT reactor has the potential to 357 be developed as part of the transition to more sustainable solid waste management practices and could contribute towards 358 a more complete circular economy.

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364 6. References

- 365 [1] NOAA National Centers for Environmental information, Climate at a Glance: Global Time Series.
 366 <u>https://www.ncdc.noaa.gov/cag/</u>, 2020 (accessed 25 April 2020).
- 367 [2] IPCC, Climate Change 2014: Synthesis Report. <u>https://www.ipcc.ch/report/ar5/syr/</u>, 2014 (accessed 14 February 2020).
- 369 [3] United Nations Climate Change, The Paris Agreement. <u>https://unfccc.int/process-and-meetings/the-paris-agreement/the-paris-agreement</u>, 2015 (accessed 26 March 2020).
- [4] IPCC, Special Report: Global Warming of 1.5°C. <u>https://www.ipcc.ch/sr15/chapter/spm/</u>, 2018 (accessed 25 April 2020).
- BP, Statistical Review of World Energy. <u>https://www.bp.com/en/global/corporate/energy-economics/statistical-review-of-world-energy.html</u>, 2019 (accessed 25 April 2020).
- S. Kaza, L. Yao, P. Bhada-Tata, F. Van Woerden, K. Ionkova, What a waste 2.0 : a global snapshot of solid waste management to 2050, World Bank Group, Washington, 2019. <u>https://doi.org/10.1596/978-1-4648-1329-0</u>
- [7] European Parliament and the Council of the European Union. Directive 2008/98/EC of the European Parliament and
 of the Council on waste and repealing certain Directives. <u>https://eur-lex.europa.eu/legal-</u>
 <u>content/EN/TXT/?uri=CELEX:32008L0098</u>, 2008 [accessed 14 February 2020].
- [8] T.P.T Pham, R. Kaushik, G. Parshetti, R. Mahmood, R. Blasubramanian, Food waste-to-energy conversion
 technologies: Current status and future directions, Waste Management 38 (2015) 399-408.
 <u>https://doi.org/0.1016/j.wasman.2014.12.004</u>
- [9] A. Kumar, S.R. Samadder, A review on technological options of waste to energy for effective management of municipal solid waste, Waste Management 69 (2017) 407-422. <u>https://doi.org/407-422.</u>
 <u>10.1016/j.wasman.2017.08.046</u>
- [10] R. Luque, J.G. Speight, Gasification for synthetic fuel production: fundamentals, processes and applications, 1st e,
 Woodhead Publishing, Cambridge, 2015.
- [11] L. Lombardi, E. Carnevale, A. Corti, A review of technologies and performances of thermal treatment systems for energy recovery from waste, Waste Management 37 (2015) 26-44. <u>https://doi.org/10.1016/j.wasman.2014.11.010</u>

- [12] B. Ruj, S. Ghosh, Technological aspects for thermal plasma treatment of municipal solid waste—A review, Fuel
 Processing Technology 126 (2014) 298-308. <u>https://doi.org/10.1016/j.fuproc.2014.05.011</u>
- **392** [13] A. Fridman, Plasma chemistry, 2008. <u>https://doi.org/10.1017/CBO9780511546075</u>
- [14] F. Fabry, C. Rehmet, V. Rohani, L. Fulcheri, Waste Gasification by Thermal Plasma: A Review, Waste and Biomass
 Valorization 5 (2013) 421-439. <u>https://doi.org/10.1007/s12649-013-9201-7</u>
- [15] D.H. Shin et al., A pure steam microwave plasma torch: Gasification of powdered coal in the plasma, Surface & Coatings Technology 228 (2013) 520-523. <u>https://doi.org/10.1016/j.surfcoat.2012.04.071</u>
- [16] L. Tang, H. Huang, H. Hao, K. Zhao, Development of plasma pyrolysis/gasification systems for energy efficient and environmentally sound waste disposal, Journal of Electrostatics 71 (2013) 839-847.
 <u>https://doi.org/10.1016/j.elstat.2013.06.007</u>
- [17] C.J. Lupa, R.S. Wylie, A. Shaw, A. Al-Shamma'a, J.A. Sweetman, B.M.J Herbert, Experimental analysis of biomass
 pyrolysis using microwave-induced plasma, Fuel Processing Technology 97 (2012) 79-84.
 <u>https://doi.org/10.1016/j.elstat.2013.06.007</u>
- [18] C.J. Lupa, R.S. Wylie, A. Shaw, A. Al-Shamma'a, J.A. Sweetman, B.M.J Herbert, Gas evolution and syngas heating
 value from advanced thermal treatment of waste using microwave-induced plasma, Renewable Energy 50 (2013)
 1065-1072. <u>https://doi.org/10.1016/j.renene.2012.09.006</u>
- 406 [19] H. Sekiguchi and T. Orimo, Gasification of polyethylene using steam plasma generated by microwave discharge,
 407 Thin Solid Films 457 (2004) 44-47. <u>https://doi.org/10.1016/j.tsf.2003.12.035</u>
- 408 [20] Y.C. Lin, T.Y. Wu, S.R. Jhang, P.M. Yang, Y.H. Hsiao, Hydrogen production from banyan leaves using an atmospheric-pressure microwave plasma reactor, Bioresource Technology 161 (2014) 304-309.
 410 <u>https://doi.org/10.1016/j.biortech.2014.03.067</u>
- [21] K.C. Lin, Y.C. Lin, Y.H. Hsiao. Microwave plasma studies of Spirulina algae pyrolysis with relevance to hydrogen production, Energy 64 (2014) 567-574. <u>https://doi.org/10.1016/j.energy.2013.09.055</u>
- [22] Y.C. Lin, T.Y. Wu, W.Y. Liu, Y.H. Hsiao, Production of hydrogen from rice straw using microwave-induced pyrolysis, Fuel 119 (2014) 21-26. <u>https://doi.org/10.1016/j.fuel.2013.11.046</u>
- [23] S.J. Yoon and J.G. Lee, Hydrogen-rich syngas production through coal and charcoal gasification using microwave
 steam and air plasma torch, International Journal of Hydrogen Energy 37 (2012) 17093-17100.
 <u>https://doi.org/10.1016/j.ijhydene.2012.08.054</u>
- [24] S.J. Yoon,J. Goo Lee, Syngas production from coal through microwave plasma gasification: Influence of oxygen,
 steam, and coal particle size, Energy and Fuels 26 (2012) 524-529. <u>https://doi.org/10.1021/ef2013584</u>
- [25] Y.C. Hong et al., Syngas production from gasification of brown coal in a microwave torch plasma. Energy 47 (2012)
 36-40. <u>https://doi.org/10.1016/j.energy.2012.05.008</u>
- [26]G. Sturm, A. Munoz, P. Aravind, G. Stefanidis, Microwave-Driven Plasma Gasification for Biomass Waste
 Treatment at Miniature Scale, IEEE Transactions on Plasma Science 44 (2016) 670-678.
 <u>https://doi.org/10.1109/TPS.2016.2533363</u>
- [27] E. Delikonstantis et al., Biomass gasification in microwave plasma: An experimental feasibility study with a side stream from a fermentation reactor, Chemical Engineering and Processing Process Intensification 141 (2019).
 https://doi.org/10.1016/j.cep.2019.107538
- [28] G.S. Ho, H.M. Faizal, F.N. Ani, Microwave induced plasma for solid fuels and waste processing: A review on affecting factors and performance criteria, Waste Management 69 (2017) 423-430.
 <u>https://doi.org/10.1016/j.wasman.2017.08.015</u>
- [29] A. Sanlisoy and M.O. Carpinlioglu, Preliminary measurements on microwave plasma flame for gasification. Energy,
 Ecology and Environment 3 (2018) 32-38. <u>https://doi.org/10.1007/s40974-017-0063-x</u>
- [30] A. Sanlisoy and M.O. Carpinlioglu, Microwave Plasma Gasification of a Variety of Fuel for Syngas Production,
 Plasma Chemistry and Plasma Processing 39 (2019) 1211-1225. <u>https://doi-org/10.1007/s11090-019-10004-x</u>
- [31]H.S Uhm, Y.H. Na, Y.C. Hong, D.H. Shin, C.H. Cho, Y.K. Park, High-Efficiency Gasification of Low-Grade Coal by Microwave Steam Plasm, Energy Fuels 28 (2014) 4402-4408. <u>https://doi.org/10.1021/ef500598u</u>
- [32]H.S. Uhm, Y.H. Na, Y.C. Hong, D.H. Shin, C.H. Cho, Production of hydrogen-rich synthetic gas from low-grade
 coals by microwave steam-plasmas, International Journal of Hydrogen Energy 39 (2014) 4351-4355.
 <u>https://doi.org/10.1016/j.ijhydene.2014.01.020</u>
- [33] J. Heberlein, A.B. Murphy, Thermal plasma waste treatment, Journal of Physics D: Applied Physics 41 (2008).
 <u>https://doi.org/10.1088/0022-3727/41/5/053001</u>
- [34] U. Arena, Process and technological aspects of municipal solid waste gasification A review, Waste Management 32 (2012) 625-639. <u>https://doi.org/10.1016/j.wasman.2011.09.025</u>
- [35] T.M. Ismail, A. Ramos, M. Abd El-Salam, E. Monteiro, A. Rouboa, Plasma fixed bed gasification using an Eulerian
 model, International Journal of Hydrogen Energy 44 (2019) 28668-28684.
 <u>https://doi.org/10.1016/j.ijhydene.2019.08.035</u>

- [36] S. Vecten, M. Wilkinson, A. Martin, A. Dexter, N. Bimbo, R. Dawson, B. Herbert, Experimental study of steam and carbon dioxide microwave plasma for advanced thermal treatment application, Energy 207 (2020) 118086. https://doi.org/10.1016/j.energy.2020.118086
- [37] L. Su, R. Kumar, B. Ogungbesan, M. Sassi, Experimental investigation of gas heating and dissociation in a microwave
 plasma torch at atmospheric pressure, Energy conversion and management 78 (2014) 695-703.
 https://doi.org/10.1016/j.enconman.2013.12.001
- [38] P.V. Aravind, W. de Jong, Evaluation of high temperature gas cleaning options for biomass gasification product gas
 for Solid Oxide Fuel Cells, Progress in Energy and Combustion Science 38 (2012) 737-764.
 https://doi.org/10.1016/j.pecs.2012.03.006
- [39] H.S. Uhm, J.H. Kim, Y.C. Hong, Disintegration of water molecules in a steam-plasma torch powered by microwaves,
 Physics of Plasmas 14 (2007). <u>https://doi-org/10.1063/1.2749225</u>
- [40] S.K. Hyoung, H.S. Uhm, Y.C. Hong, E.H. Choi, Disintegration of Carbon Dioxide Molecules in a Microwave Plasma Torch, Scientific Reports 5 (2015). <u>https://doi.org/10.1038/srep18436</u>
- [41] F. Saleem, J. Harris, K. Zhang, A. Harvey, Non-thermal plasma as a promising route for the removal of tar from the product gas of biomass gasification A critical review, Chemical Engineering Journal 382 (2020).
 <u>https://doi.org/10.1016/j.cej.2019.122761</u>
- [42] N. Abdoulmoumine, S. Adhikari, A. Kulkarni, S. Chattanathan, A review on biomass gasification syngas cleanup,
 Applied Energy 155 (2015) 294-307. <u>https://doi.org/10.1016/j.apenergy.2015.05.095</u>
- [43] J.F. De La Fuente, A.A. Kiss, M.T. Radoiu, G.D. Stefanidis, Microwave plasma emerging technologies for chemical processes, Journal of Chemical Technology & Biotechnology 92 (2017) 2495-2505. https://doi-org/10.1002/jctb.5205
- [44] A.S. Gilmour, Klystrons, traveling wave tubes, magnetrons, crossed-field amplifiers, and gyrotrons. Artech House;
 2011.
- [45] O.Z. Sharaf, M.F. Orhan, An overview of fuel cell technology: Fundamentals and applications, Renewable and
 Sustainable Energy Reviews 32 (2014) 810-853. <u>https://doi.org/10.1016/j.rser.2014.01.012</u>
- 472 [46] H.G. Song, Y.N. Chun, Microwave gasification and oxy-steam combustion for using the biomass char, The Journal
 473 of Material Cycles and Waste Management 22 (2020) 176-186. <u>https://doi.org/10.1007/s10163-019-00926-1</u>
- [47] A. Perna, M. Minutillo, A. Lubrano Lavadera, E. Jannelli, Combining plasma gasification and solid oxide cell technologies in advanced power plants for waste to energy and electric energy storage applications, Waste Management 73 (2018) 424-438. <u>https://doi.org/10.1016/j.wasman.2017.09.022</u>
- 477 [48]European Commission, EU Circular Economy Action Plan. <u>https://ec.europa.eu/environment/circular-economy/</u>,
 478 2020 (accessed 25 April 2020).