

Experimental study and numerical modeling of supercritical carbonation of steel fiber reinforced concrete

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Abstract: In this paper, supercritical carbonation tests of steel fiber reinforced concrete (SFRC) are carried out. The effects of volume fraction and length-diameter ratio of steel fiber on the carbonation depth of SFRC under supercritical conditions are studied. A novel multi-phase and multi-physics coupling model for supercritical carbonation of SFRC is proposed, which considers random distribution of coarse aggregate, porosity, and steel fibers in SFRC, as well as the distribution of interfacial transition zones (ITZ) between coarse aggregate, steel fibers and cement. The results indicate that the porosity of the SFRC is reduced by 32.3%, and its compressive strength of SFRC increases by 25.1% ~ 42.7% after supercritical carbonation treatment. When the volume fraction of steel fiber is less than 1.5%, the supercritical carbonation depth of the SFRC decreases with the increase of the volume fraction and the length to diameter ratio, respectively. The influence of the ITZ between coarse aggregate and cementitious matrix on the supercritical carbonation depth of the SFRC is found to be 2 to 6.8 times greater than that of the ITZ between steel fibers and cementitious matrix. The average carbonation depth of the SFRC increases gradually with the increase of ITZ thickness and porosity. The effect of ITZ porosity on the supercritical carbonation depth of the SFRC is more significant than that of ITZ thickness.

Keywords: Steel fiber reinforced concrete; Supercritical carbonation; Multi-phase and multi-physics coupling model; ITZ; Carbonation depth.

1. Introduction

Fiber Reinforced Concrete (FRC), recognized as an advanced engineering material, has seen extensive application in civil engineering due to its superior mechanical properties and durability [1, 2]. Compared to the conventional concrete, FRC significantly enhances tensile and flexural strengths as well as toughness by integrating steel fibers into the concrete matrix [3, 4]. This integration reduces brittleness and dry shrinkage deformation [5], delays or prevents cracking [6-8], and refines the microstructure [9, 10]. These improvements offer significant advantages to FRC in applications such as tunnel lining [11], bridge construction, airport runways, and military engineering [12]. For ordinary concrete structures, the migration of CO₂ and the carbonation reaction within the concrete convert Ca(OH)₂ into CaCO₃. This process lowers the pH of the pore solution, disrupts the passive film on the surface of the steel reinforcement, and ultimately leads to steel corrosion [13, 14]. However, SFRC is a type of cementitious composite material. Upon carbonation infiltration, CO₂ reacts with alkaline substances within the SFRC matrix and the interfacial transition zone (ITZ) between steel fibers and matrix, forming calcium carbonate with a larger molar volume than the original phase [15].

42 This process potentially reduces the pore volume and permeability of the concrete matrix and ITZ [16]. The
43 addition of steel fibers into the concrete matrix can enhance its compactness and increase resistance to CO₂
44 penetration [17, 18], thereby improving the carbonation resistance of the concrete [19]. The enhancement of
45 carbonation resistance in concrete due to steel fibers is primarily attributed to the reduction of plastic shrinkage
46 cracks and the decreased porosity [20-22]. During the carbonation reaction in a steel fiber concrete, the
47 resulting CaCO₃ fills the pore spaces [23, 24], thereby reducing porosity and CO₂ diffusion capability, which
48 in turn decreases the carbonation reaction rate [21].

49 Accelerated carbonation has been widely used to study carbonation properties of fiber-reinforced concrete
50 [25]. This process accelerates carbonation by increasing CO₂ concentration and controlling environmental
51 conditions such as temperature and humidity. Although accelerated carbonation shortens the time compared to
52 natural carbonation, it still takes several months to reach the desired carbonation depth [26, 27]. In recent years,
53 with the advancement of CO₂ storage technology [28], supercritical carbonation has attracted attention as an
54 emerging research method. This technique has been applied to treat cementitious materials [29-33] and heavy
55 metal seals [34-36]. When the temperature and pressure exceed 304.12 K and 7.38 MPa, respectively, CO₂
56 exists in a supercritical fluid state [37] with gas-liquid two-phase characteristics. This state significantly
57 enhances the diffusion and reaction rates of CO₂ within the concrete, thereby accelerating the carbonation
58 process [38, 39]. Supercritical carbonation technology has become a promising, time- and labor-saving method
59 in the study of carbonation resistance for fiber-reinforced concrete. In addition, supercritical carbonation of
60 fiber-reinforced concrete can significantly increase the strength and toughness of the material [40, 41],
61 improves pore size distribution [41], enhances permeation resistance [42], alters chemical and structural
62 properties of the concrete [39], reduces porosity, and increases material density [43, 44]. However, the addition
63 of steel fibers can increase the pathways for CO₂ penetration, reduce matrix density, and consequently decrease
64 the carbonation resistance of SFRC. It was found that adding steel fibers can change carbonation rate of
65 concrete, as observed in [20] that adding a volume fraction of 1.5% steel fibers result in the slowest
66 carbonation rate [20]. However, when the volume fraction was increased to 2%, the carbonation rate
67 accelerated [20, 45]. Thus, the optimal content of steel fiber remains a subject of debate.

68 Compressive strength and carbonation depth are primary indicators used to evaluate carbonation
69 performance of SFRC treated with supercritical CO₂ [46, 47]. Supercritical carbonation of SFRC is a complex
70 physicochemical process involving multi-physical field, coupling chemical reaction rate equations, mass
71 conservation equations, kinetic energy conservation equations, and energy conservation equations. Various
72 factors such as porosity, water content saturation, liquid permeability, and gas permeability influence the depth
73 of supercritical carbonation [40, 46]. Additionally, SFRC is a non-homogeneous material composed of phases
74 like coarse aggregate, cement matrix, steel fibers, micropores, and interfacial transition zones, all of which
75 affect the supercritical carbonation depth and compressive strength of SFRC. Consequently, supercritical
76 carbonation of SFRC is a multifaceted multiphase and multi-field coupled physicochemical reaction process.
77 Currently, both experimental and numerical simulation studies on the supercritical carbonation of ordinary
78 concrete exist. Yu et al. conducted an experimental study on the supercritical carbonation of concrete and
79 proposed a multi-field coupled model that considers physicochemical processes such as carbonation reaction,
80 gas-liquid two-phase flow in the porous medium, and CO₂ dissolution and diffusion in water [23, 48]. To
81 characterize the distribution of coarse aggregate, porosity, and interfacial transition zone in concrete, a

82 concrete multiphase model was proposed to investigate the effects of these factors on the depth of supercritical
 83 carbonation [38]. However, there are few studies on the supercritical carbonation of SFRC [23], and the
 84 strengthening mechanism of SFRC through supercritical carbonation treatment remains uncertain.

85 To elucidate the carbonation mechanism and properties of SFRC under supercritical CO₂ conditions,
 86 supercritical carbonation tests are conducted on SFRC with varying mix ratios. The changes in
 87 micro-morphology, porosity and compressive strength of SFRC specimens before and after supercritical
 88 carbonation are analyzed. Based on the previous work [38, 46], in this paper, a multifield coupling model for
 89 the supercritical carbonation of SFRC is established. To accurately represent the random distribution of coarse
 90 aggregate, porosity, and steel fibers in SFRC, as well as to account for the interfacial transition zone between
 91 the coarse aggregate, steel fibers, and the cementitious matrix, a multiphase model of SFRC is proposed.
 92 Utilizing this multifield coupling model for supercritical carbonation and the multiphase model of SFRC, a
 93 multiphase and multi-field coupling model of SFRC supercritical carbonation is developed. This model is
 94 capable of simulating the effects of multiple parameters, including aspect ratio of steel fibers, volume fraction
 95 of steel fibers, thickness and porosity of the interfacial transition zone between coarse aggregate and steel
 96 fibers, on the depth of supercritical carbonation in SFRC.

97

98 **2. Experimental investigation**

99 *2.1. Raw materials and mix proportion*

100 In accordance with the specifications for the mix proportion design of ordinary concrete [49] and the
 101 standard test methods for fiber reinforced concrete [50], the targeted compressive strength of steel
 102 fiber-reinforced concrete (SFRC) is set at 40 MPa. The cement utilized in the experiments is sourced from
 103 Ezhou, Hubei, and is classified as P.O 42.5 ordinary Portland cement. The coarse aggregates range in size from
 104 5 to 20 mm, with continuous grading, a mud (silt/clay) content of 0.41%, and bulk and specific densities of
 105 1429 kg/m³ and 2797 kg/m³, respectively. The fine aggregates consist of natural river sand, with a fineness
 106 modulus of 2.70 and bulk and specific densities of 1398 kg/m³ and 2549 kg/m³, respectively. A polycarboxylic
 107 acid high-performance water-reducing agent produced by Foshan Xinqi Tuoda New Materials Group Co., Ltd,
 108 with a water reduction rate of 30%, was used. The respective design mix proportions of the SFRC are
 109 presented in Table 1. Shear wave steel fibers with the length-diameter ratios of 30, 60, 80, manufactured by
 110 Wuhan Hansen Steel Fiber Limited Liability Company, were used to prepare the SFRC specimens. The
 111 physical and mechanical properties of the steel fibers are shown in Table 2.

112 **Table 1**

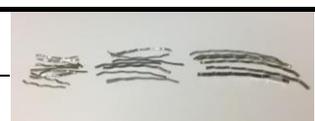
113 Mix proportions of concrete.

| Cement (kg/m ³) | Coarse aggregate (kg/m ³) | Fine aggregate (kg/m ³) | Water (kg/m ³) | Water-binder ratio |
|-----------------------------|---------------------------------------|-------------------------------------|----------------------------|--------------------|
| 445 | 1078 | 732 | 178 | 0.40 |

114 **Table 2**

115 Physical and mechanical properties of steel fibers.

| Equivalent diameter (mm) | Density (g/cm ³) | Length-diameter ratio | Length (mm) | Tensile strength (MPa) |
|--------------------------|------------------------------|-----------------------|--------------|------------------------|
| 0.55 | 7.8 | 30, 60, 80 | 16.5, 33, 44 | 1120 |



116 *2.2. Specimen preparation*

117 The performance of steel fiber-reinforced concrete is profoundly influenced by two pivotal factors:
 118 volume fraction and length-diameter ratio of steel fibers. To study carbonation performance of SFRC under
 119 supercritical conditions, a range of fiber volume fractions, namely 0%, 0.5%, 1%, and 1.5% with
 120 length-diameter ratios of 30, 60, and 80 are considered. The design of the SFRC specimens is outlined in Table
 121 3.

122 **Table 3**

123 Design details of SFRC specimens.

| Specimen | Length-diameter ratio, λ | Volume fraction, ρ | Slump (mm) |
|---------------|----------------------------------|-------------------------|------------|
| SFRC-L60-V0.0 | 60 | 0 | 34 |
| SFRC-L60-V0.5 | 60 | 0.5% | 32 |
| SFRC-L60-V1.0 | 60 | 1.0% | 31 |
| SFRC-L60-V1.5 | 60 | 1.5% | 30 |
| SFRC-L30-V1.5 | 30 | 1.5% | 32 |
| SFRC-L80-V1.5 | 80 | 1.5% | 28 |

124 Note: In Table 2, the numbers after L and V are the length-diameter ratio and steel fiber volume fraction, respectively, of a SFRC.

125 Slump is one of the methods for assessing the workability of SFRC. To ensure the workability of SFRC
 126 [50], the slump of the SFRC specimens was assessed using a standard slump cone. The slump cone was placed
 127 on a non-absorbent rigid horizontal base plate. The mixed SFRC was poured into the cone in three layers of
 128 thickness, each filling approximately one-third of the cone's height. Each of the layers underwent 25
 129 compaction cycles. Once the final layer was poured and settled, the slump cone was lifted straight up within 5
 130 seconds, allowing the concrete to settle freely. The slump values for each type of the SFRC were determined
 131 by measuring the vertical distance between the highest point of the concrete and the original cone height using
 132 a ruler, as shown in Table 3.

133 To investigate the carbonation properties of the SFRC with varying length-diameter ratios and volume
 134 fractions under supercritical CO₂ conditions, SFRC cubes with a side length of 100 mm are manufactured. Six
 135 sets of nine SFRC cubes each are made according to the specifications outlined in Table 3. Among the nine
 136 specimens of each group, three of them are tested before carbonation and another three are tested after
 137 carbonation for compressive strength. The remaining three are tested for carbonation depth. To ensure a
 138 uniform dispersion of steel fibers in the concrete and prevent fiber clumping, the mixing process followed the
 139 mix proportions of concrete in Tables 1 and the design details in Table 3. First, the aggregates, cement, and
 140 water were mixed thoroughly for 2 minutes. Next, the steel fibers were added gradually during the mixing
 141 process until all the fibers were added. This was finished by an additional 3 minutes mixing to ensure best
 142 possible uniformity. The thoroughly mixed concrete is then poured into molds. A vibration table is used to
 143 shake the molds to release any bubbles that may form in the pouring process. After casting, the molds are
 144 transferred to a wet room cured for 24 hours before being demolded. The specimens are subsequently cured in
 145 a standard curing room with a temperature of (20±2)°C and humidity of (95±3)% for 28 days. To maintain a

146 relative humidity of 70% within the SFRC specimens, the cured SFRC specimens are placed in a temperature
147 and humidity-controlled chamber for an additional 28 days.

148 2.3. Supercritical carbonation test

149 To assess the performance of SFRC under supercritical CO₂ conditions, a closed-cycle supercritical
150 carbonation system is employed as shown in Fig. 1. The system operates by compressing air with an air
151 compressor to propel a booster pump. Simultaneously, the booster pump continuously compresses and propels
152 the gas from the CO₂ cylinder into the reaction chamber until the pressure and temperature in the reaction
153 chamber surpass 7.38 MPa and 31.2°C, respectively. At this point, the CO₂ in the reaction chamber is in a
154 supercritical state, enabling supercritical carbonation of SFRC. Further details on the closed-cycle supercritical
155 carbonation system for cementitious materials can be found in the authors' previous research [51].

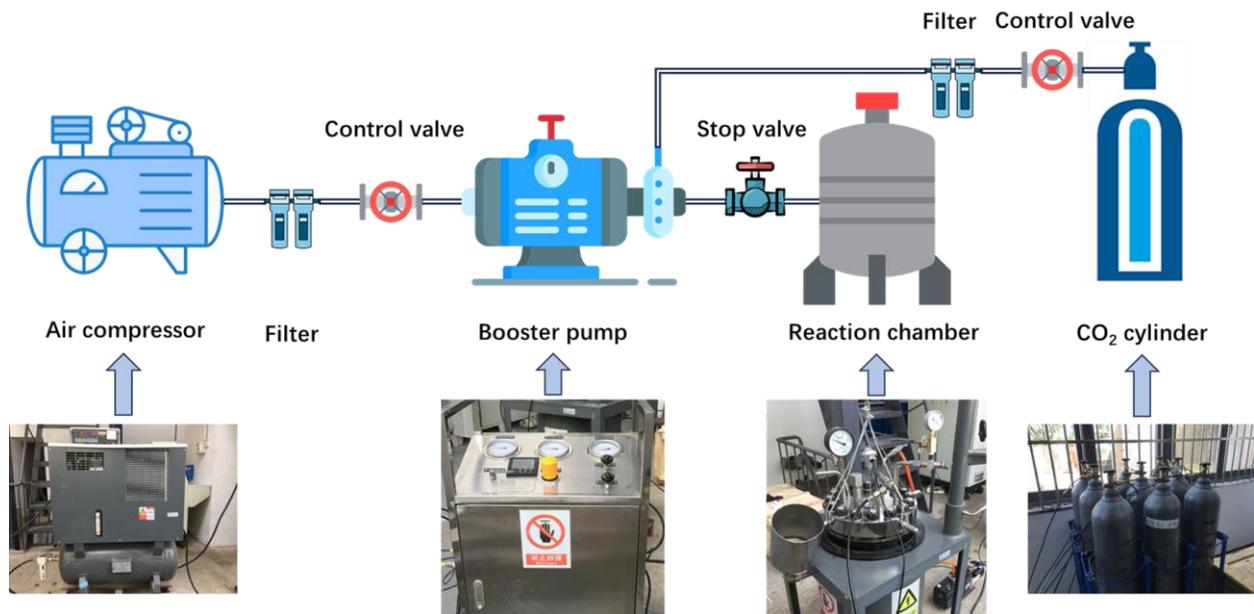


Fig. 1. Equipment connection diagram for supercritical carbonation system.

156 The supercritical carbonation process of the SFRC employing a closed-cycle supercritical carbonation
157 system is primarily structured into four distinct test phases: test preparation, CO₂ boosting, CO₂ holding, and
158 CO₂ recycling. The operational steps are summarized below:

159 (1) Test preparation phase

160 1.1) Six specimens are positioned within the reaction chamber, and the lid of the reaction chamber is
161 securely tightened.

162 1.2) The CO₂ cylinders are linked to the CO₂ gas source inlet of the booster pump, and the high-pressure
163 CO₂ outlet of the booster pump is connected to the gas delivery port of the chamber. At this stage, all the
164 switches on both the chamber and the booster pump are switched off, and the chamber is vacuumed until an
165 approximate pressure of -1.00 bar is reached.

166 (2) CO₂ boosting phase

167 2.1) All switches on reaction chamber, booster pump, and CO₂ cylinder are activated to allow CO₂ being
168 pumped into the reaction chamber driven by the pressure difference. The pressures in the reaction chamber and
169 the CO₂ cylinder are closely monitored and documented until equilibrium is reached between the two.

170 2.2) The throttle valve controlling the driving gas on the top of the booster pump is adjusted continuously
171 to allow a continuous flow of CO₂ into the chamber until the pressure reaches the designated target.

172 2.3) The temperature inside the reaction chamber is continuously regulated to attain the specified
173 temperature through water circulation and a heating rod.

174 (3) CO₂ holding phase

175 3.1) Once the pressure and temperature inside the reaction chamber has reached the designated levels, the
176 CO₂ cylinder, booster pump, and all switches on the reaction chamber are deactivated.

177 3.2) The fluctuations in temperature and pressure in the reaction chamber are closely monitored and
178 documented. In the event of a decline in air pressure in the chamber, the booster pump is reactivated. CO₂
179 injection into the reaction chamber resumes until the target pressure is reached again.

180 3.3) Supercritical carbonation of the SFRC takes place under the designated holding pressure and
181 temperature, for the specified carbonation time.

182 (4) CO₂ recycling phase

183 4.1) Upon completion of the supercritical carbonation test for the SFRC, all valves are closed.
184 Subsequently, the gas outlet of the reaction chamber is connected to the inlet of the CO₂ gas source on the
185 booster pump, followed by connecting the CO₂ gas outlet of the booster pump to the CO₂ gas cylinder.

186 4.2) All the switches on the CO₂ cylinder, booster pump, and chamber are opened. The CO₂ from the gas
187 chamber is injected into the cylinder due to the pressure difference.

188 4.3) Upon achieving equilibrium between the pressures within the chamber and the CO₂ cylinder, the
189 control valve at the booster pump is tuned to introduce the CO₂ from the chamber into the cylinder, thereby
190 initiating the process of CO₂ recovery.

191 Throughout the supercritical carbonation of the SFRC, the predetermined pressure and temperature are
192 maintained at 8.0 MPa and 40°C, respectively. The total carbonation time is 7.2 hours, including 6 hours
193 supercritical carbonation time. The air pressure and temperature in the reactor are recored at 10-minute
194 intervals encompassing CO₂ pressurization, CO₂ holding pressure, and CO₂ recovery phases. Upon the
195 conclusion of the supercritical carbonation test, the six specimens are extracted from the reaction chamber. Fig.
196 2 illustrates the temperature and pressure profiles of select specimens during the supercritical carbonation test,
197 which shows that the designed carbonation system is capable of consistently regulating the internal pressure
198 and temperature of the reaction chamber by the closed-cycle supercritical carbonation process.

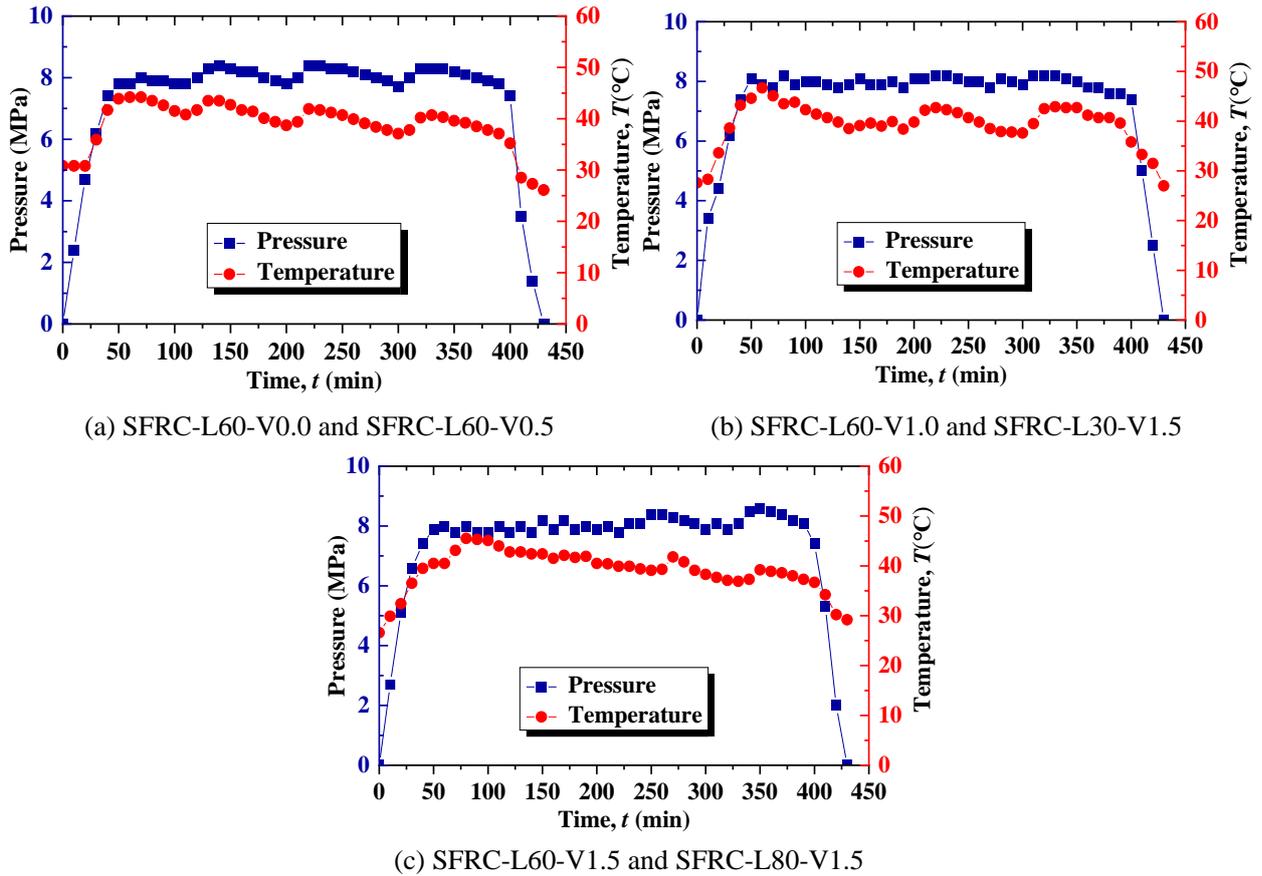


Fig. 2. Temperature and pressure in the reaction chamber during supercritical carbonation.

199 *2.4. Compressive strength testing*

200 After completing the supercritical carbonation test on the SFRC specimens, selected samples are prepared
 201 for compressive strength testing. The compressive strength of the SFRC specimens before and after
 202 supercritical carbonation is measured using a 3000 kN MTS YAW6306 testing machine, following the
 203 standard steel fiber reinforced concrete test method (CECS 13:2009) [50]. The loading rate is 1.2 MPa/s. The
 204 test system automatically records the applied load, including the ultimate load, and the corresponding
 205 displacements.

206 *2.5. Carbonation depth measurement*

207 After completing supercritical carbonation process, the SFRC specimens are removed from the reactor.
 208 Each specimen is then halved at an appropriate cutting speed to prevent fiber pullout and detachment of the
 209 interface between the cement matrix and the fibers. The surfaces of cuts are properly cleaned before
 210 phenolphthalein solution, a pH indicator, is uniformly sprayed over the cuts. The chemical reaction shows pink
 211 in alkaline (non-carbonated) concrete, while remaining colorless in areas where carbonation has occurred. The
 212 carbonation profiles of the specimens are obtained using an image scanner [23], and the mean and standard
 213 deviation of carbonation depth are then measured.

214 *2.6. Microscopic analysis*

215 *2.6.1. SEM test*

216 To explore alterations in the micro-morphology of SFRC after supercritical carbonation and investigate
217 the influence of steel fibers on the depth of carbonation, Microscopic images of the SFRC samples, both before
218 and after supercritical carbonation, are taken and examined to study the micro-morphological features, such as
219 steel fibers, matrix, interfacial transition zones (ITZs), and cracks, using a scanning electron microscope (SEM,
220 JEOL JSM-7800F). Before observation, the carbonated SFRC samples were precisely cut, broken, and cleaned
221 to ensure that the test samples retained the steel fibers and the matrix. This was done to minimize the potential
222 interference of dust on the carbonation results of the SFRC. The samples are dried up to the same weight in an
223 oven, and their surfaces are coated with ion-sputtered gold powder to increase electrical conductivity.
224 Subsequently, the samples are examined by the SEM to observe the microstructure of both the matrix and the
225 steel fibers in the samples.

226 2.6.2. MIP test

227 To evaluate the impact of supercritical carbonation on the pore structure of concrete, mercury pressure
228 tests are carried out on specimens before and after supercritical carbonation using a mercury piezometer
229 (AutoPore IV 9500). Before the MIP test, samples with and without being super-critically carbonated is
230 randomly chosen from those that has been examined by SEM. The samples are immersed in anhydrous ethanol
231 to arrest hydration and subsequently sealed to prevent further carbonation. The specimens are then subjected to
232 MIP test to study their pore size distribution, and changes in concrete porosity due to supercritical carbonation.

233 3. Numerical simulation

234 3.1. Multiphase model of SFRC

235 3.1.1. Random porosity model

236 To characterize the stochastic and inhomogeneous distribution of porosity within SFRC, and to account
237 for the correlation in the spatial distribution of porosity, it is assumed that the porosity follows a lognormal
238 distribution. Eq. (1) provides the equation for the spatial correlation function employed to describe the random
239 distribution of porosity in SFRC.

$$\phi(x, y) = \exp \left[- \left(\frac{x^2}{a^2} + \frac{y^2}{b^2} \right)^{1+r} \right] \quad (1)$$

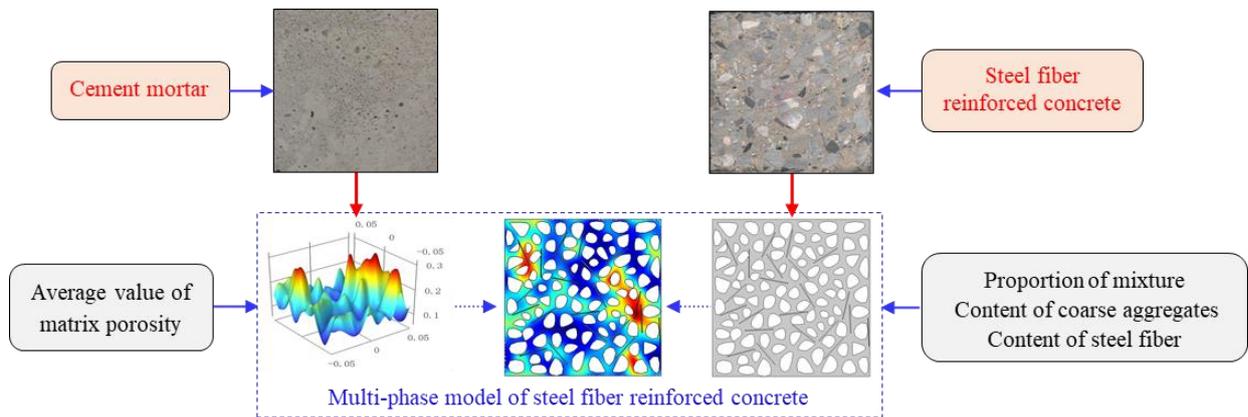
240 where a and b represent the autocorrelation lengths, assumed to be 0.01 m for both a and b [51], and r stands
241 for the roughness coefficient, set to 0 in this paper. The Fourier transform and inverse Fourier transform of Eq.
242 (1) are employed to formulate the stochastic porosity model for SFRC. Further insights into the procedure for
243 generating the random porosity model can be found in Yu [48].

244 3.1.2. 2D random steel fibers and aggregates model

245 For the grading and proportion of coarse aggregates in SFRC, the Monte Carlo method is employed to

246 generate circular aggregates randomly within specified space, from which the coordinates of the center and the
 247 area of each circular aggregate are determined. A polygonal Voronoi diagram is then created using the center of
 248 each circular aggregate as a control point. Utilizing the vertices of each of the polygonal Voronoi cells as the
 249 points for interpolation, closed B-spline curves are generated by connecting the vertices. For crushed stone
 250 aggregates, convex polygons are created by selecting nodes from the closed B-spline curves. The area is
 251 proportionally scaled down to ensure equivalence between the area of the resulting rubble and the area of the
 252 initially generated circular aggregate at the corresponding location. Based on the diameter, length, and quantity
 253 of steel fibers, the Monte Carlo method is applied also to randomly position steel fibers in the above generated
 254 aggregate model. Steel fibers are subsequently repositioned if collisions with aggregates occur. This process
 255 leads to the establishment of a two-dimensional random steel fibers and aggregates model for SFRC. Further
 256 details regarding the generation of the random aggregate model can be found in Bao [52].

257 A multiphase model, which can characterize random distribution of matrix porosity, coarse aggregate, and
 258 steel fibers in concrete is developed by the following steps: (1) incorporate the measured average porosity of
 259 the matrix into the porosity random distribution model for microscopic characterization of the random
 260 distribution of matrix porosity; (2) utilize the mix ratio, the generated aggregate and steel fibers and the
 261 characteristic parameters of SFRC as input for the random aggregate and fiber distribution model of concrete;
 262 and (3) ultimately establish a multiphase model of SFRC that considers porosity, recycled aggregate, and steel
 263 fiber distribution. The method for establishing the multiphase model of SFRC is illustrated in Fig. 3.



264
 265 **Fig. 3.** Flow chart of multi-phase modeling method of SFRC.

266 *3.1.3. Distribution of ITZs around coarse aggregates and steel fibers*

267 An ITZ exists between the coarse aggregate, steel fibers, and cement matrix [53, 54]. The presence of this
 268 ITZ also impacts the distribution of supercritical carbonation depth of concrete. To study the impact of the ITZ
 269 between coarse aggregate and steel fibers on the supercritical carbonation depth, a multiphase model is
 270 developed, encompassing the ITZs of both coarse aggregate and steel fibers. The distribution model of the ITZ
 271 is proposed based on the random model of steel fibers and aggregates.

272 The process of ITZ generation is as follows: (1) Initially, the vertex coordinates of each convex polygon
273 of the coarse aggregate and the four vertex coordinates of the steel fibers are obtained. (2) Subsequently, the
274 boundary of each convex polygon of the coarse aggregate and the rectangular steel fiber is expanded parallel to
275 the original boundary by a distance equivalent to the thickness of the ITZ. (3) The region between the extended
276 coarse aggregate boundary and the original boundary constitutes the ITZ of the coarse aggregate, while the
277 region between the extended steel fiber boundary and its original boundary forms the ITZ of the steel fiber. Fig.
278 4 illustrates the distribution of the ITZ surrounding the coarse aggregate and steel fibers, each with a thickness
279 of 100 μm .

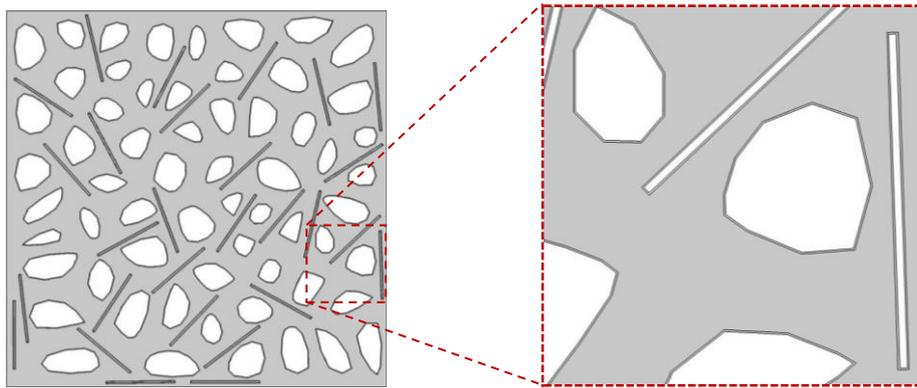


Fig. 4. Distribution of the ITZ surrounding the coarse aggregates and steel fibers.

280 3.2. Multi-physics coupling model for supercritical carbonation of SFRC

281 Supercritical carbonation of SFRC is a complex multi-physics coupled reaction process. In order to
282 describe the transport and reaction process of supercritical CO_2 in SFRC, a supercritical carbonation model for
283 SFRC is established based on previous study [23]. The model considers processes such as the carbonation
284 reaction between CO_2 and $\text{Ca}(\text{OH})_2$, gas and liquid flow in porous media, dissolution and diffusion of CO_2 in
285 the pore water film, and heat transfer in porous media. Darcy's law was employed to describe the permeation
286 process of CO_2 in SFRC under different carbonation conditions. Finally, a multi-physics coupled theoretical
287 model for the supercritical carbonation of SFRC is established. The development of the model is illustrated in
288 Fig. 5.

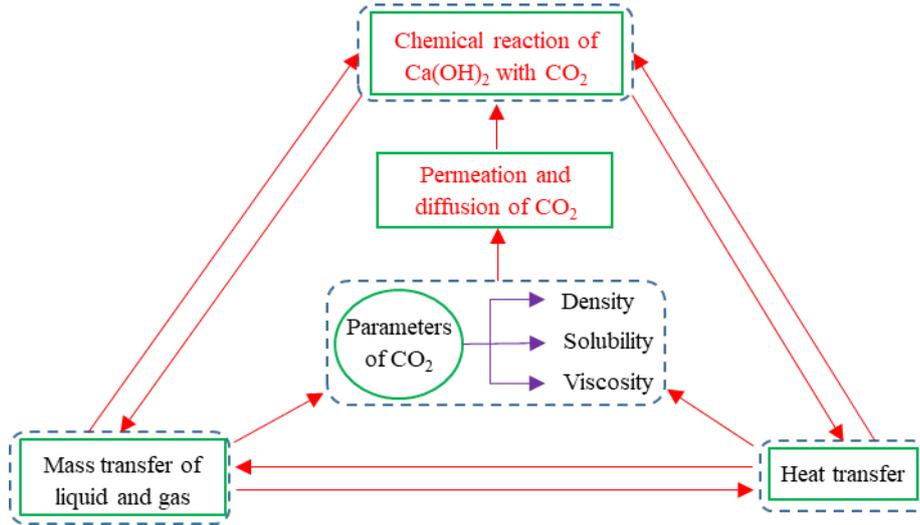


Fig. 5. Flow chart of multi-physics coupled modeling method for supercritical carbonation of SFRC.

The governing equations for the supercritical carbonation of SFRC are presented in Eqs. (2-6). The mathematical model considers the chemical reaction rate, conservation of mass in gas-liquid two-phase flow, diffusion and dispersion of CO₂ in water, conservation of energy in porous media, and solubility of CO₂ in water.

$$\frac{\partial R_c}{\partial t} = \alpha_1 \times f_1(h) \times f_2(g_v) \times f_3(R_c) \times f_4(T) \quad (2)$$

$$\frac{\partial(g)}{\partial t} = \frac{\partial(m_{co_2})}{\partial t} \quad (3)$$

$$\frac{\partial(nS_\alpha \rho_\alpha)}{\partial t} + \nabla \cdot (\rho_\alpha \mathbf{u}_\alpha) = q_\alpha \quad (4)$$

$$\mathbf{u}_\alpha = - \frac{k_0 \left(\frac{n}{n_0}\right)^3 \cdot \left(\frac{1-n_0}{1-n}\right)^2 k_{r\alpha}}{\mu_\alpha} (\nabla P_\alpha - \rho_\alpha \mathbf{g}) \quad (5)$$

$$(\rho C_q)_{eff} \frac{\partial T}{\partial t} = \nabla \cdot (k_{eff} \nabla T) - (C_g \rho_g \mathbf{u}_g + C_w \rho_w \mathbf{u}_w) \nabla T \quad (6)$$

where: R_c represents the degree of carbonation; g denotes the mass concentration of CO₂ in water; n signifies the decrease in porosity of the material during carbonation; P_α denotes the pressure in phase α , with the subscript α referring to the liquid phase (w) and the gas phase (g). Additional information about other parameters can be found in Zha [46] and Yu [48].

To solve Eqs. (2-6), the following initial and boundary conditions for the supercritical carbonation of SFRC, as outlined in Eqs. (7-9), are considered.

$$R_c = R_{c0} = 0, P_g = P_{g0}, P_w = P_{w0}, g = g_0 = 0, T = T_0, t = 0 \text{ on } \Omega \quad (7)$$

$$\vec{n} \cdot \nabla R_c = 0, \vec{n} \cdot \nabla g = 0 \text{ on } \Gamma_2 \quad (8)$$

$$P_g = P_{g,sur}, P_w = P_{w,sur}, T = T_{sur} \text{ on } \Gamma_1 \quad (9)$$

301 where: R_{c0} denotes the initial condition of carbonation degree; P_{g0} and P_{w0} represent the initial water pressure
 302 and the initial air pressure, respectively; g_0 indicates the initial dissolved concentration of CO₂ in water; T_0 is
 303 the initial temperature; \vec{n} is the normal vector of the boundary; Γ_2 and Γ_1 are the boundaries using
 304 Neumann's and Dirichlet's conditions, respectively; $P_{g,sur}$ and $P_{w,sur}$ denote the surrounding gas and liquid
 305 pressures, respectively; and T_{sur} is the ambient temperature. Details of the other parameters can also be found
 306 in Zha [46] and Yu [48].

307 3.3. Multi-phase and multi-physics coupling model for supercritical carbonation of SFRC

308 The multi-phase model and the multi-physics coupling model of supercritical carbonation of SFRC are
 309 integrated to formulate the multi-phase and multi-physics coupling model for supercritical carbonation in
 310 SFRC. SFRC can be considered as a porous medium material consisting of coarse aggregates, steel fibers, a
 311 cement matrix, and pores. The random aggregates and steel fibers model is utilized to describe the distribution
 312 of coarse aggregates and steel fibers in SFRC, respectively. The input parameters of the model, such as the mix
 313 proportion, the coarse aggregate content, and the volume fraction and length to diameter ratio of steel fibers,
 314 are Section 2.1 and 2.2.

315 The transport and reaction of CO₂ in the SFRC are influenced by the material properties, such as the
 316 initial porosity and intrinsic permeability, that need to be determined for the multi-phase and the multi-physics
 317 coupling supercritical carbonation model. Specifically, the initial average porosity of the cement matrix is
 318 determined through the MIP test, as illustrated in Table 4. The temperature and pressure data obtained during
 319 the supercritical carbonation test of the SFRC specimens are incorporated into the numerical carbonation
 320 model. The mean and variance of the carbonation depth are statistically analyzed by comparing the
 321 carbonation depth distribution obtained from both the supercritical carbonation test and the multi-phase model
 322 and the multi-physics coupling model. The multi-phase model and the multi-physics coupling model for
 323 supercritical carbonation of SFRC is then calibrated and refined.

324 **Table 4**

325 Main characteristic parameters of SFRC materials.

| Item | Value | References |
|--|---|------------|
| Intrinsic permeability, k_0 | $1.5 \times 10^{-20} \text{ m}^2$ | [55] |
| Capillary pressure curve coefficient, α | $5.3695 \times 10^{-8} \text{ Pa}^{-1}$ | [56] |
| Relative permeability coefficient, m | 0.4396 | [56] |
| Relative humidity, h_0 | 0.70 | Measured |
| Initial porosity, n_0 | 14.7% | Measured |

326

327

328 **4. Results and discussion**

329 *4.1. Compressive strength of SFRC under supercritical carbonation*

330 Fig. 6 presents the compressive strength of SFRC with a water-cement ratio of 0.4 before and after
331 supercritical carbonation, considering different volume fraction and length to diameter ratio of steel fiber.
332 Specifically, Fig. 6(a) and Fig. 6(b) illustrate the effects of volume fraction and length to diameter ratio on the
333 compressive strength of SFRC before and after supercritical carbonation, respectively. Before supercritical
334 carbonation, the compressive strength of the SFRC is increased by 2.6%, 12.5%, and 18.4% when steel fibers
335 of 0.5%, 1.0%, and 1.5% in volume fraction is, respectively, added to the concrete. When the length to
336 diameter ratio of the steel fibers is 60 and 80, the respective compressive strength of the SFRC is 9.5% and
337 11.3% higher than when the ratio is 30. This enhancement can be attributed to the bridging effect of steel
338 fibers within the concrete matrix, which restrains crack initiation and propagation under loading, thereby
339 improving load-bearing capacity [57, 58]. After supercritical carbonation, the compressive strength of the
340 SFRC shows a significant increase of 25.1% ~ 42.7% compared to that before carbonation. This improvement
341 is likely due to the deposition of CaCO₃ within internal pores and at the ITZs between steel fibers and the
342 cement matrix [59]. The filling of these voids densifies the SFRC matrix and ITZs, thereby enhancing its
343 overall compressive strength.

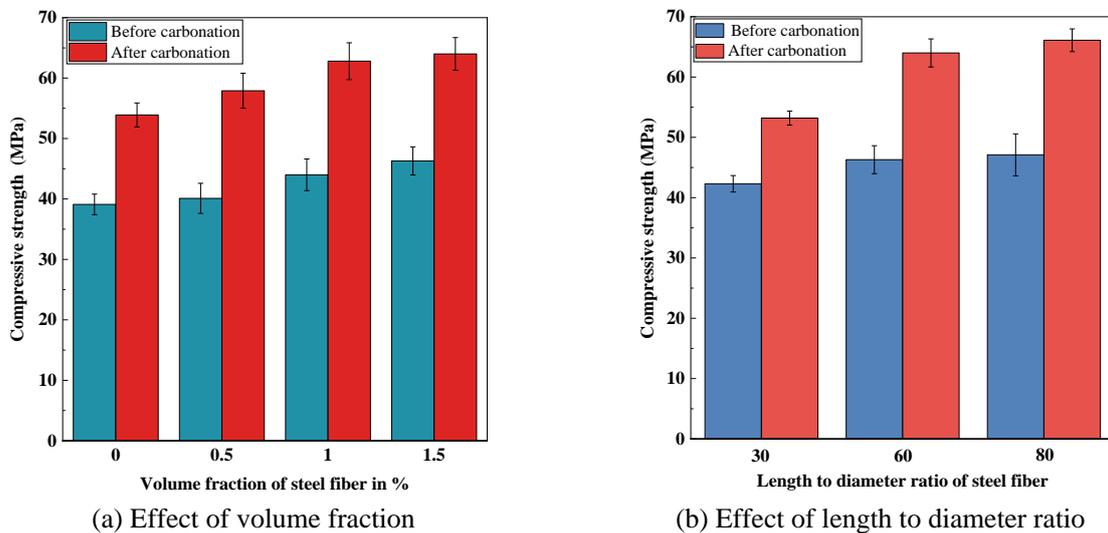


Fig. 6. Effect of volume fraction and length to diameter ratio of steel fiber on the compressive strength of SFRC.

344 *4.2. Microstructure evolution of SFRC under supercritical carbonation*

345 *4.2.1. Morphology evolution of SFRC*

346 The microstructure of the cementitious matrix and cracks before and after supercritical carbonation of the
347 concrete without steel fibers are shown in Fig. 7a and b at a magnification of 1000 times. It can be observed
348 that micro-cracks and tiny pores are present within the matrix before carbonation (Fig. 7a). However, although

349 micro-cracks and small pores still exist within the matrix after carbonation, both their quantity and size are
 350 reduced, and the cement matrix as a whole is denser (Fig. 7b). This indicates that the porosity or cracks inside
 351 the matrix can be reduced by supercritical carbonation treatment. To evaluate the evolution of the SFRC
 352 microstructure before and after supercritical carbonation treatment, the SFRC specimen (SFRC-L60-V1.5)
 353 with a volume fraction of 1.5% is selected for the analysis. To observe the steel fibers, cracks, and the cement
 354 matrix simultaneously under the scanning electron microscope (SEM), a magnification factor of 200 is chosen.
 355 The SEM images of SFRC-L60-V1.5 before and after supercritical carbonation are shown in Fig. 7 (c) and (d),
 356 respectively. It can be observed that micro-cracks appear along the narrow gap between the matrix and the
 357 steel fibers in the non-carbonated samples. Although cracks are also observed in the area between the matrix
 358 and steel fibers in the carbonated samples, the crack length is reduced, and the matrix and steel fibers are
 359 firmly bonded together. This suggests that porosity or cracks between the matrix and steel fibers can be
 360 reduced by supercritical carbonation. The weakness of the interfacial transition zones (ITZs) between cement
 361 matrix and fibers [10, 60] and between cement matrix and aggregate [38] may serve as a shortcut for CO₂
 362 transport, which leads to a more rapid carbonation in the ITZ. In addition to the porosity, coarse aggregate and
 363 ITZ, the presence of steel fibers in the SFRC may also contribute to the stochastic nature of the carbonation
 364 boundaries.

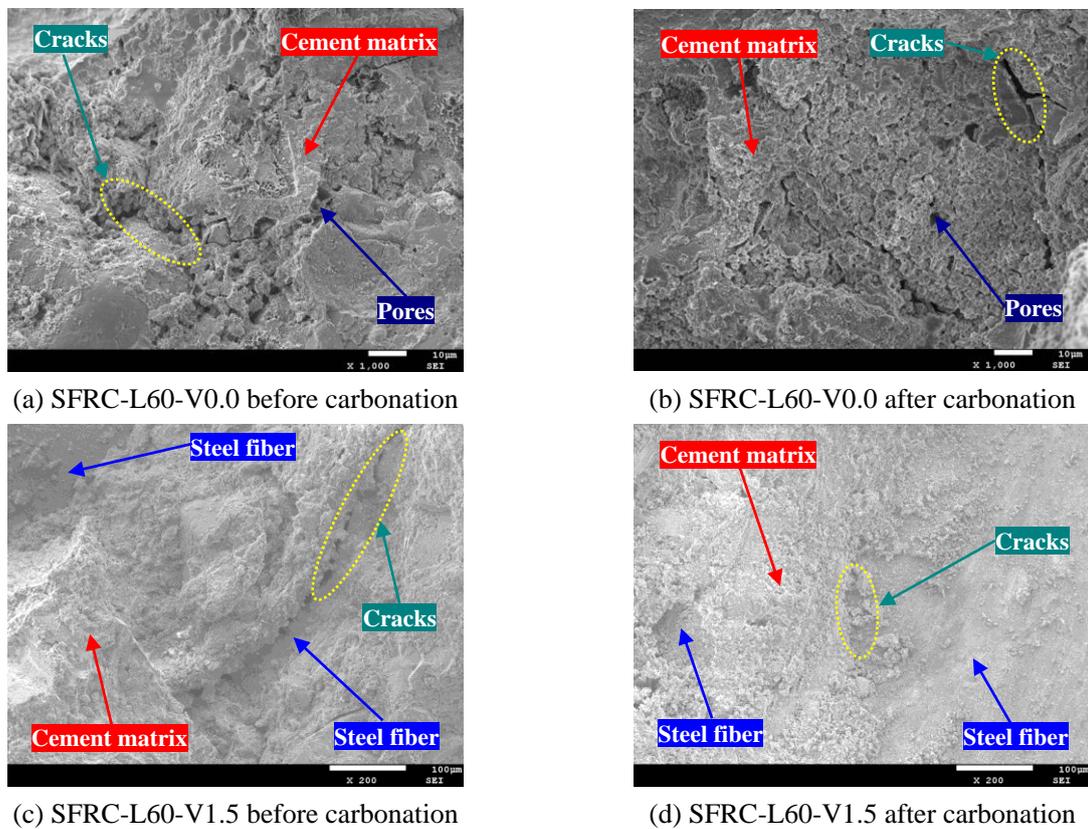


Fig. 7. SEM images of SFRC with and without steel fibers before and after supercritical carbonation.

366 To evaluate the change in porosity of the SFRC before and after supercritical carbonation and to calibrate
 367 the parameters for the numerical model, Mercury Intrusion Porosimetry (MIP) tests are conducted on the
 368 SFRC matrix before and after supercritical carbonation. The obtained Mercury porosimetry curves for
 369 SFRC-L60-V1.5 are shown in Fig. 8. The porosity of the SFRC matrix before supercritical carbonation
 370 treatment is 14.7%, and the porosity after treatment is 9.1%. The porosity of the SFRC after supercritical
 371 carbonation treatment is reduced by 32.3%. This reduction is primarily due to the reaction of $\text{Ca}(\text{OH})_2$ and
 372 C-S-H in the pores of the cement matrix with CO_2 to form CaCO_3 , which blocks the pore structure [61]. The
 373 initial average porosity and the coefficient of variation of porosity are 14.7% and 0.4 [51], respectively, and
 374 these two parameters will be input to the numerical model.

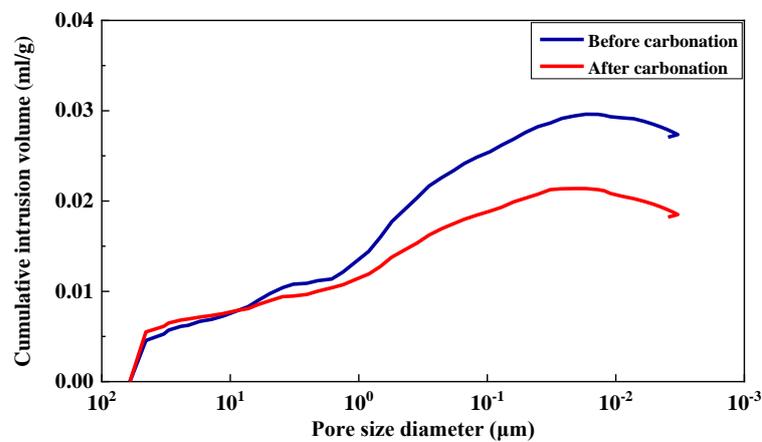
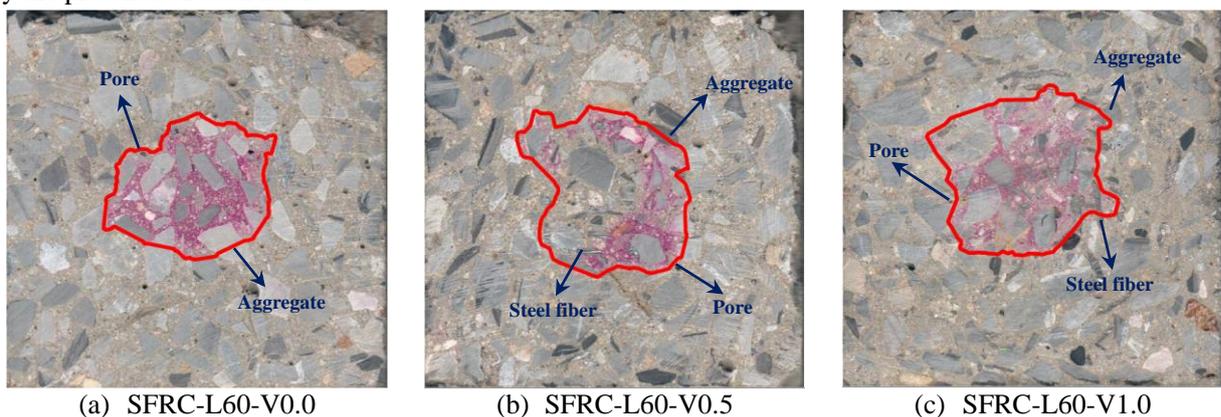
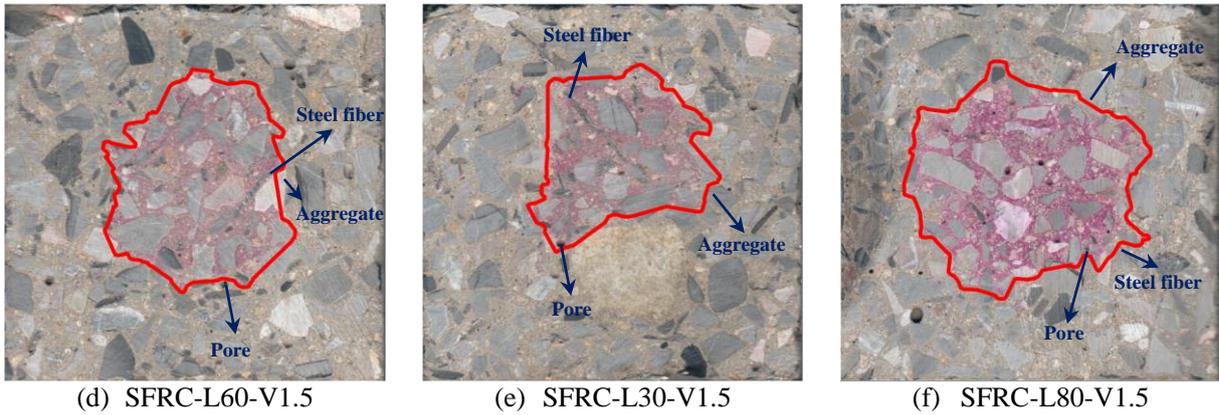


Fig. 8. Mercury porosimetry curves for SFRC-L60-V1.5 before and after supercritical carbonation.

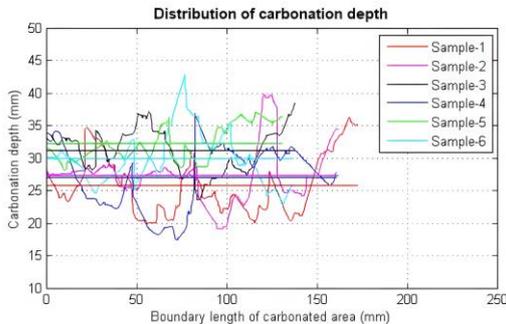
375
 376 *4.3. Experimental supercritical carbonation results of SFRC and validation of numerical models*
 377 *4.3.1. Supercritical carbonation depth of SFRC*
 378 In order to elucidate the variation of carbonation depth of the SFRC under supercritical CO_2 conditions,
 379 the carbonation depth of the SFRC specimens is measured and analyzed. The carbonation contours of selected
 380 SFRC specimens are illustrated in Fig. 9. It is evident that the randomness of concrete carbonation boundaries
 381 can be influenced not only by the random distribution of coarse aggregate and matrix pores [51, 52] but also
 382 by the presence of steel fibers.





(d) SFRC-L60-V1.5 (e) SFRC-L30-V1.5 (f) SFRC-L80-V1.5
Fig. 9. Supercritical carbonation depth distribution of SFRC with different volume fraction and length-diameter ratio of steel fibers.

383 To assess the impact of steel fibers on carbonation within the SFRC, MATLAB R2014a for statistical
 384 analysis of carbonation depth in cementitious materials is utilized for the statistical analysis of supercritical
 385 carbonation depth. The statistical analysis of SFRC carbonation depth distribution curves along the
 386 carbonation area is shown in Fig. 10. Fig. 10 (a) shows the distribution curves of the carbonation depth along
 387 the carbonation area, where the horizontal lines are the average. The results of average, maximum, minimum
 388 carbonation depth and variance of carbonation depth of the SFRC are shown in the table in Fig. 10 (b). The
 389 mean values of carbonation depth and coefficient of variation after completion of the supercritical carbonation
 390 tests on the SFRC specimens are shown in Table 5.



(a) Distribution curves of carbonation depth

| Statistical results on the carbonation depth of cementitious materials | | | | |
|--|--------------------|--------------------|--------------------|-----------------------------|
| | Average value (mm) | Maximum value (mm) | Minimum value (mm) | Variance (mm ²) |
| Sample-1 | 25.7172 | 36.2288 | 19.9153 | 19.6612 |
| Sample-2 | 27.2709 | 39.8169 | 19.2220 | 16.6891 |
| Sample-3 | 31.1499 | 38.4279 | 23.5808 | 11.1544 |
| Sample-4 | 27.0550 | 36.7713 | 17.4888 | 24.0181 |
| Sample-5 | 32.2142 | 37.0450 | 28.0514 | 6.9879 |
| Sample-6 | 29.9225 | 42.8904 | 22.8438 | 16.7747 |
| Average value | 28.8883 | 38.5301 | 21.8503 | 15.8809 |

(b) Statistical analysis of carbonation depth

Fig. 10. Statistical analysis of supercritical carbonation depth of SFRC.

391 4.3.2. Validation of the multi-phase and multi-physics coupling model

392 The multi-phase and multi-physics coupling model is applied to simulate the supercritical carbonation
 393 process of the SFRC. The mixing proportion and properties of the SFRC in Table 1-3 are used in the model. To
 394 eliminate the effect of random distribution of porosity on the carbonation depth, it is assumed that the porosity
 395 of the SFRC cement paste before supercritical carbonation is uniformly distributed and is 14.7%. Table 5
 396 presents the results of numerical simulations for the mean values and the coefficient of variation of carbonation
 397 depth after supercritical carbonation. The simulations include three aspects: (1) carbonation without
 398 considering the ITZ between aggregates, steel fibers, and the cement matrix; (2) carbonation considering the
 399 ITZ between aggregates and the cement matrix only, but not between steel fibers and the cement matrix; and (3)

400 carbonation considering the ITZ between aggregates and the cement matrix, as well as between steel fibers and
 401 the cement matrix simultaneously. By comparing the mean values of the carbonation depth and the coefficient
 402 of variation obtained from experiments and numerical simulations, it can be found that the results obtained by
 403 the numerical model are close to the experimental values, proving the validity of the model.

404 Table 5 shows that under supercritical CO₂ conditions, the average carbonation depth of the SFRC
 405 decreases by 5.7% to 16.6% as the volume fraction of steel fibers increases, while keeping the length to
 406 diameter ratio of the steel fibers. The increase in steel fiber volume fraction reduces plastic shrinkage cracks
 407 and lowers porosity, limiting the extension and connectivity of cracks. This limits carbonation pathways,
 408 enhances the overall density of the concrete, reduces porosity, and lowers the permeability of CO₂, thereby
 409 significantly improving the concrete's resistance to carbonation. When the volume fraction of steel fiber is
 410 fixed, the average carbonation depth of the SFRC under supercritical CO₂ conditions decreases by 5.4% to
 411 13.1% as the length to diameter ratio of the steel fibers increases. Adding steel fibers with higher length to
 412 diameter ratio improves concrete microstructure, increases density, and reduces porosity. This, in turn,
 413 decreases the permeability of CO₂ and reduces carbonation depth.

414 As shown in Table 5, when examining solely the effect of the ITZ between coarse aggregate and cement
 415 matrix on carbonation depth, it is observed that the carbonation depth of the SFRC increases by 2.4% to 4.8%.
 416 When considering the effect of the ITZ between steel fiber and cement matrix alone, the carbonation depth of
 417 the SFRC is found to be increased by 0.5% to 2.3%. When evaluating the combined effects of the ITZs
 418 between the coarse aggregate and the cementitious matrix, and between the steel fiber and the cementitious
 419 matrix on carbonation depth, it is found that the carbonation depth of the SFRC is increased by 3.3% to 7.1%.
 420 The impact of the ITZ between the coarse aggregate and the cementitious matrix on the carbonation depth is
 421 found to be 2 to 6.8 times greater than that of the ITZ between the steel fibers and the cementitious matrix.

422 **Table 5**

423 Comparison of experimental and numerical results of supercritical carbonation depth of SFRC.

| Specimens number | Test results | | Simulation results | | | | | |
|------------------|----------------|--------|--|--------|--|--------|--|--------|
| | | | $t_{CC}=0\mu\text{m}, t_{SC}=0\mu\text{m}$ | | $t_{CC}=100\mu\text{m}, t_{SC}=0\mu\text{m}$ | | $t_{CC}=100\mu\text{m}, t_{SC}=100\mu\text{m}$ | |
| | D_{ave} (mm) | CV_D | D_{ave} (mm) | CV_D | D_{ave} (mm) | CV_D | D_{ave} (mm) | CV_D |
| SFRC-L60-V0.0 | 32.58 | 0.17 | 32.46 | 0.03 | 33.23 | 0.02 | 33.52 | 0.01 |
| SFRC-L60-V0.5 | 30.73 | 0.15 | 30.92 | 0.07 | 31.94 | 0.08 | 32.09 | 0.07 |
| SFRC-L60-V1.0 | 30.17 | 0.11 | 29.72 | 0.43 | 30.9 | 0.36 | 31.3 | 0.42 |
| SFRC-L60-V1.5 | 27.17 | 0.18 | 28.14 | 0.66 | 29.48 | 0.61 | 30.14 | 0.62 |
| SFRC-L30-V1.5 | 28.73 | 0.13 | 29.8 | 0.06 | 30.91 | 0.04 | 31.37 | 0.04 |
| SFRC-L80-V1.5 | 24.96 | 0.14 | 27.16 | 0.53 | 28.26 | 0.44 | 28.8 | 0.39 |

424 Remarks: D_{ave} and CV_D and are average carbonation depth and the coefficient of variation, respectively. t_{CC} and t_{SC} are the
 425 thicknesses of the ITZ between coarse aggregate and cementitious matrix and the ITZ between steel fibres and cementitious
 426 matrix, respectively.

427

428 4.4. Effect of volume fraction and length to diameter ratio of steel fiber on supercritical carbonation depth

429 To further study the evolution of carbonation depth of SFRC under supercritical CO₂ conditions with
430 different volume fractions and length to diameter ratios of steel fibers, steel fiber volume fractions of 0%, 0.5%,
431 1.0%, and 1.5%, and length to diameter ratios of 30, 60, and 80 are selected. The ITZ thickness between the
432 cement matrix, coarse aggregate, and steel fibers is uniformly set to be 100 μ m. The porosity of the ITZ is set to
433 be 1.5 times that of the cement matrix. Numerical simulations of the multi-phase model for the SFRC are
434 conducted for the above volume fractions and length to diameter ratios. For each case, three multiphase models
435 are randomly generated and coupled with the SFRC supercritical carbonation multi-physics model. The
436 average carbonation depth from the three models is then calculated as the numerically simulated supercritical
437 carbonation depth of the SFRC. As shown in Fig. 11, the presence of steel fibers significantly increases the
438 irregularity of carbonation depth, and carbonation boundary irregularities are evident at the interfaces between
439 the cement matrix, coarse aggregate, and steel fibers.

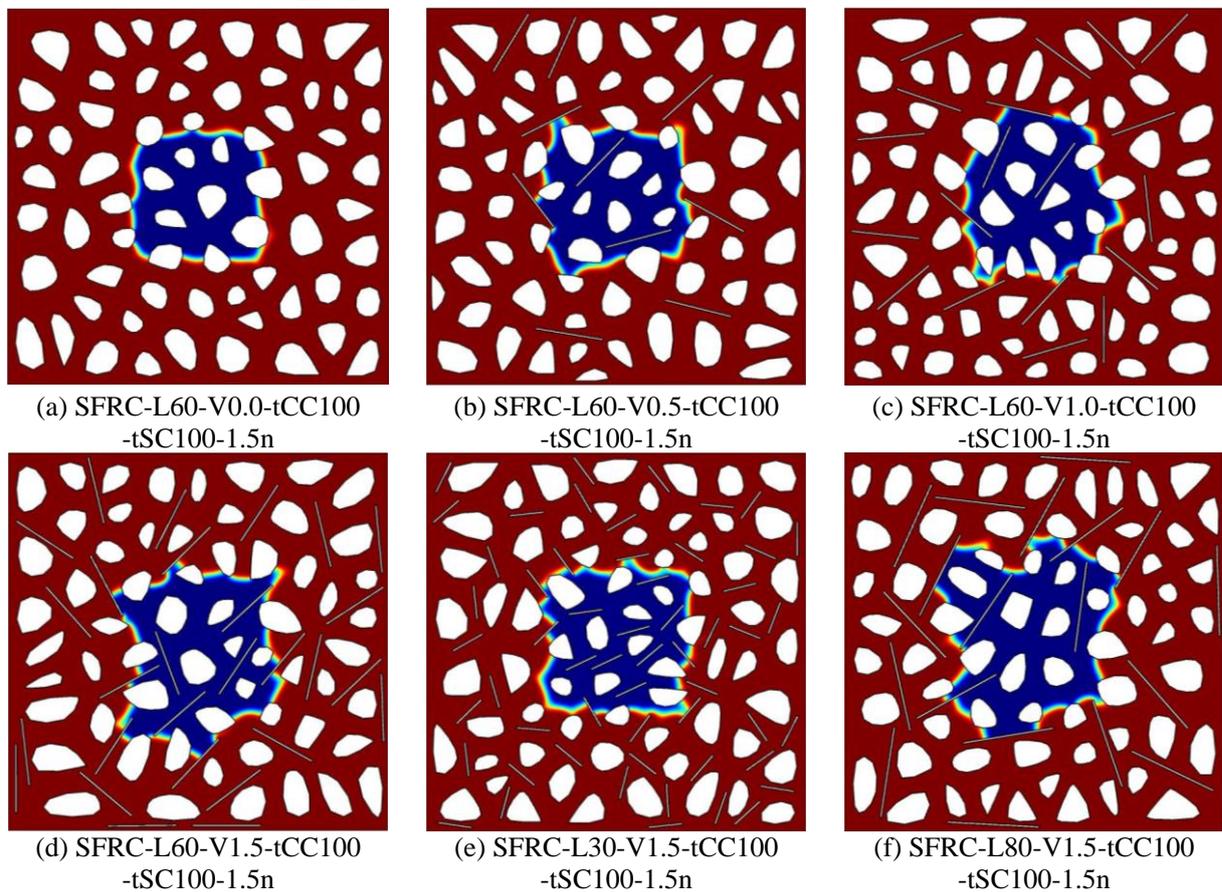


Fig. 11. Supercritical carbonation profiles of SFRC with different volume fraction and length to diameter ratio of steel fiber.

440 Note: Taking SFRC-L60-V1.5-tCC100-tSC100-1.5n as an example, tCC100 and tSC100 represent the thickness of the ITZ
441 between the cementitious matrix and coarse aggregate, and between the cementitious matrix and steel fibers, respectively, both
442 measuring 100 mm, 1.5n denotes the ITZ porosity, which is 1.5 times the porosity of the cementitious matrix.

443 The effects of different volume fractions and length to diameter ratios of steel fibers on the supercritical
444 carbonation depth of the SFRC are shown in Fig. 12. Fig. 12(a) demonstrates the effect of various steel fiber

445 volume fractions on the carbonation depth when the length to diameter ratio of the steel fibers is 60. It can be
 446 observed that as the volume fraction of steel fibers increases, the average carbonation depth of the SFRC
 447 gradually decreases. Specifically, for every 5% increase in the volume fraction of steel fibers, the carbonation
 448 depth decreases by an average of 3.5%. However, the standard deviation of the carbonation depth continuously
 449 increases. When the steel fiber volume fraction is not greater than 2%, the improvement in the concrete's
 450 carbonation resistance due to the increased steel fiber volume fraction is attributed to the reduction in concrete
 451 plastic shrinkage cracks and the decrease in porosity. After the carbonation reaction, the generated CaCO_3 fills
 452 the pores, reducing porosity and decreasing the diffusion capability of CO_2 , thereby slowing down the
 453 carbonation reaction.

454 Fig. 12(b) shows the effect of different length to diameter ratios of steel fibers on the carbonation depth
 455 when the steel fiber volume fraction is 1.5%. It can be seen that as the length to diameter ratio of the steel
 456 fibers increases, the average carbonation depth of the SFRC gradually decreases. Specifically, for every 20-30
 457 increase in the length to diameter ratio, the carbonation depth decreases by an average of 3.9%-4.4%. However,
 458 the standard deviation of the carbonation depth first increases and then decreases. Steel fibers with a higher
 459 length to diameter ratio bridge cracks more effectively, thus limiting crack propagation, improving the
 460 microstructure and density of the concrete, and ultimately reducing carbonation depth. Additionally, steel
 461 fibers with a larger length to diameter ratio are more likely to form a randomly distributed network structure
 462 within the concrete, effectively enhancing tensile strength in all directions and improving overall carbonation
 463 resistance.

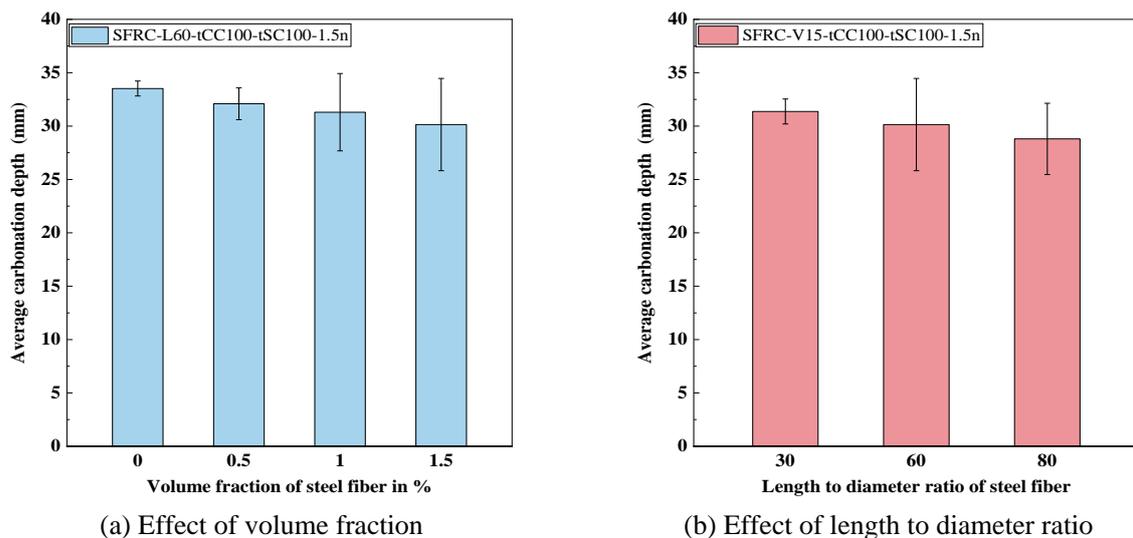


Fig. 12. Effect of volume fraction and length to diameter ratio of steel fiber on supercritical carbonation depth of SFRC.

464 *4.5. Effect of thickness and porosity of ITZ on supercritical carbonation depth*

465 The thickness and porosity of the ITZ between the cement matrix, coarse aggregate, and steel fibers are
 466 two significant factors that influence the carbonation process of the SFRC. Generally, the impact of ITZ on the

467 supercritical carbonation depth of SFRC cannot be studied separately through experiments. Moreover, due to
 468 the ITZ thickness being only a few tens of micrometers, it is challenging to cast specimens with varying ITZ
 469 thicknesses to experimentally study their effect on the supercritical carbonation depth of SFRC. Given the
 470 experimental difficulties, this section uses numerical methods to investigate the influence of ITZ on the
 471 supercritical carbonation depth of SFRC. To clarify the influence of ITZ thickness and porosity between the
 472 cement matrix, coarse aggregate, and steel fibers on the carbonation depth of SFRC under supercritical CO₂
 473 conditions, ITZ thicknesses of 0 μ m, 50 μ m, and 100 μ m, and ITZ to cement matrix porosity ratios of 1.0, 1.5,
 474 and 2.0 are selected. The volume fraction and length to diameter ratio of steel fiber are uniformly set to 1.5%
 475 and 80, respectively. For each ITZ thickness and porosity, three multiphase models are randomly generated and
 476 coupled with the SFRC supercritical carbonation multi-physics model. The average carbonation depth of the
 477 three models is taken as the numerically simulated supercritical carbonation depth of the SFRC. Fig. 13 shows
 478 some of the numerical simulation results, which shows the effect of the ITZ thickness and porosity between the
 479 cement matrix, coarse aggregate, and steel fibers on the distribution of SFRC supercritical carbonation depth.

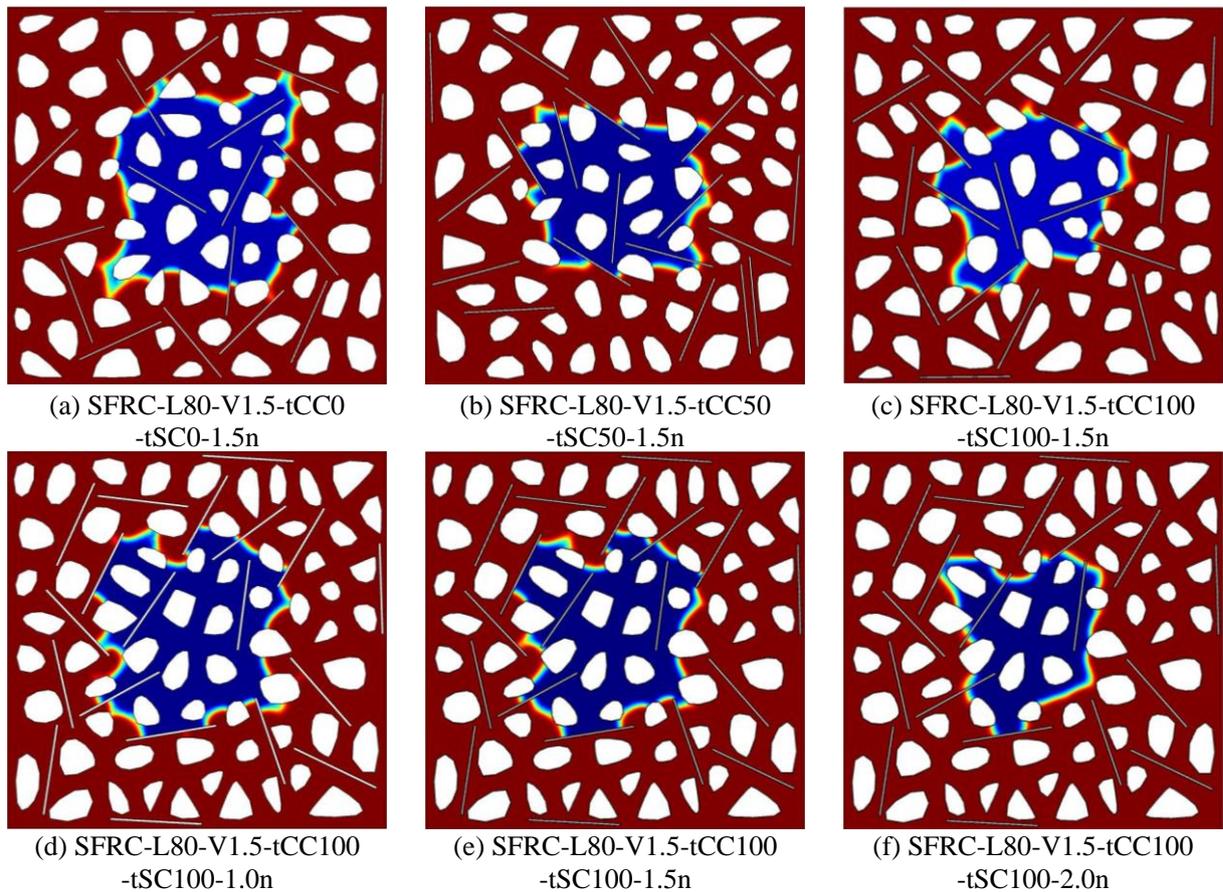


Fig. 13. Supercritical carbonation profiles of SFRC with different thickness and porosity of ITZ.

480 The effects of different ITZ thicknesses and porosities on the supercritical carbonation depth of the SFRC
 481 are shown in Fig. 14. Fig. 14(a) demonstrates the impact of various ITZ thicknesses on carbonation depth
 482 when the porosity ratio between the ITZ and the cement matrix is 1.5. It can be observed that as the ITZ
 483 thickness increases, the average carbonation depth gradually increases. A thicker ITZ, due to its high porosity,

484 low alkalinity and tendency to form cracks, provides more pathways and channels for the carbonation reaction,
 485 thereby increasing the carbonation depth. The average supercritical carbonation depth for ITZ thicknesses of
 486 50 μm and 100 μm increases 2.2% and 6.0%, respectively, compared to the model with an ITZ thickness of
 487 0 μm .

488 Fig. 14(b) shows the impact of different porosity ratios between the ITZ and the cement matrix on the
 489 carbonation depth when the ITZ thickness between the cement matrix, coarse aggregate, and steel fibers is
 490 uniformly set to 100 μm . As the porosity ratio increases, the average carbonation depth of the SFRC gradually
 491 increases. Specifically, for every 0.5 increase in the porosity ratio, the carbonation depth increases by an
 492 average of 6.0% to 9.8%. An ITZ with higher porosity forms more interconnected pores and microcracks,
 493 which provides additional pathways for gases and liquids, thereby increasing the permeability of the concrete.
 494 This allows CO₂ to more easily diffuse through these pores and cracks into the concrete, where it reacts with
 495 Ca(OH)₂ to form CaCO₃, accelerating the carbonation process. Moreover, a highly porous ITZ promotes
 496 moisture migration, which is a crucial factor in the carbonation reaction. Moisture migration can carry more
 497 CO₂ into the concrete, accelerating the carbonation reaction and further increasing the carbonation depth [62].
 498 The impact of ITZ porosity on the supercritical carbonation depth of SFRC is more pronounced compared to
 499 the impact of ITZ thickness.

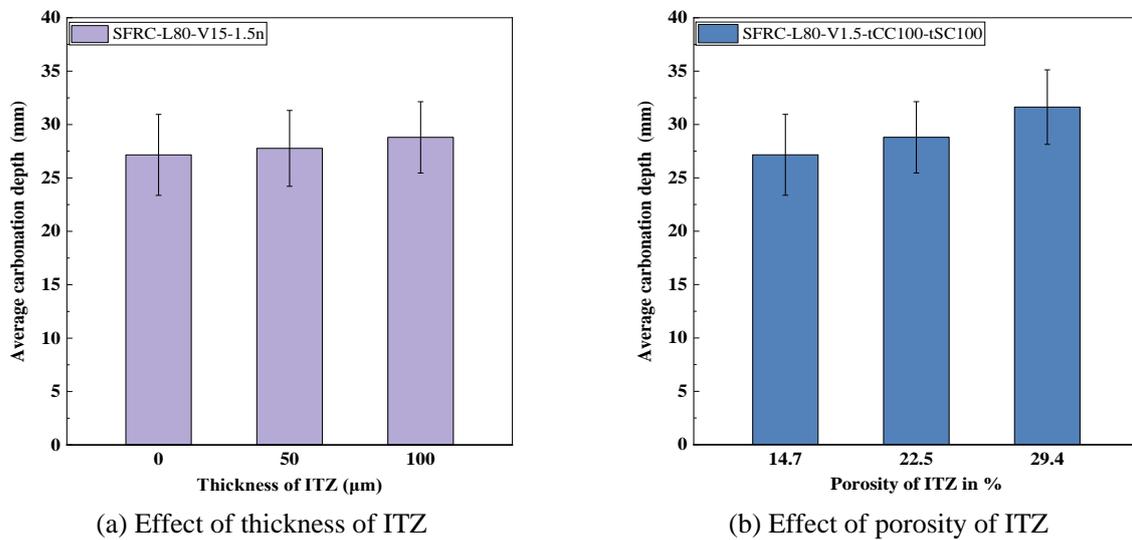


Fig. 14. Effect of thickness and porosity of ITZ on supercritical carbonation depth of SFRC.

500
 501 **5. Conclusions**

502 This paper presents a comprehensive experimental and numerical investigation into supercritical
 503 carbonation of SFRC. The primary conclusions are as follows:

- 504 (1) After supercritical carbonation treatment, the porosity of the SFRC is reduced by 32.3%. The ITZ
 505 between the steel fibers and the cement matrix becomes denser, and the porosity decreases. When the volume
 506 fraction of steel fibers is less than 1.5%, adding steel fibers can enhance carbonation resistance of the concrete.
- 507 (2) After supercritical carbonation, the compressive strength of SFRC increased by 25.1% ~ 42.7%

508 compared to that before carbonation.

509 (3) The proposed multi-phase and multi-physics coupling model for supercritical carbonation of SFRC
510 can simulate random distribution of coarse aggregate, porosity, and steel fibers, along with ITZ between coarse
511 aggregate and steel fibers and cementitious.

512 (4) When the volume fraction of steel fiber does not exceed 1.5%, the average supercritical carbonation
513 depth of SFRC decreases by a maximum of 16.6% to 13.1% as the volume fraction and length to diameter
514 ratio increases, respectively.

515 (5) The effect of the ITZ between the coarse aggregate and the cementitious matrix on supercritical
516 carbonation depth of SFRC is found to be 2 to 6.8 times greater than that of the ITZ between the steel fibers
517 and the cementitious matrix.

518 (6) The average carbonation depth of the SFRC gradually increases with the increase of ITZ thickness and
519 porosity. The effect of ITZ porosity on the supercritical carbonation depth of SFRC is more pronounced
520 compared to the impact of ITZ thickness.

521

522 **Data Availability Statement**

523 Some or all data, models, or code that support the research findings of this study are available on request.

524

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529

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