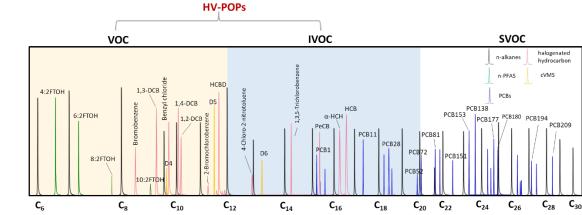
1	Overlooked Highly-Volatile Persistent Organic
2	Pollutants in the Atmosphere
3	
4	
5	Shizhen Zhao ¹ , Kevin C. Jones ² , Roland Weber ³ , Yuwei Xiao ^{1,4} , Gan Zhang ^{*1}
6	
7 8 9	¹ State Key Laboratory of Organic Geochemistry and Guangdong Key Laboratory of Environmental Resources Utilization and Protection, Guangzhou Institute of Geochemistry, Chinese Academy of Sciences, Guangzhou 510640, China
10	² Lancaster Environment Centre, Lancaster University, Lancaster, LA1 4YQ, UK
11	³ POPs Environmental Consulting, 73527 Schwäbisch Gmünd, Germany
12	⁴ University of Chinese Academy of Sciences, Beijing 100049, China
13	
14	*Corresponding author:
15	Gan Zhang
16	Email: <u>zhanggan@gig.ac.cn</u>

18 The atmosphere plays an important role in the cycling of persistent organic pollutants 19 (POPs) in the Earth surface system. It is also an ideal monitoring matrix as it provides 20 essential information on the distribution, sources, and transmission of POPs, to 21 support the risk assessments and the effectiveness evaluation of POPs management 22 policies. While many POPs in the atmosphere are considered 'semi-volatile', implying 23 dynamic partitioning between the gas and particle phases, and having the tendency 24 for temperature-driven air-surface exchanges, some emerging organic pollutants or 25 POPs that are newly listed in the Stockholm Convention occur in the atmosphere are 26 either highly volatile (cf. VOCs) or are with low or extremely low volatility. Currently, 27 highly volatile POPs (HV-POPs), which predominantly occur in the gas phase, are not 28 well-studied in POPs research.

29 In atmospheric chemistry, volatile components are classified on a continuum from Extremely low-volatility organic compounds (ELVOCs), low-volatility organic 30 31 compounds (LVOCs) to semi-volatile organic compounds (SVOCs), intermediate-32 volatility organic compounds (IVOCs), and volatile organic compounds (VOCs).¹ For POPs, their volatilities actually covers a wide range from ELVOCs to VOCs, currently 33 34 without finer operational boundaries being adopted by the scientific community. This vague boundary impedes a thorough comprehension of the atmospheric fate of POPs, 35 36 and their associated health risks. It also results in a disconnect between POPs and 37 secondary organic aerosols (SOA), to which the hydrophilic oxidized products of POPs could contribute. Therefore, we propose to incorporate HV-POPs into the realm of 38 39 atmospheric chemistry, with a specific subdivision to delineate their volatility.

40 In Figure 1, a chromatogram map shows chromatographic peaks of several POPs 41 overlaid with homologues of normal alkanes, based on real laboratory data. For 42 instance, the effective saturation concentration (C^*) within the IVOCs range is defined 43 as 300<C*<3×10⁶ µg/m³, with a saturation vapour pressure at 298 K ranging from 0.01-44 10 Pa, corresponding to the saturation vapour pressures of C₁₂-C₂₀ *n*-alkanes. The POPs that fall within the volatility range from IVOCs to VOCs are defined as HV-POPs. This 45 46 hexachlorobutadiene (HCBD), includes pentachlorobenzene (PeCB), and 47 hexachlorobenzene (HCB), along with emerging contaminants such as volatile 48 methylsiloxanes (VMS) and fluorotelomer alcohols (FTOHs). In addition to the 49 identified compounds in Figure 1, ultrashort-chain PFAS, such as trifluoroacetic acid 50 (TFA) and its ester derivatives, and very short-chain chlorinated paraffins (vSCCPs, C6-51 C9), could also be HV-POPs. It is noteworthy that many 'traditional' legacy POPs, such 52 as α -hexachlorocyclohexane (α -HCH) and 'light' congeners of polychlorinated 53 biphenyls (PCBs), polybrominated diphenyl ethers (PBDEs) and polychlorinated 54 naphthalenes (PCNs) are also found in the HV-POPs band. Compared to legacy POPs, 55 research on HV-POPs in the atmosphere is currently scarce and dispersed.

56 HV-POPs can be sampled from air using active and passive methodologies. The active 57 air samplers include the low- or high-volume samplers with filters and cartridges, and 58 the denuder samplers. The former is easy to deploy, while the denuder sampler is 59 more efficient at trapping the gas phase of HV-POPs, reducing sampling artifacts. The 60 passive air sampler (PAS) is easy to prepare and deploy, requiring no electricity, making it preferable for remote regions and surveys at multiple sites. The strong absorbents 61 62 such as XAD, Tenax resins and activated carbon are usually employed to capture VOCs. 63 The PAS using XAD sorbent impregnated polyurethane foam (PUF) sampler (SIP-PAS) has been adapted for sampling HV-POPs. However, it is reported that there is 64 breakthrough of HCBD from PUF plugs, which are widely used to capture SVOC-POPs, 65 leading to an underestimation of HCBD in Arctic air.² Specific quality control and quality 66 assurance (QA/QC) practices are needed for target HV-POPs throughout the laboratory 67 procedures to ensure reliable recoveries. Air sampling of HV-POPs is easily normalized 68 69 with established standard operating protocols, making monitoring data from various 70 laboratories more comparable.



71



Figure 1. The chromatogram map overlaid by typical persistent organic pollutants 73 (POPs) and C₆-C₃₀ *n*-alkanes using a DB-5MS GC column (60 m \times 0.25 mm \times 0.25 μ m).

74 Several HV-POPs have served as vital raw materials and solvents for the chemical industry, with widespread applications, large production and significant disposed 75 76 waste volumes. We give examples in China. First, chlorobenzenes (CBs) are used 77 extensively in pesticide and dye manufacturing, and in the electronics industry as raw materials and additives. More than half of the global production and consumption of
CBs is in China. Second, HCBD is used as a chemical solvent, heat transfer agent and
hydraulic system liquid, and can be unintentionally emitted from chlorinated solvent
production (e.g. chloromethane and perchloroethylene). Nearly a thousand tonnes of
HCBD emissions were estimated for 2016 in China. Third, methylsiloxanes, essential in
organic synthesis, are predominantly manufactured in China, with a global annual
production of 8-10 million tonnes.³

Because of their high volatility and large production and consumption volumes, HV-85 86 POPs can occur in the atmosphere at high levels relative to other POPs, and are likely 87 to transport to remote places such as the Arctic. Our recent analysis of HV-POPs in 88 urban air samples from Guangzhou, China, revealed concentrations of CBs, HCBD, FTOHs, and VMS in the ng- μ g/m³ range. This is similar to, or higher than the polycyclic 89 90 aromatic hydrocarbons (PAHs), a well-known group of air toxics. HCBD concentrations 91 in areas surrounding perchloroethylene production could range from μg to mg/m³, exceeding its occupational exposure limit of 210 µg/m³.⁴ Abundant evidence shows 92 that HCBD, HCB and α -HCH are widespread in the Arctic. 93

94 Their high volatility enables HV-POPs to reside in the atmosphere for a long time, 95 favoring their reactions with free radicals. This can lead to their mineralization and 96 transformation into hydrophilic oxidized products, which can be removed by 97 atmospheric deposition. For example, there is potential for HCB to be transformed to 98 more water-soluble pentachlorophenol (PCP); and for FTOH to PFOA. For VMSs at 99 higher concentrations in indoor air, the continuously produced silanol products may 100 participate in the formation of SOA. Modellers may treat HV-POPs as 'long-life' gases, 101 disregarding their multimedia exchanges. Observation-based inverse models can be applied to estimate their primary emissions, thereby supporting chemicals 102 103 management and policy making.

104 Inhalation may be the key pathway for HV-POPs to enter wildlife and human food 105 chains. Most HV-POPs exhibit low K_{OW} (<10⁵) and high K_{OA} (>10⁶), making them prone 106 to bioaccumulate in respiring animals, including humans.⁵ HCBD can form toxic 107 metabolites in the human body, and the U.S. Environmental Protection Agency 108 classifies it as a potential human carcinogen (Group C). PeCB, HCB, perfluorochemicals, 109 and their precursors are widely detected in human breast milk and serum, and capable 110 of accumulating in the human placenta.

111 Future research may benefit from integrating the HV-POPs concept into the regulatory

- 112 frameworks of the Stockholm Convention. Robust global collaboration is essential for
- standardized monitoring, observation-based inverse modeling, mitigation strategies in
- the technosphere, as well as direct and indirect atmospheric transport studies at
- 115 regional and global scales.

116 References

(1) Chang, X.; Zhao, B.; Zheng, H.; Wang, S.; Cai, S.; Guo, F.; Gui, P.; Huang, G.; Wu, D.;
Han, L.; Xing, J.; Man, H.; Hu, R.; Liang, C.; Xu, Q.; Qiu, X.; Ding, D.; Liu, K.; Han, R.;
Robinson, A. L.; Donahue, N. M. Full-volatility emission framework corrects missing
and underestimated secondary organic aerosol sources. *One Earth* **2022**, *5* (4), 403412.

122 (2) Balmer, J. E.; Hung, H.; Vorkamp, K.; Letcher, R. J.; Muir, D. C. G.
123 Hexachlorobutadiene (HCBD) contamination in the Arctic environment: A review.
124 *Emerging Contaminants* 2019, *5*, 116-122.

(3) Mojsiewicz-Pieńkowska, K.; Krenczkowska, D. Evolution of consciousness of
 exposure to siloxanes—review of publications. *Chemosphere* 2018, *191*, 204-217.

(4) Kong, Q.; Wang, Y.; Yang, X. A Review on Hexachloro-1,3-butadiene (HCBD): Sources,
Occurrence, Toxicity and Transformation. *Bulletin of Environmental Contamination and Toxicology* 2020, *104* (1), 1-7.

- 130 (5) Kelly, B. C.; Ikonomou, M. G.; Blair, J. D.; Morin, A. E.; Gobas, F. Food web-specific
- biomagnification of persistent organic pollutants. *Science* **2007**, *317* (5835), 236-239.

132