

# MXene incorporated nanofluids for energy conversion performance augmentation of a concentrated photovoltaic/thermal solar collector

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## Abstract

This research work introduces emerging two-dimensional MXene ( $\text{Ti}_3\text{C}_2$ ) and Therminol<sup>®</sup>55 oil-based mono and hybrid nanofluids for concentrated photovoltaic/thermal (CPV/T) solar systems. This study focuses on the experimental formulation, characterization of properties, and performance evaluation of the nanofluid-based CPV/T system. Thermo-physical (conductivity, viscosity, and rheology), optical (UV-vis and FT-IR), and stability (Zeta potential and TGA) properties of the formulated nanofluids are characterized at 0.025-0.125 wt.% concentrations of dispersed particles using experimental analysis. By suspending the nanomaterials photo-thermal energy conversion is improved considerably, up to 85.98%. The thermal conductivity of pure oil is increased by adding  $\text{Ti}_3\text{C}_2$  and CuO nanomaterials. The highest enhancements of up to 84.55 and 80.03% are observed for the TH-55/ $\text{Ti}_3\text{C}_2$  and TH-55/ $\text{Ti}_3\text{C}_2$ +CuO nanofluids, respectively. Furthermore, dynamic viscosity decreased dramatically over the temperature range investigated (25-105°C), and the nanofluid exhibited dominant Newtonian flow behavior as viscosity remained nearly constant up to a shear rate of  $100\text{s}^{-1}$ . Numerical simulations of the experimentally evaluated nanofluids are performed to evaluate the effect on a CPV/T collector using a three-dimensional transient model. The numerical analysis revealed significant improvements in thermal and electrical energy conversion performance, as well as cooling effects. At a concentrated solar irradiance of 5000

33  $\text{W/m}^2$  and an optimal flow rate of 3 liters/min, the highest thermal and electrical energy  
 34 conversion efficiency enhancements are found to be 12.8 and 2%, respectively.

35 **Keywords:** Nanofluid, Thermal-optical Properties, Stability, MXene, Therminol<sup>®</sup>55, CPV/T  
 36 Solar Collector.

### 12 Nomenclature

13 $A_c$	Area of collector ( $\text{m}^2$ )	<i>Subscripts</i>	
14 $c_p$	Specific heat ( $\text{J/kg.K}$ )	<i>amb</i>	Ambient
15 $FF$	Field factor	<i>bf</i>	Base fluid
16 $G$	Solar radiation intensity ( $\text{W/m}^2$ )	<i>el</i>	Electrical
17 $h$	Convective heat transfer coefficient ( $\text{W/m}^2.\text{K}$ )	<i>in</i>	Inlet
18 $I_{sc}$	Short circuit current (A)	<i>out</i>	Outlet
19 $V_{oc}$	Open circuit voltage (V)	<i>s</i>	Solid particle
20 $k_{bf}$	Thermal conductivity of base fluid ( $\text{W/m.K}$ )	<i>th</i>	Thermal
21 $k_{nf}$	Thermal conductivity of nanofluid ( $\text{W/m.K}$ )	<i>el</i>	Electrical
22 $k_s$	Thermal conductivity of nanoparticle ( $\text{W/m.K}$ )	<i>nf</i>	Nanofluids
23 $P_{el}$	Electrical power output (W)	<i>Abbreviations</i>	
24 $P_{th}$	Thermal output (W)	NF	Nanofluid
25 $T$	Temperature (K)	TH-55	Therminol <sup>®</sup> 55
26 $A$	Absorptivity	FTIR	Fourier-transform infrared spectroscopy
27 <i>Greeks</i>		SEM	Scanning electron microscopy
28 $\zeta$	Zeta potential, mV	FTIR	Fourier-transform infrared spectroscopy
29 $\sigma$	Stefan Boltzmann Constant $\text{W}/(\text{m}^2.\text{K}^4)$	TGA	Thermogravimetric analysis
30 $\rho$	Density, $\text{kg/m}^3$	UV-	Ultraviolet-visible spectroscopy
31 $\eta$	Efficiency	Vis	Visible spectroscopy
32 $\Phi$	Nanoparticle weight fraction	EVA	Ethylene Vinyl Acetate
		PV/T	Photovoltaic thermal
		SI	Solar irradiance
		LPM	Litter per minute

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### 38 1. Introduction

39 Inadequate renewable energy supplies have turned into a sophisticated obstacle to meeting the  
 40 world's ever-growing energy demand. The depletion of carbon-based fossil fuel supplies and

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3 41 their ongoing threat to the environment are compelling researchers to seek renewable energy  
4 42 sources that will be sufficient to meet potential energy needs. Given that energy demand will  
5 43 continue to rise in the future, one way to mitigate energy shortages is to increase reliance on  
6 44 sustainable renewable energy sources such as solar energy. It is one of the most convenient,  
7 45 pollutant-free, clean, and abundantly available energy resources [1]. As a result, numerous  
8 46 experimental and systematic studies are being conducted to develop renewable solar-based  
9 47 energy conversion systems for thermal and electrical applications. Solar collector technologies  
10 48 such as concentrated photovoltaic/thermal (CPV/T) allow us to capture and transform solar  
11 49 irradiation, resulting in increased thermal and electrical energy production. Additionally, the  
12 50 CPV/T solar system needs fewer PV cell surface areas due to the concentrated optical design,  
13 51 and construction and maintenance costs are low as well. The hybrid CPV/T device is a low-  
14 52 emission technology that is commonly used in industrial and residential applications [2]. A  
15 53 CPV/T is a hybrid device that combines a concentrated photovoltaic (CPV) panel with a  
16 54 thermal collector to provide a potential cooling effect. It is an advanced variant of a traditional  
17 55 CPV collector that does not include a cooling unit, which results in a rise in cell temperature at  
18 56 higher irradiation intensities and a decrease in device performance [3]. In CPV/T, the  
19 57 collector's effective concentrating device concentrates incoming solar irradiation onto the  
20 58 photovoltaic panel's surface. Solar irradiance then rapidly heats the photovoltaic cells by  
21 59 producing a high heat flux [4]. As concentrated solar radiation strikes the surface of the CPV/T  
22 60 system's solar cell integrated PV plate, approximately 40% of the incident solar energy is  
23 61 absorbed and converted to electrical energy by the PV unit, while the remaining heat can be  
24 62 transferred to the collector's working fluid in the thermal cooling unit [5]. In comparison to  
25 63 traditional CPV, non-concentrating hybrid PV/T, and standalone PV devices, hybrid CPV/T  
26 64 collectors produce more electrical power and thermal energy simultaneously [6]. However, the  
27 65 inefficiency of common heat transfer fluids (HTFs) in converting thermal energy to heat limits  
28 66 the cooling and overall efficiency of the PV/T system, thus limiting the output of CPV/T [7].  
29 67 Therefore, to augment the energy efficiency of solar collectors, such as PV/T and CPV/T,  
30 68 traditional HTF should be replaced with a working fluid that possesses efficient thermal energy  
31 69 conversion properties.

32 70 In this framework, due to their superior optical, thermal, and chemical properties, nanofluids  
33 71 (NFs) can be considered a possible substitute for conventional HTF in solar energy  
34 72 technologies [8]. Said, et al. [9] investigated the efficiency and cooling effects of traditional  
35 73 fluids and nanofluids in hybrid PV/T systems. After reviewing a large number of studies, they  
36 74 concluded that NFs could be utilized to significantly boost the cooling power and efficiency of

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3 75 the device. Bellos, et al. [10] reviewed the implementation of NFs on concentrating solar  
4 76 systems, including CPV/T, PV/T, compound parabolic collector (CPC), evacuated tube  
5 77 collector (ETC) and so on. They reported significant benefits from using NFs in the solar  
6 78 system rather than conventional HTFs. NFs/nano-colloids are highly engineered dispersions of  
7 79 solid nanomaterials with a nanoscale (<100 nm) in a liquid base fluid. Due to the advanced  
8 80 properties of dispersed nanoparticles, stable NFs exhibit exceptional photo-thermal properties  
9 81 in comparison to base fluids (BFs) [11]. In an experimental investigation, Qu, et al. [12]  
10 82 examined the photo-thermal energy transformation characteristics of aqueous/CuO+MWCNT  
11 83 NFs for solar energy harvesting applications. The NFs exhibited significant absorption  
12 84 properties and attained a 14.1°C higher temperature compared to the base fluid. Furthermore,  
13 85 NFs showed advanced optical behavior in response to incoming solar irradiation at low particle  
14 86 loading. Karami, et al. [13] examined the effects of CuO NFs based on ethylene glycol and  
15 87 water on a direct absorption solar collector (DASC) for domestic water heating at different NP  
16 88 volume fractions. They reported a growth in collector efficiency of up to 17% because of the  
17 89 NF's promising properties. Despite numerous studies establishing the specific properties of  
18 90 NFs on a theoretical and experimental level, the long-term stability of suspended NPs is the  
19 91 primary impediment to their commercial production in the industrial energy sector.  
20 92 Furthermore, several factors influence NF stability, including base fluids, NPs (concentration,  
21 93 size, and geometry), temperature, inter-molecular/chemical interactions, and so on [14, 15].  
22 94 Recent experimental/numerical studies on solar energy conversion technology have drawn  
23 95 attention to the potency of NFs in enhancing performance in a variety of concentrating [16]  
24 96 and non-concentrating solar systems by utilizing them as working fluids [17]. Nonetheless,  
25 97 only a few studies on the implementation of NF-based hybrid PV/T or CPV/T devices at  
26 98 concentrated irradiance have been conducted. Han, et al. [18] examined the performance  
27 99 augmentation of a hybrid PV/T collector employing the propylene glycol-based Ag/CoSO<sub>4</sub> NF  
28 100 optical filter. They observed that inclusion of the optical filter leads to excellent absorbance,  
29 101 transmittance, and an increment in overall photo-thermal conversion efficiency. Hemmat Esfe,  
30 102 et al. [19] evaluated the effectiveness of NF-based collectors by analyzing a wide range of  
31 103 literature that used pure fluids and NF as the working fluid in hybrid PV/T collectors. They  
32 104 discovered that NF-based systems outperform conventional fluids in terms of electrical  
33 105 efficiency (EE), PV panel cooling, and thermal efficiency (TE). In a Multiphysics simulation  
34 106 of the hybrid CPV/T collector using the full coupling method, Ju, et al. [20] showed that the  
35 107 Therminol®VP-1 based indium tin oxide (ITO) NF filter achieved significantly higher powers  
36 108 and efficiencies along with excellent absorption characteristics than that of the BF. Huaxu, et

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3 109 al. [21] carried out a cost-effective outdoor experimental investigation with an NF-based  
4 110 spectral splitting CPV/T system. The study discovered that the CPV/T system had 47%  
5 111 enhanced photo-thermal efficiency along with 3.8% higher energy conversion efficiency than  
6 112 a conventional CPV collector at a very cheap cost relative to Au, Ag and polypyrene NPs. An,  
7 113 et al. [22] performed an experimental analysis on energy efficiency of hybrid CPV/T collector  
8 114 using Oleylamine-Cu<sub>0</sub>S<sub>5</sub> NF. They reported 17.9% improved thermal efficiency relative to  
9 115 without NF filtration, along with significant EE of the hybrid CPV/T collector.

15 116 Due to their superior properties, nanomaterials dispersed in colloidal suspensions provide the  
16 117 remarkable development of unique thermal and optical properties of NFs. Different types of  
17 118 materials are extensively examined in the literature, including metal/metal oxides (Cu, Si, Ti,  
18 119 CuO, Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, TiO<sub>2</sub> and so on), carbon-based nanotubes (CNT) and two-dimensional (2D)  
19 120 graphene NPs [23]. Among a wide range of NPs, carbon-based 2D materials [24] and CNTs  
20 121 exhibit greater effectiveness because of their larger surface area and chemical structure relative  
21 122 to non-carbon-based particles [25]. Recent investigations with NF suspensions, due to the  
22 123 superior characteristics of emerging advanced 2D nanomaterials have attracted unprecedented  
23 124 attention. MXenes are the latest discovery in the innovative two-dimensional family, invented  
24 125 back in 2011 by Naguib, et al. [26]. MXenes exhibit remarkable mechanical strength [27] and  
25 126 an unique multilayered chemical structure [28] accompanied by distinct thermal [29], optical  
26 127 [30], photo-thermal [31], electrical [32] and magnetic properties [33] relative to conventional  
27 128 nanomaterials. MXenes are comprised of transition metal carbides, nitrides and carbonitrides.  
28 129 MXenes are produced from three-dimensional M<sub>n+1</sub>AX<sub>n</sub> (M stands for transition metals, A  
29 130 represents group-A element, X denotes carbon or/and nitrogen and n = 1, 2 or 3) phase  
30 131 employing selective etching method to eliminate 'A' layers [34]. Over 70 MAX phases have  
31 132 been identified so far, and a large portion of them are converted to MXenes through etching  
32 133 approaches. The MXenes are being studied extensively due to their remarkable properties, and  
33 134 a few of them are being used in applications such as, but not limited to, energy storage and  
34 135 conversion systems [35]. Pang, et al. [36] comprehensively reviewed advances with MXene-  
35 136 based studies on energy conversion and storage applications, including their mechanical,  
36 137 thermo-chemical, and optical properties.

53 138 In the present study, we experimentally formulated Therminol<sup>®</sup>55 oil-based mono and hybrid  
54 139 NFs dispersing synthesized 2D Ti<sub>3</sub>C<sub>2</sub> nanosheets and spherical CuO nanoparticles for the first  
55 140 time to date. Prepared NF samples are characterized to evaluate their thermo-physical, optical,  
56 141 and chemical properties. Thermal conductivity, specific heat capacity, viscosity, UV-Vis,  
57 142 TGA, FT-IR and stability analysis are carried out to assess the potency of the NF for hybrid

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3 143 CPV/T solar systems. The prepared NF's improved thermo-optical properties (i.e., thermal  
4 144 conductivity and solar absorbance) indicate its potential application in a hybrid CPV/T system.  
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6 145 The thermal conductivity of the NFs is augmented with the addition of nanomaterials and  
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8 146 elevated temperatures for both variants. In addition, the dynamic viscosity of the NFs decreased  
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10 147 radically at elevated temperatures, which makes the nanofluid particularly suitable for high  
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12 148 temperature applications like the CPV/T solar collector. In the numerical part of this work,  
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14 149 performance (thermal, electrical, and cooling) of the nanofluid-based hybrid CPV/T system is  
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16 150 reported as new findings and compared with the base fluid operated system. By adding  $Ti_3C_2$   
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18 151 and CuO nanomaterials, the highest thermo-optical, thermal, and electrical energy conversion  
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20 152 efficiencies are augmented by 85.98, 13 and 2.8%, respectively.

## 21 153 2. Methodology

### 22 154 2.1. Materials

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25 155 To experimentally formulate the NF for solar thermal application, medium temperature range  
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27 156 industry-standard synthetic Therminol<sup>®</sup>55 (TH-55) oil was purchased from EASTMAN and  
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29 157 used as pure base fluid. Emerging two-dimensional MXene nanoflakes are synthesized from  
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31 158 the three-dimensional MAX phase and used as NP to prepare the fluid. Specific characteristics  
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33 159 of TH-55 are provided in **Table 1**. CuO nanoparticles are used to prepare hybrid NF. The  
34  
35 160 particles purchased from US Research Nanomaterials. The specifications of synthesized  
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37 161 MXene and purchased CuO particles are depicted in **Table 2**.

### 38 162 2.2. Formulation of $TH-55/Ti_3C_2$ and $TH-55/Ti_3C_2+CuO$ nanofluids

39  
40 163 The synthesis of  $Ti_3C_2$  nanosheets from its three-dimensional MAX phase is performed using  
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42 164 the same method reported in our previous research [37].  $TH-55/Ti_3C_2$  NF samples are prepared  
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44 165 experimentally in two steps at three different particle loadings (0.025, 0.075, and 0.125 wt.%)  
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46 166 of synthesized  $Ti_3C_2$  nanoflakes. The two-step method is suitable in terms of commercial  
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48 167 aspects and industrial large-scale production of NF. The Schematic of MXene synthesis and  
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50 168 nanofluid is presented in **Figure S1** (See supplementary material). In the first step, the  
51  
52 169 estimated amount of synthesized two-dimensional  $Ti_3C_2$  nanosheets is dispersed into pure TH-  
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54 170 55 as weight fractions. To make the suspension homogenous, several mechanical stabilization  
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56 171 techniques are employed in the second step of the formulation process. Immediately after the  
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58 172 addition of solid particles into the oil, all the samples are stirred with a magnetic stirrer for  
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60 173 about 30 to 60 minutes at 700 to 900 rpm and 80°C. For further stabilization, NFs are sonicated  
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175 at a high frequency (1200 W, 20 kHz) utilizing an ultrasonic homogenizer (FS-1200N) for 30



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3 175 minutes. To ensure effective stabilization, the sonication is carried out at a temperature of  
4 176 around 80°C. This technique provides a uniform dispersion of MXene nanosheets in the base  
5 177 fluid by distributing the particles evenly throughout the suspension. Hybrid TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO  
6 178 nanofluids are formulated similarly using the same method and dispersing Ti<sub>3</sub>C<sub>2</sub>+CuO  
7 179 nanocomposite in powder form at a weight ratio of 1:1 into pure TH-55 oil at the same three  
8 180 particle loadings as solo NF. The same stabilization techniques are used after the addition of  
9 181 particles to produce stable dispersion of particles in hybrid NFs as described for solo NFs in  
10 182 the above.

### 183 **2.3. Morphology, optical and chemical structure characterization**

184 Nanofluid samples are characterized by employing several instruments for optical, chemical  
185 and morphological investigation. The surface morphology of Ti<sub>3</sub>C<sub>2</sub> flakes was inspected using  
186 TESCAN-VEGA3, a Scanning Electron Microscope (SEM). The equipment can provide high  
187 resolution SEM imaging for coated metal samples under a stable electron beam using a LaB<sub>6</sub>  
188 filament. Magnified SEM images of synthesized Ti<sub>3</sub>C<sub>2</sub> nanosheets illustrate the particles'  
189 distinct thin multi-layered structure in **Figure 1(a-b)**. Thin layered nanosheets of Ti<sub>3</sub>C<sub>2</sub> have  
190 been found to be more stable and transparent than two-dimensional graphene nanomaterials  
191 Zhou, et al. [38]. The amount of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> flakes synthesized is dependent on the etching  
192 conditions and etchant used Shen, et al. [39]. Additionally, HF-etched Ti<sub>3</sub>C<sub>2</sub> exhibits a  
193 characteristic accordion-like morphology with a less compact structure between the layers than  
194 other etching methods [40]. The obtained results, which include Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> flakes from SEM  
195 images, demonstrate the efficient conversion of two-dimensional Ti<sub>3</sub>C<sub>2</sub> from three-dimensional  
196 MAX-phase (Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>). The spherical structure of the CuO nanoparticles is also identified in  
197 **Figure 1(c-d)**. FESEM and HRTEM analysis are used to further investigate the morphology  
198 of the Ti<sub>3</sub>C<sub>2</sub> nanostructure [41]. The FESEM images (**Figure S2, a-b**) confirms the thin multi-  
199 layered flake-like structure of the Ti<sub>3</sub>C<sub>2</sub>. In addition, high-resolution HRTEM images (**Figure**  
200 **S2, c-d**) represent a typical Ti<sub>3</sub>C<sub>2</sub> nanosheet which is incredibly thin and flexible. A hexagonal  
201 based crystal is exposed with the analogous atomic configuration relative to original MAX-  
202 phase (Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>) in the grain boundaries [42] (as depicted in **Figure S2, d**). Hence, HRTEM  
203 and FESEM images provide more compelling evidence for the material's transformation from  
204 three-dimension to two-dimension.

205 Fourier transform infrared spectroscopy (FT-IR) is conducted to identify the existing chemical  
206 functional groups and observe the chemical composition in the suspension of Therminol®55  
207 with Ti<sub>3</sub>C<sub>2</sub> and CuO nanoparticles. The experiment was performed using a high-performance

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3 208 Spectrum-Two™ FT-IR spectrometer from Perkin Elmer. The instrument can measure at the  
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5 209 highest resolution of  $0.5\text{ cm}^{-1}$  for spectral range of  $350\text{ to }8300\text{ cm}^{-1}$  using single channel  
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7 210  $\text{LiTaO}_3$  detector. Transmittance and chemical compositions of TH-55/ $\text{Ti}_3\text{C}_2$  and TH-  
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9 211 55/ $\text{Ti}_3\text{C}_2$ +CuO NFs are examined utilizing the FT-IR for the wavenumber range of  $450\text{ to }4000$   
10  
11 212  $\text{cm}^{-1}$  at  $0.2\text{ cm/s}$  scanning speed. Lambda 750 (Parkin Elmer) is employed to accomplish  
12  
13 213 absorbance characteristic of the formulated NFs using a monochromatic light source of  $860$   
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15 214 nm. All samples are measured at room temperature for  $200\text{ to }800\text{ nm}$  wavelengths at a  
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17 215 scanning speed of  $266.75\text{ nm/min}$ . Suspension stability of the NFs is examined by using an  
18  
19 216 electrophoresis technique to measure the electrophoretic mobility of dispersed particles in  
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21 217 terms of Zeta potential. Anton Paar's LITESIZER 500 is operated to conduct the assessment  
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23 218 using dynamic, electrophoretic, and static light scattering techniques. It offers good  
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25 219 measurement accuracy for samples with a size range of  $3.8\text{ nm to }100\text{ }\mu\text{m}$ . All the equipment  
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27 220 is calibrated before assessing the NF samples, and protocols were repeated several times to  
28  
29 221 ensure consistent precision of the measurements.

#### 222 **2.4. Characterization of thermal conductivity and viscosity**

30  
31 223 To obtain the thermal conductivity (TC) of TH-55/ $\text{Ti}_3\text{C}_2$  and TH-55/ $\text{Ti}_3\text{C}_2$ +CuO NFs,  
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33 224 TEMPOS (form Meter Group USA) is operated from  $30\text{ to }70\text{ }^\circ\text{C}$  temperature. This equipment  
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35 225 functions under the transient hot wire method and is supplied with several sensors and a digital  
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37 226 controller with a DB-15 connector. In the present experiment with the NFs, the KS-3 sensor  
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39 227 ( $1.3\text{ mm diameter} \times 60\text{ mm length}$ ) is utilized as it is particularly suitable to assess TC in the  
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41 228 range of  $0.02\text{ to }2\text{ W/m.K}$ . TEMPOS can keep the heat constant from the source and offers  
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43 229 excellent accuracy in the measurements with an uncertainty of  $\leq \pm 10\%$ . A temperature stable  
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45 230 water-bath (MEMMERT, WNB22) is integrated during the experiment, and temperature can  
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47 231 be controlled using a knob to measure conductivities at each temperature. Before measuring  
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49 232 the conductivities of the NF samples, the equipment is calibrated with glycerin samples  
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51 233 provided by the suppliers. In this approach, a certain temperature is set to the water-bath and  
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53 234 the sample are placed into the water-bath using jigs. The KS-3 needle sensor is placed inside  
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55 235 the sample vertically and connected with the analyzer via a USB cable.

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57 236 The viscosity characteristics of pure Therminol®55 oil and the solo and hybrid NFs are studied  
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59 237 using the MCR-92 Rheometer from Anton Paar. The measurements are conducted with respect  
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238 to a temperature scale of  $20\text{ to }105\text{ }^\circ\text{C}$  and shear-rate from  $30\text{ to }100\text{ s}^{-1}$ . The instrument offers  
239 very good accuracy (precision:  $\pm 1\%$ ) for a wide range of viscosity from  $-40\text{ to }400\text{ }^\circ\text{C}$ . The



device is calibrated with water before performing the test with oil samples. 60 ml of each sample is required to measure viscosity with MCR-92.

#### 2.4.1. Specific heat capacity

Due to measurement facility limitations, the  $c_p$  of the TH-55/Ti<sub>3</sub>C<sub>2</sub> and TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO NFs is measured using the correlations widely used in reported studies [43] as follows:

$$(\rho c_p)_{nf} = \varphi(\rho c_p)_{np} + (1 - \varphi)(\rho c_p)_{bf} \quad (1)$$

Where,  $(c_p)_{np}$ ,  $(c_p)_{bf}$  and  $(c_p)_{nf}$  are the specific heat capacity (in J/kg. K) of nanomaterials, base fluid and formulated nanofluid, respectively.

The  $c_p$  of Ti<sub>3</sub>C<sub>2</sub> nanomaterial is experimentally determined and correlated as follows [41]:

$$(c_p)_{np} = -0.001T_{nf}^2 + 3.4T_{nf} + 604.5 \quad (2)$$

Where,  $T_{nf}$  is evaluated in °C.

Since the concentrations of the Ti<sub>3</sub>C<sub>2</sub> nanoparticles is very low, the density of the TH-55/Ti<sub>3</sub>C<sub>2</sub> is presumed to be constant and equal to the density of pure Therminol®55. Indeed, the effect of nanoparticle loading on the nanofluid's density can be significant at high loading values. Without considering the effect of density-shift due to the nanoparticle loadings, the specific heat of the TH-55/Ti<sub>3</sub>C<sub>2</sub> nanofluid is measured as follows.

$$(c_p)_{nf} \cong (c_p)_{np}(1 - \varphi)(c_p)_{bf} \quad (3)$$

Similar approach is followed to calculate the  $c_p$  of the hybrid NF.

The density ( $\rho_{nf}$ ) of the nanofluid is considered to be constant, and the properties were obtained from Eq. 4 [44]:

$$\rho_{nf} = (1 - \phi)\rho_{bf} + \phi\rho_s \quad (4)$$

#### 2.5. Application of TH-55/Ti<sub>3</sub>C<sub>2</sub> and TH-55/Ti<sub>3</sub>C<sub>2</sub>-CuO nanofluids on CPV/T solar system

In a concentrated photovoltaic/thermal (CPV/T) system, Fresnel lenses are used to concentrate the sunlight on each of the solar cells of a poly-crystalline-silicon PV module. Researchers used a variety of active and passive techniques to reduce the temperature rise of photovoltaic module cells. Some of them used nanofluid as a coolant at the back of the photovoltaic panel to reduce the heat and thus boost electrical and thermal efficiency. This section conducts a numerical analysis of the newly produced TH-55/MXene-based nanofluid and its hybrid variant TH-55/MXene+CuO in the CPV/T system. The proposed work utilizes a CPV/T system to compare the performance of mono and hybrid Therminol®55-based nanofluids to that of base fluid alone, numerically. The problem under investigation is presented in **Figure 2**. This investigation considers a large photovoltaic module with 72 polycrystalline silicon cells (each cell has an area of 0.024 m<sup>2</sup>). In India, the average solar radiation is 1000 W/m<sup>2</sup>. To achieve

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3 269 solar irradiation of up to  $5000 \text{ W/m}^2$ , each lens must have a surface area of  $5 \times 0.024 \text{ m}^2$ , taking  
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5 270 the total surface area of the 72 lenses to  $72 \times 0.122 \text{ m}^2$ . The computational domain for numerical  
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7 271 simulation is the total area of the solar cells ( $1.73 \text{ m}^2$ ). **Table 3** summarizes the physical  
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9 272 properties of the PVT module's various layers.

### 10 273 *2.5.1 Numerical modelling of CPV/T solar system*

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13 274 The solar collector under investigation is a 300-watt photovoltaic module made up of four  
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15 275 layers: a photovoltaic solar cell, EVA (Encapsulated vinyl acetate) on both sides of the  
16  
17 276 photovoltaic cell, and a tedlar plate. A serpentine copper tubing heat exchanger is placed  
18  
19 277 underneath the photovoltaic module (**Figure 2**). The PV cells, EVA, and tedlar layers are each  
20  
21 278  $0.3\text{mm}$ ,  $0.5\text{mm}$ , and  $0.1\text{mm}$  thick. The remaining specifications are identical to those of the  
22  
23 279 photovoltaic plate, i.e. ( $1955\text{mm} \times 982\text{mm}$ ). The three-dimensional (3D) numerical analysis is  
24  
25 280 carried out using the Finite Element Method-based Multiphysics Software COMSOL.  
26  
27 281 COMSOL's CFD and heat transfer modules are used to determine the CPV/T system's output  
28  
29 282 parameters. In this study, 3D transient model is preferred because the flow pattern is 3D in real  
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31 283 cases. Moreover, the 3D simulations show the interaction of individual components with their  
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33 284 surroundings. The nanofluid flow is presumed to be constant, three-dimensional,  
34  
35 285 incompressible, and laminar. The transmissivity of EVA is assumed to be approximately 100%,  
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37 286 dust's effect on the absorptivity of the PV surface is assumed to be negligible, and temperature  
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39 287 variation along the module's thickness is assumed to be zero. Additionally, it is assumed that  
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41 288 the base fluid contains a homogeneous mixture of nanoparticles (i.e., no particle  
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43 289 sedimentation). In this study, TH-55/ $\text{Ti}_3\text{C}_2$  and hybrid variant TH-55/ $\text{Ti}_3\text{C}_2$ +CuO based  
44  
45 290 nanofluids with varying nanoparticle concentration are used. Thermal conductivity, which is  
46  
47 291 proportional to the weight fraction at various temperatures, is fitted to a third order polynomial  
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49 292 using regression analysis and then incorporated into the COMSOL through a user defined  
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51 293 function (UDF).  
52  
53 294 Regression analysis of experimental data is used to model temperature dependence on the  
54  
55 295 viscosity of TH-55/ $\text{Ti}_3\text{C}_2$  and hybrid variant of TH-55/ $\text{Ti}_3\text{C}_2$ +CuO nanofluids. Like thermal  
56  
57 296 conductivity, the viscosity correlations are integrated into COMSOL's CFD environment  
58  
59 297 through the UDF and used for simulation purposes. For solid domain in the PV/T device, heat  
60  
298 conduction equations are used to account for heat transfer. Heat transmission from the surface  
299 of the photovoltaic panel to the flow channel is established using the heat conduction equation  
300 shown below in Eq.5 (a-d) [51].

$$\rho_c \delta_c C_{pc} \frac{dT_c}{dt} = \alpha_p G - E_e - h_{panel-ted}(T_{penal} - T_{ted}) + k_c \delta_c \left( \frac{\partial^2 T_c}{\partial x^2} + \frac{\partial^2 T_c}{\partial y^2} + \frac{\partial^2 T_c}{\partial z^2} \right) \quad (5-a)$$

301 It reflects the heat transfer between the photovoltaic panel and the tedlar plate. Other thermal  
 302 energy equations for additional layers can be written similarly. Here,  $\alpha_p$  represents the panel's  
 303 absorptivity,  $G$  is the irradiance,  $E_e$  is the electrical energy output and  $h_{panel-ted}$  is the heat  
 304 transfer coefficient between PV module and tedlar plate. Correspondingly, other heat transfer  
 305 coefficients between the layers are specified in Eq. 5-b, 5-c and 5-d. Specifications of the  
 306 CPV/T collector are listed in **Table 3**.

307 From tedlar to serpentine tubing:

$$\rho_{td} \delta_{td} C_{ptd} \frac{dT_{td}}{dt} = -h_{penal-ted}(T_p - T_{td}) - h_{ted-tubing}(T_{ted} - T_{tubing}) + k_{td} \delta_{td} \left( \frac{\partial^2 T_{td}}{\partial x^2} + \frac{\partial^2 T_{td}}{\partial y^2} + \frac{\partial^2 T_{td}}{\partial z^2} \right) \quad (5-b)$$

308 From serpentine tubing to nanofluid:

$$\rho_{tb} \delta_{tb} P dy C_{ptb} \frac{dT_{tb}}{dt} = -h_{ted-tubing}(T_{ted} - T_{tubing}) - h_{tubing-nf} P dy (T_{tubing} - T_{nf}) + k_{tb} \delta_{tb} \left( \frac{\partial^2 T_{tb}}{\partial x^2} + \frac{\partial^2 T_{tb}}{\partial y^2} + \frac{\partial^2 T_{tb}}{\partial z^2} \right) \quad (5-c)$$

309 Where  $P$  is the periphery of the tube. For working fluid in serpentine channel:  
 310

$$\rho_f A_f dy C_f \frac{dT_f}{dt} = h_{tubing-nf} P dy (T_{tubing} - T_{nf}) \quad (5-d)$$

311 Moreover, Eqs.6-8 describes the mass and momentum and energy equations for transient  
 312 laminar fluid flow. For flow in the collector, the coupled heat transfer equation is used, as both  
 313 conduction and convection are considered for the nanofluids in Eq.6-8.

314 Continuity:

$$\frac{\partial \rho_f}{\partial t} + \nabla \cdot (\rho_f V_f) = 0 \quad (6)$$

315 Momentum:

$$\rho_f \left( \frac{\partial V_f}{\partial t} + (V_f \cdot \nabla) V_f \right) = -\nabla P + \nabla \cdot (\mu_f \nabla V_f) \quad (7)$$

316 Energy:

$$\rho_f C_{Pf} \frac{\partial T_f}{\partial t} + \rho_f C_{Pf} V_f \cdot \nabla T_f = \nabla \cdot (k_f \nabla T_f) \quad (8)$$

317 The Nusselt number (Nu) for different flow regime between fluid and tube can be expressed as  
 318 Eq.9 [45]:

$$\text{Re} < 2300, \text{Nu} = 4.364 \text{ and } \text{Re} > 2300, \text{Nu} = 0.0234 \text{Re}^{0.8} \text{Pr}^{0.4} \quad (9)$$

319 In equation 12, Reynolds number (Re) and Prandtl Number (Pr) can be calculated using Eq.10  
320 and 11 [45]:

$$Pr = \mu_f C_{pf} / k_f \quad (10)$$

$$Re = \rho_f v_f^D / \mu_f \quad (11)$$

321 The electrical modelling of the PV cell is developed using MATLAB/Simulink using the model  
322 of single PV cell.

323 The output electrical output and thermal energy are calculated by equation 12 & 13:

$$P_{el} = V_{oc} * I_{sc} * FF \quad (12)$$

$$P_{th} = \dot{m} C_p (T_{out} - T_{in}) \quad (13)$$

324 here,  $\dot{m}$  is the mass flow rate of the nanofluid.

325 Eq. 14 and 15 determine the electrical and thermal efficiency, respectively.

$$\eta_{el} = \frac{P_{el}}{G * A_c} \quad (14)$$

$$\eta_{th} = \frac{P_{th}}{G * A_c} \quad (15)$$

### 326 2.5.2. Boundary Conditions

327 Proper boundary conditions were used throughout the domain in accordance with the physics  
328 of the problem. The boundary condition that is applied across the top and bottom layers of the  
329 photovoltaic module is represented by Eq. 16:

$$-n \cdot q = h_c (T_{amb} - T_s) \quad (16)$$

330 Where  $\mathbf{n}$  is the surface normal and  $T_{amb}$  and  $T_s$  are the surrounding environment and surface  
331 temperatures, correspondingly. For the fluid domain, the inlet boundary condition is specified  
332 as velocity of the inlet along x-axis i.e,  $u=U_o$ ,  $v=0$ ,  $w=0$  and  $T=T_o$ , for solid boundaries no-slip  
333 condition is used ( $u=v=w=0$ ), however, at the outlet, zero pressure boundary condition is used  
334 ( $p=0$ ). For solid-fluid interface heat flux continuity at the interface is used as equation 17:

$$\left( \frac{\partial T_s}{\partial n} \right)_f = \frac{k_s}{k_f} \left( \frac{\partial T_s}{\partial n} \right)_s \quad (17)$$

335 Adiabatic boundary condition is applied for the solid walls of the device. Furthermore, the  
336 bottommost plate of the solar PV/T collector remains isolated (see **Figure S3**).

### 337 2.5.3. Meshing and Grid Independency

338 The PVT module was meshed in COMSOL Multiphysics® using the built-in physics-controlled  
339 mesh sequence setting shown in **Figure 3(a-c)**. It consists of tetrahedral and triangular mesh  
340 elements at the sub-domain and at the boundary, respectively. The number of mesh elements

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3 341 increases at each boundary so that the heat transfer and flow fields can be resolved accurately.  
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5 342 For grid independency simulation at  $1000 \text{ W/m}^2$  and a mass flow rate of 3LPM is performed  
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7 343 using water as a coolant with different mesh size (from coarser to finer) shown in **Table 4**. The  
8  
9 344 initial layer thickness is set to 1/50 of the element's size at that boundary. The meshing along  
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11 345 the thickness of the collector and diameter of the tube is presented in **Figure 3(b-c)** [46]. It was  
12  
13 346 observed that there was no further change in panel temperature and outlet fluid temperature  
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15 347 values after mesh no.5. Thus, mesh no.5 is selected for simulation purpose. **The solution**  
16  
17 348 **method of the governing equations using COMSOL Multiphysics modelling package is shown**  
18  
19 349 **in FigureS4.**

#### 20 350 **2.5.4. Validation**

21 351 The results of the solar cell temperature at an irradiation intensity of  $1000 \text{ W/m}^2$  and several  
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23 352 flow rates (0.5 to 4 LPM) were obtained from the present numerical model, validated with  
24  
25 353 Nasrin, et al. [47]. **Table 5** expresses this validation and provides a very good accord with the  
26  
27 354 experimental findings.

### 28 355 **3. Results and discussion**

#### 29 356 **3.1. FT-IR analysis**

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31 357 **Figure 4** represents FT-IR spectra of TH-55 and TH-55/ $\text{Ti}_3\text{C}_2$  NF at 0.025-0.125 wt.% particle  
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33 358 loadings of  $\text{Ti}_3\text{C}_2$  to assess existing chemical bonds in the fluids. The study is performed for  
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35 359 each sample from  $450$  to  $4000 \text{ cm}^{-1}$  wavenumber at room temperature. Both solo and hybrid  
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37 360 NFs depicted similar profile of the spectrums with no major deviation in the absorption peaks  
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39 361 as particles are suspended in identical base fluid for both types of nanofluid. The  
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41 362 indistinguishable spectrums are observed with both oil and the NF samples which convey  
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43 363 indication of no chemical reaction between the fluids and particles except physical interactions.  
44  
45 364 Samylingam, et al. [46] have attained similar outcomes by investing palm oil as base fluid as  
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47 365 well as adding solid particles at different concentrations. They reported a chemically unique  
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49 366 FT-IR spectrum for each of the individual fluids. Therefore, the present optical results with  
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51 367 TH-55/ $\text{Ti}_3\text{C}_2$  NFs confirm that the fluids are consistent in terms of chemical interactions.

52 368 Several absorption peaks are noticed from the mixture of TH-55 oil and nanomaterial in the  
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54 369 spectra absorbing IR bands. Key absorption peaks were observed in the spectrums at  
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56 370 wavenumbers of  $2923 \text{ cm}^{-1}$ ,  $2856 \text{ cm}^{-1}$ ,  $1459 \text{ cm}^{-1}$ ,  $1376 \text{ cm}^{-1}$ ,  $825 \text{ cm}^{-1}$  and  $701 \text{ cm}^{-1}$ . The  
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58 371 corresponding peaks at  $2923 \text{ cm}^{-1}$  and  $1459 \text{ cm}^{-1}$  revealed the presence of  $\text{CH}_3$  and  $\text{CH}_2$   
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60 372 stretching bond in the synthetic oil chain of TH-55 respectively [46]. The bands at  $1459 \text{ cm}^{-1}$



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3 373 and 1376  $\text{cm}^{-1}$  are identified as C=C and C=O stretches of the carbon skeleton due to  
4 374 dominance of the oil in the mixture [48]. The peaks at low-frequency of 701  $\text{cm}^{-1}$  is attributed  
5 375 to C-H bending vibration of the suspension [46]. No discrepancy is observed in terms of  
6 376 deviation of absorption peaks changes in FT-IR spectrum profiles among the oil and NFs. The  
7 377 spectrums of samples with different concentrations of  $\text{Ti}_3\text{C}_2$  flakes are undistinguishable due  
8 378 to excellent chemical stability and consistency of the dispersion. The conclusive results imply  
9 379 that the formulated NFs are chemically stable.

### 16 380 3.2. UV-vis absorbance assessment

18 381  $\text{Ti}_3\text{C}_2$  and CuO nanoparticles are suspended into TH-55 to advance the absorption aptitude of  
19 382 the fluids so that it can improve the efficiency and cooling capacity of the hybrid CPV/T  
20 383 system. The optical absorption characteristics of formulated TH-55/ $\text{Ti}_3\text{C}_2$  and TH-  
21 384 55/ $\text{Ti}_3\text{C}_2$ +CuO NFs are examined by analyzing UV-vis spectrums presented in **Figure S5**. The  
22 385 analysis was performed with TH-55 and several concentrations (0.025, 0.075 and 0.125 wt.%)  
23 386 of  $\text{Ti}_3\text{C}_2$  and hybrid  $\text{Ti}_3\text{C}_2$ +CuO particles for a wavelength range of 200 to 800 nm as it contains  
24 387 over 80% of total solar radiation emitted from the sun. According to the law of Beer-Lambert  
25 388 ( $A = \log_{10} \frac{I_0}{I} \alpha C$ , where  $I$  and  $I_0$  is the intensity of the incident and transmitted light,  
26 389 respectively), the absorption property of fluid will be amplified due to addition of particles into  
27 390 it and absorbance should be improved with increasing weight loadings of particles dispersed  
28 391 into the fluid [49]. **Figure S5** exhibits absorption of TH-55/ $\text{Ti}_3\text{C}_2$  and TH-55/ $\text{Ti}_3\text{C}_2$ +CuO NFs  
29 392 intensified notably at different peaks and in visible wavelength scale relative to pure TH-55.  
30 393 Absorbance enhanced rapidly at 200-250 nm and observed high light absorbance up to 450 nm  
31 394 wavelength. At higher wavelength (450-800 nm), absorbance remained steady for all the fluids.  
32 395 Nevertheless, absorbance of NF samples remained greater than Therminol<sup>®</sup>55 oil. Gulzar, et  
33 396 al. [50] observed analogous trend investigating several Therminol<sup>®</sup>55 based NFs.  
34 397 The obtained average increments in absorbance for TH-55/ $\text{Ti}_3\text{C}_2$  NF at 0.025, 0.075 and 0.125  
35 398 wt.% are 37.23, 61.72 and 80.16% respectively. For the hybrid NFs, the percentage  
36 399 augmentations are similar, being 32.75, 66.55 and 85.98% at the same weight loadings as solo  
37 400 NFs. The intermolecular homogeneousness among TH-55,  $\text{Ti}_3\text{C}_2$  and CuO as well as the two  
38 401 dimensional structure of  $\text{Ti}_3\text{C}_2$  nanoflakes with larger surface area lead to advanced light-to-  
39 402 heat (photo-thermal) energy conversion of the NFs [46]. The results from the UV-vis analysis  
40 403 suggest superior absorption capability of the formulated NFs and undoubtedly validates their  
41 404 potential implementation on direct absorption solar collectors, for instance, hybrid CPV/T solar  
42 405 system. MXenes are reported to have better absorption capability compared to conventional

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3 406 carbon nanotube based materials due to its distinct layered structure. Li, et al. [51] reported  
4 407 enhanced photo-thermal energy conversion efficiency of layered  $Ti_3C_2$  nanomaterial. Wang, et  
5 408 al. [31] observed that MXene based NF exhibits 4.34% higher photo-thermal energy  
6 409 conversion efficiency relative to Graphene at 20 ppm.

### 10 410 **3.3. Stability of TH-55 with nanoparticles**

11 411 The homogeneous of the suspension or the nanofluid's long-term stability is a critical  
12 412 precondition for industrial applications. In general, the instability of nanofluid is caused by  
13 413 several attraction and repulsion forces acting at the solid-fluid interface of the suspension.  
14 414 When attraction forces dominate, the nanoparticles tend to form clusters and degrade the  
15 415 homogeneity and thermo-optical properties of the suspension. Zeta potential ( $\zeta$ ) analysis is an  
16 416 effective way to determine the dispersion stability of nanomaterials in a fluid medium. In this  
17 417 method, the electrical potential difference (i.e. Zeta potential,  $\zeta$ ) is measured in mV at the  
18 418 electric double layer (EDL) formed at the solid-fluid interface of the suspension [52]. Higher  
19 419 values of  $\zeta$  implies higher mobility i.e., dispersion stability of the particles in the fluid medium.  
20 420 The  $\zeta$  value beyond  $\pm 60$  mV represents excellent stability of suspension whereas, values below  
21 421  $\pm 15$  mV indicates unstable or poor dispersion of nanoparticles in the mixture [53].

22 422 The suspension stability of prepared NFs is characterized by varying particle wt.% (0-0.125)  
23 423 and temperature (25 and 80°C). The obtained  $\zeta$  results (depicted in **Figure 5**) suggest good  
24 424 stability of the formulated fluids, particularly for mono TH-55/ $Ti_3C_2$  NF at 0.025wt.%. It is  
25 425 also clear that the addition of more particles (0.075 and 0.125wt.%) deteriorates dispersion  
26 426 homogeneity as  $\zeta$  drops due to formation of clusters in the suspensions. Nevertheless, NF  
27 427 samples with single NPs are found to be more stable than hybrid ones. On the 4<sup>th</sup> day of  
28 428 preparation, absolute  $\zeta$  value of 79.39, 64.83, 44.17 mV and 54.36, 41.74, 27.29 mV are  
29 429 measured for TH-55/ $Ti_3C_2$  and TH-55/ $Ti_3C_2$ +CuO NFs, respectively. In terms of temperature,  
30 430 NFs are observed to be more stable at high temperatures than the lower ones. At 80°C, both  
31 431 mono and hybrid NFs exhibited absolute  $\zeta$  value above 30 mV, while comparatively lower  $\zeta$  is  
32 432 noted at 25°C for all three wt.%. This is due to particle addition movement (i.e., Brownian  
33 433 motion), which results in a decrease in Van der Waals attraction potential at higher  
34 434 temperatures. Other studies examining oil-based NFs confirm the associated effects of particle  
35 435 loading and temperature on stability [37, 48].

### 3.4. Thermal conductivity of TH-55 with nanoparticles

**Figure 6** depicts experimental results on the thermal conductivity (TC) variation of TH-55/Ti<sub>3</sub>C<sub>2</sub> and hybrid TH-55/Ti<sub>3</sub>C<sub>2</sub>-CuO NFs at 0.025-0.125 wt.% nanoparticle concentration against increasing temperature (30 to 70 °C). The TC values of the NFs are measured experimentally within a standard deviation of less than ±0.0025. The addition of Ti<sub>3</sub>C<sub>2</sub> and CuO particles into the base fluid provides a significant improvement in the TC of the fluids. Furthermore, as temperature was raised, TC of the NFs was observed to improve, whereas TC of the TH-55 oil alone deteriorated. **Figure 6(a-b)** depicts the percentage linear increment of TC for formulated NFs with the inclusion of nanoparticles as well at increasing temperatures. The enhancement is calculated using the equation  $\left(\frac{k_{nf} - k_{bf}}{k_{bf}}\right) \times 100$  where,  $k_{nf}$  is TC of NF and  $k_{bf}$  is TC of base fluid TH-55. At 70°C, the maximum augmentations in TC obtained by suspending the particles are 84.55% and 80.03% for TH-55/Ti<sub>3</sub>C<sub>2</sub> and hybrid TH-55/Ti<sub>3</sub>C<sub>2</sub>-CuO NFs, respectively. The remarkable improvement of TC for all the NFs is due to the distinct thin two-dimensional layered structure of Ti<sub>3</sub>C<sub>2</sub> flakes, their high surface area to volume ratio, stable dispersion, and movement of the suspended particles into the TH-55 [54]. In the case of hybrid NF, TC increases are attributed to the incorporation of highly heat conductive Ti<sub>3</sub>C<sub>2</sub> and CuO nanomaterials into pure oil, as well as intermolecular movements among the particles and their large conductive surface area [55]. Added particle wt.% is observed to have a noteworthy impact on TC as higher wt.% yields more conductivity of the NFs. However, the addition of nanomaterial showed better TC augmentation at lower wt.% with hybrid NF, being 18.16% and 25.88% at 0.025 wt.% and 30°C for sole and hybrid NFs, respectively. Thus, it is evident that the dispersed solid concentration of the particles offers an effective heat transport aptitude in comparison with pure base fluid.

TC and heat transfer characteristics of NF are a function of temperature variation of the system. The phenomenon of TC improvement with increasing temperature is associated with particle Brownian motion and kinetic energy, which increases heat transport through dispersed nanomaterials [56, 57]. As Brownian motion intensifies at higher temperatures, it enables heat to be transported more efficiently from one particle to another utilizing large surface area of Ti<sub>3</sub>C<sub>2</sub> nanoflakes and spherical CuO particles. Furthermore, tinny Ti<sub>3</sub>C<sub>2</sub> flakes with negligible thickness (<10 nm) are able to randomly move with high energy within the oil medium [58]. For TH-55 oil with the highest weight concentration (0.125 wt.%) of Ti<sub>3</sub>C<sub>2</sub> flakes, TC is intensified by 4-8% for each 10°C rise in NF temperature. TC of the NFs intensified more at higher temperatures due to effective Brownian movement at elevated temperatures. A similar

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3 469 trend is noticed for hybrid loading of  $Ti_3C_2$  and CuO nanoparticles in thermal oil except at  
4 470 0.125 wt.% conductivity values of hybrid NF are lower relative to that of solo NF samples. The  
5 471 outcomes can be ascribed to extra-large surface area of  $Ti_3C_2$  particles, aggregation of particles  
6 472 at higher loadings and different size and shape of two types of particles dispersed together into  
7 473 the oil. Similar findings are reported by Yu, et al. [59] analyzing NF formulated from ethylene  
8 474 glycol and two-dimensional graphene nanosheets. It is important to note that the inherent TC  
9 475 of  $Ti_3C_2$  is much lower than graphene. However, due to stable dispersion and compatibility  
10 476 with solvents, it can offer potential TC in suspension. As intensified TC of working fluid can  
11 477 produce enhanced thermal performance of the system, the formulated NFs can be potential  
12 478 HTFs to utilize in thermal systems like CPV/T systems.

13 479 From the obtained results, it is evident to state that the impact of wt.% is dominant over the  
14 480 effect of increasing temperatures on TC of the prepared NFs. Several previous studies with oil-  
15 481 based NF and  $Ti_3C_2$  particles have yielded results that are consistent with those obtained in this  
16 482 experiment. [60] investigated colloidal conductivity of ethylene glycol (EG) based multi-  
17 483 layered and single layered  $Ti_3C_2$  nanofluid. They reported that adding  $Ti_3C_2$  sheets produced  
18 484 the highest 64.9% augmented TC relative to EG. Samyilingam, et al. [46] experimentally  
19 485 estimated TC of palm oil-based NFs adding 0.01 to 0.2 wt.%  $Ti_3C_2$  nanomaterials for  
20 486 temperature range of 25-70 °C. Their results revealed that the NF obtained a 68.5% increment  
21 487 in TC compared to palm oil.

### 22 488 **3.5. Dynamic viscosity of TH-55 with nanoparticles**

23 489 Dynamic viscosity ( $\mu$ ) is one of the key properties of NF, which is defined by the resistance  
24 490 force that causes fluid deformation in the reverse path of flow. It has a substantial effect on  
25 491 heat transfer by convection in fluids during the application of fluids on thermal systems. This  
26 492 is mainly due to its impact system's pumping power. Lower  $\mu$  of working fluids results in less  
27 493 pumping power being required to operate solar thermal systems [61]. The  $\mu$  of synthesized NFs  
28 494 is measured at considered concentrations (0-0.125wt.%) over a range of temperature (25-  
29 495 105°C) and shear-rate (0-100s<sup>-1</sup>). The  $\mu$  of TH-55 varies with the addition of  $Ti_3C_2$  and  
30 496  $Ti_3C_2$ +CuO nanocomposite, increasing with inclusion of solid nanomaterials and decreasing  
31 497 with rising temperatures (**Figure 7**). However, the growth is marginal in contrast to the  
32 498 significant decrease in all concentrations at higher temperatures. The drop in  $\mu$  is due to weaker  
33 499 intermolecular interaction (i.e., adhesion forces) between the fluid and solid materials and  
34 500 higher molecular movement at elevated temperatures. Among the NF samples, 0.025 wt.% of

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3 501 NP loading resulted in the lowest  $\mu$  of 1.75 and 2.64 mPa.s at 105°C, while 0.125 wt.% resulted  
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5 502 in the highest  $\mu$  of 5.71 and 6.31 mPa.s for mono and hybrid NFs, respectively.

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7 503 Rheology is the study of flow behavior of fluids streaming against applied tensions. To  
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9 504 characterize the rheological behavior of experimentally formulated NFs, variation in  $\mu$  of the  
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11 505 NFs is observed for corresponding value of shear-rate at constant temperature of 25 and 50°C  
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13 506 (**Figure 8**). The results suggest that Newtonian flow characteristic is dominant in TH-55/Ti<sub>3</sub>C<sub>2</sub>  
14  
15 507 and hybrid TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO NFs as  $\mu$  remained constant over the range of shear-rate (0-100s<sup>-</sup>  
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17 508 <sup>1</sup>) except little fluctuation at very low of shear-rates up to 20s<sup>-1</sup>. Furthermore, the rheological  
18  
19 509 property remained constant regardless of temperature variation. Newtonian shear behavior of  
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21 510 the NFs is a result of spindle rotation and decorated fluid molecules. Hence, further increment  
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23 511 in shear-rate will not alter the  $\mu$  of the fluids. The obtained results are in accord with other oil-  
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25 512 based NFs reported at similar operating conditions [62, 63]. Since these NFs exhibit  
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27 513 prominently low  $\mu$  and Newtonian flow characteristics at elevated operating temperatures,  
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29 514 these mono and hybrid NFs can be an efficient working fluid in the CPV/T solar system.

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### 516 **3.6. Performance assessment of hybrid CPV/T system**

#### 517 **3.6.1. Temperature of the PV panel**

518 To keep the PV component temperature within the acceptable range, this study used two  
519 nanofluid-based coolants. The effects of various operational parameters on cell efficiency were  
520 investigated under the influence of five effective suns on a tightly packed concentrated  
521 photovoltaic/thermal device. Simulations were performed at 1000-5000 W/m<sup>2</sup> and with  
522 varying nanomaterial concentrations and flowrates. The temperature distribution among  
523 different layers (PV cell, EVA, tedlar and back plate) of the PV pane at 5000 and 1000 W/m<sup>2</sup>  
524 is presented in **Figure S6**. The obtained temperature distribution data show that temperature of  
525 the layers is significantly higher at highest solar irradiance intensity (5000 W/m<sup>2</sup>) than that of  
526 lowest irradiance (1000 W/m<sup>2</sup>). It was observed that the of PV cell temperature becomes  
527 highest compared to EVA, tedlar and back plate layers of the system. Tedlar temperature is  
528 lower than PV cell and adjacent EVA layer due to the cooling effect of the heat exchanger  
529 connected to the back plate.

530 **Figure S7** shows temperature contours of the PV surface and fluid flow through the serpentine  
531 tubes of the thermal collector at a flow rate of 3 LPM and different solar irradiance (1000 and  
532 5000 W/m<sup>2</sup>). It is noticed from the obtained contours that the average surface temperature of  
533 the PV panel is much higher, about 60°C at concentrated intensity of 5000 W/m<sup>2</sup> relative to



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3 534 38°C at 1000 W/m<sup>2</sup> using the nanofluids at 0.125 wt.%. This is due to the fact that high solar  
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5 535 irradiation is captured on the panel at higher irradiation intensity. In addition, the temperature  
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7 536 augmentation of the PV panel is uniform as the radiation rises (shown in **Figure 9**).

8  
9 537 The temperature of the cooling nanofluid increases as it passes through the serpentine thermal  
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11 538 collector tubes and absorbs heat from the tedlar layer to reduce the surface temperature of the  
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13 539 PV panel connected above it. The captured heat transfers through the solid layers and nanofluid  
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15 540 via conduction and convection heat transfer mechanisms, respectively. **Figure S8** represents  
16  
17 541 the temperature contours of the outlet of the cooling nanofluids at a flow rate of 3 LPM and  
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19 542 concentration of 0.125 wt.%. It is observed that the maximum outlet temperature for the  
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21 543 nanofluid is about 32°C higher at 5000 W/m<sup>2</sup>.

22  
23 544 **Figure 9** (a-b) reveals that the average cell temperature values of the PV panel drop with rising  
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25 545 inlet fluid volume flow rate from 0.5 to 4 LPM at an irradiation level of 5000 W/m<sup>2</sup>, and at an  
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27 546 inlet fluid temperature of 30°C. As the inlet fluid flow rate increases, more heat is removed  
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29 547 from the PV unit by convection, which decreases the temperature. It is obtained that the cell  
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31 548 median temperature declines rapidly with increasing flow rates of the fluids. For  
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33 549 Therminol®55, at the flow rate of 0.5 LPM and average cell temperature of 91°C, the  
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35 550 temperature reduces steadily as the flow rate extends upward to 3 LPM. Following that, at a  
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37 551 flow rate of 4 LPM, it decreases slightly, but the pumping capacity is increased. Hence, for this  
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39 552 CPV/T collector, the optimized volume flow rate of cooling fluid is 3 LPM. This trend is same  
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41 553 for both mono and hybrid nanofluids. However, more temperature reduction is achieved in  
42  
43 554 mono nanofluid at the same concentrations of nanoparticle (0.025-0.125 wt.%). It was noticed  
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45 555 that the average solar cell temperature drops from 81 to 51°C for TH-55/Ti<sub>3</sub>C<sub>2</sub> at 0.125 wt.%  
46  
47 556 and from 79 to 56°C for TH-55/Ti<sub>3</sub>C<sub>2</sub> at the same concentration. **Figure 9**(c-d) depicts the cell  
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49 557 temperature variation with solar irradiance. Cell temperature rises reasonably with an increase  
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51 558 in irradiance intensity from 1000 to 5000 W/m<sup>2</sup> for every nanofluid sample for both mono as  
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53 559 well as hybrid nanofluid. It was observed that for each 100 W/m<sup>2</sup> rise in intensity, the average  
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55 560 cell temperature enhances by 0.85°C for Therminol®55 as cooling fluid. However, in the case  
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57 561 of mono TH-55/Ti<sub>3</sub>C<sub>2</sub> nanofluid (0.125 wt.%), it dropped to 0.55°C. The similar rate, 0.57°C  
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59 562 per 100 W/m<sup>2</sup>, is observed in the case of hybrid TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO nanofluid at 0.125wt.%.  
60  
563 This clearly reveals that the formulated nanofluids are performing better as cooling fluid  
564 relative to TH-55. We have observed Nasrin, et al. [64] showed an increase of 0.9°C for each  
565 100 W/m<sup>2</sup> growth in irradiance intensity.

### 3.8.2. Electrical efficiency

The electrical output of the concentrated PV/T system varies between 1000 and 5000 W/m<sup>2</sup> and 0.5 and 4 LPM, respectively, against solar irradiance and volume flow rate (presented in **Figure 10(a-d)**). The electrical efficiency of PV panel increases with flowrate. However, the increment was not remarkable when the flow rate enhanced from 0.5 to 4 LPM for each nanofluid sample at a fixed irradiance of 5000 W/m<sup>2</sup>. Thus, an inlet fluid flow rate of no more than 3 LPM is advantageous for the cooling system of the PV/T module. The maximum efficiencies recorded for nanofluids at maximum fraction (0.125 wt.%) are **12.8 and 12.6%** at a flowrate of 4 LPM for TH-55/Ti<sub>3</sub>C<sub>2</sub> and TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO nanofluids, respectively. Due to rising inlet volume flow of nanofluid, the cell's average temperature is reduced (as shown in **Figure 9**). Consequently, the PV/T module's current declines slightly with an evident increase in PV/T voltage which, in turn, improves the output power and electrical efficiency. **Figure 10(c-d)** confirms the variation of electrical efficiency with irradiance level. Here, the flowrate is fixed at 3 LPM and irradiance is varied up to 5000 W/m<sup>2</sup>. The electrical efficiency decreases with rising irradiation level. The electrical efficiency depreciates from **12 to 9.7%** due to increasing solar radiation for Therminol®55, from **14.1 to 12.5%** for TH-55/Ti<sub>3</sub>C<sub>2</sub> (0.125 wt.%), from **14 to 12.4%** for TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO (0.125 wt.%). Nasrin, et al. [65] revealed that the electrical efficiency for Water/Ag nanofluid in PVT system at 5000 W/m<sup>2</sup> and 3 LPM was 12.5%. This clearly indicates that the formulated nanofluids perform more effectively.

### 3.8.3. Thermal efficiency

The thermal energy efficiency increases with rising volume flow rate of the cooling fluid at a constant solar intensity of 5000 W/m<sup>2</sup> is presented in **Figure 11(a-b)**. An escalation in flow rate from 0.5 to 4 LPM augments the convective heat transfer coefficient of the working fluids. As a result, more heat is transferred at higher velocities under a given temperature difference, increasing energy efficiency. The efficiency intensifies from **58 to 65%** for Therminol®55, from **71 to 78%** for TH-55/Ti<sub>3</sub>C<sub>2</sub> (0.125 wt.%) and **70 to 77%** for TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO (0.125 wt.%). The rate of growth in energy efficiency decreases as solar irradiance increases from 1000 to 5000 W/m<sup>2</sup>, because increasing the solar intensity increases the overall amount of receiving energy from the CPV/T system (depicted in **Figure 11 (c-d)**). The thermal efficiency of the CPV/T collector decreases from **87 to 77%** for TH-55/Ti<sub>3</sub>C<sub>2</sub> (0.125 wt.%) and from **86 to 76%** for TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO (0.125 wt.%). Nasrin, et al. [64] stated a 0.3% abatement of thermal energy efficiency at each 100 W/m<sup>2</sup> growth of irradiation. In this study, using the nanofluids, **0.25% reduction** rate is found for per each 100 W/m<sup>2</sup> increment of irradiation.

### 3.8.4. Efficiency comparison

**Figure 12** (a-b), a comparison of electrical and thermal efficiency of mono and hybrid nanofluid at maximum volume fraction and at a flowrate of 3 LPM with varying irradiation level is presented. It is clear from the **Figure 12** that the mono nanofluid is performing well in comparison to hybrid one, but variation among the performance is very little. So, we can conclude that TH-55/Ti<sub>3</sub>C<sub>2</sub> operated CPV/T system is better, and it is not necessary to make it hybrid one by adding CuO in Ti<sub>3</sub>C<sub>2</sub>.

## 4. Conclusions

In this research, mono (TH-55/Ti<sub>3</sub>C<sub>2</sub>) and hybrid (TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO) nanofluids are prepared using experimentally synthesized emerging two-dimensional MXene (Ti<sub>3</sub>C<sub>2</sub>) nanomaterial and spherical CuO nanoparticles. Significant results are obtained in terms of thermal conductivity, solar absorbance, and stability behavior of the nanofluids. In the simulation-based application part of this work, we numerically evaluated the effectiveness of the synthesized nanofluids as cooling fluids on a concentrated PV/T system under transient conditions. The investigation is performed considering various nanomaterial loadings (0.025-0.125 wt.%), solar irradiations (1000-5000 W/m<sup>2</sup>), flow rates (0.5-4 LPM) and particles to evaluate cooling, electrical and thermal efficiency. The major experimental and numerical findings of this work are drawn in the following points:

- The morphological characterization of the synthesized Ti<sub>3</sub>C<sub>2</sub> nanosheets demonstrates that the MAX-phase structure was successfully transformed into the two-dimensional multi-layered structure. Zeta potential analysis discloses that the formulated nanofluids exhibit a very good level of suspension stability.
- UV-vis analysis confirms enhanced absorbance of the nanofluids. The photo-thermal energy conversion efficiency of TH-55 oil is increased by 80.16 and 85.98% by adding Ti<sub>3</sub>C<sub>2</sub> and Ti<sub>3</sub>C<sub>2</sub>+CuO nanocomposite, respectively.
- The addition of nanomaterials to the base fluid increases thermal conductivity. At 0.125 wt.%, the maximum 84.55 and 80.03% increments are measured 70 °C for TH-55/Ti<sub>3</sub>C<sub>2</sub> and hybrid TH-55/Ti<sub>3</sub>C<sub>2</sub>+CuO NFs, respectively. Furthermore, dynamic viscosity measurements showed a slight increase when adding the nanoparticles, but they dropped remarkably at elevated temperatures. Besides, the nanofluids behaved like Newtonian fluid.

- 630       ▪ Application of the nanofluids in a CPV/T solar collector produced an improved cooling  
 631 effect on the PV unit of the system as 25 and 24°C temperature drops are achieved at  
 632 0.125 wt.% using mono and hybrid nanofluids, respectively.
- 633       ▪ At 5000 W/m<sup>2</sup> solar irradiance, the highest thermal and electrical output enhancement  
 634 of the formulated nanofluid-based CPV/T system is found to be 12.8% and 2% at an  
 635 optimum flowrate of 3 LPM. Moreover, mono nanofluids performed marginally better  
 636 relative to hybrid ones.

637 Future research could lead to the practical implementation of proposed nanofluids on CPV/T  
 638 solar collectors by examining additional parameters, such as exergy analysis and estimation of  
 639 the system's pumping power.

#### 640 **Competing interests**

641 The authors declare that they have no financial or personal conflicts of interest that may seem  
 642 to have affected the work reported in this article.

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849 **Table 1.** Characteristics of TH-55 base fluid.

Properties	Observed values for therminol®55
Dielectric constant (at 23°C)	2.23
Refractive index	1.48
Flash point	193°C
Thermal conductivity (at 30°C)	0.136 W/m.K
Dynamic viscosity (at 30°C)	25.2 mPa.s

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851 **Table 2.** Specifications of dispersed nanoparticles used to formulate the nanofluids.

Particles	Dimension	Morphology	Density (g/cm <sup>3</sup> )	Purity
MXene (Ti <sub>3</sub> C <sub>2</sub> )	1-10 μm × 1 nm	two-dimensional	--	99%
CuO	10 nm	spherical	6.4	99%

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853 **Table 3.** Specifications and properties (thermal and optical) of the CPV/T system

Parameter	Values
PV Model No.	ELDORA VSP.72.AAA.03
Material	Polycrystalline silicon cell
Power	300 W
Dimensions	1955 × 982 × 36 mm
Weight of PV panel	20.5 kg
$h_{panel - tedlar}$	150 W/m <sup>2</sup> K
$h_{tedlar - tubing}$	77 W/m <sup>2</sup> K
$h_{tubing - nanofluid}$	66 W/m <sup>2</sup> K
$A_{PV}$	0.9
$A_{tedlar}$	0.5
$Emissivity_{PV}$	0.99
$k_{EVA}$	0.311 W/m.K
$k_{PV}$	148 W/m.K
$k_{Tedlar}$	0.15 W/m.K
$k_{thermalpaste}$	1.9 W/m.K
$k_{tubes}$	2700 W/m.K

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**Table 4.** Grid independency test.

<b>S. No.</b>	<b>Mesh size (elements)</b>	<b>PV Temp. (°C)</b>	<b>Deviation (%)</b>	<b>Outlet Temp. (°C)</b>	<b>Deviation (%)</b>	<b>Solution Time (s)</b>
1	$2.5 \times 10^5$	42.341	--	41.213	--	560
2	$4 \times 10^5$	43.872	1.2%	40.751	1.13%	720
3	$6 \times 10^5$	44.003	0.29%	40.254	1.23%	817
4	$8 \times 10^5$	44.118	0.26%	39.104	2.94%	1115
5	$1.5 \times 10^6$	45.200	2.3%	38.889	0.55%	1487
6	$3.5 \times 10^6$	45.201	0.002%	38.801	0.22%	1815

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**Table 5.** Validation of average cell Temperature.

<b>Flowrate (LPM)</b>	<b>Cell Temperature (°C)</b>		<b>Percentage error</b>
	<b>Present Research</b>	<b>Nasrin, et al. [47]</b>	
0.5	52.56	51.11	2.83%
1	49.85	48.04	3.70%
3	47.10	45.76	2.92%

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