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## Dynamic Simulation and Exergy Analysis of an Organic Rankine Cycle Integrated with Vapor Compression Refrigeration System

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#### 17 Abstract

18 Organic Rankine Cycle (ORC) has consistently been demonstrated to be one of the most effective and reliable methods for 19 extracting low-grade waste heat energy. Research works including simulations and experimentations with considerable 20 refrigerants are carried out to investigate the performance analyses of an integrated system. The objective of this study is to 21 analyze the integrated vapor compression refrigeration system (VCRS)-ORC system, for which a custom model is built on 22 MATLAB-Simulink, and refrigerant properties are imported from the Coolprop database. This methodology allowed the authors 23 to perform dynamic analysis of the system for varying VCRS evaporator temperature, VCRS load, and VCRS condenser 24 temperature, and that too with different refrigerants considered making a total of 150 combinations. Research works use 25 Engineering Equation Solver (EES) or MATLAB to simulate systems, which has limited capabilities in terms of dynamic 26 analysis and the number of refrigerants selected for the analysis. This however is not the case with the methodology followed in 27 this research work. The model developed for the dynamic simulation and performance assessment of the integrated VCRS-ORC 28 system is verified. For dynamic simulation, the VCRS load, evaporator, and condenser temperatures are varied from 1 TR to 10 29 TR, 0  $^{\circ}$ C to 10  $^{\circ}$ C, and 45  $^{\circ}$ C to 60  $^{\circ}$ C, respectively. From the exergy analysis, it is found that, in comparison to all combinations, 30 the best performance is obtained for the R141b-R1234ze(Z) pair as the ORC and VCRS refrigerant respectively. The 31 performance parameters for this combination give a maximum system overall exergy efficiency of 33.045 %, a maximum net 32 Coefficient of Performance of 4.593, and the least exergy destruction of 2.591 kW. The net Coefficient of Performance increases 33 from 3.9 to 5.5 as the evaporator temperature rises, but the turbine work decreases from 84.5 W to 78.5 W. However, these 34 parameters show an inverse trend with a rise in the condenser temperature. Based on the simulations, the integrated vapor 35 compression refrigeration system - Organic Rankine Cycle system has a 10.62 % higher Coefficient of Performance than the 36 refrigeration system for the R141b-R1234ze(Z) refrigerant pair.

Keywords: Organic Rankine Cycle; waste heat; exergy analysis; dynamic simulation; vapor compression refrigeration system;
 exergy efficiency

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#### 39 1. Introduction

The dependence of humans on the consumption of energy has increased drastically during the last few 40 41 vears. It causes a huge consumption of fossil fuels which creates certain environmental issues. According 42 to the survey by Mahmoudi et al. [1], about 50 % of the world's energy is available in the form of waste 43 heat. The various sources of heat energy like internal combustion engines, household waste, geothermal, 44 solar radiation, steam and gas turbines, etc. are available. This waste heat can be recovered and used to 45 produce a useful power output. Wahile et al. [2], suggested that there are many other ways to utilize this 46 waste heat by using the phase change materials. However, the use of an ORC offers certain advantages over other waste heat recovery systems like a requirement of no land, large potential utilization factor, is 47 48 of manufacture, easy availability of components, flexibility, low maintenance requirement, safe, cheaper, 49 showing better thermal performance, etc. [3]. As organic fluids have a lower boiling point than water, 50 this allows recovering energy from low-temperature waste heat resources. The thermal efficiency of ORC 51 is a function of working fluid temperature, the temperature of the heat source, and the sink. In general, its 52 value lies between 2 % to 19 %. Malwe et al. [4] conducted a comprehensive review of the usage of an 53 ORC as a method of waste heat recovery, citing it as one of the most prominent and efficient methods. 54 The results of an exergetic analysis of heat extraction from a VCRS using 29 distinct potential ORC fluids 55 and an integrated VCRS-ORC system are presented. In this paper, VCRS and ORC systems are integrated to have energy conservation by tapping the waste heat. The objective of the study is to make use of the 56 waste heat from the refrigeration system's condenser as a means of waste heat recovery to run an ORC; 57 thereby producing a power output from the waste heat. Aside from that, further ORC characteristics such 58 59 as working fluid selection, thermodynamics, and environmental impact on working fluid performance, 60 among others, are examined. 61



Fig. 1. A schematic representation of Dincer's six-step approach [4]

Fig. 2. A systematic representation of exergization [4]

Dincer et al. [5] emphasize the significance and necessity of doing an exergy analysis in any system. The major goal of this article is to introduce some thermodynamic dimensions to create a platform for exergization as Dincer's six-step approach is shown in Fig. 1. Exergization (Fig. 2) is the process of applying exergy analysis to improve the design and analysis of a thermal system to get an efficient system [6]. Several types of research have worked in the fields of dynamic simulation, modeling, and performance evaluation of an integrated VCRS-ORC system employing energy and exergy studies as a tool. It's further broken down into exergy analysis as a tool and ORC as a waste heat recovery method.

#### 69 1.1. Exergy analysis as a tool

70 Dincer et al. [7] have examined the role of exergy and its application by developing the AIDA concept which is relevant to all systems. Malwe et al. applied exergy analysis as a tool to evaluate the 71 72 performances of thermal engineering systems like internal combustion engines [8], cryocoolers, etc. to evaluate the performances of the systems [9]. Ahamed et al. [10] examined the potential/scope of research 73 74 work in the field of second analysis of a VCRS for various applications. Exergy efficiency can be improved by lowering the temperature difference between the evaporator and condenser, as well as 75 76 ensuring that the refrigerant is subcooled by about 5 <sup>o</sup>C. By blending these hydrocarbons in various 77 percentages with R134a (as a baseline), the performance of a VCRS can be improved. Compressors have 78 the highest exergy destruction of all the components, indicating that there is room for improvement in 79 exergy efficiency. Using nano lubricants and nanofluids instead of conventional refrigerants in the system 80 is another way to reduce compressor exergy destruction. Anand et al. [11] conducted a refrigerant literature review that included simulated case studies. The goal of this review is to learn about the various 81 82 software programs used to specify refrigeration systems. A solution heat exchanger can be used to improve the performance of an existing system. A suitable output can also be obtained by delivering solar 83 streams at temperatures of roughly 100 °C to a vapor absorption refrigeration system. Borikar et al. [12] 84 85 performed an experimental assessment on a domestic refrigerator to measure energy consumption.

Malwe et al. carried out exergy analysis [13], and management of a simple VCRS [14]. Moreover, a 86 simulation work involving the energy and exergy assessments of a multi-stage VCRS suggesting the 87 88 component-wise exergy destructions and efficiencies formulations is carried out. [15]. Jemma et al. [16] 89 investigate the usage of R1234ze(E) as a refrigerant alternative to R134a. The results demonstrate that the 90 compressor, condenser, expansion valve, and evaporator are in decreasing order of exergy destruction. 91 R1234 has performance characteristics that are similar to R134a. However, as compared to R134a, the COP and exergy efficiency of R1234ze as a refrigerant is higher. R1234ze(E) has been determined to be 92 superior to R134a in terms of performance and environmental friendliness. Babiloni et al. [17] executed 93 94 the exergy analysis of a small VCRS is done using R513A as a new alternative refrigerant to R134a.

95 Compressor, evaporator, expansion valve, and condenser are in decreasing sequence of exergy 96 destruction rate. In comparison to R-134a, Ozgur et al. [18] employed R1234yf as an alternative 97 refrigerant in the refrigeration system, and the results reveal a lower exergy destruction value. However, 98 one of the main issues of R1234vf is its safety. Geete et al. [19] have analyzed a VCRS using the concept 99 of minimization of entropy and exergy destruction. The VCRS has the worst performance with a 100 combination of 85 % R134a and 15 % R152a due to higher entropy generation and exergy destruction. As a result, a blend of 85 % R134a and 15 % R290, or R600a alone, is used as a replacement for R134a. Roy 101 et al. [20]have thermodynamically analyzed a VCRS with R134a as a refrigerant. The modified system 102 103 achieves a 3.7 % and 3.5 % increase in COP and exergy efficiency, respectively. Gaurav et al. [21] 104 conducted the first and second analyses of a VCRS with various refrigerants such as R134a, R1234vf, and R1234ze, as well as their mixtures. Among the 31 different sets, the best performance and alternative to 105 R134a is obtained for R134a/R1234yf (90%/10%), N13a [R134a/R1234yf/R1234ze (42%/18%/40%)], 106 R152a and N13b [R134a/R1234ze (42%/58%)]. From an energy consumption standpoint, Shikalgar et al. 107 [22] used a water-cooled and a hot wall air-cooled condenser alternately for a domestic refrigerator type, 108 109 and different metrics such as COP, exergy efficiency, and so on are investigated. The refrigerator with the 110 water-cooled condenser is 7 to 9 % more efficient than the refrigerator with the other type.

111 Agarwal et al. [23] used EES to investigate the energy and exergy of a simple VCRS with R1234ze 112 and R1234yf as R134a replacements. For similar operating conditions, R1234yf performs marginally 113 worse than R134a. R1234ze is a remarkable alternative to R134a, outperforming it by 1.8 % and 1.87 % 114 in terms of exergy efficiency and COP, respectively when compared to R134a. Kumar et al. [24] 115 demonstrated an exergy approach for assessing a vapor compression refrigeration system. For R11 and R12, respectively, exergy efficiencies of 49 % and 50 % were found. Fazar et al. [25] preferred R290 to 116 replace R410a in a split air conditioning system. Instead of R410a, R290 refrigerant is used, resulting in a 117 refrigerant need of about 45 to 55 % of the full charge of R410a, and the COP rises from 4.6 to 4.9, and 118 compression work is reduced by 35.7 %. Arora et al. [26] investigated the performance of an actual vapor 119 refrigeration system using R502, R404a, and R507a as refrigerants. According to the results of the 120 investigation, R507 is a great refit to R502 when compared to R404a.R1234yf and R1234ze are labeled 121 "fourth-generation refrigerants" by Yataganbaba et al. [27] since they have zero ODP and are employed 122 123 as substitutes to R-134a in a two-evaporator VCRS with an evaporator pressure regulator. The results show that a multi-evaporator VCRS has a higher exergy efficiency than a single-evaporator system. 124 125 Nikolaidis et al. [28] investigated a two-stage VCRS with a flash intercooler using a technology method 126 using R22 refrigerants. By lowering the individual irreversibilities in the condenser and evaporator, the 127 overall irreversibility rate of the plant is reduced by 2.40 and 2.78 times, respectively, compared to the 128 original system. Anjum et al. [29] conducted a first and second law analysis of waste heat extraction from

the intercooler of a multi-stage VCRS utilizing ammonia refrigerant. For a -40 <sup>o</sup>C evaporator temperature, a maximum COP of 3.087 is obtained. COP, on the other hand, decreases when the condenser temperature rises and the evaporator temperature falls. Dwinanto et al. [30] provide a step-by-step guide on designing a refrigeration system.

#### 133 *1.2. Waste heat recovery using ORC*

Quoilin et al. [31], described a technological solution for ORC, a list of ORC manufacturers over the 134 globe, etc. with a focus on the temperature levels [32]. Franchetti et al. [33] specified selection of 135 working fluid is the basic step in ORC design. Lakew et al. [34] designed an ORC to achieve either a 136 137 maximum power output or to have a smaller size of components. Six working fluids namely R123, R134a, R245fa, R227ea, R123, R134, R290, and n-pentane are considered. R227ea and R245fa produce 138 maximum power for heat source temperatures in the range of 80  $^{\circ}C - 160 ^{\circ}C$  and 160  $^{\circ}C - 200 ^{\circ}C$ 139 respectively. Rajabloo [35] reveal that the performance improvement in the ORC occurs with both - pure 140 and mixtures of working fluids due to an increase in the mass flow rate of hot source fluid. Yaday, et al. 141 [36] explored working fluid selection from an ORC in an Indian region of 75 °C to 80 °C. R245fa 142 outperforms all other refrigerants in terms of ORC efficiency. Tahir et al. [37] built an ORC with working 143 fluid- R245fa having an objective function of producing lower than 1 kW of turbine power. The cold 144 source and hot source temperatures are maintained in the temperature in the range of 60  $^{\circ}C - 100 ^{\circ}C$  and 145  $10 \, {}^{0}\text{C} - 30 \, {}^{0}\text{C}$  respectively. It is advised to make use of a largely efficient pump to have positive heat 146 recovery. Caceres et al. [38] did the optimization of the organic Rankine cycle to select the most suitable 147 organic fluid out of 39 different combinations of working fluids. Dai et al. [39] explored the outcome of 148 149 performing parameters on ORC using 10 different organic fluids. The sizing of the turbine varies directly 150 to the inlet-specific volume. Kaushik et al. [40] introduced a Canopus heat exchanger across the compressor and condenser with R134a and R507a as working fluids to enhance the system's COP. Sarkar 151 152 et al. [41] have demonstrated a pinch point method for the analysis of an ORC for maximum heat recovery. Bonk et al. [42] presented a paper on designing a micro-ORC system of 1 kW power output. Ali 153 154 [43] did effective thermal management of the electronic devices using Nickel foam along with paraffin wax by increasing the surface area of heat flow. Additionally, Shahsavar et al. [44] have investigated the 155 156 laminar forced convection problem and improved with the use of holes.

From an efficiency point of view, Novec 649 is considered a novel organic fluid with a GWP of 158 1.1, which is comparatively lower than R245fa. Scagnolatto et al. [45] simulated a small-scale ORC (10 159 kW) for which R123 gives the highest exergy efficiency among the other selected refrigerants due to the 160 presence of a recuperator. Saha [46] assured a waste heat utilization which becomes the need of an hour 161 from environmental and energy conservation concerns. An ORC with a capacity of 20 kW was

162 numerically examined by Kong et al. [47] utilizing R 245fa as the refrigerant in the combined saturated 163 steam/hot water resource and has the highest exergy efficiency. The exergy efficiency varies in direct 164 proportion to the temperature of the heat source. Ozdemir [48]suggested that the power output of the 165 regenerative ORC is higher than that of the simple/basic ORC. The thermal energy efficiency and exergy efficiency of regenerative ORC are 23.1 % and 69.9 % higher than those of simple ORC, respectively. An 166 exergy analysis of an ORC using a geothermal heat source at 125 °C with four refrigerants was performed 167 by Bademlioglu et al. [49].R123 is preferable to R152. Shet et al. [50] investigated an exergetic and 168 energetic analysis of the  $CO_2$  refrigeration system (transcritical mode) using an EES. Chen et al. 169 170 [51]showed the thermal efficiency with R32 is slightly higher than that with the  $CO_2$  system. An ORC 171 device with maximum power output as an objective feature was considered by Oyewumni et al. [52]. A system by Cayer et al. [53] running on a Rankine transcritical cycle draws larger power to the (simple) 172 subcritical cycle. The ratio of the evaporator to the P<sub>cri</sub> for corresponding ORC fluid should lie in the 173 174 range of 4 to 5 to achieve this result. Rawat et al. [54] focused on the energy crisis as a serious problem faced across the globe because of an increase in the day-by-day population which directly and indirectly 175 hampers the environment and the economy. Aphornratana et al. [55] used heat at roughly 60 °C low-grade 176 thermal energy employing R22 and R134a as refrigerants, the output of this integrated system produces a 177 cooling effect at roughly -10 °C, and the coefficient of performance climbs from 0.1 to 0.6. Saleh 178 179 [56]suggested that R602 emerges as the most significant fluid for the integrated system for the temperature range of 70 to 110 °C, out of 14 various refrigerants investigated. Malwe et al. [57] showed 180 181 that R123 and R450a refrigerants are the best performing pair for the VCRS and ORC systems, 182 respectively. The combined system's COP rises to 3.88 (compared to 3.17) with an overall exergy efficiency of roughly 18 % when ORC is used instead of a basic refrigeration system. Moles et al. 183 [58]suggested that the combined system's thermal COP rises from 0.7 to 1.10 when the ORC-VCRS 184 temperature rises and the condenser temperature falls. Over the other refrigerants, R1234ze(E) produces 185 the best results. Wang et al. [59], [60]created a prototype with a 5-kW refrigeration capacity. The 186 refrigerating effect and COP obtained in the experiments were 4.4 kW and 0.48, respectively. Kim et al. 187 [61] have constructed a cogeneration plant employing low-grade heat energy as input to the system. Asim 188 et al. [62] and Imran et al. [63]carried out energy, exergy and economic analysis of a combined VCRS 189 190 and ORC called integrated air conditioning Organic Rankine Cycle system using 36 various sets of 191 refrigerants. Asim et al. [64] evaluated a unique VCRS-ORC waste heat recovery system for tapping condenser heat of around 50 °C. The system under consideration has a refrigerating capacity of 35 kW. 192 193 R600a-R141b produces the highest results (thermal efficiency = 3.05 %) and is the best choice for an integrated system. When ORC is used as part of an integrated VCRS-ORC system, the COP is 12.5 % 194 195 greater than when utilizing a conventional VCRS system.

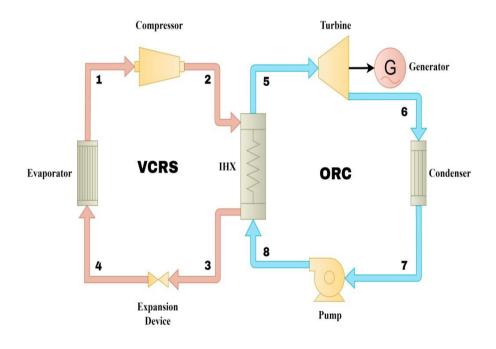
From the extensive literature survey done, it is found that most of the researchers have carried out the 196 static simulations on the integrated VCRS-ORC system with only a few refrigerants considered for the 197 198 research. This may be due to the limitations of the software used for simulation purposes. The necessity to perform this research work is to investigate the novel integration of a Vapor Compression 199 200 Refrigeration System (VCRS) to an Organic Rankine Cycle (ORC) for low-grade waste heat recovery for 201 improving the overall VCRS system efficiency. This creates a necessity to carry out research work on more refrigerants, with a larger number of VCRS-ORC refrigerant combinations, and that too with 202 dynamic simulations. 203

#### 204 1.3. Novelties of the research work

To analyze the integrated VCRS-ORC system, a custom model of the system was built on MATLAB-205 206 Simulink and refrigerant properties were imported from the Coolprop database. This methodology allowed the authors to perform a dynamic analysis of the system for varying VCRS evaporator 207 temperature, VCRS load, and VCRS condenser temperature for 150 combinations of different refrigerant 208 types. Dynamic simulations with MATLAB-Simulink and Coolprop database allow for the incorporation 209 210 of 150 combinations of the simulation results. This is unlike the other research works using EES or MATLAB to simulate systems, which have limited capabilities in terms of dynamic analysis and the 211 number of refrigerants selected for the analysis. The model developed for the dynamic simulation and 212 213 performance assessment of the integrated VCRS-ORC system is verified with the [64] research paper. The objective of this study is to perform both – the energy and exergy analysis of the VCRS-ORC 214 system for all the 150 combinations of selected refrigerants. In conclusion in this research work, the 215 216 authors found that for a VCRS-ORC integrated system, the refrigerant R1234ze(Z) in VCRS and R141b in ORC have minimal energy and exergy destruction and will be the best combination for this integrated 217 218 system. The authors also found from the system dynamic analysis that, work output from ORC can be 219 maximized by decreasing VCRS condenser temperature and increasing VCRS load. Decreasing VCRS 220 condenser load is also found to decrease overall system exergy destruction.

#### 221 2. System Description

The integrated VCRS-ORC system is divided into two parts, the VCRS part and the ORC part. The ORC part also consists of four major components, evaporator, expander, condenser, and pump. The working fluid recovers the low-grade waste heat from the condenser of VCRS, making it increase its enthalpy. This energy is recovered in the expander by making the high pressure and high temperature working fluid expand and generate power. This low-pressure working fluid is then condensed in the condenser and pumped back to high pressure in the pump to complete the ORC cycle. The schematic of 228 the integrated ORC-VCRS system is shown in Fig. 3(a) and the T-S diagram of the VCRS and ORC 229 systems is shown in Fig. 3(c) and Fig. 3 (d) respectively. Working fluid selection for both VCRS and 230 ORC systems is a crucial factor for this research work since the properties of working fluid will define the 231 energy and exergy efficiencies of the system as well as have significant importance in figuring out the safety and environmental impact of the integrated VCRS-ORC system. For this research work, we have 232 233 considered 19 working fluids, selected following their feasibility with the system, availability, and environmental impact. We have also selected some new and less researched working fluids to compare 234 235 their performance on the system against conventional working fluids. All 19 fluids are enlisted in Table 1 236 with their properties, safety, and environmental impact factors. A total of 10 working fluids were 237 considered for VCRs and 15 working fluids were considered for ORC. The fluids labeled as VCRS/ORC in the table were used in both VCRS and ORC systems. Table 2 summarizes the physical, and 238 239 environmental properties of 20 refrigerants considered for the performance analysis and dynamic 240 simulation of the integrated VCRS-ORC system.



(a)

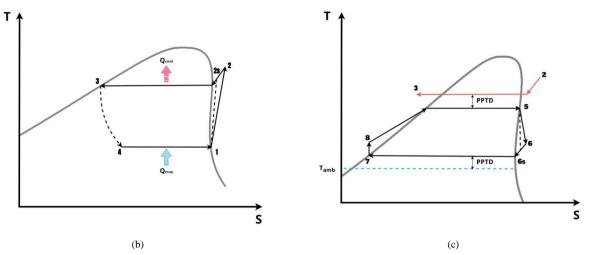


Fig. 3. (a) Schematic of the integrated VCRS-ORC system (b) T-s chart for the VCRS (c) T-s chart for the ORC system

## 241 **3. Mathematical Model**

The mathematical model for dynamic simulation and performance assessment of an integrated VCRS-ORC system is developed on MATLAB-Simulink (R2021a), with a fluid property database being imported from Coolprop [65]. The thermal analysis of the integrated system involves both energy and exergy analyses. This section presents all the energy and exergy equations used for the simulation model. However, for the ease of simulation and simplicity of the model, the following assumptions are considered:

#### 248 *3.1. Assumptions*

- Frictional, heat, kinetic, and potential energy losses and pressure drops across individual components are neglected.
- For the VCRS system, the refrigerant enters the compressor in a dry-saturated state and exits the condenser in a wet-saturated state.
- The throttling process in the expansion device of the VCRS system is considered to be isenthalpic.
- For the ORC system, refrigerant exits the turbine in a dry-saturated state and enters the pump in a wet-saturated state [4].

#### 256 3.2. Parameters considered

The parameters considered for performing energy and exergy analysis of the VCRS, ORC, and integrated VCRS-ORC system include various COPs ( $COP_{vcrs}$  and  $COP_{net}$ ), thermal efficiency ( $\eta_{th}$ ), component-wise and overall system exergy destruction (Ex<sub>dest</sub>), various exergy efficiencies ( $\eta_{ex,vcrs}$ ,  $\eta_{ex,orc}$ ,  $\eta_{ex,net}$ ).

## 261 *3.3. Energy and exergy analysis*

The energy analysis is carried out using the principles of the first law of thermodynamics. For the VCRS and ORC systems, calculating the  $COP_{vcrs}$  and  $\eta_{th}$  is the ultimate aim of the energy analysis. The energy analysis is based on the fundamentals of the second law of thermodynamics (eq. 1 and eq. 2).  $\eta_{ex,vcrs}$  and  $\eta_{ex,orc}$  are the final expected outcomes using the same. For the integrated VCRS-ORC system,  $COP_{net}$ , and  $\eta_{ex,net}$  are calculated. The formulation for the same is tabulated in Table 3:

267 Specific exergy flow of a component is given by [5]:

$$ex = (h - h_0) - T_0(s - s_0)$$
(1)

268 The general exergy balance in rate form: [16]:

$$\sum_{in} \left( 1 - \frac{T_0}{T_i} \right) \dot{Q}_i + \dot{W}_{in} + \sum_{in} \dot{m}_i e x_i = \sum_{out} \left( 1 - \frac{T_0}{T_i} \right) \dot{Q}_i + \dot{W}_{out} + \sum_{out} \dot{m}_i e x_i + \dot{E} x_D \tag{2}$$

Table 1. Operating conditions of the system

Parameter	Units	Value
Load on the evaporator	TR	1
Ambient temperature	$^{0}C$	25
Evaporator temperature	$^{0}C$	0
Condenser temperature	$^{0}C$	45
Compressor isentropic efficiency	%	70
ORC pump isentropic efficiency	%	80
ORC turbine isentropic efficiency	%	80
Effectiveness of IHX		0.75
Pinch point temperature difference	$^{0}C$	5

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Table 2. Properties of the working fluids/refrigerants considered for the simulation and performance assessment of the integrated VCRS-ORC system

System	Refrigerant Group	Refrigerant Name	Refrigerant Type	Molecular mass (M)	Critical temperature	Critical Pressure	ODP	GWP (for 100	ASHRAE Safety	ASHRAE Flammability	ASHRAE Toxicity	
				(kg/mol)	$(T_{cr}) (^{0}C)$	(P <sub>cr</sub> ) (bar)		years)				
VCRS	HCFC	R22	Wet	0.0864	96.29	49.9	0.055	1760	A1	Non-Flammable	Lower Toxicity	
VCRS	HC	R290	Wet	0.0440	96.7	42.48	0	3.3	A3	Higher Flammability	Lower Toxicity	
VCRS	HFC	R407C	Wet	0.0862	86.05	46.34	0	1774	A1	Non-Flammable	Lower Toxicity	
VCRS	HFO	R1234ze(Z)	Dry	0.1140	150.27	35.30	0	6	A2L	Lower Flammability	Lower Toxicity	
ORC	HCFC	R123	Isentropic	0.1529	183.68	36.62	0.02	79	B1	Non-Flammable	Higher Toxicity	
ORC	HCFC	R141b	Isentropic	0.1169	204.2	42.50	0.12	725	A2	Lower Flammability	Lower Toxicity	
ORC	HFC	R227ea	Dry	0.1700	102.8	29.80	0	3350	A1	Non-Flammable	Lower Toxicity	
ORC	HFC	R245fa	Dry	0.1340	154.05	36.40	0	858	B1	Non-Flammable	Higher Toxicity	
ORC	HFO	R1233zd(E)	Dry	0.1304	166.6	36.23	0	1	A1	Non-Flammable	Lower Toxicity	
ORC	HFC	R245ca	Dry	0.1340	174.42	39.25	0	716	<b>B</b> 1	Non-Flammable	Higher Toxicity	
ORC	HCFC	R124	Dry	0.1364	122.28	36.24	0.02	527	A1	Non-Flammable	Lower Toxicity	
ORC	HFC	R152a	Wet	0.0660	113.26	45.17	0	138	A2	Lower Flammability	Lower Toxicity	
ORC	HFC	R32	Wet	0.0520	78.11	57.82	0	677	A2L	Lower Flammability	Lower Toxicity	
VCRS/ORC	HFC	R134a	Isentropic	0.1020	101.21	40.59	0	1300	A1	Non-Flammable	Lower Toxicity	
VCRS/ORC	HFC	R404A	Wet	0.0976	72.14	37.35	0	3922	A1	Non-Flammable	Lower Toxicity	
VCRS/ORC	HC	R600a	Dry	0.0581	134.7	36.40	0	3	A3	Higher Flammability	Lower Toxicity	
VCRS/ORC	HFC	R410A	Wet	0.0725	70.17	47.70	0	2088	A1	Non-Flammable	Lower Toxicity	
VCRS/ORC	HFO	R1234ze(E)	Dry	0.1140	109.52	36.34	0	6	A2L	Lower Flammability	Lower Toxicity	
VCRS/ORC	HFO	R1234yf	Dry	0.1140	94.85	33.82	0	4	A2L	Lower Flammability	Lower Toxicity	

		Energy Analysis			Exergy Analysis	
	Compressor	$h_2 = \left(\frac{h_{2s} - h_1}{\eta_{is}}\right) + h_1$	(3)	Compressor	$Ex_{comp} = m_{vcrs}[(h_1 - h_2) - T_0(s_1 - s_2)] - W_{comp}$	(18)
		$W_{comp} = m_{vcrs} \left( h_2 - h_1 \right)$	(4)			
	Condenser	$Q_{vcrs,cond} = m_{vcrs} \left( h_2 - h_3 \right)$	(5)	Condenser	$Ex_{vcrs,cond} = (m_{vcrs}[(h_2 - h_3) - T_0(s_2 - s_3)]) + (m_{orc}[(h_8 - h_5) - T_0(s_8 - s_5)])$	(19)
				Expansion valve	$Ex_{exp} = m_{vcrs}[(h_3 - h_4) - T_0(s_3 - s_4)]$	(20)
VCRS	Evaporator	$Q_{vcrs,evap} = m_{vcrs} (h_1 - h_4)$	(6)	Evaporator	$Ex_{vcrs,evap} = (m_{vcrs}[(h_4 - h_1) - T_0(s_4 - s_1)])$	(21)
					$-\left(\left(1-\frac{T_0}{T_4}\right)Q_{vcrs,evap}\right)$	
	COP	$Q_{vcrs,evap}$	(7)	Exergy destruction	$Ex_{vcrs} = Ex_{comp} + Ex_{vcrs,cond} + Ex_{exp} + Ex_{vcrs,evap}$	(22)
		$COP_{vcrs} = \frac{Q_{vcrs,evap}}{W_{comp}}$		Exergy efficiency	$\eta_{ex,vcrs} = 1 - \left(\frac{Ex_{vcrs}}{W_{comp}}\right)$	(23)
	Mass flow rate	$m_{orc} = \left(\frac{Q_{vcrs,cond} \cdot \epsilon_{hx}}{h_t - h_t}\right)$	(8)	Evaporator	$Ex_{orc,evap} = (m_{orc}[(h_8 - h_5) - T_0(s_8 - s_5)])$	(24)
		$(n_1 n_4)$			+ $(m_{vcrs}[(h_2 - h_3) - T_0(s_2 - s_3)])$	
	Evaporator	$Q_{orc,evap} = m_{orc} (h_5 - h_8)$	(9)			
	Turbine	$h_6 = h_5 - [(h_5 - h_{6s}) \cdot \eta_{is}]$	(10)	Turbine	$Ex_{turb} = m_{orc}[(h_5 - h_6) - T_0(s_5 - s_6)] + W_{turb}$	(25)
		$W_{turb} = m_{orc} (h_5 - h_6)$	(11)			
ODC	Pump	$h_8 = \left(\frac{h_{8s} - h_7}{\eta_{is}}\right) + h_7$	(12)	Pump	$Ex_{pump} = m_{orc}[(h_7 - h_8) - T_0(s_7 - s_8)] - W_{pump}$	(26)
ORC		$W_{pump} = m_{orc} \left( h_8 - h_7 \right)$	(13)			
	Condenser	$Q_{orc,cond} = m_{orc} \left( h_6 - h_7 \right)$	(14)	Condenser	$Ex_{orc,cond} = (m_{orc}[(h_6 - h_7) - T_0(s_6 - s_7)])$	(27)
					$+\left(\left(1-rac{T_0}{T_7} ight)Q_{orc,cond} ight)$	
	Net work	$W_{net} = W_{turb} - W_{pump}$	(15)	Exergy destruction	$Ex_{orc} = Ex_{orc,evap} + Ex_{turb} + Ex_{pump} + Ex_{orc,cond}$	(28)
	Thermal	$n_{\star} = \frac{W_{turb}}{W_{turb}}$	(16)	Exergy efficiency	$\eta_{ex,orc} = \frac{W_{net}}{m_{vcrs}[(h_2 - h_3) - T_0(s_2 - s_3)]}$	(29)
	Efficiency	$\eta_{th} = \frac{W_{turb}}{Q_{orc,evap}}$				
Integrated VCRS-	Net COP	$COP_{net} = \frac{Q_{vcrs,evap}}{W_{comp} - W_{net}}$	(17)	Overall/net exergy destruction	$Ex_{net} = Ex_{vcrs} + Ex_{orc}$	(30)
ORC				Overall/net exergy	$\eta_{ex,net} = \frac{Q_{vcrs,evap} \left(1 - \frac{T_0}{T_4}\right)}{W_{comp} - W_{net}}$	(31)
system				efficiency	$\eta_{ex,net} = \frac{1}{W} - \frac{1}{W}$	

Table 3. Formulae used for energy and exergy analysis for the integrated VCRS-ORC system [64], [5], [16]

274

#### 275 *3.4. Dynamic simulation model*

In this research work, a dynamic simulation model of the integrated VCRS-ORC system is developed to analyze the system over a comprehensive range of input conditions and to scale up the number of working fluids being studied. This simulation model is developed on MATLAB-Simulink, enabling us to use its time step feature for dynamically simulating the system. To calculate working fluid properties during the simulation of different states, this model is integrated with Coolprop open-source database.

The simulation model is mainly divided into three parts, the VCRS model, ORC model, and energy and exergy analysis model. The simulation starts with the VCRS model, where first the VCRS working fluid is defined, followed by the calculation of its state points for the respective input conditions. This data is forwarded to the ORC model, where its state points for its respective input conditions are calculated. Finally, all the calculated state points data are forwarded to the final energy and exergy analysis model, where all equations from Table 3 are used to perform component-wise energy and exergy analysis.

288 To simulate multiple fluids at once, we have implemented a matrix type calculation layout for the 289 model, where instead of calculating for one fluid at an instance, the model calculates state properties of 290 all the considered fluids at once and stores the data of each fluid in specially allotted locations for that 291 data in a matrix. To interact with data of another state matrix, the model uses matrix rules to perform 292 mathematical operations. This approach allowed us to study much more working fluids than previously done. For a fixed ORC working fluid, the performance assessment of the integrated VCRS-ORC system 293 using exergy analysis as a tool is done for 10 different VCRS refrigerants (R22, R134a, R290, R404A, 294 295 R407C, R600a, R410A, R1234ze(E), R1234ze(Z), and R1234yf), and all the performance parameters like COP<sub>vcrs</sub>, COP<sub>net</sub>, W<sub>turb</sub>, Ex<sub>net</sub>, and various exergy efficiencies are calculated individually for a particular 296 VCRS refrigerant. So, for one ORC working fluid and for 10 VCRS refrigerants, 10 sets of the results are 297 298 obtained. This process is repeated for 15 corresponding ORC fluids (R600a, R123, R141b, R227ea, R245fa, R1234ze(E), R134a, R245ca, R124, R152a, R404A, R410A, R32, R1233zd(E), and R1234yf). 299 300 Consequently, 150 (15 ORC refrigerants  $\times$  10 VCRS refrigerants) sets of the results are obtained for the 301 above said ORC-VCRS refrigerant combinations.

For dynamically simulating the model, input parameters vary linearly over a range and magnitudes of these parameters were stepped through this range by Simulink to have a dynamic image of the model. In this research work, VCRS evaporator temperature, VCRS condenser temperature, and load on VCRS were varied over a range of 0°C - 10°C, 45°C - 60°C, and 1 TR – 10 TR respectively for all the considered working fluids. Fig. 4 shows the general flow process of the dynamic simulation model. The input parameters are depicted in orange color. VCRS model takes in T<sub>1</sub>, T<sub>3</sub>, Q<sub>evap</sub>, and  $\eta_{is}$  as input parameters to calculate state -1 to state - 4. The VCRS model gives  $m_{vcrs}$  as its output, and it is used along with PPTD, T<sub>3</sub>, T<sub>amb</sub>,  $\eta_{is}$ , and  $e_{hx}$  as input parameters for the ORC model. The outputs of both the VCRS and ORC models are used in the energy and exergy analysis model to calculate COP<sub>net</sub>,  $W_{net}$ ,  $\eta_{th}$ , COP<sub>net</sub>,

311  $\eta_{ex}$ , and  $Ex_{net}$ .

#### 312 3.5. Model verification

The integrated VCRS-ORC system dynamic simulation model is verified with the model in the research work of [64]. For verification, similar input parameters present in the above-said research work were replicated in the current model and the obtained results were compared. The VCRS working fluid is considered R22 for the model verification with all VCRS input parameters listed in Table 4.

Fig. 5 (a), (b), and (c) show the model verification between the present study and research work by [64]. The verification is done by varying the condenser temperature for calculating the corresponding variation in the compressor work,  $COP_{vcrs}$ , and refrigerating effect. The verification results reveal that the percentage error between the present study and research work referred for compressor work,  $COP_{vcrs}$ , and refrigerating effect are 7.28 %, 12.76 %, and 0.09 % respectively. Since the performance curves show similar trends to that of referred work with a considerable deviation, the model is said to be verified for this study.



Table 4 . VCRS model validation input parameters [64]

Values
-20 °C
30 °C to 70 °C
0.7
5 °C

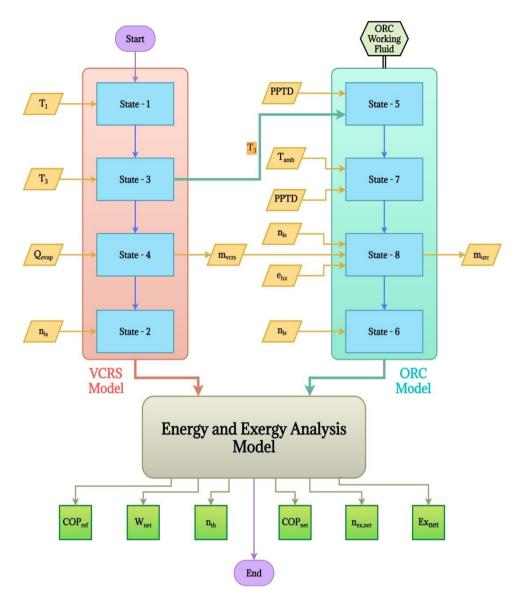


Fig. 4. Flow chart for the dynamic simulation of the integrated VCRS-ORC system

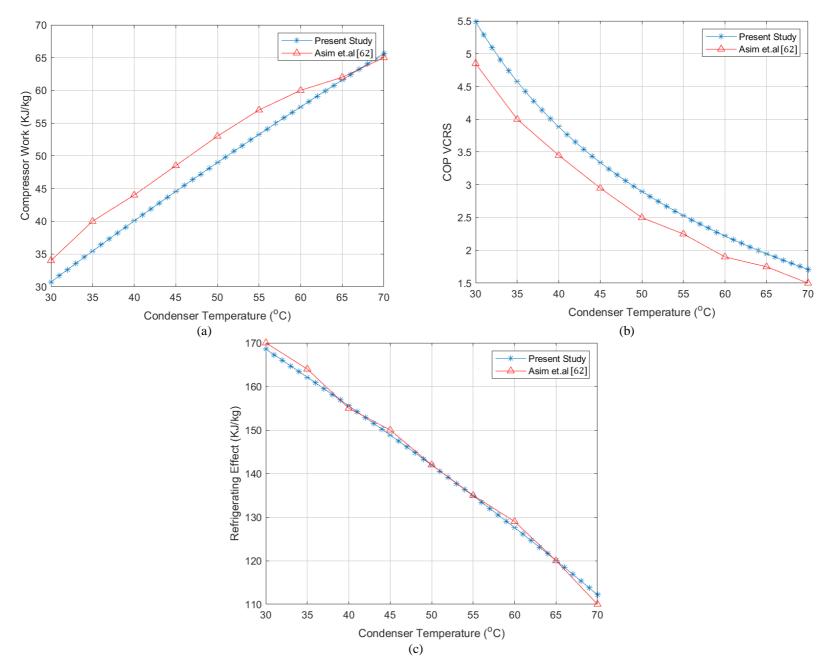


Fig. 5. Model verification between the present study and a reference paper by varying the condenser temperature for the variation in (a) compressor work (b) COP of VCRS (c) refrigerating

#### 329 4. Results and Discussions

This section deals with the simulation results followed by the discussions using various graphs for the 330 331 performance analysis of the integrated VCRS-ORC systems. For the simulation purpose, Table 5 gives the values of the various performance parameters like COP, turbine work output, exergy efficiency, and 332 333 exergy destructions for the corresponding set of refrigerant combinations. The dynamic simulation is performed for the specific set of conditions as mentioned in Table 5. A total of 20 refrigerants are 334 335 considered for the dynamic simulation of the integrated vapor compression refrigeration system - Organic Rankine Cycle system. For a fixed ORC working fluid, 10 different VCRS refrigerants (R22, R134a, 336 R290, R404A, R407C, R600a, R410A, R1234ze(E), R1234ze(Z), and R1234vf) are used for the analysis. 337 This process is repeated for 15 corresponding ORC fluids (R600a, R123, R141b, R227ea, R245fa, 338 R1234ze(E), R134a, R245ca, R124, R152a, R404A, R410A, R32, R1233zd(E), and R1234yf). 339 Consequently, 150 sets of the results are obtained for the above said ORC-VCRS refrigerant 340 combinations. In comparison to all ORC-VCRS refrigerants combinations, the best performance is 341 obtained for the R141b - R1234ze(Z) pair as the ORC and VCRS refrigerant respectively. All the 342 performance parameters for this combination gives maximum values (COP<sub>vcrs</sub>= 4.152, COP<sub>net</sub>= 4.593, 343  $\dot{\eta}_{ex,vers}$  = 29.871 %, and  $\dot{\eta}_{ex,net}$  = 33.045 %); whereas the least exergy destruction (Ex<sub>net</sub> = 2.591 kW) is 344 obtained for this combination. For this reason, various performance graphs are plotted and discussed for 345 the R141b – R1234ze(Z) pair only. 346

From Table 5, among the different ORC-VCRS refrigerants combinations, the best performance in comparisons among the allis obtained for the R141b–R1234ze(Z) combination as the ORC and VCRS refrigerant respectively. The reason being for the R141b–R1234ze(Z) combination, all the performance parameters maximum values are obtained ( $COP_{vers}$ = 4.152,  $COP_{net}$ = 4.593,  $\dot{\eta}_{ex,vers}$ = 29.871 %, and  $\dot{\eta}_{ex,net}$  = 33.045 %). This combination shows the least Ex<sub>net</sub> of 2.591 kW among the all. Due to these reasons, the R141b – R1234ze(Z) combination is considered the best performance pair, and the various performance graphs are plotted and discussed as shown below.

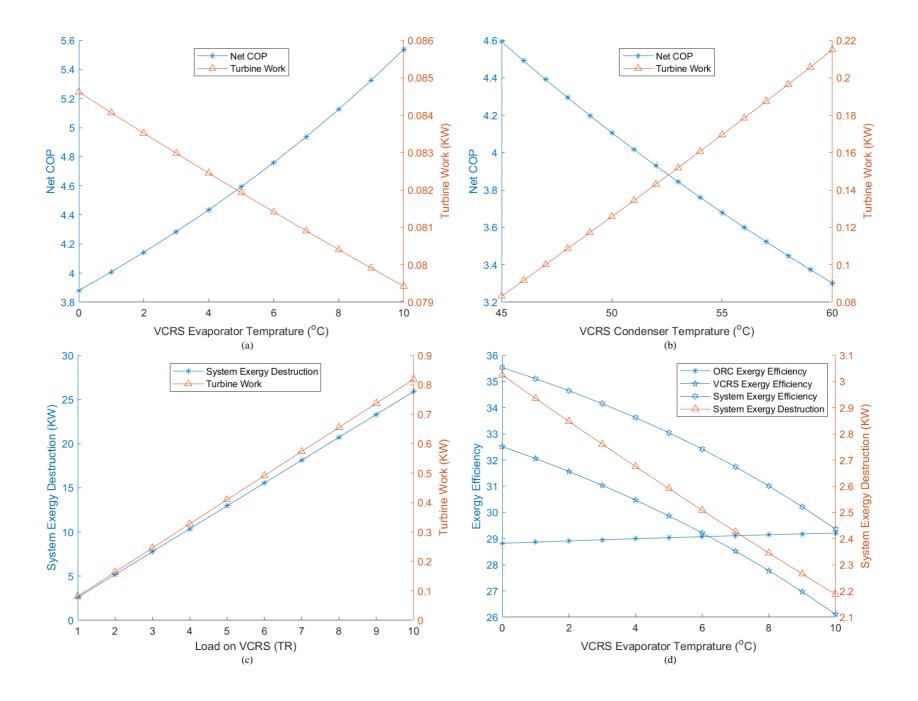
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Table 5. Performance analysis of the integrated VCRS-ORC system for different sets of refrigerant combinations

Sr. No.	ORC working fluid	VCRS working fluid	COP <sub>vcrs</sub>	COP <sub>net</sub>	W <sub>turb</sub> (kW)	ή <sub>th,orc</sub> (%)	ή <sub>ex,vcrs</sub> (%)	ή <sub>ex,orc</sub> (%)	Ex <sub>net</sub> (kW)	ή <sub>ex,net</sub> (%)
1.	it offiting frate	R22	3.905	4.288	0.083	2.518	28.09	26.428	2.778	30.85
2.		R134a	3.866	4.242	0.084	2.518	27.812	28.22	2.76	30.52
3.		R1940 R290	3.787	4.15	0.084	2.518	27.246	28.209	2.807	29.853
,. I.		R404A	3.35	3.639	0.086	2.518	24.469	27.831	3.12	26.607
5.		R404/1 R407C	3.229	3.499	0.087	2.518	27.199	24.092	3.343	30.831
5.	R600a	R407C R600a	3.963	4.357	0.083	2.518	28.509	28.884	2.686	31.346
7.		R000a R410A	3.547	3.868	0.085	2.518	25.554	25.917	3.013	27.932
7. 3.		R1234ze(E)	3.842	4.214	0.085	2.518	27.64	28.774	2.759	30.318
). ).		R1234ze(Z)	4.152	4.583	0.082	2.518	29.871	28.405	2.599	32.969
,. 10.		R1234yf	3.642	3.979	0.082	2.518	26.203	28.874	2.881	28.627
11.		R125491 R22	3.905	4.295	0.083	2.496	28.09	26.884	2.001	30.902
12.		R134a	3.866	4.293	0.083	2.490 2.496	28.09	28.707	2.771	30.902
12. 13.		R134a R290	3.787	4.249	0.083	2.490 2.496	27.812	28.696	2.755	29.902
13. 14.		R290 R404A	3.35	4.130 3.644	0.085	2.490 2.496	27.240 24.469	28.312	2.8 3.113	29.902
5.		R404A R407C	3.33	3.504	0.085	2.490 2.496	27.199	28.512	3.335	
	R123					2.496 2.496			5.555 2.679	30.876
l6.		R600a	3.963	4.365	0.082		28.509	29.383		31.4
17.		R410A	3.547	3.874	0.084	2.496	25.554	26.364	3.006	27.976
8.		R1234ze(E)	3.842	4.221	0.083	2.496	27.64	29.271	2.752	30.368
19.		R1234ze(Z)	4.152	4.591	0.082	2.496	29.871	28.895	2.592	33.028
20.		R1234yf	3.642	3.985	0.084	2.496	26.203	29.372	2.874	28.672
21.		R22	3.905	4.298	0.083	2.503	28.09	27.018	2.77	30.917
22.		R134a	3.866	4.252	0.083	2.503	27.812	28.85	2.752	30.587
23.		R290	3.787	4.158	0.083	2.503	27.246	28.839	2.8	29.917
24.		R404A	3.35	3.646	0.086	2.503	24.469	28.453	3.112	26.659
25.	R141b	R407C	3.229	3.506	0.086	2.503	27.199	24.63	3.335	30.889
26.		R600a	3.963	4.367	0.083	2.503	28.509	29.529	2.678	31.416
27.		R410A	3.547	3.875	0.085	2.503	25.554	26.496	3.005	27.989
28.		R1234ze(E)	3.842	4.223	0.083	2.503	27.64	29.417	2.751	30.383
29.		R1234ze(Z)	4.152	4.593	0.082	2.503	29.871	29.039	2.591	33.045
30.		R1234yf	3.642	3.987	0.084	2.503	26.203	29.519	2.874	28.686
31.		R22	3.905	4.277	0.083	2.508	28.09	25.748	2.784	30.772
32.		R134a	3.866	4.232	0.083	2.508	27.812	27.494	2.766	30.444
33.		R290	3.787	4.139	0.084	2.508	27.246	27.484	2.813	29.779
34.		R404A	3.35	3.631	0.086	2.508	24.469	27.116	3.126	26.549
35.	R227ea	R407C	3.229	3.492	0.087	2.508	27.199	23.472	3.349	30.765
36.	122700	R600a	3.963	4.346	0.083	2.508	28.509	28.142	2.692	31.266
37.		R410A	3.547	3.859	0.085	2.508	25.554	25.251	3.019	27.868
38.		R1234ze(E)	3.842	4.204	0.083	2.508	27.64	28.034	2.765	30.242
39.		R1234ze(Z)	4.152	4.57	0.082	2.508	29.871	27.674	2.604	32.881
40.		R1234yf	3.642	3.97	0.084	2.508	26.203	28.131	2.887	28.559
41.		R22	3.905	4.292	0.083	2.491	28.09	26.695	2.772	30.88
42.		R134a	3.866	4.247	0.083	2.491	27.812	28.505	2.754	30.551
13.		R290	3.787	4.154	0.083	2.491	27.246	28.494	2.802	29.882
14.		R404A	3.35	3.642	0.085	2.491	24.469	28.113	3.114	26.631
15.	D2456-	R407C	3.229	3.502	0.086	2.491	27.199	24.335	3.337	30.858
46.	R245fa	R600a	3.963	4.361	0.082	2.491	28.509	29.176	2.68	31.378
17.		R410A	3.547	3.871	0.084	2.491	25.554	26.179	3.007	27.958
48.		R1234ze(E)	3.842	4.218	0.083	2.491	27.64	29.065	2.753	30.347
		R1234ze(Z)	4.152	4.587	0.082	2.491	29.871	28.692	2.593	33.003
19.										

Sr.	ORC	VCRS	COP <sub>vcrs</sub>	COP <sub>net</sub>	W <sub>turb</sub>	$\acute{\eta}_{th,orc}$	$\acute{\eta}_{ex,vcrs}$	ή <sub>ex,orc</sub>	Ex <sub>net</sub>	ή <sub>ex,net</sub>
No.	working fluid	working fluid								
51.		R22	3.905	4.284	0.084	2.53	28.09	26.182	2.782	30.821
52.		R134a	3.866	4.238	0.084	2.53	27.812	27.957	2.764	30.493
53.		R290	3.787	4.146	0.084	2.53	27.246	27.947	2.811	29.826
54.		R404A	3.35	3.636	0.087	2.53	24.469	27.572	3.124	26.586
55.	$D_{1024-4}(E)$	R407C	3.229	3.496	0.087	2.53	27.199	23.868	3.347	30.807
56.	R1234ze(E)	R600a	3.963	4.353	0.084	2.53	28.509	28.615	2.69	31.317
57.		R410A	3.547	3.864	0.086	2.53	25.554	25.676	3.017	27.909
58.		R1234ze(E)	3.842	4.21	0.084	2.53	27.64	28.506	2.763	30.29
59.		R1234ze(Z)	4.152	4.578	0.083	2.53	29.871	28.14	2.603	32.937
60.		R1234yf	3.642	3.976	0.085	2.53	26.203	28.605	2.885	28.602
61.		R22	3.905	4.283	0.084	2.549	28.09	26.12	2.784	30.814
62.		R134a	3.866	4.238	0.085	2.549	27.812	27.891	2.766	30.486
63.		R290	3.787	4.145	0.085	2.549	27.246	27.88	2.814	29.819
64.		R404A	3.35	3.635	0.087	2.549	24.469	27.507	3.127	26.581
65.	D124-	R407C	3.229	3.496	0.088	2.549	27.199	23.811	3.349	30.801
66.	R134a	R600a	3.963	4.352	0.084	2.549	28.509	28.547	2.693	31.31
67.		R410A	3.547	3.864	0.086	2.549	25.554	25.615	3.019	27.903
68.		R1234ze(E)	3.842	4.209	0.085	2.549	27.64	28.439	2.765	30.283
69.		R1234ze(Z)	4.152	4.577	0.083	2.549	29.871	28.073	2.605	32.929
70.		R1234yf	3.642	3.975	0.086	2.549	26.203	28.537	2.888	28.596
71.		R22	3.905	4.293	0.082	2.485	28.09	26.761	2.771	30.888
72.		R134a	3.866	4.248	0.082	2.485	27.812	28.576	2.753	30.558
73.		R290	3.787	4.155	0.083	2.485	27.246	28.565	2.8	29.889
74.		R404A	3.35	3.643	0.085	2.485	24.469	28.182	3.113	26.636
75.	R245ca	R407C	3.229	3.503	0.086	2.485	27.199	24.396	3.335	30.864
76.	K245Ca	R600a	3.963	4.363	0.082	2.485	28.509	29.248	2.679	31.385
77.		R410A	3.547	3.872	0.084	2.485	25.554	26.244	3.006	27.964
78.		R1234ze(E)	3.842	4.219	0.083	2.485	27.64	29.137	2.752	30.355
79.		R1234ze(Z)	4.152	4.589	0.081	2.485	29.871	28.763	2.592	33.012
80.		R1234yf	3.642	3.984	0.084	2.485	26.203	29.238	2.874	28.66
81.		R22	3.905	4.287	0.083	2.519	28.09	26.366	2.779	30.842
82.		R134a	3.866	4.241	0.084	2.519	27.812	28.154	2.761	30.513
83.		R290	3.787	4.149	0.084	2.519	27.246	28.143	2.808	29.846
84.		R404A	3.35	3.638	0.086	2.519	24.469	27.766	3.121	26.602
85.	R124	R407C	3.229	3.498	0.087	2.519	27.199	24.035	3.343	30.825
86.		R600a	3.963	4.356	0.083	2.519	28.509	28.816	2.687	31.339
87.		R410A	3.547	3.867	0.085	2.519	25.554	25.856	3.013	27.926
88.		R1234ze(E)	3.842	4.213	0.084	2.519	27.64	28.706	2.76	30.311
89.		R1234ze(Z)	4.152	4.582	0.082	2.519	29.871	28.338	2.599	32.961
90.		R1234yf	3.642	3.978	0.085	2.519	26.203	28.806	2.882	28.62
91.		R22	3.905	4.289	0.084	2.546	28.09	26.463	2.78	30.854
92.		R134a	3.866	4.243	0.085	2.546	27.812	28.258	2.762	30.524
93.		R290	3.787	4.15	0.085	2.546	27.246	28.247	2.81	29.856
94.		R404A	3.35	3.639	0.087	2.546	24.469	27.869	3.123	26.61
95.	R152a	R407C	3.229	3.5	0.088	2.546	27.199	24.124	3.345	30.835
96.		R600a	3.963	4.358	0.084	2.546	28.509	28.923	2.689	31.35
97.		R410A	3.547	3.868	0.086	2.546	25.554	25.952	3.015	27.936
98.		R1234ze(E)	3.842	4.215	0.085	2.546	27.64	28.812	2.762	30.321
99.		R1234ze(Z)	4.152	4.583	0.083	2.546	29.871	28.442	2.601	32.973
100.		R1234yf	3.642	3.98	0.086	2.546	26.203	28.912	2.884	28.63

Sr.	ORC	VCRS	COP <sub>vcrs</sub>	COP <sub>net</sub>	W <sub>turb</sub>	$\dot{\eta}_{\text{th,orc}}$	$\eta_{ex,vcrs}$	$\dot{\eta}_{ex,orc}$	Exnet	ή <sub>ex,net</sub>
No.	working fluid		• ••••	net	uit	latore	100,000	penjore	-1101	leating
101.		R22	3.905	4.25	0.085	2.558	28.09	24.038	2.805	30.578
102.		R134a	3.866	4.205	0.085	2.558	27.812	25.668	2.787	30.254
103.		R290	3.787	4.114	0.085	2.558	27.246	25.658	2.835	29.597
104.		R404A	3.35	3.611	0.088	2.558	24.469	25.314	3.148	26.402
105.	R404A	R407C	3.229	3.473	0.088	2.558	27.199	21.913	3.371	30.6
106.	K404A	R600a	3.963	4.318	0.085	2.558	28.509	26.272	2.713	31.066
107.		R410A	3.547	3.836	0.087	2.558	25.554	23.573	3.04	27.706
108.		R1234ze(E)	3.842	4.178	0.085	2.558	27.64	26.172	2.786	30.054
109.		R1234ze(Z)	4.152	4.54	0.084	2.558	29.871	25.836	2.626	32.662
110.		R1234yf	3.642	3.946	0.086	2.558	26.203	26.262	2.909	28.389
111.		R22	3.905	4.262	0.088	2.643	28.09	24.806	2.807	30.665
112.		R134a	3.866	4.217	0.088	2.643	27.812	26.488	2.789	30.339
113.		R290	3.787	4.125	0.088	2.643	27.246	26.478	2.837	29.678
114.		R404A	3.35	3.62	0.091	2.643	24.469	26.124	3.15	26.467
115.	R410A	R407C	3.229	3.481	0.091	2.643	27.199	22.613	3.373	30.674
116.		R600a	3.963	4.331	0.087	2.643	28.509	27.112	2.715	31.156
117.		R410A	3.547	3.846	0.089	2.643	25.554	24.327	3.042	27.778
118.		R1234ze(E)	3.842	4.189	0.088	2.643	27.64	27.008	2.788	30.138
119.		R1234ze(Z)	4.152	4.554	0.087	2.643	29.871	26.661	2.628	32.76
120.		R1234yf	3.642	3.957	0.089	2.643	26.203	27.102	2.911	28.465
121.		R22	3.905	4.276	0.088	2.648	28.09	25.692	2.799	30.765
122.		R134a	3.866	4.231	0.088	2.648	27.812	27.434	2.781	30.438
123.		R290	3.787	4.138	0.088	2.648	27.246	27.424	2.829	29.773
124.		R404A	3.35	3.63	0.091	2.648	24.469	27.056	3.142	26.544
125.	R32	R407C	3.229	3.491	0.091	2.648	27.199	23.421	3.365	30.76
126.		R600a	3.963	4.345	0.087	2.648	28.509	28.08	2.707	31.259
127.		R410A	3.547	3.858	0.09	2.648	25.554	25.195	3.034	27.862
128.		R1234ze(E)	3.842	4.203	0.088	2.648	27.64	27.973	2.78	30.236
129.		R1234ze(Z)	4.152	4.569	0.087	2.648	29.871	27.614	2.62	32.874
130.		R1234yf	3.642	3.969	0.089	2.648	26.203	28.07	2.903	28.553
131.		R22	3.905	4.294	0.083	2.496	28.09	26.791	2.772	30.891
132.		R134a	3.866	4.248	0.083	2.496	27.812	28.608	2.754	30.561
133.		R290	3.787	4.155	0.083	2.496	27.246	28.597	2.801	29.892
134.		R404A	3.35	3.643	0.085	2.496	24.469	28.214	3.114	26.639
135.	R1233zd(E)	R407C	3.229	3.503	0.086	2.496	27.199	24.423	3.336	30.867
136.		R600a	3.963	4.363	0.082	2.496	28.509	29.281	2.68	31.389
137.		R410A	3.547	3.872	0.084	2.496	25.554	26.273	3.007	27.967
138.		R1234ze(E)	3.842	4.22	0.083	2.496	27.64	29.17	2.753	30.358
139.		R1234ze(Z)	4.152	4.589	0.082	2.496	29.871	28.795	2.593	33.016
140.		R1234yf	3.642	3.984	0.084	2.496	26.203	29.271	2.875	28.663
141.		R22	3.905	4.277	0.085	2.556	28.09	25.759	2.789	30.773
142.		R134a	3.866	4.232	0.085	2.556	27.812	27.506	2.771	30.445
143.		R290	3.787	4.139	0.085	2.556	27.246	27.495	2.818	29.781
144.		R404A	3.35	3.631	0.088	2.556	24.469	27.127	3.131	26.549
145.	R1234yf	R407C	3.229	3.492	0.088	2.556	27.199	23.482	3.354	30.766
146.	-	R600a	3.963	4.346	0.084	2.556	28.509	28.153	2.697	31.267
147.		R410A	3.547	3.859	0.086	2.556	25.554	25.261	3.024	27.869
148.		R1234ze(E)	3.842	4.204	0.085	2.556	27.64	28.045	2.77	30.243
149.		R1234ze(Z)	4.152	4.571	0.084	2.556	29.871	27.685	2.61	32.882
150.		R1234yf	3.642	3.97	0.086	2.556	26.203	28.143	2.893	28.56
		The best combination	on of refrigerar	it pair		Maximum	value	M	inimum val	le



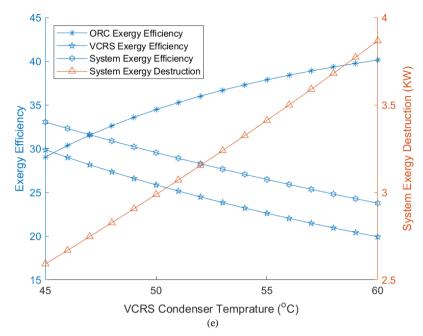


Fig. 6. (a) Variation of net COP and turbine work against the VCRS evaporator temperature (b) Variation of net COP and turbine work against the VCRS condenser temperature (c) Variation of system exergy destruction and turbine work against load on VCRS (d) Variation of exergy efficiencies of ORC, VCRS and combined system, and system exergy destruction against the VCRS evaporator temperature (e) Variation of exergy efficiencies of ORC, VCRS and combined system and system exergy destruction against the VCRS condenser temperature

364 The performance of the integrated VCRS-ORC system is further assessed using various performance 365 curves among the calculated parameters. Fig. 6 (a) and (b) show the variation for net COP, and turbine 366 work against the VCRS evaporator and condenser temperatures respectively. It is found that, with an increase in the VCRS evaporator temperature (varied from 0 °C to 10 °C), the net COP increases (from 367 368 3.9 to 5.5); the turbine work output of the ORC however decreases (from 84.5 W to 78.5 W) correspondingly. This is because net COP is proportional to the evaporator temperature and consequently 369 it increases. Although, the net COP of the system decreases due to an increase in the VCRS evaporator 370 temperature; thus, and the overall power reduces. Assuming the pump work constant, this causes a 371 decrease in the turbine work power output. Exactly reverse trend/variation is shown by the net COP and 372 ORC turbine work output for an increase in condenser temperature. With an increase in the VCRS 373 condenser temperature (varied from 0  $^{\circ}$ C to 10  $^{\circ}$ C), the net COP decreases (from 4.6 to 3.3); the turbine 374 work output of the ORC however increases (from 80 W to 220 W) correspondingly. 375

Fig. 6 (c) depicts the variation of the system exergy destruction and the ORC turbine work output by varying the load on the refrigeration system. The results reveal that, as the load increases (from 1 TR to 10 TR), both – the system exergy destruction and the ORC turbine work output linearly increase from 2.5 kW to 25 kW, and 0.1 kW to 0.8 kW respectively. At higher loads, higher entropy generation takes place inside the system, which reduces the exergy potential due to the exergy destruction. However, the net COP of the system increases correspondingly, which causes an increase in the ORC turbine work output. 382 Fig. 6 (d) shows the variation of exergy efficiencies of ORC, VCRS, and combined system, and system 383 exergy destruction against the VCRS evaporator temperature. For an increase in the VCRS evaporator temperature (from 0 °C to 10 °C), the VCRS, ORC, and overall integrated system exergy efficiencies 384 385 decrease (from 32.5 % to 26 %), increases slightly (from 28.8 % to 29.2 %), and decreases (from 35.5 % to 29.5 %) respectively; the overall system exergy destruction, however, decreases from 3 kW to 2.2 kW. 386 387 The VCRS exergy efficiency is indirectly proportional to the VCRS evaporator temperature. However, it does not affect to a considerable extent the ORC exergy efficiency and thus it shows a slight increase 388 389 only. The system exergy efficiency is the combined effect of the above-two efficiencies.

390 Fig. 6 (e) shows the variation of the above-mentioned performance parameters with the variation in VCRS condenser temperature. For an increase in the VCRS condenser temperature (from 45 °C to 60 °C). 391 392 the VCRS, ORC, and overall integrated system exergy efficiencies decrease (from 30 % to 22.5 %). increase (from 29 % to 39 %), and decreases (from 33 % to 26 %) respectively; the overall system exergy 393 destruction, however, increases from 2.6 kW to 3.85 kW. Higher the condenser temperature, the higher 394 the system exergy destruction takes place. Similar to the earlier curves, the VCRS exergy efficiency is 395 396 indirectly proportional to the VCRS condenser temperature. However, it varies directly with the ORC 397 exergy efficiency and thus it shows a linear increase. The system exergy efficiency is the combined effect 398 of the above-two efficiencies.

## 399 **5. Conclusion**

This research work has performed a dynamic simulation and performance assessment of an integrated vapor compression refrigeration system - Organic Rankine Cycle system using the exergy analysis as a tool. The system taken under consideration includes a refrigeration system installed to produce a cooling effect and preserve the raisins. The waste heat rejected from the condenser of a refrigeration plant is utilized as a potential/input to drive an organic Rankine Cycle system. This system which is supposed to be installed is modeled for the corresponding input and operating conditions. The conclusion of the paper can be summarized as follows:

From the performance graphs, with an increase in the VCRS evaporator temperature (varied from 0 °C to 10 °C), the net COP increases (from 3.9 to 5.5), and the W<sub>turb</sub> decreases (from 84.5 W to 78.5 W) correspondingly. However, a reverse trend is shown by the net COP and ORC turbine work output for an increase in the condenser temperature. With an increase in the VCRS condenser temperature (varied from 0 °C to 10 °C), the net COP decreases (from 4.6 to 3.3), and the W<sub>turb</sub> 412 increases (from 80 W to 220 W) correspondingly.

- As the load on the VCRS increases (from 1 TR to 10 TR), both Ex<sub>net</sub> and the W<sub>turb</sub> linearly
   increase from 2.5 kW to 25 kW, and 0.1 kW to 0.8 kW respectively.
- For an increase in the VCRS evaporator temperature (from 0 °C to 10 °C), the exergy efficiencies for VCRS, ORC, and overall integrated system decrease (from 32.5 % to 26 %), increase slightly (from 28.8 % to 29.2 %), decreases (from 35.5 % to 29.5 %) respectively, and the Ex<sub>net</sub> decreases from 3 kW to 2.2 kW.
- For an increase in the VCRS condenser temperature (from 45 °C to 60 °C), the exergy efficiencies for VCRS, ORC, and overall integrated system decrease (from 30 % to 22.5 %), increases (from 29 % to 39 %), decrease (from 33 % to 26 %) respectively, and the overall Ex<sub>net</sub> increases from 2.6 kW to 3.85 kW.

Hence, from this research work, it can be concluded that, for performing the waste heat recovery utilizing the heat rejected from the condenser of a VCRS (operating range 60 °C to 70 °C), R141b– R1234ze(Z)stands as the optimum ORC-VCRS refrigerant pair, considering the thermal performance and exergy assessment.

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## 432 **Conflict of Interest**

433 The authors declare no conflict of interest.

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## 439 Nomenclature

Abbreviation	15	Greek Symbols				
ORC	Organic Rankine Cycle	η	efficiency			
VCRS	Vapor Compression Refrigeration System	€	effectiveness			
TR	Tons of Refrigeration	Subscrip	ots			
COP	Coefficient of Performance	cri	critical			
AIDA	Analysis, Implement, Design, and Assessment	vcrs	vapor compression refrigeration system			
GWP	Global Warming Potential	th	thermal			
CFC	Chlorofluorocarbon	ex,vcrs	exergy of vapor compression refrigeration system			
HCFC	Hydrochlorofluorocarbon	ex,orc	exergy of organic Rankine cycle			
EU	European Union	net	net/overall			
HVAC	Heating, Ventilation, and Air Conditioning	ex,net	net exergy efficiency			
TEWI	Total Equivalent Warming Index	ex	exergy			
EES	Engineering Equation Solver	0	dead state			
ASHRAE	American Society for Heating, Ventilation, and Air Conditioning	in	inlet			
IHX	Internal Heat Exchanger	i	i <sup>th</sup> state			
ODP	Ozone Depletion Potential	out	outlet			
HFO	Hydrofluoroolefin	D	destruction			
MATLAB	Matrix Laboratory	S	saturated			
TEQIP	Technical Education Quality Improvement Programme	comp	compressor			
AICTE	All India Council for Technical Education	cond	condenser			
NDF	National Doctoral Fellowship	evap	evaporator			
Parameters		is	isentropic			
Р	Pressure	th	thermal			
h	Enthalpy	turb	turbine			
Т	Temperature	exp	expansion device			
S	Entropy					
Q	Heat					
W	Work					
m	Mass flow rate					
Ex	Exergy					

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