# Effect of phase transformation on mechanical properties of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> high entropy alloy coatings processed by laser cladding

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## **Abstract:**

Eight Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> high entropy alloy (HEA) coatings were fabricated by laser cladding with different laser scanning speed. The mechanical properties caused by phase transformation and microstructure evolution of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings were investigated. The experimental results showed that the volume fraction of the FCC phase and the BCC phase in each coating were different depending on laser scanning speed. High laser scanning speed will increase the number of BCC phase. BCC phase was composed of alternate A2 and B2 structure formed by spinodal decomposition. All coatings exhibited a polycrystalline structure composed of uniform equiaxed grains. The grain size of equiaxed grains reduced from 276µm to 42µm with increased laser scanning speed. The phase-mechanical properties connection at nano-scale were established by mapping and elastic modulus mapping. nano-hardness Furthermore, the micromechanical properties of individual FCC phase and BCC phase were studied by analyze nanoindentation data statistically. The BCC phases were found to have a

higher nano-hardness and elastic modulus than the FCC phases. Therefore, the increment of the BCC phase could significantly enhance the strength and wear resistance of coatings. The optimum wear resistance was obtained in V17, owning a higher volume fraction of BCC (>90%) and finer grains (75 $\mu$ m). In addition, the strengthening mechanism has been discussed. Grain boundary strengthening makes a great contribution to the excellent performance of coatings.

**KEYWORDS:** high entropy alloys; laser cladding; Nanoindentation; Mechanical properties

#### Introduction

High-entropy alloys (HEAs) has received widespread attention since it was first proposed by Yen et al. in 2004 [1]. It was defined as an alloy that included five or more elements with an equimolar or near equimolar ratio, and the content of each element was between 5% and 35% [2]. For phase composition, the presence of the high entropy effect and the sluggish diffusion effect promote the formation of single-phase solid solution and nano-sized precipitates rather than brittle phases or other intermetallic compounds in traditional metallurgical theory [3–5]. Numerous studies have implied that HEAs often have excellent mechanical, physical, and chemical properties, such as high-temperature wear resistance [6], high strength and ductility [7] and excellent corrosion resistance [8].

Al-Co-Cr-Fe-Ni system high-entropy alloy system has been mostly studied due to excellent comprehensive property, low cost and huge potential in future applications. The phases usually formed in this system include disordered solid solution FCC phase, disordered solid solution BCC phase (A2) and ordered solid solution BCC phase (B2) [9-14]. According to previous research reports, the mechanical properties of Al-Co-Cr-Fe-Ni HEAs were found strongly depend on the volume fraction and morphology of FCC, BCC and B2 phase. Zhang et al. proposed that the strength of HEAs was related to A2/B2 morphology, AlCoCrFe<sub>2.5</sub>Ni coating with B2 structure made by A2 matrix and B2 precipitates showed higher strength and better wear resistance than that with B2 matrix and A2 precipitates [15]. The volume fraction of the constituent phases of the Al<sub>x</sub>CoCrFeNi HEAs were measured, the deformation mechanism causing the difference in mechanical properties varied according to the volume fraction of BCC. [13]. Furthermore, Al<sub>1.8</sub> alloy and Al<sub>2.0</sub> alloy with smaller feature size of the spinodal structure exhibited increased hardness compared with Al<sub>1.2</sub> alloy and Al<sub>1.5</sub> alloy [14]. Therefore, the phase and microstructure are desired to be adjusted to obtain alloys with outstanding comprehensive properties. Zhang et al. added Cr element (a BCC stabilizer) to single FCC-structured  $Al_7Co_{24}Cr_{21}Fe_{24}Ni_{24}$  to formed BCC phase,  $(Al_7Co_{24}Cr_{21}Fe_{24}Ni_{24})_{74}Cr_{26}$  alloy displyed the best performance, having high yield strength (1649 MPa) and fracture strength (2830 MPa) and good compressive plastic strain of 24.9% [16]. By adjusting molar ratios of Al: Co of Al<sub>x</sub>CrCo<sub>2-x</sub>FeNi HEA, the phase switched from FCC +  $\sigma$  + disordered BCC + ordered BCC structures to  $\sigma$  + disordered BCC + ordered BCC structures, and then to disordered BCC + ordered BCC structures, and then Al<sub>1.6</sub>CrCo<sub>0.4</sub>FeNi alloy with high wear resistance and Al<sub>1.4</sub>CrCo<sub>0.6</sub>FeNi alloy with combinations of strength and ductility were manufactured respectively[17]. Qin et al. added Co element to the AlCoCrFeNi, it was found that the addition of Co element increased the volume fraction of the FCC phase having more slip directions and thus improved the plasticity of AlCoCrFeNi [18]. In addition to adding constituent elements of AlCoCrFeNi alloy, such as Co and Cr mentioned above. It can also add new elements to the base AlCoCrFeNi alloy to adjust the phase and microstructure. Liu et al. [19] added Si to AlCoCrFeNi, six AlCoCrFeNiSix laser cladding coatings were synthesized, x = 0, 0.1, 0.2, 0.3, 0.4 and 0.5 (x: molar ratio), the addition of Si atoms affects the microstructure of the coating and causes the dislocation density to increase. They found that the microhardness and dislocations density increased with Si content linearly. Liu et al. [20] also added Ti element to AlCoCrFeNi high-entropy alloy. The in-situ TiC particles generated in the coating have high stability at elevated temperatures, so that AlCoCrFeNiTi0.8 HEA coatings still have good wear resistance at high temperatures of 1200°C. The above researches are all adding the alloying element into the base alloy to change the evolution of phase and microstructure. However, there is another way to modify the phase and microstructure, that is, to change the solidification behavior by changing the process parameters. Laser cladding is known as a typical non-equilibrium process because of its rapid heating and cooling rates, which will lead to a significant difference in solidification behavior [21,22]. And it has been determined that laser cladding coating can effectively improve the performance of metal substrates [23]. Therefore, laser cladding method was chosen to prepare high-entropy alloy coatings in this paper. In the current work, eight coatings with different volume fractions and morphologies of FCC and BCC phase were manufactured by the laser cladding. And the mechanical properties of the different

phases and the influence of the phase transformation on the mechanical properties of coatings were systematically investigated.

#### 2. Experimental procedures

#### 2.1Preparation of the coatings

The Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings were prepared by laser cladding. Commercial HEA powder containing 16.8% Al, 20.74% Co, 20.49%Cr, 21.28% Fe and 20.70%Ni (at. %) with high purity (>99.99%) were used as raw materials, and the particle sizes of the powders were in the range of 15-53 $\mu$ m. The HEA powders were preset onto the surface of a Q235 steel substrate with dimensions of 50 mm×50 mm×10 mm, and the thickness of the powder layer was 1mm. Before presetting powder, the surface of each substrate was mechanically polished and cleaned with alcohol to remove all oil and impurities.

Laser cladding was carried out by IPG YLS-5000 fiber system with a protective gas system. Eight HEA laser cladding coatings were prepared in an argon atmosphere with a flow rate of 15 L/min. The laser scanning speed was 7mm/s, 9mm/s, 11mm/s, 13mm/s, 15mm/s, 17mm/s, 19mm/s and 21mm/s for eight HEA coatings, thus, named as V7, V9, V11, V13, V15, V17, V19 and V21 in turn. The laser power was selected as 3000W. The laser cladding parameters and the chemical composition of the HEA powders were shown in Table 1.

Coating	Powder composition(at%)					Laser	Laser Scanning	Thickness
	Al	Co	Cr	Fe	Ni	(W)	Speed (mm/s)	(mm)
V7						3000	7	1
V9						3000	9	1
V11						3000	11	1
V13	16.90	5.80 20.74 2	20.49	21.28	20.70	3000	13	1
V15	10.80					3000	15	1
V17						3000	17	1
V19						3000	19	1
V21						3000	21	1

Table 1 The processing parameters of laser cladding.

#### 2.2Characterization of the coatings

X-ray diffractometry (XRD) (D/max-RB, Rigaku Corp, Japan) with Cu K $\alpha$  irradiation ( $\lambda$ =0.154060 nm) at a scanning speed of 4°/min was used to identify the phases of the coatings. The cross-section of the coating was ground, polished, cleaned, and then etched to observe the microstructure. The microstructure of the coatings was observed by optical microscopy (OM, VHX-5000) and scanning electron microscopy (SEM, TESCAN VEGA 3) and transition electron microscope (TEM, FEI TECNAI G2 S-TWIN F20) with an EDS detector and high-resolution camera. TEM samples are prepared by the focused ion beam (FIB) technique (FIB, FEI Helios 600).

The middle area of the coating cross-section, an area of  $72 \times 72 \mu m$ , was pressed into the indenter at the maximum load of 10mN using the nanoindentation method. Berkovich indenter with a tip radius of 100 nm was used. The holding time under peak load was 10 s. 100 indentations were made on each coating. Keysight Nano Indenter G200 was employed in the current study to perform the nanoindentation test. The distance between the two indentations should be 20-30 times the indentation depth. To ensure the accuracy of the measured value, the distance between the indentations was set to  $8\mu m$ , forming a  $10 \times 10$  indentation matrix (100 indentations), as shown in Fig. 1.

The friction and wear properties of the coatings were tested by the MFT-5000 multifunctional friction and wear tester (USA, RETC) with the test parameters: the load 40 N, the friction and wear radius 3 mm, the rotation velocity 100 r/min, the friction time 60 min and the test temperature RT (room temperature) and 800 °C respectively.



						-	-	-	-	-		-
	72	ŀ				*		٠		*		
	64	ŀ	٠		۸	۸	۸		۸		۸	
•	56	ŀ		٠								
	48	ŀ								۸		۸
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	8	ŀ	۸						۸	٠		
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		L	1		1					-		-
			0	8	16	24	32	40	48	56	64	72

Fig. 1. Schematic diagram of  $10 \times 10$  matrix indentations.

#### **3 Results**

#### **3.1 Phases analysis**

The XRD patterns of raw powder and eight Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings processed by different laser scanning speed were shown in Fig. 2(a). Two solid solution phases, face-centered cubic (FCC) solid solution and body-centered cubic (BCC) solid solution were identified from all XRD patterns. When the laser scanning speed was not more than 9 mm/s (i.e. V7 and V9), the relative intensity of the FCC reflection peak was much stronger than the intensity of the BCC reflection peak. However, as the laser scanning speed increased, the relative intensity of the FCC diffraction peak gradually decreased, and it almost vanished in V21. When the laser scanning speed was larger than 11mm/s, the XRD patterns began to exhibit a strong BCC diffraction peak and a weaker FCC diffraction peak. The intensity evolution of the corresponding peaks of the FCC phase and BCC is related to the volume fraction of these two phases, which will be discussed below. Noticeably, a weak diffraction peak near 30 degrees 20 angle appeared in all coatings and raw powder, which corresponded to ordered BCC structure (B2 superstructure). It indicated that the BCC structure in the coatings may consist of disordered BCC phases (A2) and ordered BCC phases (B2) or single ordered BCC phases (B2) [14]. The following research will give a detailed introduction.

The diffraction patterns between 42 and 46 degrees were shown in Fig. 2(b). Compared with the diffraction pattern of raw powder, the  $(111)_{FCC}$  diffraction peak in coatings moved to a lower 2 $\theta$  angle. The possible reason was that more elements entering the basic close-packed FCC structure, causing the lattice to expand [24]. In V7-V13, the  $(110)_{BCC}$  diffraction peak of the coatings moved to a higher 2 $\theta$  angle than the raw powder. Moreover, the shift angle of  $(110)_{BCC}$  diffraction peak increases as the scanning speed increases, and the shift angle in V13 reaches the maximum value, which is 0.073° higher than the raw powder, as shown in Fig. 3. In V15 and V19, the shift angle of the  $(110)_{BCC}$  diffraction peak is significantly smaller than that of V13,

which is 0.009° and 0.005° higher than that of the raw powder, respectively. In addition, in V17 and V21, the (110) BCC peak shifted to a lower 2θ angle than the raw powder. It showed that the shift of the (110)<sub>BCC</sub> peak was very complicated in the current coatings, which may be related to the complex lattice distortion in the HEA. The crystal lattice of a HEA is not a perfect solid solution lattice. Different from traditional alloys, there is no clear difference between solvent elements and solute elements in HEAs [25]. A possible schematic diagram of the distorted lattice of the Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings was shown in Fig. 3. The balls of different colors represent the atoms of the five elements Fe, Co, Cr, Ni and Al randomly distributed in the lattice sites. Al atoms with the largest atomic radius may cause lattice expansion.





Fig. 2. XRD analyses of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings with different laser scanning speed: (a) The XRD patterns of the V7-V21 and powder; (b) The diffraction patterns between 42 and 46 degrees.

#### 3.2. Microstructural evolution

Fig. 4 showed the microstructure of V7, V9, V11, V13, V15, V17, V19 and V21. All coatings exhibited polycrystalline structures composed of equiaxed grains [26]. Two precipitation phases, white nanoprecipitates located in the interior and grain boundaries of the equiaxed matrix grains and Widmanstätten side plate originating from grain boundaries, were formed in the V7 and V9 (see Fig .4(a) and (b)). Then, after the laser scanning speed increased to more than 9mm/s, there were only the white nanoprecipitates formed in V11-V21. The Widmanstätten side plate structures found in V7 and V9 were FCC phases, which have also been reported in other literature [14,27]. Moreover, Widmanstätten side plates of FCC structure in V9 were finer than V7. It can be seen that the laser scanning speed showed a significant effect on the microstructure. The equiaxed grains size of all coatings were shown in Fig. 5. They were reduced from 276µm to 42µm, indicating that increased laser scanning speed had a positive effect on grain refinement when its values varied in the range of 7mm/s-21mm/s. Optical micrographs from the middle parts of the V7, V13, V17 and

V21 were shown in Fig. 6, V21 with the maximum laser scanning speed exhibited the most refined grains.



Fig. 4. SEM micrographs of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings with different laser scanning speed: (a) and (b) are V7 and V9 with nanoprecipitates and Widmanstätten side plate; (c)-(h) are V11-V21 with nanoprecipitates only.



Fig. 5. The equiaxed grain size of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings at different laser scanning speed.



Fig. 6. OM micrographs of middle parts of coatings: (a) V7, (b) V13, (c) V17 and (d) V21.

In order to further reveal the microstructure of the coatings in detail, high-magnification secondary electron images of V7, V13 and V17 were shown in Fig. 7. It showed that the equiaxed matrix grains consisted of nano-scale two alternating bright and dark phases, formed by spinodal decomposition [13, 28–30]. And V7 showed a finer decomposition than V13 and V17, this may be attributed to sufficient solidification time in V7. For further identification, V17 was analyzed by TEM. The SADPs analysis for Region A, C, and D of the equiaxed matrix grains demonstrated that it had a B2 superlattice structure, as shown in Fig. 8. Fig. 9 showed a TEM

dark-field (DF) image, the corresponding high-resolution TEM (HRTEM) images and Fast Fourier Transform (FFT) patterns of V17. The microstructure in Fig. 9(a) was composed of alternate bright and dark two phases, the same as in Fig. 7(f). HRTEM image and the corresponding Fast Fourier Transform (FFT) patterns identified that the dark phase was disordered BCC phase (A2) and the bright phase was ordered BCC phase (B2). More precisely, the volume fractions of A2 and B2 were identified as 55% and 45%, respectively. This A2/B2 morphology with A2 particles distributed in the B2 matrix has been previously observed by Zhang et al. [15].

Furthermore, as shown in Fig. 9(b) and (c), the boundary between A2 and B2 were separated by a yellow dashed line. It can be seen that the A2 and B2 phases were rather co-continuous. The coexistence of A2 and B2 will simultaneously improve the plasticity and strength of the coating. From the Inverse Fast Fourier Transform (IFFT) results in the insets in Fig. 9(d) and (e), it can be seen that the intensity of two adjacent atoms along the [110] direction of the B2 structure was different. The reason is that atoms in different sublattices of the B2 structure occupy specific positions, which results in different average atomic scattering factors at the positions of the crystal lattice, and therefore, leads to different observed intensities. However, for the A2 structure, the intensity was the same, because the atoms of each element in the disordered solid solution have the same probability of occupying each lattice site [31].



Fig. 7. (a), (b) and (c): High magnification scanning electron micrograph of V7, V13 and V17, respectively; (d), (e)and (f): enlarged view of yellow box in Fig. 7(a), (b) and (c), respectively.

Point	Crystal structure	Al	Со	Cr	Fe	Ni
E	BCC, matrix	15.1	13.96	16.91	41.07	12.94
F	FCC, nanoprecipitates	0.3	5.51	70.1	21.92	2.14
G	BCC, matrix	15.32	14.74	19.02	40.44	10.45
Н	BCC, matrix	14.66	14.83	19.87	39.89	10.72

Table 2 TEM-EDS results of V17 (at%).



Fig. 8. SADPs of the matrix phase and nanoprecipitates phase in V17: (a) The TEM morphology of V17 near the grain boundary (Bright field image); (b), (c), (d), (e) and (f) are SADPs of region A, region B, region C and region D, respectively.



Fig. 9. HRTEM image of the matrix phase of V17: (a): Darkfield image taken from the  $(100)_{B2}$  reflection marked by a red circle in the SADP (inset); (b): HRTEM image along the <001> direction showing the interface (yellow dashed line) between A2 and B2; (c): Enlarged image of the blue box in Fig.9(b); (d) and (e): FFT patterns from Fig. 9(c).

To further identify nanoprecipitates observed in Fig. 4 and Fig. 7. The selected area electron diffraction at the region included nanoprecipitates were performed, and this region was marked as B in Fig. 8(a). In the SADP of Region B, two sets of patterns were found, and then they were calibrated separately. The results were shown in Fig.8(e) and (f), they were FCC structure and B2 superlattice lattice structure respectively [32]. It also learned that the crystallographic-orientation relationship of the two structures was  $(110)_{B2} \parallel (1\overline{11})_{FCC}$  and  $[001]_{B2} \parallel [01\overline{1}]_{FCC}$ . Region B contains nanoprecipitates and matrix phase (equiaxed matrix grains). The matrix phase is a B2 superlattice lattice structure, therefore, the calibrated FCC structure corresponds to the nanoprecipitates phase. The chemical composition test results of E, F, G, and H in Fig. 8 were shown in Table 2. Fe and Cr segregation was found to exist in the nanoprecipitate according to TEM-EDS results of F in Table 2. Therefore, the nanoprecipitates were Cr-rich FCC precipitate phases. The higher Fe content of the point E G and H are 41.07 %, 40.44 % and 39.89%, respectively. The higher Fe content may be due to the melting and mixing of the substrate and the coating.

#### 3.3 Nanoindentation

The nano-hardness contour map of V7, V13 and V17 were shown in Fig. 10 (a1), (b1), (c1), which drawn using Origin Pro 9.1 software. A lot of blue areas appeared in the nano-hardness contour map of V7, of which the nano-hardness value was below 3.6GPa. The average nano-hardness value of V7 was 5.4GPa, and the average nano-hardness value of V13 was 6.6GPa. Compared with V7, the average hardness value of V13 increased significantly with the increase of laser scanning speed, and the hardness value increased by 1.2GPa. V17 and V13 have a similar nano-hardness

distribution, and the nano-hardness values are both between 3.6-9.9GPa. But the average nano-hardness value of V17 was slightly smaller than that of V13, which was 6.4GPa. The results show that as the laser scanning speed increases, the nano-hardness of the coatings shows an overall increasing trend, which was also found in previous studies[33-34]. The elastic modulus contour map of V7, V13 and V17 were shown in Fig. 10 (a2), (b2) and (c2), respectively. The elastic modulus mapping of V17 clearly showed that the elastic modulus value of one-third of the mapping area was below 196GPa. The average elastic modulus of V7, V13, and V17 were 213.6MPa, 213.3MPa, 202.0MPa, respectively, which gradually decreased with the increase of laser scanning speed. Both the distribution of nano-hardness and elastic modulus of V7, V13 and V17 were not uniform, indicating that the three coatings were not homogenous. However, the Vickers hardness values of all eight coatings did not change significantly as the laser speed increased (see Fig. 11). All hardness values were around 480HV. Similar hardness values may be attributed to the fact that all coatings have equiaxial grains with BCC structure as the matrix phase. but the dilution ratio has little effect on the hardness.

Through the nanoindentation test, the difference in micromechanical properties was characterized. The contour map shows the unevenness at nano-scale of the nano-hardness and elastic modulus of the coating, indicating that the mechanical properties of the different phases of the coating are different.



Fig. 10. Nano-hardness contour map of (a1) V7, (b1) V13 and (c1) V17; Elastic

modulus contour map of (a2) V7, (b2) V13 and (c2) V17.



Fig. 11. Microhardness of  $Al_{16.80}Co_{20.74}Cr_{20.49}Fe_{21.28}Ni_{20.70}$  HEA coatings.

Coatings	Area	Phases	Displacement (nm)	Hardness (H) (GPa)	Elastic Modulus (E) (GPa)	Number of Indents (n)	CV Hardness (%)	CV Elastic Modulus (%)	Reduced Elastic Modulus (E <sub>r)</sub> (GPa)	H/E <sub>r</sub>	${\rm H}^{3}/{\rm E_{r}}^{2}$
	W	FCC	$340 \pm 40$	$3 \pm 1$	$200\pm30$	33			182	0.015	0.000675
V7	NP-M	FCC-BCC	$280 \pm 10$	$5 \pm 1$	$225\pm35$	16	40	9	202	0.022	0.002469
	Μ	BCC	$250 \pm 10$	$7 \pm 1$	$240\pm30$	40			213	0.029	0.005955
	GB	FCC	$310 \pm 40$	$5 \pm 1$	$176\pm28$	21			163	0.028	0.004035
V13	NP-M	FCC-BCC	$262 \pm 10$	$6 \pm 1$	$219\pm21$	37	24	15	197	0.027	0.004504
	Μ	BCC	$239 \pm 15$	$8\pm 2$	$242\pm34$	43			214	0.033	0.008743
	GB	FCC	$314 \pm 42$	$5 \pm 1$	$169\pm33$	26			158	0.029	0.004377
V17	NP-M	FCC-BCC	$262 \pm 11$	6 ± 1	$217\pm31$	40	24	16	196	0.028	0.004587
	М	BCC	$238 \pm 18$	$8 \pm 2$	$237 \pm 44$	33			211	0.034	0.009115

Table 3 Hardness and Elastic modulus of different phases in V7, V13 and V17

The mechanical properties of the individual phase of coatings were systematically studied. The nanoindentation responses of V7 and V13 were shown in Fig. 12 and Fig. 14, respectively. Fig. 12 showed the SEM morphology of V7 after indentation. There were three characteristic phases in V7, Widmanstätten (W), Matrix (M) and the mixed structure of Nanoprecipitates and Matrix (NP-M). The high magnification SEM images of the representative indents of W, M and NP-M were shown in Fig. 12 (b), (c) and (d), which were marked by the yellow boxes in Fig. 12(a). It was worth noting that the nanoprecipitates were observed at the indent of NP-M (see Fig. 12(c)). The load-displacement curves of them were shown in Fig. 13. Under the same load condition (10mN), W had the deepest indentation depth, NP-M had the middle indentation depth, and M had the smallest indentation depth. 33 indents were measured in the W phase, 16, 40 indents were also measured in the other two phases M, NP-M, and similar indentation depth results were obtained (Table 3). Thus, W was identified as softer than M and NP-M; while M was the hardest phase and had better resistance to deformation. In summary, the hardness of different phases was very different. It may be attributed to the different crystal structures of the three phases. From the above chapter, W, M and NP-M phases were FCC structure, BCC structure and FCC-BCC structure respectively. BCC structure generally has a larger hardness value and superior strength than the FCC phase [36,37]. Interestingly, the difference in nano-hardness of the three phases can be reflected by the indentation size effect (ISE), which reveals that the reduction in the size of the indentation leads to an increase in nano-hardness and a decrease in plasticity [38]. As shown in Fig. 12 (a), the indentation area of W phase was larger than that of M phase and NP-M phase. Table 3 showed the exact value of elastic modulus, E (GPa) and hardness, H (GPa) for W, M and NP-M. The values of elastic modulus and nano-hardness were calculated by the Oliver-Pharr method [39]. The difference in elastic modulus of each phases was not as obvious as that in hardness value, easy to identify from the coefficient of variation (CV) of hardness and elastic modulus listed in Table 3.



Fig. 12. (a) SEM morphology of V7 after indentation, and the high magnification SEM images of the representative indents of (b) Widmanstätten (W), (c)Nanoprecipitates and Matrix mixed structure (NP-M) and (d) Matrix (M) (marked by boxes in Fig. 12(a)).



# Fig. 13. Load-displacement curves of the representative indents of Widmanstätten (W), Matrix (M) and Nanoprecipitates and matrix mixed structure (NP-M) in V7

The SEM morphologies of V13 after indentation were shown in Fig .14. The three characteristic phases, Grain Boundary (GB), Matrix (M) and Nanoprecipitates and Matrix mixed structure (NP-M) were FCC, BCC and FCC-BCC structure respectively. Grain boundaries and nanoprecipitates after indentation can be clearly observed in Fig. 14 (b) and (c). The load-displacement curves were shown in Fig. 15. GB displayed the largest displacement, followed by M and NP-M. This is consistent with the research results of V7 that the FCC structure has a larger indentation depth than the BCC structure. But it was noticed that the indentation depth of GB was less than W (see Fig. 13).

The exact values of the elastic modulus and nano-hardness of each phase in V17 were listed in Table 3. The characteristic phases of it were the same as those of V13, so detailed introduction will not be given here. Multiple indentations were performed on these three characteristic phases of V13 and V17, respectively, the CV of hardness and elastic modulus were listed in Table 3.



Fig. 14. (a) SEM morphology of V13 after indentation, and the high magnification SEM images of the representative indents of (b) Grain Boundary (GB), (c) Nanoprecipitates and Matrix mixed structure (NP-M) and (d) Matrix (M) (marked by boxes in Fig. 14(a)).



Fig. 15. Load-displacement curves of the representative indents of Grain Boundary

(GB), matrix (M) and Nanoprecipitates and Matrix mixed structure (NP-M) in V13.

Moreover, the load-displacement curves corresponding to V7 and V13 were not smooth especially in M phases and NP-M phases (see Fig.13 and Fig.15), which is a phenomenon of the serrations [40]. It indicates that a deformation mechanism that hinders the movement of dislocations may be formed in the coatings. The possible strengthening mechanisms are dispersion strengthening in the mixed structure of nanoprecipitates phases and matrix phases, and solid solution strengthening in the matrix phases.

The nanoindentation results in the literature were depicted in Fig. 16 [16,41–43]. It should be noted, here, we only focus on nano-hardness and elastic modulus of Al-Co-Cr-Fe-Ni HEAs. The comparison results obviously revealed that Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings have the best combination of namo-hardness and elastic modulus combination of the nano-hardness and elastic modulus superior to that of other Al-Co-Cr-Fe-Ni HEAs. This may be due to the special microstructure of the coating in this experiment. in which the nanoprecipitates are distributed in the equiaxed matrix grains with A2 and B2 structures.



Fig. 16 Nano-hardness and elastic modulus of FCC and BCC phases compared with those of other Al-Co-Cr-Fe-Ni HEA alloys [16,41–43]. Open and solid symbols represent the value of FCC and BCC, respectively. The open and closed symbols of the same color were from the same material.

#### 3.4 Wear test

 $H/E_r$  and  $H^3/E_r^2$  are the suitable parameters for evaluating wear resistance, and  $H^3/E_r^2$  can reflect the ability to resist plastic deformation [44]. Here,  $E_r$  refers to reduced elastic modulus, which is calculated by the following formula:  $(E_r)^{-1} = (E_s)^{-1}(1 - v_s^2) + (E_i)^{-1}(1 - v_i^2)$ , where  $E_s$  and  $E_i$  are the elastic modulus of the specimen and indenter, respectively;  $v_s$  and  $v_i$  were the Poisson's ratio of the specimen and indenter, respectively; The  $E_i$  and  $v_i$  were 1147GPa and 0.3 in turn. The calculated values of  $E_r$ ,  $H/E_r$  and  $H^3/E_r^2$  were exhibited in Table 3. The  $H/E_r$  and  $H^3/E_r^2$  value of each phase of V17 was the highest among the three coatings. According to the mixing principle, V17 should have the best wear resistance.

The wear performance of the coatings was evaluated by the dry sliding friction test. The V7, V13, and V17 were subjected to dry sliding friction tests both at room temperature and 800°. The friction coefficient curves were shown in Fig. 17 (a) and (b), respectively. The room temperature friction coefficients calculated statistically of V7, V13, and V17 were 0.73, 0.49 and 0.51, respectively. The friction coefficients at 800°C were 0.18, 0.12 and 0.10, respectively. The friction coefficient at room temperature is 4.06, 4.08, and 5.10 times of the friction coefficient at high temperature in turn. The high-entropy alloy coating has a smaller coefficient of friction at high temperatures. Moreover, regardless of whether it is at room temperature or 800°C, the friction coefficient of the coating gradually decreases with the increase of the laser scanning speed, and the friction coefficient of V13 was much smaller than V7.

In order to measure wear resistance more accurately, the wear profile of the coating was measured [45]. 30 wear profiles were measured for each wear sample to ensure the authenticity of the data, and then the Origin software was used to average multiple curves to get the ultimate wear profile curve, as shown in Fig. 17(c). It can be seen from the wear profile that, from V7 to V17, the width and depth of the wear track of the coatings gradually decrease, indicating that the amount of wear volume

loss is gradually reduced. In order to obtain more accurate wear data, The wear rate was calculated according to the following formula: P = V/LS, where P is wear rate (mm<sup>3</sup> N<sup>-1</sup>·m<sup>-1</sup>), V is wear volume (mm<sup>3</sup>), L is load (N), and S is the total sliding distance (m) [46]. the wear rate of V7, V13 and V17 were calculated as 0.000552mm<sup>3</sup>N<sup>-1</sup>m<sup>-1</sup>, 0.000331mm<sup>3</sup>N<sup>-1</sup>m<sup>-1</sup>, 0.000175mm<sup>3</sup>N<sup>-1</sup>m<sup>-1</sup>, respectively.

Combining the lower friction coefficient and lower wear rate, V17 has excellent wear resistance. The wear mechanism of V17 was studied. Wear debris and shallower furrows indicated that the wear mechanism of the coating is abrasive wear (Fig. 18(b)). In addition, as shown in Fig. 18(c), deep spalling and tearing were observed, indicating that fatigue wear and slight adhesive wear had occurred [47]. In short, the coating has better wear resistance at high temperatures, and the wear resistance of V17 is the best among the three coatings, which was consistent with the predicted results by  $H/E_r$  and  $H^3/E_r^2$ .







4 5 6 7 8 9 22 23 24 25 26 27 28 29 30 32 33 34 43 



Fig. 17. (a) Room temperature friction and wear curve; (b) High temperature (800°C) friction and wear curve; (c) Wear scar profile curve of V7, V13 and V17 at 800°C and (d) The wear rate of V7, V13 and V17 at 800°C.



Fig. 18. The wear morphology of V17.

## 4. Discussion

#### 4.1 Phase transformation

In order to predict phase formation in multi-component HEAs, some parameters had been published. From the perspective of thermodynamics, Zhang et al. [4] proposed two parameters  $\Delta H_{mix}$  (enthalpy of mixing) and  $\delta$  (atomic radius difference) to evaluate the solid solution forming ability,  $\Delta H_{mix}$  and  $\delta$  are expressed as Equation (1) and (2):

$$\Delta H_{mix} = \sum_{i=1, i\neq j}^{n} 4c_i c_j \Delta H_{ij}^{mix} \tag{1}$$

$$\delta = \sqrt{\sum_{i=1}^{n} c_i \left(1 - \frac{r_i}{\bar{r}}\right)^2} \tag{2}$$

$$\overline{r} = \sum_{i=1}^{n} c_i r_i \tag{3}$$

where  $c_i$  and  $c_j$  are the atomic percentage and  $\Delta H_{ij}^{mix}$  is the enthalpy of mixing between the *ith* and *jth* elements, and  $r_i$  is the Goldschmidt atomic radius of the *ith* element. The parameter  $\delta$  characterizes the atomic size mismatch, which produces the local elastic strain and determines the system topological instability. And the parameter  $\Delta H_{mix}$  reflects the tendency of forming stable intermetallic compounds [27]. The standard for forming a mixture of ordered and disordered solid solutions are  $-20 < \Delta H_{mix} < 0 \text{ KJ mol}^{-1}$ ,  $5\% < \delta < 6.6\%$  [4]. r and  $\Delta H_{ij}^{mix}$ listed in Table 4 and Table 5 were substituted into the above formulas. And then the  $\Delta H_{mix}$  and  $\delta$  of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings were calculated as  $-11.23 \text{ KJ mol}^{-1}$  and 5.43% respectively, which may fall into the ordered and disordered solid solutions FCC phase, solid solution BCC phase. It shows that this standard can accurately predict the phase formation of this alloy system.

Element	$\Delta H_{mix}(kJ/mol)$								
	Fe	Co	Cr	Ni	Al				
Fe									
Со	-1								
Cr	-1	-4							
Ni	-2	0	-7						
Al	-11	-19	-10	-22					

Table 4 The values of mixing enthalpy of element pairs

Table 5 Characteristics of each element used in this paper

Element	Atomic Radius (Å)	Crystal	VEC	Melting point(°C)
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Fe	1.241	BCC/FCC	8	1538
Co	1.251	HCP	9	1492
Cr	1.249	BCC	6	1890
Ni	1.246	FCC	10	1453
Al	1.432	BCC	3	660

Furthermore, a new parameter,  $\Omega$ , expressing the competition between entropy and enthalpy, had been proposed by Yang et al [48]. And the author gives the criteria for forming the single stabilized solid solution,  $\Omega \ge 1$ ,  $\delta \le 6.6\%$ .  $\Omega$  were defined as:

$$\Omega = \frac{T_m \Delta S_{mix}}{|\Delta H_{mix}|} \tag{4}$$

$$T_m = \sum_{i=1}^n c_i \left( T_m \right)_i \tag{5}$$

$$\Delta S_{mix} = -R \sum_{i=1}^{n} c_i Lnc_i \tag{6}$$

where  $T_m$  is the average melting temperature,  $(T_m)_i$  is the melting point of the *ith* element,  $\Delta S_{mix}$  is the mixing entropy and R (8.314 *KJ mol<sup>-1</sup>*) is the gas constant. The  $\Omega$  of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> was calculated as 1.71 listed in Table 6. In short, the  $\delta$  and  $\Omega$  values of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> coatings are 5.43% and 1.71. According to  $\Omega$  and  $\delta$  criteria, the single-phase solid solution will be generated. However, the coatings consist of disordered solid solution FCC phase, solid solution BCC phase with ordered B2 structure. Therefore, the  $\Omega$  and  $\delta$  criteria is not appropriate to predict the phase formation in this HEA system.

It should be noted that for laser cladding, due to the dilution from the substrate, using the present criterions suitable to the alloys processed by the traditional casting and smelting method will cause the phase prediction results to be inaccurate. Juan et al. [49] proposed the modified criterions of the solid-solution formation ability in the multi-component laser-clad coatings :  $10.8 \le \Delta \text{Smix} \le 16.2 \text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ ,  $-17 \le \Delta$  Hmix  $\le 7 \text{ kJ} \cdot \text{mol}^{-1}$ ,  $0 \le \delta \le 14$ . According to this standard, combined with Table 6, the coating can form a solid solution, which is consistent with the phase structure of the

coatings.

In addition, Guo et al. [50] proposed a physical parameter, valence electron concentration (*VEC*), to predict the phase stability for FCC and BCC phases in HEAs. When  $VEC \ge 8.0$ , only FCC phase exists; when  $6.87 \le VEC < 8.0$ , mixed FCC and BCC phases will co-exist; only BCC phase exists when VEC < 6.87. The *VEC* of the system is calculated by the following equation:

$$VEC = \sum_{i=1}^{n} c_i \left( VEC \right)_i \tag{7}$$

where  $c_i$  is the atomic percentage and  $(VEC)_i$  is the VEC of the *i*th element. The calculated VEC value of Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings is 7.37, falling into the region of mixed FCC and BCC phases. And the coating is a mixture of FCC and BCC phases, implying that the phase stability for Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEAs can be predicted by VEC.

The above parameters can predict whether to generate solid solutions or ordered solid solutions for Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings, as well as the structure of solid solutions, such as FCC and BCC. But the volume fraction of each phase cannot be predicted. According to the results obtained by the above chapters, we found that the increase in laser scanning speed promotes the formation of the BCC phase, as shown in Fig. 2(a) and Fig. 4, which is consistent with the results reported by M.D. et al. [51]. Wang et al. [52] and Cui et al. [28] also have a similar research report. They proposed that high volume fraction of the BCC phase can be attributed to the higher scanning speeds and high cooling rates caused by the former. The volume fraction of the BCC phase calculated by Image J software was exhibited in Fig. 19 [43], which increases with the increase of the laser scanning speed. And the coatings can be divided into three types according to the volume fraction of BCC: (1) < 80%; (2) 80%-90%; (3) >90\%. The corresponding morphologies of the three coatings were also shown in the insets in Fig. 19, in which the white sphere represents the nanoprecipitates phase, and the white needle-shaped represents the Widmanstätten side plates. When the volume fraction of BCC is <80%, the coatings include

Widmanstätten, tnanoprecipitates and matrix; When the volume fraction is 80%-90%, only nanoprecipitates and matrix exists; When the volume fraction is >90%, the phase composition is the same as the former, but the number and size of the nanoprecipitates are reduced. It can be understood by considering the phase transformation process which requires the cooperative diffusion of elements to achieve the equilibrium partitioning among different phases, but severe lattice distortion occurred in the HEAs in fact hindering the atomic movement by which the diffusion happens. Therefore, elements cannot complete the transition to the equilibrium state within a limited cooling time and have no time to form the second phase. This evolution of the phase structure is attributed to the sluggish diffusion effect of the HEA.

Table 6 Results of,  $\Delta S_{mix}$ ,  $\Delta H_{mix}$ ,  $\delta$  and VEC of HEA coatings

Alloy composition	$\Delta S_{mix}$	$\Delta H_{mix}$	δ	Ω	VEC
	(J/K•mol)	(J/K•mol)	(%)		
$Al_{16.80}Co_{20.74}Cr_{20.49}Fe_{21.28}Ni_{20.70}$	13.35	-11.23	5.43	1.71	7.37



Fig. 19. The volume fraction of BCC as a function of laser scanning speed.

4.2 Relationship between the phase transformation and mechanical properties

The laser scanning speed significantly improves the strength of the HEA coatings.

With the increase of laser scanning speed, the nano-hardness exhibited a significant increase trend overall, from 5.4GPa to 6.6GPa and then to 6.4GPa. It may be the phase transformations that cause the increase of average nano-hardness from V7 to V17. Fig. 19 showed that the increase of volume fraction of the BCC phase from V7 to V17. Since the nano-hardness values of the BCC phase ( $7 \pm 1$ GPa in V7,  $8 \pm 2$ GPa in V13, and  $8 \pm 2$ GPa in V17) are much greater than those of the FCC phase ( $3 \pm 1$ GPa in V7,  $5 \pm 1$ GPa in V13, and  $5 \pm 1$ GPa in V17), the increased number of BCC phase will lead to an increase in the nano-hardness of the whole coating. Consistent with nano-hardness, wear resistance was also significantly improved (see Fig. 17). The increase in wear resistance and strength can be attributed to many strengthening mechanisms, such as grain boundary hardening, solid solution strengthening and second phase strengthening.

Grain boundary hardening exists in this HEA and it is related to the undercooling. The total undercooling is defined as a sum of four different contributions [53].

$$\Delta T = \Delta T_t + \Delta T_c + \Delta T_r + \Delta T_k \qquad (8)$$

where  $\Delta T_t$  represents the thermal undercooling,  $\Delta T_c$  the constitutional undercooling,  $\Delta T_r$  the curvature undercooling, and  $\Delta T_k$  the kinetic undercooling. In the laser cladding process, kinetic undercooling is related to laser scanning speed which can be determined by the following methods [54,55]:

$$\Delta T_k = \frac{v_s}{\lambda} \tag{9}$$

$$v_s = av \tag{10}$$

$$\lambda = \frac{\Delta H_f v_0}{k_B T_L^2} \tag{11}$$

$$\sigma_{\rm v} = \sigma_0 + k D^{-1/2} \tag{12}$$

where v is laser scanning speed,  $v_s$  is the growth velocity of solidification front, a is a constant defined by us (>1),  $\lambda$  is the interfacial kinetic coefficient,  $v_0$  is the speed of sound (m/s),  $k_B$  is the Boltzmann constant,  $\Delta H_f$  is the standard molar enthalpy of formation, and  $\Delta H_f$  is the liquidus temperature (K). From equation (9) and (10), increased laser scanning speed will greatly increase the kinetic undercooling in the melt pool. According to the solidification theory [56], a large kinetic undercooling will lead to a significant increase in the nucleation rate. A large amount of nucleation in a limited time leads to insufficient time for grain growth, thus, refining the grains of HEA coatings [57]. As mentioned above (see Fig. 5), the grain size keeps decreasing with increasing laser scanning speed. Therefore, grain boundary strengthening occurs in the coating, having a great contribution to the excellent performance of coatings. At the same time, grain boundary strengthening can be calculated quantitatively with the Hall-Patch formula (Eq. 12) [58]. The specific inquiry is expected in the following research.

# 5. Conclusions

Low-cost Al<sub>16.80</sub>Co<sub>20.74</sub>Cr<sub>20.49</sub>Fe<sub>21.28</sub>Ni<sub>20.70</sub> HEA coatings were processed by laser cladding. The phase formation, microstructure evolution, and mechanical behaviors of these HEA coatings were investigated. The following conclusions were drawn as follows:

- (1) All alloys were typical polycrystalline structures composed of equiaxed matrix grains with BCC structure. White nanoprecipitates are distributed in the grain boundary and inside the grain as a precipitated phase. And the nanoprecipitates were Cr-rich FCC precipitate phases. The BCC structures in V17 were composed of 55% B2 structure and 45% A2 structure.
- (2) The evolution of the phase and microstructure of coatings has been discovered. As the laser scanning speed increases, the volume fraction of

the BCC structure gradually increases. The coatings can be divided into three types according to the volume fraction of BCC: (1) V7-V9 (<80%): Widmanstätten side plate (FCC), nanoprecipitates phase (FCC) and matrix phase (BCC); (2) V11-V15 (80%-90%): nanoprecipitates phase (FCC) and matrix phase (BCC); (3) V17-V21(>90%): less and finer nanoprecipitates phase (FCC) and matrix phase (BCC). The grain size of equiaxed grains reduced from 276µm to 42µm with increased laser scanning speed.

- (3) Nanoindentation measurements were performed on V7, V13 and V17, respectively. The nano-hardness values of the BCC phase ( $7 \pm 1$ GPa in V7,  $8 \pm 2$ GPa in V13, and  $8 \pm 2$ GPa in V17) were much greater than those of the FCC phase ( $3 \pm 1$ GPa in V7,  $5 \pm 1$ GPa in V13, and  $5 \pm 1$ GPa in V17).
- (4) The coating has better wear resistance at high temperatures, and the wear resistance of V17 is the best among the three coatings, consistent with the predicted results by  $H/E_r$  and  $H^3/E_r^2$ .
- (5) The excellent mechanical properties for V17 lie in grain boundary strengthening, increased laser scanning speed will greatly increase the kinetic undercooling in the melt pool, which increases the nucleation rate and refine the grain.

#### Acknowledgments

This research was supported by Foundation of Natural Science Foundation of China (52075317, 51905333, 51805316), the Royal Society through International Exchanges 2018 Cost Share (China) scheme (IEC\NSFC\181278), Shanghai Sailing Program (19YF1418100), Shanghai Science and Technology Committee Innovation Grant (17JC1400600, 17JC1400601, 19511106400, 19511106402), Karamay Science and Technology Major Project (2018ZD002B), Aid for Xinjiang Science and Technology Project (2019E0235), Shanghai local colleges and universities capacity building special plan project (19030501300).

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#### **Declaration of interests**

 $\boxtimes$  The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

□The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: