

Subscriber access provided by Lancaster University Library

Characterization of Natural and Affected Environments

Public water supply is responsible for significant fluxes of inorganic nitrogen in the environment

Matthew J Ascott, Daren C Gooddy, and Ben W.J. Surridge

Environ. Sci. Technol., Just Accepted Manuscript • DOI: 10.1021/acs.est.8b03204 • Publication Date (Web): 19 Oct 2018

Downloaded from http://pubs.acs.org on October 23, 2018

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

1	Public water supply is responsible for significant fluxes of inorganic nitrogen in the environment
2	
3	Matthew J. Ascott ^{1*} , Daren C. Gooddy ¹ , and Ben W.J. Surridge ²
4	
5	
6	¹ British Geological Survey, Maclean Building, Crowmarsh, Oxfordshire, United Kingdom, OX10 8BB
7	² Lancaster Environment Centre, Lancaster University, Lancaster, United Kingdom, LA1 4YQ
8	
9	
10	*Corresponding author: Matthew J. Ascott, email: matta@bgs.ac.uk, tel: +44 (0) 1491 692408
11	
12	
13	
14	
15	
16	
17	
18	
19	

20 Abstract

21	Understanding anthropogenic disturbance of macronutrient cycles is essential for assessing risks
22	facing ecosystems. For the first time, we quantified inorganic nitrogen (N) fluxes associated with
23	abstraction, mains water leakage and transfers of treated water related to public water supply. In
24	England, the mass of nitrate-N removed from aquatic environments by abstraction (ABS-NO ₃ -N) was
25	estimated to be 24.2 kt N/yr. This is equal to six times estimates of organic N removal by abstraction,
26	15 times in-channel storage of organic N and 30 times floodplain storage of organic N. ABS-NO $_3$ -N $$ is
27	also between 3-39% of N removal by denitrification in the hydrosphere. Mains water leakage of
28	nitrate-N (MWL-NO $_3$ -N) returns 3.62 kt N/yr to the environment, equating to approximately 15% of
29	ABS-NO ₃ -N . In urban areas, MWL-NO ₃ -N can represent up to 20% of total N inputs. MWL-NO ₃ -N is
30	predicted to increase by up to 66% by 2020 following implementation of treated water transfers.
31	ABS-NO ₃ -N and MWL-NO ₃ -N should be considered in future assessments of N fluxes, in order to
32	accurately quantify anthropogenic disturbances to N cycles. The methodology we developed is
33	transferable, using widely-available datasets and could be used to quantify N fluxes associated with
34	public water supply across the world.

43 Abstract Art

44



60 **1 Introduction**

61

62 Eutrophication associated with anthropogenic disturbance to nitrogen (N) and phosphorus (P) cycles 63 within aquatic ecosystems is a widespread environmental challenge that has been recognised for over 50 years¹. In response, substantial efforts have been made to reduce both point² and diffuse³ 64 65 source nutrient inputs to the environment, for example through implementation of national and international legislation (e.g the USA Clean Water Act⁴, EU Water Framework⁵, Nitrates⁶ and Urban 66 Wastewater Treatment⁷ Directives). Despite such efforts, there are multiple instances in which the 67 68 reductions in riverine concentrations and fluxes of N and P after the implementation of mitigation activities have been lower than expected⁸. The release of nutrients, both from previously 69 70 unidentified sources and from legacy stores (e.g. groundwater, soils or the riparian zone⁹), is a 71 potentially important contributor to unexpected trajectories of catchment nutrient balances. In this 72 context, we sought to constrain the previously unquantified influence of public water supply on 73 inorganic N fluxes.

74 Recent research has suggested that the water treatment cycle (water abstraction, treatment, 75 distribution for public water supply and subsequent leakage) may have significant impacts on both 76 nutrient sources and nutrient retention, particularly in urban areas. For example, previous research 77 suggests that leakage of phosphate-dosed mains water (MWL-P) can be an important source of P in 78 urban environments, with MWL-P fluxes now reaching up to 24% of sewage treatment P loads in some catchments¹⁰⁻¹². Methods have also been developed to quantify N fluxes to groundwater 79 from mains water leakage at the city-scale¹³⁻¹⁵, suggesting that approximately 36% of the total N 80 81 loads in urban areas could be attributed to this source.

83 In the context of aquatic ecosystems, retention (also referred to as losses, sinks or removal) can be 84 conceptualised as transient (e.g. nutrient storage within floodplains or groundwater) or permanent N removal (e.g. denitrification)¹⁶. Based on the difference between N fluxes from terrestrial to 85 aquatic ecosystems and N fluxes at the tidal limit of the fluvial system, Worrall, et al. ¹⁷ concluded 86 87 that substantial in-stream N retention occurs in Great Britain. Whilst detailed understanding of in-88 stream retention mechanisms at the national-scale remains to be developed, it has been postulated 89 that these losses of N are primarily due to permanent N removal via denitrification¹⁷. However, the 90 processes associated with public water supply may also contribute significantly to N retention within aquatic ecosystems. For example, Finlay, et al. ¹⁸ estimated the flux of organic N removed by water 91 92 abstraction, suggesting that this flux is greater than N retention associated with transient floodplain 93 storage.

94 However, important gaps remain in our understanding of how macronutrient fluxes are perturbed by the processes associated with public water supply. Whilst Finlay, et al. ¹⁸ calculated organic N 95 96 removal, the mass of inorganic N removed during abstraction and the significance of this N retention 97 process relative to other forms of in-stream retention have not been quantified to date. Further, 98 previous methods developed for nutrient source apportionment either do not consider mains leakage^{19, 20}, or use simplistic approaches¹³⁻¹⁵ based on export coefficients and literature values 99 100 rather than observed leakage and concentration data. Finally, potential changes in inorganic N fluxes 101 due to mains water leakage associated with future treated drinking water transfers have not been 102 quantified. Water transfers are considered essential if future water demand is to be met, but the water quality implications of water transfers remain poorly constrained²¹. If effective, integrated 103 104 approaches to the management of nutrient sources and fate are to be developed, it is essential that 105 future nutrient fluxes associated with water transfers are quantified and the locations where these 106 fluxes are significantly enhanced in the future are identified. Within this context, in the research 107 reported here we hypothesized that:

108	٠	Abstraction of raw untreated water removes a significant, but spatially heterogeneous, mass
109		of nitrate (ABS-NO $_3$ -N) from the aquatic environment (defined here as groundwater, rivers
110		and coastal marine environments);
111	٠	A considerable proportion of ABS-NO $_3$ -N is returned to the environment through leakage of
112		treated mains water containing nitrate (MWL-NO $_3$ -N), particularly within urban areas;
113	•	Treated drinking water transfers will substantially alter the future magnitude and spatial
114		distribution of MWL-NO ₃ -N fluxes.

Materials and Methods 115 2

2.1 Study Area 116

117

118 The research reported here was developed at the national-scale for the country of England (Figure 119 1). Water is supplied by 21 water companies across England. Water company areas are divided into water resource zones (WRZs) within which water supplies are largely self-contained²². These WRZs 120 121 are further divided into water supply zones (WSZs) which correspond to areas with uniform water quality, supplying up to 100,000 people²³. WSZs in England derived in this research (see supporting 122 123 information) and principal regions referred to in this research are shown in Figure 1.

124

2.2 Quantification of ABS-NO₃-N and MWL-NO₃-N 125

126

2.2.1 ABS-NO₃-N 127

128

129 We derived ABS-NO₃-N using estimates of the rates at which treated water enters the distribution 130 network from water treatment works (referred to herein as "distribution input") and raw water 131 quality data for each WSZ. Distribution input rates are not reported nationally at the WSZ level, so

Environmental Science & Technology

these were derived as follows using estimates of treated water supplied to customers (referred to herein as "water supplied"). The total distribution input, DI (m^3/day) into a WSZ was divided into water supplied, S (m^3/day), and leakage, L (m^3/day):

$$DI = S + L \tag{1}$$

S for each WSZ was extracted from published data submitted by the drinking water regulator for England to the European Commission²⁴. For 12% of WSZs, values of S were not reported by the drinking water regulator. For these WSZs, S was estimated using WRZ-level estimates of per capita water consumption (PCC, L/h/d) published by Water UK ²¹ and the total population (P, unitless) for each WSZ area, based on 1 km gridded population density mapping²⁵:

$$S = \frac{PCC \cdot P}{1000} \tag{2}$$

140 Because WSZ-level leakage rates were not available, these were derived based on published WRZ-

141 level leakage rates²¹. Assuming that the percentage leakage rate in a WSZ is equal to the percentage

142 rate in the corresponding WRZ, L was derived as a fraction of the DI:

$$L = DI \cdot f_{WRZ} \tag{3}$$

where f_{WRZ} (unitless) is the fractional leakage rate of the WRZ. Combining equations (1) and (2), DI
was derived as:

$$DI = \frac{S}{1 - f_{WRZ}} \tag{4}$$

145

As no national scale datasets on WSZ raw water quality were available, mean raw water nitrate concentrations for each WSZ were estimated using a heuristic approach based on data held in the English environmental regulator's water quality monitoring database. Following previous studies of national pollutant transport¹¹ and groundwater use maps²⁶, WSZs were classified as either groundwater or surface water sources using national-scale hydrogeological mapping²⁷. If a WSZ is underlain by a moderate or highly productive aquifer, then the WSZ was considered to be a 152 groundwater WSZ. If underlain by a low productivity aquifer or rocks with essentially no 153 groundwater resource, then the WSZ was considered to be a surface water WSZ. For each WSZ, raw 154 water nitrate concentrations were estimated using the following rules: (1) only samples taken within 155 a WSZ; (2) only samples that correspond to the derived WSZ source water (i.e. groundwater or 156 surface water); (3) only samples taken in 2015 for direct comparability with drinking water quality 157 data for the same year (see section 2.2.2 and supporting information); and (4) only samples for 158 routine environmental monitoring, to avoid samples taken specifically for pollution investigations 159 unduly impacting the analyses. Where no water quality samples were present within a WSZ (4.5% of 160 WSZs in England by area) the mean raw water nitrate concentration of the nearest groundwater or 161 surface water sample point was used. Where WSZs are fed by existing water transfers, the receiving 162 WSZ was assigned the raw water nitrate concentration of the donor WSZ. After estimating DI and 163 raw water nitrate concentrations (C_{R} , mg N/L) for each WSZ, the mass of nitrate-N removed from the 164 aquatic environment through water abstraction, ABS-NO₃-N (kg N/day), was calculated as:

$$ABS-NO_3-N = \frac{C_R \cdot DI}{1000}$$
(5)

165

166 2.2.2 MWL-NO₃-N167

We calculated MWL-NO₃-N at the WSZ level using existing datasets for nitrate concentrations in treated drinking water and the water supply rates and WRZ-level leakage rates detailed in section 2.2.1. Combining equations (3) and (4), L was calculated as:

$$L = \frac{S \cdot f_{WRZ}}{1 - f_{WRZ}} \tag{6}$$

171 Nitrate concentration data were extracted from national-scale databases of WSZ drinking water 172 quality reports for 2015 (see supporting information). All water company laboratories are required 173 to be accredited by the UK Accrediation Service to the Drinking Water Testing Specification²⁸. This includes interlaboratory proficiency testing for all parameters including nitrate concentrations, and thus any differences in nitrate concentrations between water companies associated with different laboratory methods were assumed to be minimal. Using the mean treated drinking water nitrate concentration for each WSZ for 2015, C_T (mg N/L), MWL-NO₃-N_{wsZ} (kg N/day) was calculated as:

$$MWL-NO_3-N_{WSZ} = \frac{C_T \cdot L}{1000} \tag{7}$$

178

179 2.2.3 Changes in MWL-NO₃-N associated with future transfers of treated drinking water 180

181 Transfers of treated drinking water at the WRZ-level that are planned to be implemented by 2020 182 were provided by the environmental regulator for England (Figure S1). These transfers are all located 183 in South East and Eastern England, where current water stress is greatest²¹. To assess the impact of 184 future transfers on the spatial distribution of MWL-NO₃-N, the WSZ-level MWL-NO₃-N_{WSZ} estimates 185 derived in section 2.2.2 were aggregated to the WRZ level (MWL-NO₃-N_{WRZ}, kg N/day) to obtain a 186 baseline flux for 2015:

$$MWL-NO_3-N_{WRZ} = \sum MWL-NO_3-N_{WSZ}$$
(8)

The impact of aggregation of WSZ MWL-NO₃-N fluxes to WRZ level, in addition to impacts of using WRZ percentage leakage rates at the WSZ level and the impacts of estimating raw water nitrate concentrations using national environmental regulator water quality databases, are considered in section 4.3.

191

For each water transfer, the future leakage rate for the transferred water in the receiving WRZ was assumed to equal the existing leakage rate in the receiving zone. This is a reasonable assumption because all water companies in England have current leakage rates at or below the sustainable economic level of leakage²⁹ and leakage rates are not forecast to change substantially over the next 25 years^{21, 30}. The volumetric leakage rate for a transfer, L_T (m³/day), can therefore be estimated as:

$$L_T = DI_T \cdot f_R \tag{9}$$

where f_R is the leakage fraction (unitless) for the receiving WRZ and DI_T (m³/day) is the additional 197 198 distribution input associated with the transfer. The nitrate concentration in the transferred water in 199 the future was assumed to equal to the current drinking water nitrate concentrations in the donor 200 WRZ. Nitrate concentrations in drinking water in England are consistently below the European drinking water standard, with 99.99% of test samples compliant³¹. Consequently, there is unlikely to 201 202 be any additional treatment for nitrate removal associated with future treated water transfers. The 203 flux of N from mains water leakage associated with an individual planned transfer, MWL-NO₃-N_T (kg N/day), was estimated as: 204

$$MWL - NO_3 - N_T = L_T \cdot C_{T-D} \tag{10}$$

where C_{T-D} is the mean treated drinking water nitrate concentration in the donor WRZ (mg N/L). Some WRZs are expected to receive a number of different water transfers in the future. In these cases, the individual transfers were combined to derive the total additional flux of N from mains leakage resulting from all transfers, MWL-NO₃-N_{WRZ-T} (kg N/day):

$$MWL - NO_3 - N_{WRZ-T} = \sum MWL - N_T \tag{11}$$

209 Using the baseline WRZ-level estimates of MWL-NO₃-N_{WRZ} for 2015, the percentage change in MWL-

210 NO_3-N_{WRZ} expected after the implementation of the planned transfers by 2020, $\Delta MWL-NO_3-N_{WRZ}$

211 (unitless) was calculated as:

$$\Delta MWL - NO_3 - N_{WRZ} = \frac{MWL - NO_3 - N_{WRZ-T} + MWL - NO_3 - N_{WRZ}}{MWL - NO_3 - N_{WRZ}} \cdot 100$$
(12)

213 2.3 Comparison between ABS-NO₃-N, MWL-NO₃-N and estimates of other N 214 sources and N retention in the aquatic environment

215

216 Our estimates of ABS-NO₃-N were compared with previous, national-scale estimates of the retention of organic N via water treatment for public supply, floodplain storage and in-channel storage¹⁸. We 217 218 also compared ABS-NO₃-N with estimates of N removal via denitrification in the entire English hydrosphere (groundwater, surface water and the coastal marine environments)³² and total in-219 stream losses (assumed to be dominated by denitrification)¹⁷. Organic N retention by abstraction 220 221 was reported as a per-capita flux which was transformed to a total flux for England using the current population of England³³. All other fluxes were reported either as an absolute flux or flux per unit 222 223 area. Where necessary, fluxes at the scale of England were calculated using the percentage of the 224 UK land area which England comprises (54%).

225

226 To quantify the significance of MWL-NO₃-N, this flux was compared with other datasets aggregated 227 to both the national and the WRZ level. To compare with previous estimates of urban N fluxes, we 228 used estimates of N fluxes to groundwater at 1 km grid scale from water mains and the sewer network derived by the UK Environment Agency ³⁴ following the export coefficient approach of 229 230 Lerner ¹⁴. We also used previous catchment-scale estimates of both urban diffuse (roads, residential 231 zones, open urban spaces, industrial and commercial areas) N fluxes to rivers and from all N sources provided under the SEPARATE framework²⁰. We calculated ratios of both current and future (post-232 233 transfers) MWL-NO₃-N made in our research to these estimates of N inputs from other sources.

234 **3 Results**

235

236 **3.1 ABS-NO₃-N**

237

238 Figure 2 reports the spatial distribution of ABS-NO₃-N across England. Blank areas denote locations 239 in which no public water supplies are present (typically in sparsely populated upland areas in the 240 north of England) or where data were unavailable (a small number of WSZs in southern England). 241 Substantial spatial variation in the distribution of ABS-NO₃-N is revealed, with ABS-NO₃-N broadly 242 greater in the Midlands and the South East than in the far North and Southwest of England. There 243 are also additional hotspots of high ABS-NO₃-N in Yorkshire and the Humber, the West and East 244 Midlands and near to London. The broad spatial trends and the hotspots are primarily related to 245 variations in ABS-NO₃-N associated with high nitrate concentrations in some raw groundwaters and 246 surface waters.

247

248 Table 1 reports national-scale estimates of ABS-NO₃-N for England in comparison to previous studies 249 estimating N retention in aquatic environments in England. For 2015, we estimated total ABS-NO₃-N 250 for England to be 24.2 kt N. Finlay, et al. ¹⁸ suggest that the total mass of organic N removed during 251 water abstraction for public supply in the UK is 0.07 kg N/h/yr, equating to a total flux in England of 252 3.9 kt N/yr. ABS-NO₃-N therefore removes approximately six times more N from the environment as 253 nitrate than is associated with abstraction of organic N. ABS-NO₃-N is also 30 and 15 times greater 254 than N storage in floodplains and in channels as organic N respectively. The range of estimates for 255 the magnitude of denitrification within aquatic environments in England is substantial, from 62 kt N/yr for the complete hydrosphere (surface water, groundwater and marine coastal environments)³² 256 to 716 kt N/yr within the stream network alone¹⁷. Based on this range of available estimates, ABS-257

258	NO_3 -N is equivalent to between 3 – 39% of current estimates of denitrification in aquatic
259	environments.

260 **3.2 MWL-NO₃-N**

261

The distribution of MWL-NO₃-N across WSZs in England is reported in Figure 3. Substantial spatial variation is also revealed in the MWL-NO₃-N fluxes, reflecting both the spatial distribution of leakage rates and drinking water nitrate concentrations. Significant N fluxes from mains leakage are present in urban areas with high leakage rates such as London, the West Midlands and the Humber. Relatively high MWL-NO₃-N fluxes are also observed in Eastern England and along parts of the south coast, where particularly high nitrate concentrations have been observed in drinking water.

268

Table 2 reports the total MWL-NO₃-N flux for England alongside previous national-scale N flux estimates. The estimate of national-scale MWL-NO₃-N in our research (3.62 kt N/yr) is significantly below previous estimates of this flux made by the Environment Agency ³⁴ and relatively similar to estimates of fluxes from sewer leakage (4.07 kt N/yr). Based on previous N source apportionment studies (SEPARATE²⁰), the return of MWL-NO₃-N to the environment is small (c. 1%) at the nationalscale relative to all other N sources, but much more significant (c. 160%) relative to urban diffuse N sources.

276

Figure 4 reports our estimates of MWL-NO₃-N as a percentage of the total N flux from all sources derived from Zhang, et al. ²⁰, aggregated to the WRZ level. MWL-NO₃-N is generally small relative to total N fluxes, at 2 - 5% of the total flux in the majority of WRZs in Southeast England and the Midlands. MWL-NO₃-N is <1% in large WRZs in Northern, Southwest and Eastern England, where agricultural land is extensive and urban areas are small relative to the total land area. However, in London and the surrounding area where urban land area is more extensive, $MWL-NO_3-N$ grows in significance. For example, $MWL-NO_3-N$ fluxes are estimated to be c. 15%, 16% and 20% of total N sources in WRZs in London, the south coast and Kent (east of London) respectively.

3.3 Changes in MWL-NO₃-N associated with future treated water transfers

286

287 Figure 5 reports the percentage change in the MWL-NO₃-N flux in receiving WRZs after 288 implementation of planned treated water transfers by 2020. In some WRZs on the south coast of 289 England, MWL-NO₃-N fluxes are predicted to increase by up to 66% compared to 2015 levels, 290 although implementation of water transfers is generally predicted to increase MWL-NO₃-N fluxes by 291 a relatively small amount (5 – 15%) compared to the 2015 baseline. MWL-NO₃-N remains small 292 relative to other N sources in the majority of WRZs where transfers are implemented. However, in 293 WRZs to the west of London, MWL-NO₃-N is expected to rise after the implementation of planned 294 water transfers to the equivalent to up to 20% of all N sources (Figure S2).

295 4 Discussion

4.1 ABS-NO₃-N: A significant sink for N within the environment?

297

298 Nitrogen retention within the aquatic environment contributes to the differences between the N flux from the terrestrial biosphere and the fluvial flux of N to the oceans³⁵. Assuming net growth and 299 decay of riverine biota at the national-scale is negligible¹⁷, denitrification has previously been 300 301 considered the primary mechanism for N removal in aquatic ecosystems^{36, 37}. Only recently has 302 research begun to quantify direct N removal from the environment through anthropogenic processes such as water abstraction¹⁸. 303 In the context of N retention, a significant unique 304 contribution of our research is the quantification of nitrate removed from aquatic environments by 305 water abstraction, which is a significantly greater N flux than the estimate of organic nitrogen 306 removal by abstraction, storage within floodplain and in-channel environments by Finlay, et al. ¹⁸.
307 The wide range of estimates for the magnitude of aquatic denitrification means that quantifying the
308 relative importance of ABS-NO₃-N is challenging. Further research is needed to better constrain
309 these highly variable estimates to more accurately evaluate the significance of ABS-NO₃-N.
310 Nevertheless, given that the upper limit for ABS-NO₃-N represents c. 40% of denitrification in the
311 hydrosphere of England, including ABS-NO₃-N in future catchment N budgets appears to be
312 important.

313

314 Whilst ABS-NO₃-N removes significantly more N than is associated with abstraction of organic N, and 315 may also be significant relative to the N flux associated with denitrification, ABS-NO₃-N may only 316 represent a transient retention process. The ABS-NO₃-N flux can be divided into: (1) N removed by 317 water treatment processes; (2) N lost through mains water leakage; and (3) N supplied within 318 treated water that reaches households. Mains water leakage will return N directly into the 319 environment, although it is likely to be delayed before finally reaching receiving waters. Using 320 drinking water nitrate concentrations and the WSZ supply rates estimated in this research, we 321 estimate that 13.5 kt N/yr enters treated public water supply. Based on the difference between ABS-322 NO_3-N and the sum of MWL-NO₃-N and N in treated public water supply, we estimate that 7.1 kt 323 N/yr is removed by the treatment of raw water before distribution. Much of the N that enters treated water will ultimately move into the wastewater network³⁸. Waters that are highly enriched 324 325 in nitrate produced by nitrate removal from raw water through ion exchange during drinking water 326 treatment will also enter the wastewater network, although in coastal areas direct discharge to the 327 marine environment may occur³⁹. Following wastewater treatment, N will be partitioned between solid phase waste sent to landfill or returned to agricultural land and dissolved or particulate N 328 329 discharged directly to receiving waters. Whilst these processes are likely to have varying timescales 330 for return of N into the environment, critically they are all likely to be substantially longer than the mean residence time for water within UK rivers (26.7 hrs¹⁸). Therefore, whilst ultimately a transient
 retention processes, ABS-NO₃-N will significantly alter the spatial and temporal distribution of N
 inputs into aquatic ecosystems.

4.2 MWL-NO₃-N: A significant N source in urban areas

335

336 At the national-scale, our research suggests that MWL-NO₃-N is equal to approximately 15% of ABS-337 NO₃-N. This is likely to be reasonable because leakage rates in England have been reported to be up to 20% of water entering the distribution network¹⁰ and treated waters show substantial variation in 338 339 nitrate concentrations. MWL-NO₃-N is small compared to all other N sources in large WRZs, where 340 agriculture is the most significant N source. However, in urban areas MWL-NO₃-N may account for 341 up to 20% of all N loads. These findings are broadly consistent with initial city-scale studies reported previously for Nottingham, UK¹⁵. Previous work that adopts relatively simple estimates of MWL-NO₃-342 343 N using export coefficient approaches³⁴ substantially overestimates MWL-NO₃-N compared to our 344 approach which uses observed leakage and concentration data. The flux of N associated with MWL 345 should be considered in more detail in urban N source apportionment studies. The methodology we 346 report here is transferable and could be applied where the necessary data are available, delivering a 347 significant increase in accuracy and spatial resolution of estimates of MWL-NO₃-N compared to 348 alternative approaches.

349

Previous research has suggested that incorporation of the environmental cost of P fluxes from mains water leakage within methodologies to estimate the sustainable economic level of leakage (SELL) could result in more stringent leakage targets, and consequently lower P fluxes to the environment¹². Extending the SELL approach to consider the environmental impact of MWL-NO₃-N would also be of environmental benefit. However, whilst MWL-NO₃-N represents a substantial N flux to the environment, concentrations in mains leakage are lower than in sewer leakage¹³. Consequently, mains leakage to groundwater may dilute high nitrate sewer leakage. Integrating N sources into SELL would require an evaluation of the tradeoff between the negative environmental impact of MWL-NO₃-N as an N source and any environmental benefit associated with dilution of sewer N leakage. Additional nitrate removal from raw waters to reduce MWL-NO₃-N fluxes whilst continuing to dilute sewer N leakage is unlikely to be a viable solution, given that concentrations of nitrate in drinking water within England are already below the European drinking water standard³¹.

362

363 It should be noted that MWL-NO₃-N, rather than representing a new input of reactive N to the 364 terrestrial environment (as would be associated with the application of inorganic N fertilisers or N 365 fixation for example), instead represents a return of N to the environment after water abstraction. 366 Return of N via MWL therefore represents a potential alteration to the distribution and residence 367 time of N in different environmental compartments. In this context, it should also be noted that the 368 ultimate fate of MWL-NO₃-N currently remains uncertain. The MWL-NO₃-N flux may be mediated by 369 the soil microbial community and by plants following leakage. MWL-NO₃-N may also be subject to 370 denitrification, although this is unlikely near to the site of a leak as treated water has been shown to be low in organic carbon²³ and is unlikely to be anaerobic due to oxidation processes during water 371 treatment (e.g. ozonation producing dioxygen⁴⁰) and extensive contact with the atmosphere⁴¹. It is 372 373 likely that the fate of MWL-NO₃-N can be broadly divided between fluxes to groundwater, surface 374 water and the sewer network. Based on the national-scale hydrogeological mapping described in 375 section 2.2.1, the MWL-NO₃-N flux on moderate and highly productive aquifers is estimated to be 1.67 kt N/yr. Given the long residence times in the unsaturated and saturated zones^{42, 43}, it is likely 376 377 that legacy MWL-NO₃-N from previous decades has accumulated within both groundwater and the 378 vadose zone, as has been observed for nutrient loads from agricultural land⁴⁴⁻⁴⁶. Moreover, 379 historical leakage rates have been significantly higher than at present, with rates falling by a third

between 1994 and 2015^{47} . Consequently, the rate of MWL-NO₃-N accumulation within the vadose zone and groundwater is likely to have been greater in the past compared to the present day.

382

383 Our research predicts relatively small changes in the spatial distribution of MWL-NO₃-N fluxes 384 associated with treated drinking water transfers. These transfers are fully costed, planned and will 385 be implemented by water utilities by 2020 in England. Larger-scale transfers which may result in 386 larger differences in water quality (and changes in MWL-NO₃-N fluxes) are currently being 387 evaluated²¹, although these transfers are yet to be formally included in the water resources plans. 388 However, it should be noted that for both planned and speculative transfers of treated water, 389 implementation of additional treatment for nitrate removal is unlikely because concentrations are 390 already below the European drinking water limit. With mitigation measures to control other N pollution sources being implemented⁴⁸, and limited changes planned for mains water leakage 391 392 rates³⁰, it is likely that the relative significance of MWL-NO₃-N will increase in the future, as has been reported for MWL-P in the Thames catchment³⁰. 393

394

4.3 Local and global research priorities for ABS-NO₃-N and MWL-NO₃-N 396

The unique contribution of this study is the first national-scale estimates of N fluxes from abstraction and mains water leakage using observed nitrate concentration and leakage data, both before and after the implementation of future water transfers. There are a number of outstanding research priorities within this field which remain to be addressed. To date the only published national estimates of aquatic N retention by abstraction are for organic N compounds¹⁸, and comparing ABS-NO₃-N with other temporary retention processes that influence inorganic N would be helpful. Our research has quantified the flux of nitrate associated with water treatment processes as the 404 difference between ABS-NO₃-N and the sum of MWL-NO₃-N and N in treated water supplied. Direct 405 quantification of N removal during water treatment at the national-scale using observed 406 concentration and flow data for process losses could be used to further validate the approach 407 reported here. Water abstraction for public supply is also likely to be associated with significant 408 fluxes of other inorganic N species, for example meaning that quantifying the mass of N removed via 409 abstraction of ammonium may also be important. Further research should also evaluate the 410 ultimate fate of MWL-NO₃-N and the timescales for re-release of N to the environment from 411 transient stores associated with the water treatment cycle.

412

413 The research reported here at the national-scale highlights the particular importance of MWL-NO₃-N 414 in urban areas. However, due to the dearth of data at the WSZ level, there are a number of 415 uncertainties in this study which should be considered when evaluating future research priorities. It 416 should be noted that WSZ level leakage rates were estimated by applying percentage leakage rates 417 from the corresponding WRZ. Whilst treated drinking water nitrate concentrations are reported at 418 the WSZ level, there is likely to be some uncertainty in estimates of raw water nitrate concentrations 419 as these have been made based on environmental regulator water quality databases and a heuristic 420 assessment of water sources based on hydrogeological mapping and existing water transfers. 421 Treated drinking water transfers are reported at the WRZ level and thus the impact of transfers on 422 MWL-NO₃-N fluxes has been reported at the WRZ level in this research. As a consequence of these 423 factors, it is plausible that there may be more detailed local variability in MWL-NO₃-N, ABS-NO₃-N 424 and the impact of treated water transfers in urban areas than reported in this research. Additional 425 local-scale work should focus on these areas, drawing on datasets such as detailed raw water quality 426 data, District Metering Area leakage data and information on water transfers.

428 Further research to quantify both ABS-NO₃-N and MWL-NO₃-N fluxes globally is also likely to be of 429 importance. For example, water abstraction for all uses at the global-scale has been estimated at c. 4000 km³/yr in 2010⁴⁹. Using a conservative assumption that nitrate concentrations in raw 430 431 abstracted water are half of the World Health Organisation guideline value (i.e. 5.65 mg N/L), global 432 ABS-NO₃-N would remove 22.6 Tg N/yr from the aquatic environment. This is equivalent to between 5 and 39% of denitrification in soils, groundwater and riparian zones globally⁵⁰. Global leakage rates 433 434 have been estimated at 32 billion m^3/yr^{51} . Assuming the same nitrate concentration in leakage as in 435 treated drinking water, leakage of N would contribute 0.525 Tg N/yr into the environment. This equates to approximately 1% of the global leaching and riverine N flux⁵². Whilst small in absolute 436 437 terms, this ratio at the global-scale is similar to that reported above for England. Given the 438 significance of MWL-NO₃-N in urban areas in England we report above, it is plausible that MWL-NO₃-439 N may be a significant N source in urban areas across the world. Further, both ABS-NO₃-N and MWL-440 NO₃-N may be more significant N fluxes in developing countries where nitrate concentrations and leakage rates⁵¹ are likely to be higher than those reported here for England⁵³. Therefore, accurately 441 442 quantifying N fluxes associated with public water supply represents an important challenge to be 443 addressed across the globe. 444

445

446

447 **5** Acknowledgements

- 449 This research was funded by the British Geological Survey's Groundwater Directorate National
- 450 Capability programme (Natural Environment Research Council). This paper is published with
- 451 permission of the Executive Director, British Geological Survey (NERC).

453 **6** Supporting Information

454

455 Methodology for Water Supply Zone mapping and extraction of treated drinking water nitrate

456 concentration data

- Figure S1 Location of planned treated drinking water transfers by 2020 and receiving water resource
 zones (WRZs) in England
- Figure S2 MWL-NO₃-N as a percentage of all N sources (as derived from) in receiving WRZs following implementation of treated drinking water transfers by 2020

461

462

463 **7 References**

464

Vollenweider, R. A. Scientific fundamentals of the eutrophication of 1. 465 lakes and flowing waters, with particular reference to nitrogen and 466 phosphorous as factors in eutrophication; Organisation for Economic Co-467 operation and Development Paris, France, 1968; p 193. 468 Jarvie, H. P.; Neal, C.; Withers, P. J. A., Sewage-effluent phosphorus: A 469 2. greater risk to river eutrophication than agricultural phosphorus? Sci. Total 470 Environ. 2006, 360, (1), 246-253. 471 3. Cherry, K.; Shepherd, M.; Withers, P.; Mooney, S., Assessing the 472 effectiveness of actions to mitigate nutrient loss from agriculture: A review 473 of methods. Sci. Total Environ. 2008, 406, (1), 1-23. 474 United States Environmental Protection Agency Clean Water Act. 4. 475 http://www2.epa.gov/laws-regulations/summary-clean-water-act 476 European Union, Directive 2000/60/EC of the European Parliament 5. 477 and of the Council establishing a framework for the Community action in the 478 field of water policy. European Union: Brussels, 2000. 479

European Union, Council Directive 91/676/EEC of 12 December 1991 6. 480 concerning the protection of waters against pollution caused by nitrates from 481 agricultural sources European Union: Brussels, 1991. 482 7. European Union Council Directive of 21 May 1991 concerning urban 483 *waste water treatment (91/271/EEC)*; European Union: Brussels, 1991. 484 Sharpley, A.; Jarvie, H. P.; Buda, A.; May, L.; Spears, B.; Kleinman, P., 485 8. Phosphorus Legacy: Overcoming the Effects of Past Management Practices 486 to Mitigate Future Water Quality Impairment. J. Environ. Qual. 2013, 42, 487 (5), 1308-1326.488 Havgarth, P. M.; Jarvie, H. P.; Powers, S. M.; Sharpley, A. N.; Elser, J. J.; 9. 489 Shen, J.; Peterson, H. M.; Chan, N.-I.; Howden, N. J.; Burt, T., Sustainable 490 phosphorus management and the need for a long-term perspective: The 491 legacy hypothesis. Environ. Sci. Technol. 2014, 48, (15), 8417-8419. 492 Gooddy, D. C.; Lapworth, D. J.; Ascott, M. J.; Bennett, S. A.; Heaton, T. H. 10. 493 E.; Surridge, B. W. J., Isotopic Fingerprint for Phosphorus in Drinking Water 494 495 Supplies. Environ. Sci. Technol. 2015, 49, (15), 9020-9028. Ascott, M. J.; Gooddy, D. C.; Lapworth, D. J.; Stuart, M. E., Estimating 496 11. the leakage contribution of phosphate dosed drinking water to 497 environmental phosphorus pollution at the national-scale. Sci. Total 498 Environ. 2016, 572, 1534-1542. 499 12. Gooddy, D. C.; Ascott, M. J.; Lapworth, D. J.; Ward, R. S.; Jarvie, H. P.; 500 Bowes, M. J.; Tipping, E.; Dils, R.; Surridge, B. W. J., Mains water leakage: 501 Implications for phosphorus source apportionment and policy responses in 502 catchments. Sci. Total Environ. 2017, 579, 702-708. 503 Wakida, F. T.; Lerner, D. N., Non-agricultural sources of groundwater 504 13. nitrate: a review and case study. *Water Res.* **2005**, *39*, (1), 3-16. 505 Lerner, D. N., Estimating Urban Loads of Nitrogen to Groundwater. 14. 506 Water Environ. J. 2003, 17, (4), 239-244. 507 Lerner, D. N.; Yang, Y.; Barrett, M. H.; Tellam, J., Loadings of non-15. 508 agricultural nitrogen in urban groundwater. IAHS Publ. 1999, 117-124. 509 Bruna, G.; Paul, P.; Gilles, B.; Fayçal, B.; Josette, G.; Luis, L., The role of 16. 510 water nitrogen retention in integrated nutrient management: assessment 511 in a large basin using different modelling approaches. Environ. Res. Lett. 512 **2015,** *10*, (6), 065008. 513 17. Worrall, F.; Davies, H.; Burt, T.; Howden, N. J.; Whelan, M. J.; Bhogal, 514 A.; Lilly, A., The flux of dissolved nitrogen from the UK—Evaluating the role 515 of soils and land use. Sci. Total Environ. 2012, 434, 90-100. 516 Finlay, N. C.; Johnson, K.; Worrall, F., The role of water treatment 517 18. abstraction in the flux and greenhouse gas emissions from organic carbon 518 and nitrogen within UK rivers. Water Resour. Res. 2016, 52, (10), 8190-519 8201. 520

521 19. Comber, S. D.; Smith, R.; Daldorph, P.; Gardner, M. J.; Constantino, C.; Ellor, B., Development of a chemical source apportionment decision 522 support framework for catchment management. Environ. Sci. Technol. 523 **2013**, 47, (17), 9824-9832. 524 Zhang, Y.; Collins, A. L.; Murdoch, N.; Lee, D.; Naden, P. S., Cross sector 20. 525 526 contributions to river pollution in England and Wales: Updating waterbody scale information to support policy delivery for the Water Framework 527 Directive. *Environ. Sci. Policy* **2014**, *42*, (0), 16-32. 528 21. Water UK *Water resources long-term planning framework*; London, 529 2016. 530 22. Environment Agency Water Resources Planning Guideline: The 531 *Technical Methods and Instructions*; Environment Agency: Bristol, UK, 2012. 532 23. Drinking Water Inspectorate *Guidance on implementing the Water* 533 Supply (Water Ouality) Regulations in England and Wales: Drinking Water 534 Inspectorate: London, 2017; p 99. 535 536 24. Eionet UK Drinking Water Data Return 2011-2013 Revised Version. http://cdr.eionet.europa.eu/gb/eu/dwd/envs0x9ra 537 25. Reis, S.; Liska, T.; Steinle, S.; Carnell, E.; Leaver, D.; Roberts, E.; Vieno, 538 M.; Beck, R.; Dragosits, U., UK Gridded Population 2011 based on Census 539 2011 and Land Cover Map 2015. In NERC Environmental Information Data 540 Centre: 2017. 541 UK Groundwater Forum The Use of Groundwater. 26. 542 http://www.groundwateruk.org/Image-Gallery.aspx 543 British Geological Survey Hydrogeological maps. 544 27. http://www.bgs.ac.uk/research/groundwater/datainfo/hydromaps/home. 545 <u>html</u> (29/01/2015), 546 United Kingdom Accrediation Service LAB 37: Accreditation 547 28. Requirements for Sampling and Testing in Accordance with the Drinking 548 *Water Testing Specification (DWTS)*; United Kingdom Accrediation Service: 549 London, UK, 2013; p 13. 550 29. Chartered Institute for Water and Environmental Management Water 551 distribution system leakage in the UK. <u>http://www.ciwem.org/wp-</u> 552 content/uploads/2016/03/Water-distribution-network-leakage-in-the-553 UK.pdf 554 30. Ascott, M. J.; Gooddy, D. C.; Lapworth, D. J.; Davidson, P.; Bowes, M. J.; 555 Jarvie, H. P.; Surridge, B. W. J., Phosphorus fluxes to the environment from 556 mains water leakage: Seasonality and future scenarios. Sci. Total Environ. 557 **2018,** *636*, 1321-1332. 558 31. Drinking Water Inspectorate Nitrate. 559 http://dwi.defra.gov.uk/consumers/advice-leaflets/nitrate.pdf 560 Leip, A.; Achermann, B.; Billen, G.; Bleeker, A.; Bouwman, A.; de Vries, 32. 561 562 W.; Dragosits, U.; Doring, U.; Fernall, D.; Geupel, M., Integrating nitrogen

fluxes at the European scale. In *The European Nitrogen Assessment*, Sutton, 563 M.; Howard, C.; Erisman, J. W.; Billen, G.; Bleeker, A.; Greenfelt, P.; van 564 Grinsven, H.; Grizzetti, B., Eds. Cambridge University Press: Cambridge, UK, 565 2011; pp 345 - 376. 566 33. Office for National Statistics Population of England reaches 55 567 million. 568 https://www.ons.gov.uk/peoplepopulationandcommunity/populationand 569 migration/populationestimates/bulletins/annualmidyearpopulationestima 570 tes/latest#population-of-england-reaches-55-million 571 34. Environment Agency Updating NVZ Urban N to Groundwater layer; 572 Environment Agency,: Bristol, UK, 2013; p 17. 573 Bernot, M. J.; Dodds, W. K., Nitrogen retention, removal, and 35. 574 saturation in lotic ecosystems. *Ecosystems* **2005**, *8*, (4), 442-453. 575 Durand, P.; Breuer, L.; Johnes, P., Nitrogen processes in aquatic 36. 576 ecosystems. In *The European Nitrogen Assessment*, Sutton, M. A.; Howard, C. 577 578 M.; Erisman, J. W.; Billen, G.; Bleeker, A.; Greenfelt, P.; van Grinsven, H.; Grizzetti, B., Eds. Cambridge University Press: Cambridge, 2011. 579 37. Alexander, R. B.; Böhlke, J. K.; Boyer, E. W.; David, M. B.; Harvey, J. W.; 580 Mulholland, P. J.; Seitzinger, S. P.; Tobias, C. R.; Tonitto, C.; Wollheim, W. M., 581 Dynamic modeling of nitrogen losses in river networks unravels the 582 coupled effects of hydrological and biogeochemical processes. 583 Biogeochemistry 2009, 93, (1), 91-116. 584 38. Sedlak, R. I., Phosphorus and nitrogen removal from municipal 585 *wastewater: principles and practice.* CRC press: 1991. 586 Crittenden, J. C.; Trussell, R. R.; Hand, D. W.; Howe, K. J.; 587 39. Tchobanoglous, G., MWH's water treatment: principles and design. John 588 Wiley & Sons: 2012. 589 Camel, V.; Bermond, A., The use of ozone and associated oxidation 40. 590 processes in drinking water treatment. Water Res. 1998, 32, (11), 3208-591 3222. 592 41. Goncharuk, V.; Bagrii, V.; Mel'nik, L.; Chebotareva, R.; Bashtan, S. Y., 593 The use of redox potential in water treatment processes. J. Water Chem. 594 Techno. 2010, 32, (1), 1-9. 595 42. Gooddy, D. C.; Darling, W. G.; Abesser, C.; Lapworth, D. J., Using 596 chlorofluorocarbons (CFCs) and sulphur hexafluoride (SF6) to characterise 597 groundwater movement and residence time in a lowland Chalk catchment. 598 *J. Hydrol.* **2006**, *330*, (1–2), 44-52. 599 Loáiciga, H. A., Residence time, groundwater age, and solute output in 43. 600 steady-state groundwater systems. Adv. Water Resour. 2004, 27, (7), 681-601 688. 602

603	44. Ascott, M. J.; Gooddy, D. C.; Wang, L.; Stuart, M. E.; Lewis, M. A.; Ward,
604	R. S.; Binley, A. M., Global patterns of nitrate storage in the vadose zone.
605	Nat. Commun. 2017, 8, (1), 1416.
606	45. Wang, L.; Stuart, M.; Lewis, M.; Ward, R.; Skirvin, D.; Naden, P.; Collins,
607	A.; Ascott, M., The changing trend in nitrate concentrations in major
608	aquifers due to historical nitrate loading from agricultural land across
609	England and Wales from 1925 to 2150. <i>Sci. Total Environ.</i> 2016, <i>542</i> , 694-
610	705.
611	46. Ascott, M. J.; Wang, L.; Stuart, M. E.; Ward, R. S.; Hart, A.,
612	Quantification of nitrate storage in the vadose (unsaturated) zone: a
613	missing component of terrestrial N budgets. <i>Hydrol. Processes</i> 2016, <i>30</i> ,
614	(12), 1903-1915.
615	47. Ofwat Leakage.
616	https://www.ofwat.gov.uk/consumerissues/rightsresponsibilities/leakage
617	Δ
618	48. Environment Agency Final Water Resources Planning Guideline;
619	DEFRA: Bristol, 2016; p 38.
620	49. FAO AQUASTAT website.
621	http://www.fao.org/nr/water/aquastat/water_use/index.stm
622	50. Bouwman, A. F.; Beusen, A. H. W.; Griffioen, J.; Van Groenigen, J. W.;
623	Hefting, M. M.; Oenema, O.; Van Puijenbroek, P. J. T. M.; Seitzinger, S.; Slomp,
624	L. P.; Stehfest, E., Global trends and uncertainties in terrestrial
625	denitrification and N(2)U emissions. <i>Philos. Trans. R. Soc., B</i> 2013, 368,
626	(1021), 20130112 .
627 628	Developing Countries; World Bank: Washington DC, 2006.
629	52. Fowler, D.; Coyle, M.; Skiba, U.; Sutton, M. A.; Cape, J. N.; Reis, S.;
630	Sheppard, L. J.; Jenkins, A.; Grizzetti, B.; Galloway, J. N., The global nitrogen
631	cycle in the twenty-first century. Philos. Trans. R. Soc., B 2013, 368, (1621),
632	20130164.
633	53. World Health Organisation <i>Guidelines for Drinking-water Quality</i> ;
634	World Health Organisation: Geneva, Switzerland, 2011.
635	
636	
637	
638	
639	
640	

Figures



647

648 Figure 1 Water Supply Zones (WSZs) and principal regions in England



650

651 Figure 2 Distribution of ABS-NO₃-N in England for 2015



655 Figure 3 MWL-NO₃-N fluxes for WSZs in England for 2015



657 Figure 4 Percentage of MWL-NO₃-N to all N sources (as derived from SEPARATE²⁰) at the WRZ scale in England for 2015

656

659



662 Figure 5 Percentage change in MWL-NO₃-N flux compared to 2015 baseline in receiving WRZs following implementation of 663 treated drinking water transfers by 2020

671 **8 Tables**

672

Table 1 Total flux of nitrate removed from the aquatic environment by abstraction (ABS-NO₃-N) in England for 2015 in

674 comparison to previous estimates of aquatic N retention (organic N retention and storage) and removal processes

675 (denitrification)

Flux Name	Reference	Flux Value (kt N/yr)	ABS-NO₃- N/Flux Value (-)
ABS-NO ₃ -N	This study	24.2	-
Organic N retention by abstraction	Finlay et al. (2016)	3.9	6.21
Organic N storage in floodplains	Finlay et al. (2016)	0.8	30.03
Organic N storage in channel	Finlay et al. (2016)	1.6	15.02
Denitrification in the hydrosphere (rivers,			
groundwater and coastal marine	Leip et al. (2011);		
environments)	Worrall et al. (2012)	62 -716	0.39 – 0.03

676

677

678

679 Table 2 Total flux of nitrate entering the environment from mains water leakage (MWL-NO₃-N) in England for 2015 in

680 comparison to previous estimates of N sources (MWL-NO₃-N, N from leaking sewers, all N sources, urban diffuse N sources).
 681 MWL-NO₃-N/Flux Value refers to the MWL-NO₃-N estimate made in this study.

Flux Name	Reference	Flux Value (kt N/yr)	MWL-NO ₃ -N/Flux Value (-)
MWL-NO ₃ -N	This study	3.62	-
MWL-NO3-N	Environment Agency (2013)	20.31	0.178
SEWER-N	Environment Agency (2013)	4.06	0.891
All N sources	Zhang et al. (2014)	277	0.013
Urban diffuse N sources	Zhang et al. (2014)	2.31	1.567

682

683

684

685

686