1	Developing Global Pedotransfer Functions to Estimate Available Soil Phosphorus
2	
3	Yones Khaledian ¹ , John N. Quinton ² , Eric C. Brevik ³ *, Paulo Pereira ⁴ , Mojtaba
4	Zeraatpisheh⁵
5	
6	1 – Department of Agronomy, Iowa State University, Ames, IA, USA
7	2 – Lancaster Environment Centre, Lancaster University, Lancaster, UK
8	3 – Department of Natural Sciences, Dickinson State University, Dickinson, ND, USA;
9	Eric.Brevik@dickinsonstate.edu
10	4 – Environmental Management Center, Mykolas Romeris University, Ateities g. 20, LT-08303 Vilnius, Lithuania
11	5 – Department of Soil Science, College of Agriculture, Isfahan University of Technology, 84156-83111 Isfahan, Iran
12	* - corresponding author
13	
14	Abstract
15	There are a large number of investigations that estimate available soil phosphorous (P),
16	but a paucity of global data on available soil P. One significant modern challenge is developing
17	low cost, accurate approaches to predict available soil P that are useful to scientists around the
18	world. We conducted a global meta-analysis using data on available soil P from 738 sites, 640 in
19	the USA and 149 in 14 other countries. Four different methods of determining available soil P,
20	New Zealand (NZ), acid oxalate, Bray and Mehlich 3 were represented in the dataset. Inputs
21	evaluated for inclusion in the pedotransfer functions to predict available soil P were clay (C), fine
22	silt, (FSi) coarse silt (CSi), very fine sand (VFS), fine sand (FS), medium sand (MS), coarse sand (CS),
23	very coarse sand (VCS), organic carbon (OC), pH, calcium (Ca), magnesium (Mg), potassium (K),

24 iron (Fe), aluminum (Al), and manganese (Mn). Available soil P was estimated for: 1) the entire 25 dataset, 2) only the USA, and 3) the non-USA dataset. The best models to estimate available soil 26 P were obtained for the NZ method (using the co-variates C, FSi, CSi, VFS, MS, CS, OC, Fe, Al, Mn, 27 Ca, Mg, and pH) and for the acid oxalate method (using the co-variates C, FSi, Fe, Al, Mn, Ca, and 28 Mg). Although estimation of available soil P determined with the acid oxalate method was poor for the entire dataset, good estimates were obtained for the USA and non-USA datasets 29 separately. Models for the Bray and Mehlich 3 methods only predicted available soil P well for 30 the non-USA dataset. Using pedotransfer function models to estimate available soil P could 31 provide an efficient and cost effective way to estimate global distributions of a soil property that 32 is important for a number of agricultural and environmental reasons. 33

34

35 Key Words: Genetic Algorithm; Soil Properties; Simulation; Statistical Models

36

37 1. Introduction

Soil phosphorus (P) is essential to living organisms and often a limiting factor for plant growth and biomass production in forests, grasslands, and croplands (Trichet et al., 2009; Smith et al., 2015). It is also a limiting element for algal growth, thus contributing to eutrophication and subsequent effects on ecosystem services associated with freshwater (Carpenter et al., 2008).

Soil P is among the most limited nutrients in the soil system because much of it is in forms that are not available to plants. This has resulted in the soil P cycle being strongly affected by human management and land use, especially in agricultural areas where P has been used intensively to fertilize the soils in industrialized agricultural systems to boost the availability of P to crops (Bryant et al., 2006). The use of P fertilizers has resulted in large soil P surpluses in
countries such as China, India, and the United States (West et al., 2014) as well as parts of Europe
(Lemercier et al., 2008, Follain et al., 2009), creating a risk of ecosystem contamination.
Moreover, P is indispensable in physiological and biochemical processes (Simpson et al., 2011).
Hence, the availability of P controls the growth and development of crops (Wyngaard et al., 2016).

In soils, P is derived mainly from weathering of the primary mineral apatite (Zhou et al., 52 53 2018) and the addition of inorganic and organic fertilizer. Total P content is high in most soils, however, only a small portion (about 13%; Sayers et al., 2008) is available for agricultural plants 54 and microorganisms, because most P is bound strongly in soil particles and incompletely 55 56 weathered material or is occluded as a secondary mineral. It is estimated that worldwide soil P content (0-50 cm) is 3.6±3 Pg in labile form, 8.6±6 Pg in organic form, 12.2±8 Pg in occluded form, 57 58 and 3.2±2 Pg in secondary pools (Yang et al., 2013). The average total P in soils ranges from 200 mg/kg in older/highly weathered soils to 800 mg/kg in younger/less developed soils (ljaz et al., 59 2017); the average amount of organic P in soils ranges between 30% and 65% of the total P 60 (Condron and Tiessen, 2005). The concentration of P in the soil solution is influenced by 61 interactions between physical, chemical and biological processes (Hinsinger, 2013; Messiga et al., 62 2015). P adsorption and desorption are complex processes that depend on the charge 63 64 characteristics of the mineral and organic surfaces and the concentration and forms of P in the soil solution (Haygarth et al. 2005). These processes are highly pH dependent, as this alters both 65 the characteristics of the phosphate ions in solution and the charge characteristics of the soil 66 67 surfaces. In soils with high aluminum (AI) and iron (Fe) oxide contents P may be strongly bound

to mineral surfaces. However, in soils rich in carbonates (CaCO₃) both surface reactions between
P and precipitation of P-CaCO₃ compounds occur (von Wandruszka, 2006) and various Caphosphate minerals form (Beauchemin et al., 2003), making the P unavailable to plants (Haygarth
et al., 2013).

72 Since P is often a limiting soil factor for plant growth and excess P can have negative impacts on the environment, better ways to estimate soil P levels and spatial distribution would 73 be useful to help scientists and stakeholders manage this important element, especially in 74 75 developing nations (Borggaard et al., 2004, Krause et al., 2016). This latter point is extremely important because of the lack of resources to conduct frequent and thorough soil 76 characterizations in these areas (Khaledian et al., 2017a, b). One option for estimating available 77 78 soil P contents is to use pedotransfer functions. These consist of equations or sets of equations that allow the value of a soil property to be estimated from other properties that can be 79 80 determined with simpler, faster, more efficient and/or less expensive techniques or that is 81 already available due to previous work. Pedotransfer functions have usually been developed 82 from multiple linear regression (MLR) models. Recently some functions have been developed using genetic algorithms (GA) to optimize the coefficients that may have been utilized in MLR 83 84 (Hosseini et al., 2016; Khaledian et al., 2017a).

There have been previous attempts to develop pedotransfer functions related to available soil P, however, these have been limited in their geographic extent. Kleinman et al. (1999) developed a pedotransfer function to estimate available soil P using 59 soil samples in the USA and achieved R²=0.91 for multivariate pedotransfer functions containing Al, Fe, soil organic matter, and pH as input variables. Likewise, researchers in the UK (Withers et al., 2017, R² from

0.64-0.87), Brazil (Valadares et al., 2017b, R² = 0.73 and 0.94), Scandinavia (Mundus et al., 2017, 90 91 R² ranged from 0.46 to 0.83 using a Mitscherlich equation) and Iran (Seilsepour et al., 2008, R²=0.92; Keshavarzi et al., 2016, R²= 0.61 for training set and R²= 0.50 for testing set; Hosseini et 92 al., 2017, R² from 0.77-0.91) estimated available soil P based on several soil properties. However, 93 94 in each of these studies the samples came from a limited geographic distribution and focused on 95 a single available soil P test. Other studies involving pedotransfer functions for P-related issues have focused on soil phosphate adsorption capacity (Borggaard et al., 2004), available phosphate 96 97 ions (Achat et al., 2011), phosphate adsorption-desorption curves (Peña and Torrent, 1990; Scheinost and Schwertmann, 1995), P mobilization (Borda et al. 2010), or remaining P (Cagliari et 98 99 al., 2011). With the exception of Borggaard et al. (2004), which included a wide geographic range 100 of samples (Canada, Denmark, Ghana, and Tanzania), these were all local (field scale) or regional 101 (single country or a portion of a country) studies.

In this paper we aim 1) to provide the first pedotransfer functions for the global estimation of
available soil P, utilising a dataset that consisted of available soil P from world-wide locations and
to evaluate pedotransfer functions for multiple soil P tests for the global data and national
data sets from the global, USA, and non-USA locations.

106

107 **2. Materials and Methods**

108 2.1. Data Collection

109 We collected data using the USA Natural Resources Conservation Services (NRCS) and 110 National Cooperative Soil Survey (NCSS) database (<u>http://ncsslabdatamart.sc.egov.usda.gov/</u>) 111 targeting the collection of data generated between 2000 and 2014. This process resulted in 738 available soil P samples, 640 in the USA and 149 in other nations; Afghanistan, Canada, Chile,
China, Costa Rica, Denmark, Ghana, Guatemala, Mongolia, Nigeria, Poland, Puerto Rico, Russia
and Spain (14 countries) (Figure 1).

115 The following information was available for each sampling point: clay (C), fine silt (FSi), 116 coarse silt (CSi), very fine sand (VFS), fine sand (FS), medium sand (MS), coarse sand (CS), very 117 coarse sand (VCS), organic carbon (OC), pH, calcium (Ca), magnesium (Mg), potassium (K), Fe, Al, and manganese (Mn). Soil texture was determined with sieve and pipette (NRCS analysis code 118 119 3A1a). Soil OC was measured with acid dichromate digestion and FeSO₄ titration (analysis code 120 6A1c). Soil pH was determined using a 1:1 water to soil ratio (analysis code 4C1a2a). Fe, Al, and 121 Mn were determined with ammonium oxalate extraction (analysis codes 6C9b, 6G12b, and 6D5b, 122 respectively). Ca, Mg, and K were determined as NH₄OAC extractable bases (analysis codes 6N2e, 6O2d, and 6Q2b, respectively). Four different approaches for determining available P were 123 124 investigated, the New Zealand (NZ) (analysis code 6S4), acid oxalate (analysis code 4G2), Bray 125 (analysis code 4D3a1) and Mehlich 3 (analysis code 4D6a1) methods.

Prior to data analysis, observations were grouped into categories depending on their location. Owing to the large number of potential variables being evaluated to create the pedotransfer functions, particular attention was given to determining variables that were related to available soil P. Therefore, before obtaining the pedotransfer formula, we omitted the variables that did not have a significant effect on estimation of soil P using a stepwise regression model.

132

133 **2.2. Statistical Methods - Data analysis.**

After selecting the variables that most influenced the prediction of available soil P, we tested the fit of linear (y = ax + c) and nonlinear quadratic regression ($y = ax^2 + bx + c$) formulas for 1) the entire dataset, 2) the USA, and 3) the other countries (14 countries). Following that, the formulas were optimized by genetic algorithms, and finally the fitted graphs prepared by test samples. All analysis was performed using XLSTAT publisher (Data Analysis for Microsoft Excel, Addinsoft, Paris, France)

Genetic algorithms solve complex optimization issues based on natural genetics; see 140 141 Holland (1975) for details and Cropper and Comerford (2005) and Johari et al. (2010) for soil science applications. The algorithm used included three basic operators: the selection, crossover 142 143 and mutation (Cieniawski et al., 1995) (Figure 2). Genetic algorithms are based on the concept of 144 evolution by natural selection as solutions are evolved in a stochastic, iterative manner. The algorithm procedure consists of the following steps (Nelson and Odeh, 2009): 1. Start at initial 145 146 time (t= 0); 2. Initialize a population of individuals (rules) P(t); 3. Evaluate the fitness of each 147 individual by evaluating how well a rule predicts the distribution using a random subset of 148 observations (training dataset) and save the best individuals in a rule archive; 4. Test against fitness criterion and terminate this rule archive if the criterion is met; otherwise 5. Increase time 149 150 counter; 6. Create a new set of individuals using the rule archive and random generators; 7. Apply 151 heuristic operators to population; 8. Go back to 3.

For all methods and datasets about 70% of the data was randomly selected to train the models and the remaining data (around 30%) was used as a validation dataset. The one exception was the data for the other countries determined with the Mehlich 3 method, which only had 12 samples. Therefore, the entire other countries Mehlich 3 dataset was used for full-cross validation in this one instance. Finally, in order to find the influential variables in different subsets, the correlation (Pearson) of variables was considered. Significant differences were considered at a p<0.05.

159

160 **3. Results**

To find the best regression model(s) to predict soil P by different extraction methods, six 161 regression equations (linear and non-linear equations) were evaluated for all datasets. Table 1 162 163 gives the values of the fit indictors with the pedotransfer formulas. The NZ method had promising 164 R^2 and RMSE values for all evaluated datasets (Table 1); however, a more accurate model was produced for the USA than for the entire dataset and the other locations' data (Table 1). The 165 166 formulas for linear regression in Table 1 indicate that when predicting available soil P around the globe, the importance of AI, Fe and Mn should be considered, because higher R² and RMSE values 167 168 were achieved when including them. The most influential variables in the nonlinear and linear 169 regression is AI (Table 2). In the USA, AI and Fe played influential roles in estimating available soil P, while including Mn in the regression achieved higher coefficients in the other nations' 170 equations. The R² values for all nonlinear models in the three datasets (R²=0.78, 0.82 and 0.84 171 for the entire dataset, the USA, and the other nations, respectively) were better than for the 172 173 linear models, but the linear regression for the USA was also promising (R^2 = 0.77).

The acid oxalate method had higher R² and lower RMSE values for the linear regressions for the entire dataset and the USA locations, and for both the linear and nonlinear regressions for the other country locations. However, R² and RMSE values for the non-linear regression were not significant for the USA dataset (Table 1). Al played an important role in predicting available soil P in almost all the formulas. The importance of Ca, Al and Mn in the entire dataset and Ca, Mg and Al in other parts of the world is clear (Table 1, 2). The nonlinear and linear models (R²=0.62 and 0.85) for other countries and the linear (R²=0.60) and nonlinear (R²=0.42) models for the USA showed promising results, but the entire dataset models did not. Linear regression models estimated available soil P better than the nonlinear regression models for the entire dataset (Table 1). Nevertheless, accuracy indicators (R² and RMSE, Table 1) showed that the input variables could not adequately predict available soil P using the entire dataset.

185 Results from the Bray and Mehlich 3 methods showed that apart from the USA dataset 186 for the Mehlich 3 method and the other country locations for the Bray method, the R² and RMSE 187 values for the other regressions were not promising (Table 1, 2).

188

189 **4. Discussion**

190 Compared to previous work, the study has considered samples from a wide geographic 191 distribution. The fact that available P across a wide range of soil and environmental settings could be predicted using a number of relatively stable soil properties, such as sand, silt, and clay, along 192 193 with other soil properties that could potentially change over shorter time periods, such as pH, 194 extractable Ca, and total organic carbon provides the opportunity to make rapid regional and global assessments of P availability, something that is of use to both agricultural and 195 196 environmental planners. Although different methods have been developed to test for available 197 soil P, often because of the different soil conditions and properties that are encountered and different crops that are produced in various parts of the world, relationships with good predictive 198

ability were developed for both the NZ and acid oxalate available P methods. However,relationships for the Mehlich 3 test were poorer.

The pedotransfer functions allowed identification of the key variables controlling P 201 202 availability for large scale assessments. Except for the Bray method, at least one of Al, Fe, and 203 Mn were used in all the pedotransfer functions to predict available soil P (Table 1), indicating the 204 importance of these variables. Available soil P is known to be closely related to the concentration of Fe and Al (Vincent et al., 2012; Seguel et al., 2015) as Al and Fe oxides are the main phosphate 205 adsorbents in soils (Borggaard, 1990, 2002). Borggaard et al. (2004) found close relationships 206 207 between amounts of adsorbed phosphate and soil contents of some forms of Al and Fe and created pedotransfer functions for predicting adsorbed phosphate from Al and Fe oxide contents 208 209 that were quantified by selective extractions. The relationship between available soil P and Fe 210 was weaker in this study than for Al, agreeing with the findings of Karlsson et al. (2008) and 211 Turner et al. (2003).

212 Soil texture is another important factor for predicting available soil P in the pedotransfer functions (Table 1). Texture can substantially affect the chemical characteristics of soil, including 213 the formation of stable P-Al-organic matter bonds and P leaching from soil (Negassa and 214 Leinweber, 2009; Sugihara et al., 2012), which in turn is linked to plant available P. The influence 215 216 of fine particles such as C, FSi and CSi were important in all pedotransfer functions except those 217 for the Mehlich 3 method (Table 1). Clay content was significantly correlated to P in almost all the models, as P can be adsorbed on the surfaces of clay minerals (Shen et al., 2011; Ulén and 218 219 Snäll, 2007). Clay mineralogy is also important for determining the P-sorption capability of clays. 220 Clay minerology was not a variable investigated in this study and represents a potential direction

of future research. For the Mehlich 3 method, coarse soil texture components, such as CS, VCS
 and CSi were important in the prediction equations for the USA dataset. This may be because soil
 texture is important in determining P leaching from soils through its influence on soil hydrology
 (Negassa and Leinweber, 2009).

The importance of Ca and Mg in the entire dataset and in other parts of the world for the acid oxalate method is interesting, given that the acid oxalate method is typically used in acidic soils. This could be because the acid oxalate method causes excessive solubilization of non-labile P pools, (Six et al., 2012; Valadares et al., 2017a), as e.g. Ca-P, therefore Ca shows amongst the predictor variables because Ca from Ca-P is being released into solution. Mg is probably a predictor variable given that Ca and Mg have similar chemical behavior in soils (Havlin et al., 2005).

In our study soil OC was used in the pedotranfer functions for all available soil P methods, except for the acid oxalate method. Soil OC can influence available soil P by providing binding sites (Kang et al., 2009), and the transformation of organic P strongly influences overall P bioavailability in soils (Shen et al., 2011). Yang et al. (2013) note that organic P is a major part of the global soil P pool.

It is clear that agricultural practices will have a significant influence over available P and, as we have not included them as variables in our work, will be one of the major sources of uncertainty in the study. Given that fertilizer application accounts for 42 % of total P inputs to agricultural ecosystems, or 25 % of total plant P uptake in fertilized soils (Wang et al., 2010), fertilizer inputs of P are likely to account for a meaningful amount of the variation not explained by the pedotransfer models presented here. Therefore future attempts to develop available P pedotransfer functions may benefit from finding a way to account for fertilizer inputs. Our
 pedotransfer functions also did not incorporate crops, something that represents a limitation for
 the equations.

246

247 **5. Conclusions**

This research fills a clear gap in our ability to estimate available soil P at a global scale. Pedotransfer functions were developed for the common available soil P tests for global datasets Non-linear functions provided the best predictions of available soil P for global and non-global datasets. Including fertilizer and soil management practices may improve these predictions further.

253 Our work has demonstrated that the development of pedotransfer functions for available 254 soil P over large areas using routinely collected soil survey data is possible. This could help 255 address issues such as food security by providing initial estimates of soil fertility and global 256 environmental concerns related to available soil P.

257

258 References

Achat, D.L., Augusto, L., Morel, C., Bakker, M.R., 2011. Predicting available phosphate ions from
 physical-chemical soil properties in acidic sandy soils under pine forests. Journal of Soils
 Sediments 11, 452–466.

Beauchemin, S., Hesterberg, D., Chou, J., Beauchemin, M., Simard, R.R., Sayers, D.E., 2003.
 Speciation of phosphorus in phosphorus-enriched agricultural soils using x-ray absorption

- 264 near-edge structure spectroscopy and chemical fractionation. Journal of Environmental
 265 Quality 32, 1819-1819.
- Borda, T., Withers, P.J.A., Sacco, D., Zavattaro, L., Barberi, E., 2010. Predicting mobilization of
 suspended sediments and phosphorus from soil properties: a case study from the
 northwest Po valley, Piemonte, Italy. Soil Use and Management 26, 310–319.
- Borggaard, O.K., 1990. Dissolution and Adsorption Properties of Soil Iron Oxides. DSc Thesis.
 Royal Veterinary and Agricultural University, Copenhagen.
- 271 Borggaard, O.K., 2002. Soil Chemistry in a Pedological Context, 6th ed. DSR Forlag, Frederiksberg.
- Borggaard, O.K., Szilas, C. Gimsing, A.L., Rasmussen L.H. 2004. Estimation of soil phosphate
 adsorption capacity by means of a pedotransfer function. Geoderma 118, 55–61.
- Bryant, R., Gburek, W., Veith, T., Hively, W., 2006. Perspectives on the potential for
 hydropedology to improve watershed modeling of phosphorus loss. Geoderma, 131, 299–
 307.
- Cagliari, J., Veronez, M.R., Alves, M.E., 2011. Remaining phosphorus estimated by pedotransfer
 function. Revista Brasileira de Ciencia do Solo 35, 203–212.
- Carpenter, S.R., 2008. Phosphorus control is critical to mitigating eutrophication. Proceedings of
 the National Academy of Sciences of the United States of America 105, 11039–11040.
- 281 Condron L.M., Tiessen H., 2005. Interactions of organic phosphorus in terrestrial ecosystems. In:
- Turner B.L., Frossard E., Baldwin D.S. (eds): Organic Phosphorus in the Environment.
 Wallington, CAB International, 295–307.

284	Cropper, W., Comerford, N., 2005. Optimizing simulated fertilizer additions using a genetic
285	algorithm with a nutrient uptake model. Ecological Modelling 185, 271–281.

Follain, S., Schvartz, C., Denoroy, P., Villette, C., Arrouays, D., Walter, C., Lemercier, B., Saby,

- 287 N.P.A. 2009. From quantitative to agronomic assessment of soil available phosphorus
- content of French arable topsoils. Agronomy for Sustainable Development 29, 371-380.
- Havlin, J.L., Beaton, J.D., Tisdale, S.L., Nelson, W.L. 2005. Soil fertility and fertilizers, 7th Ed.
 Pearson-Prentice Hall, Upper Saddle River, NJ.
- Haygarth, P., Condron, L., Heathwaite, A.L., Turner, B., Harris, G., 2005. The phosphorus transfer
- 292 continuum: linking source to impact with an interdisciplinary and multi-scaled approach.
 293 Science of the Total Environment 344, 5–14.
- Haygarth, P.M., Bardgett, R.D., Condron, L.M., 2013. Nitrogen and phosphorus cycles and their
 management, Soil Conditions and Plant Growth. Blackwell Publishing Ltd, pp. 132–159.
- Hinsinger, P., 2013. Plant-induced changes in soil processes and properties. In: Gregory, P.J.,
- 297 Nortcliff, S. (Eds.), Soil Conditions and Plant Growth. Blackwell Publishing Ltd., Oxford.
- Hosseini, M., Agereh S.R., Khaledian Y., Zoghalchali H.J., Brevik E.C., Naeini S.A.R.M. 2017.
- Comparison of multiple statistical techniques to predict soil phosphorus. Applied Soil
 Ecology 114, 123–131.
- Hosseini, M., Naeini, S.A.R.M., Dehghani A.A., Khaledian, Y., 2016. Estimation of soil mechanical
 resistance parameter by using particle swarm optimization, genetic algorithm and
 multiple regression methods. Soil and Tillage Research 157, 32–42.
- Ijaz, W., Ahmed, M., Asim, M., Aslam, M., 2017. Models to Study Phosphorous Dynamics Under
 Changing Climate, Quantification of Climate Variability, Adaptation and Mitigation for

306	Agricultural Sustainability, in: Ahmed, M., Stockle, C.O. (eds.), Quantification of Climate
307	Variability, Adaptation and Mitigation for Agricultural Sustainability. Springer, Cham. pp.
308	371-386.
309	Johari, A., Javadi, A., Habibagahi, G.A., 2010. Modelling the mechanical behaviour of unsaturated
310	soils using a genetic algorithm-based neural network. Computers and Geotechnics 38, 2–
311	13.
312	Kang, J., Hesterberg, D., Osmond, D.L., 2009. Soil organic matter effects on phosphorus sorption:
313	A path analysis. Soil Science Society of America Journal 73, 360–366.
314	Karlsson, T., Persson, P., Skyllberg, U., Morth, C.M., Giesler, R., 2008. Characterization of Iron (III)
315	in organic soils using extended X-ray absorption fine structure spectroscopy.
316	Environmental Science and Technology 42, 5449–5454.
317	Keshavarzi, A., El-Sayed, E.O., Bateni, S.M., Pradhan, B., Vasu, D., Bagherzadeh, A. 2016. Modeling
318	of available soil phosphorus (ASP) using multi-objective group method of data handling.
319	Modeling Earth Systems and Environment 2, 157. doi:10.1007/s40808-016-0216-5
320	Khaledian, Y., Brevik, E.C., Pereira, P., Cerdà, A., Fattah, M.A., Tazikeh, H., 2017a. Modeling soil
321	cation exchange capacity in multiple countries. Catena 158, 194–200.
322	Khaledian, Y., Kiani, F., Ebrahimi, S., Brevik, E.C., Aitkenhead-Peterson, J., 2017b. Assessment and
323	monitoring of soil degradation during land use change using multivariate analysis. Land
324	Degradation and Development 28, 128–141.

325	Kleinman, P.J., Bryant, R., Reid, W., 1999. Development of pedotransfer functions to quantify
326	phosphorus saturation of agricultural soils. Journal of Environmental Quality 28, 2026-
327	2030.

- Krause, A., Nehls, T., George, E., Kaupenjohann, M., 2016. Organic wastes from bioenergy and
 ecological sanitation as a soil fertility improver: a field experiment in a tropical Andosol.
 SOIL 2, 147–162.
- Lemercier, B., Gaudin, L., Walter, C., Aurousseau, P., Arrouays, D., Schvartz, C., Saby, N.P.A.,
 Follain, S., Abrassart, J., 2008. Soil phosphorus monitoring at the regional level by means

of a soil test database. Soil Use and Management 24, 131–138.

- Messiga, A.J., Ziadi, N., Mollier, A., Parent, L.E., Schneider, A., Morel, C., 2015. Process-based mass-balance modeling of soil phosphorus availability: Testing different scenarios in a long-term maize monoculture. Geoderma 243-244, 41–49.
- 337 Mundus, S., Carstensen, A., Husted, S., 2017. Predicting phosphorus availability to spring barley

338 (*Hordeum vulgare*) in agricultural soils of Scandinavia. Field Crops Research 212, 1–10.

- 339 Negassa, W., Leinweber, P., 2009. How does the Hedley sequential phosphorus fractionation
- reflect impacts of land use and management on soil phosphorus: a review. Journal of Plant
 Nutrition and Soil Science 172, 305–325.
- Nelson, M., Odeh, I., 2009. Digital soil class mapping using legacy soil profile data: a comparison
- of a genetic algorithm and classification tree approach. Soil Research, 47(6), 632-649.
- Peña, F., Torrent, J., 1990. Predicting phosphate sorption in soils of Mediterranean regions.

345 Fertilizer Research 23, 173–179.

347	Reconciling changing concepts of soil phosphorus behaviour with agronomic information
348	FAO Fertilizer and Plant Nutrition Bulletin. Food and Agriculture Organization of the
349	United Nations, Rome.
350	Scheinost, A.C., Schwertmann, U., 1995. Predicting phosphate adsorption-desorption in a soi
351	scape. Soil Science Society of America Journal 59, 1575–1580.
352	Seguel, A., Barea, J.M., Cornejo, P., Borie, F., 2015. Role of arbuscular mycorrhizal symbiosis ir
353	phosphorus-uptake efficiency and aluminum tolerance in barley growing in acid soils.
354	Crop and Pasture Science 66, 696–705.
355	Seilsepour, M., Rashidi, M., Khabbaz, B.G., 2008. Prediction of soil available phosphorus based
356	on soil organic carbon. American-Eurasian Journal of Agriculture and Environmental
357	Science 4, 189–193.
358	Shen, J., Yuan, L., Zhang, J., Li, H., Bai, Z., Chen, X., Zhang, W., Zhang, F., 2011. Phosphorus
359	Dynamics: From Soil to Plant. Plant Physiology 156, 997–1005.
360	Simpson, R.J., Oberson, A., Culvenor, R.A., Ryan, M.H., Veneklaas, E.J., Lambers, H., Lynch, J.P.,
361	Ryan, P.R., Delhaize, E., Smith, F.A., Smith, S.E., Harvey, P.R., Richardson, A.E., 2011
362	Strategies and agronomic interventions to improve the phosphorus-use efficiency of
363	farming systems. Plant Soil 349, 89–120.
364	Six, L., Pypers, P., Degryse, F., Smolders, E., Merckx, R. 2012. The performance of DGT versus
365	conventional soil phosphorus tests in tropical soils - An isotope dilution study. Plant and
366	Soil 359, 267–279.

Sayers, J.K., Johnston, A.E., Curtin, D. 2008. Efficiency of soil and fertilizer phosphorus use:

367	Smith, D.R., Francesconi, W., Livingston, S.J., Huang, C., 2015. Phosphorus losses from monitored
368	fields with conservation practices in the Lake Erie Basin, USA. Ambio 44, 319–331.
369	Sugihara, S., Funakawa, S., Nishigaki, M., Kilasara, M., Kosaki, T., 2012. Dynamics of fractionated
370	P and P budget in soil under different land management in two Tanzanian croplands with
371	contrasting soil textures. Agriculture Ecosystems and Environment 162, 101–107.
372	Trichet, P., Bakker, M.R., Augusto, L., Alazard, P, Merzeau, D., 2009. Fifty years of pine fertilization
373	experiments in the Landes of Gascogne (France). Forest Science 55, 390–402.
374	Turner, B.L., Mahieu, N., Condron, L.M., 2003. Quantification of myo-inositol hexakisphosphate
375	in alkaline soil extracts by solution P-31 NMR spectroscopy and spectral deconvolution.
376	Soil Science 168, 469–478.
377	Ulén, B., Snäll, S. 2007. Forms and retention of phosphorus in an illite-clay soil profile with a
378	history of fertilisation with pig manure and mineral fertilisers. Geoderma 137, 455-465.
379	Valadares, S.V., Alvarez V, V.H., Santos, W.O., Paes, J.L.D.A., Lins, C.B. and Novais, R.F., 2017a.
380	Sensitivity of soil P availability tests to Ca-P in Oxisols. Communications in Soil Science and
381	Plant Analysis 48, 1834–1842.
382	Valadares, S.V., Cropper, W.P., Lima Neves, J.C., Leite, H.G., Barros, N.F., Gerber, S., 2017b.
383	Pedotransfer Functions to Estimate Parameters for Soil Phosphorus Models. Soil Science
384	Society of America Journal 81, 210–213.
385	Vincent, A.G., Schleucher, J., Grobner, G., Vestergren, J., Persson, P., Jansson M., Giesler R., 2012.
386	Changes in organic phosphorus composition in boreal forest humus soils: the role of iron
387	and aluminum. Biogeochemistry 108, 485–499.

- 388 Von Wandruszka, R., 2006. Phosphorus retention in calcareous soils and the effect of organic
 389 matter on its mobility. Geochemical Transactions 7, 6.
- 390 West, P.C., Gerber, J.S., Engstrom, P.M., Mueller, N.D., Brauman, K.A., Carlson, K.M., Cassidy, E.S.,
- Johnston, M., MacDonald, G.K., Ray, D.K., Siebert, S., 2014. Leverage points for improving
 global food security and the environment. Science 345, 325-328.
- 393 Withers, P.J., Hodgkinson, R.A., Rollett, A., Dyer, C., Dils, R., Collins, A.L., Bilsborrow, P.E., Bailey,
- 394 G., Sylvester-Bradley, R., 2017. Reducing soil phosphorus fertility brings potential long-
- term environmental gains: A UK analysis. Environmental Research Letters 12, p.063001.
- Wyngaard, N., Cabrera, M.L., Jarosch, K.A., Bunemann, E.K., 2016. Phosphorus in the coarse soil
- fraction is related to soil organic phosphorus mineralization measured by isotopic
 dilution. Soil Biol. Biochem. 96, 107–118.
- Yang, X., Post, W.M., Thorton, P.E., Jain, A., 2013. The distribution of soil phosphorous for global
 biogeochemical modeling. Biogeosciences 10, 2525–2537.
- Zhou, J., Bing, H., Wu, Y., Sun, H., Wang, J., 2018. Weathering of primary mineral phosphate in
 the early stages of ecosystem development in the Hailuogou Glacier foreland
 chronosequence. European Journal of Soil Science, 69(3), 450-461.

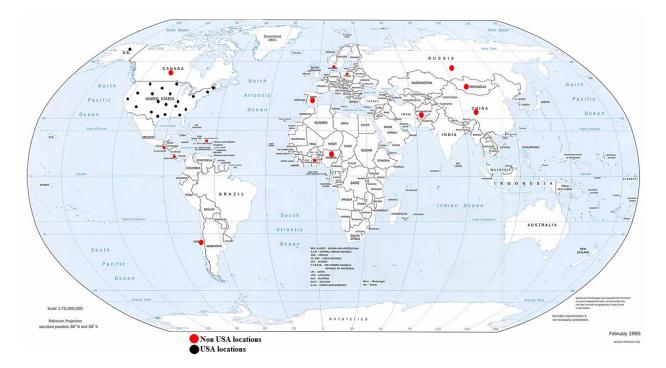


Figure 1. The location of sites that supplied data for this study.

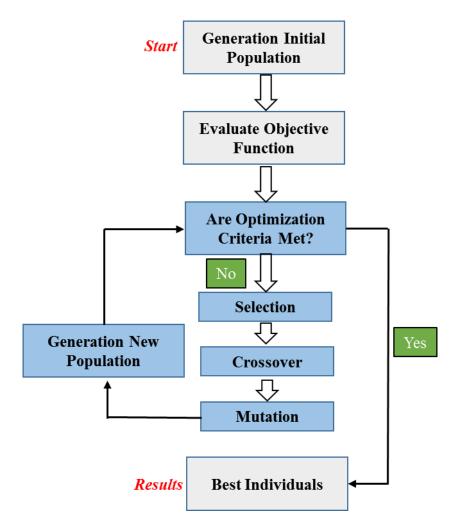


Figure 2. A flowchart showing how genetic algorithms function.

Methods	Locations	Training Set	Validation Set	Range	Regression method	R ²	RMSE	P-value	Equation
					Linear	0.69	14.59	0.00	NZ = 66.73+3.83E-02*CSi-0.73*VFS- 0.11*CS+8.93*Fe+37.30*Al+1.58E- 02*MN+0.16*Ca-4.29E-02*Mg-7.44*pH
	The Whole World	129	55	96.0	Non-Linear	0.78	13.42	0.00	NZ = 61.86+1.55*CSi-2.31*VFS- 37.26*Fe+91.39*Al+1.71E-02*Mn- 0.40*Ca+0.36*Mg-7.34*pH-3.67E- 02*CSi^2+5.46E-02*VFS^2+17.11*Fe^2- 23.05*Al^2+1.78E-07*Mn^2+5.92E- 03*Ca^2-3.50E-03*Mg^2+0.15*pH^2
	The USA				Linear	0.77	12.84	0.00	NZ = 20.64+0.13*C-0.67*VFS+0.43*MS- 0.68*CS+0.37*OC+15.14*Fe+39.66*AI
NZ	Alaska, California, Idaho, Maine, Massachusetts, Michigan, Minnesota, Montana, Nebraska, New Hampshire, New York, Oregon, Texas, Utah, Vermont, Washington and Wisconsin	95	40	96.0	Non-Linear	0.82	11.48	0.00	NZ = 44.13+6.29E-02*C- 4.48*VFS+1.12*MS-1.29*CS+2.021*OC- 3.91*Fe+75.10*Al-3.54E- 03*C^2+0.12*VFS^2-2.24E- 02*MS^2+2.77E-02*CS^2-3.867E- 02*OC^2+3.96*Fe^2-19.45*Al^2
	The Other Locations	34			Linear	0.79	13.36	0.00	NZ = 96.69+6.41E-02*C+0.62*FSi- 0.13*MS+1.63E-02*MN-12.91*pH
	Chile, Costa Rica, Denmark, Guatemala, Nigeria, Puerto Rico and Russia		15	92.0	Non-Linear	0.84	14.05	0.00	NZ = -5.45+1.12*C- 1.37*FSi+0.43*MS+0.03*Mn+18.80*pH- 9.20E-03*C^2+3.04E-02*FSi^2-1.14E- 02*MS^2-1.81E-06*Mn^2-2.53*pH^2
		236	102	1195.6	Linear	0.40	165.50	0.00	Acid Oxal = 263.56-3.43*C- 4.77*FSi+119.24*Fe+220.28*Al+5.07E- 02*Mn
Acid Oxalate	The Whole World				Non-Linear	0.39	185.30	0.00	Acid Oxal = 59.74- 2.18*C+5.03*FSi+198.87*Fe+568.73*Al+ 1.86E-02*Mn-1.09E-02*C^2-5.38E- 02*FSi^2-38.46*Fe^2- 232.38*Al^2+5.79E-06*Mn^2
	The USA	214	91	5825.8	Linear	0.60	561.64	0.92	Acid Oxal = -96.20- 5.12*C+6.21*FSi+914.60*Al+0.23*Mn+4 .77*Ca
	Alabama, Alaska, California, Colorado, Connecticut, Delaware, Florida, Georgia, Hawaii, Idaho, Illinois,				Non-Linear	0.42	296.78	0.92	Acid Oxal = -50.03- 6.32*C+5.52*FSi+1063.51*Al+0.42*Mn+

Table 1. Performance indices for each model (R², RMSE, regression method and P Value) with their equations.

	Indiana, Iowa, Maine, Maryland, Massachusetts, Michigan, Minnesota, Mississippi, Montana, Nebraska, Nevada, New Hampshire, New Jersey, New Mexico, New York, North Carolina, North Dakota, Ohio, Oklahoma, Oregon, Pennsylvania, South Carolina, South Dakota, Tennessee, Texas, Utah, Vermont, Virginia, Washington, West Virginia, Wisconsin and Wyoming								3.11*Ca+3.44E-02*C^2-4.62E-02*FSi^2- 463.31*Al^2-4.60E-05*Mn^2+6.92E- 03*Ca^2
	The Other Locations	28	12	705.1	Linear	0.62	110.20	0.00	Acid Oxal = 39.19- 4.80*C+1175.29*Al+8.33*Ca-4.08*Mg
	Chile, China, Costa Rica, Denmark, Ghana, Guatemala, Poland and Puerto Rico				Non-Linear	0.85	91.40	0.00	Acid Oxal = 33.46- 2.26*C+1276.51*Al+7.39*Ca+1.46*Mg- 3.52E-02*C^2-112.11*Al^2-4.10E- 02*Ca^2-4.01E-02*Mg^2
		101	43	91.9	Linear	0.27	19.70	0.00	Bray1 = 22.52+1.71E-03*C+7.90E- 02*FSi+0.12*CSi+0.68*FS+0.72*CS+2.19 *OC-5.10*pH
	The Whole World				Non-Linear	0.31	19.00	0.00	Bray1 = 98.09-1.04*C-0.60*FSi- 0.40*CSi+1.25*OC-16.99*pH+9.84E- 03*C^2+1.01E-02*FSi^2+6.45E- 03*CSi^2+6.43E-02*OC^2+1.00*pH^2
	The USA	68	28	72.3	Linear	0.00	9.62	0.00	Bray1 = 42.02-5.78*pH
Bray	Alaska, Delaware, Hawaii, Illinois, Louisiana, Montana, North Carolina, Oklahoma, Pennsylvania, South Dakota, Tennessee, Vermont and Virginia				Non-Linear	0.28	12.42	0.00	Bray1 = 124.16-35.31*pH+2.56*pH^2
	The Other Locations	34	14	91.9	Linear	0.64	18.74	0.09	Bray1 = 131.27-0.70*C-0.29*FSi- 1.27*CSi+1.27*VCS+2.40*OC+3.96*Mg- 17.63*pH
	Canada, Ghana, Nigeria, Poland, Costa Rica and Guatemala				Non-Linear	0.89	12.35	0.09	Bray1 = 779.56-2.02*C+4.11*FSi- 3.51*CSi+2.48*VCS+25.30*OC- 21.12*Mg-247.63*pH+1.37E-02*C^2- 9.95E-02*FSi^2+0.13*CSi^2-3.76E- 02*VCS^2- 5.64*OC^2+0.75*Mg^2+19.69*pH^2
		83	35	86.1	Linear	0.23	24.75	0.00	Mehlich = 12.09-32.35*Al- 0.41*Ca+24.58*K
Mehlich	The Whole World				Non-Linear	0.44	17.40	0.00	Mehlich = 2.92+9.06*Al+1.89*Ca- 5.35*K-20.02*Al^2-4.45E- 02*Ca^2+11.57*K^2
	The USA	74	31	86.1	Linear	0.63	19.06	0.044	Mehlich = -8.60+0.77*CSi+1.63*CS- 1.49*VCS-1.15*OC-2.14*Mg+21.96*K

California, Connecticut, Delaware, Iowa, Kansas, Michigan, Mississippi, Nevada, New Mexico, North Dakota, Oregon, Vermont, West Virginia and Wyoming				Non-Linear	0.62	13.29	0.044	Mehlich = -0.17-3.77E-02*CSi+1.46*CS- 1.89*VCS+0.18*OC- 3.82*Mg+40.47*K+0.02*CSi^2+2.48E- 02*CS^2-1.58E-02*VCS^2- 0.07*OC^2+4.30E-02*Mg^2-7.169*K^2
The Other Locations	12	-	76.6	Linear	0.36	23.76	0.044	Mehlich = 44.99-41.36*Al
Costa Rica and Poland				Non-Linear	0.38	24.70	0.044	Mehlich = 49.93-79.664*Al+30.46*Al^2

Note: clay (C), fine silt (FSi), coarse silt (CSi), very fine sand (VFS), fine sand (FS), medium sand (MS), coarse sand (CS), very coarse sand (VCS), total organic carbon (OC), extractable iron (Fe),

extractable aluminum (Al), extractable manganese (Mn), extractable calcium (Ca), extractable magnesium (Mg) and extractable potassium (K)

Table 2. Correlation (Pearson) between soil co-variates and extraction methods for different datasets. Significant differences are indicated as p < 0.05* and p <0.01**. ns, not significant

Methods	Locations	С	FSi	CSi	VFS	FS	MS	CS	VCS	OC	PH	Са	Mg	К	Fe	Al	Mn
NZ	The Whole	-	-	0.04 ^{ns}	-0.06 ^{ns}	-	-	-0.08 ^{ns}	-	-	-0.39**	-0.18**	-0.22**	-	0.61**	0.72**	0.30**
	World																
	The USA	-0.07 ^{ns}	-	-	-0.07 ^{ns}	-	0.03 ^{ns}	0.03 ^{ns}	-	0.36**	-	-	-	-	0.68**	0.81**	-
	The Other	0.35**	0.60**	-	-	-	-0.35**	-	-	-	-0.45**	-	-	-	-	-	0.61**
	Locations																
Acid	The Whole	-0.12*	0.27**	-	-	-	-	-	-	-	-		-	-	0.47**	0.46**	0.20**
Oxalate	World																
	The USA	-0.03 ^{ns}	0.22**	-	-	-	-	-	-	-	-	0.08 ^{ns}	-	-	-	0.57**	0.43**
	The Other	-0.23 ^{ns}	-	-	-	-	-	-	-	-	-	0.21 ^{ns}	0.05 ^{ns}		-	0.66**	-
	Locations																
Bray	The Whole	-0.22**	-0.23**	-0.27**	-	0.36**	-	0.17*		0.13 ^{ns}	-0.20*	-	-	-	-	-	-
	World																
	The USA	-	-	-	-	-	-	-	-	-	-0.34**	-	-	-	-	-	-
	The Other	-0.42**	-0.31*	-0.28*	-	-	-	-	0.09 ^{ns}	0.20 ^{ns}	-0.22 ^{ns}	-	-0.09 ^{ns}	-	-	-	-
	Locations																
Mehlich	The Whole	-	-		-	-	-	-	-	-	-	0.28**		0.43**	-	-0.11 ^{ns}	-
	World																
	The USA	-	-	0.26**	-	-	-	0.06 ^{ns}	-0.03 ^{ns}	-0.03 ^{ns}	-	-	0.13 ^{ns}	0.56**	-	-	-
	The Other	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-0.60*	-
	Locations																

Note: clay (C), fine silt (FSi), coarse silt (CSi), very fine sand (VFS), fine sand (FS), medium sand (MS), coarse sand (CS), very coarse sand (VCS), total organic carbon (OC), extractable iron (Fe),

extractable aluminum (Al), extractable manganese (Mn), extractable calcium (Ca), extractable magnesium (Mg) and extractable potassium (K)